

Supplementary Material for:
Accurate spin-orbit and spin-other-orbit
contributions to the g-tensor for transition metal
containing systems

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S.1 RhC test molecule

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Table 1 Gaussian 09 calculations of the g-tensor, using several basis sets and the PBE functional. In Gaussian 09 only the Z_{eff} method was used, so they can only be compared to the non-relativistic calculations of ORCA using the same spin-orbit approximation. All values are expressed as $\Delta g = g_{ortho} - g_e$ in ppt.

	ORCA, Z_{eff}		G09, Z_{eff}	
	Δg_{ortho}	Δg_{para}	Δg_{ortho}	Δg_{para}
VDZ	111.73	0.21	108.1	0.21
VDZ, 2df	91.89	0.22	91.85	0.22
VDZ, 2df, p	96.27	0.22	95.76	0.22
VDZ, ppp, p	94.03	0.22	93.62	0.22
TZV	122.04	0.20	118.84	0.20
TZV, 2df	104.28	0.21	104.32	0.21
TZV, 2df, p	106.09	0.22	105.98	0.22
TZV, ppp, p	103.08	0.22	103.00	0.22
exp	51.78	1.58	51.78	1.58

Table 2 GIPAW results of g-tensor calculation in function of the box size (in Bohr). All values are expressed as $\Delta g = g_{ortho} - g_e$ in ppt.

Box Size	V_{eff} , PM	
	Δg_{ortho}	Δg_{para}
5	-8.76	-47.01
10	40.07	-0.23
15	43.15	-0.11
15	43.15	-0.11
20	48.67	-0.10
25	51.68	-0.10
30	53.22	-0.11
35	54.06	-0.11
exp	51.78	1.58

S.2 Doublet Molecules

Molecule	d (Å)	Molecule	d (Å)
RhC	1.613	ZnAg	2.550
BO	1.204	YO	1.788
BS	1.609	PdH	1.529
AlO	1.618	CdH	1.781
GaO	1.744	CdF	2.014
ScO	1.668	CdAg	2.727
ZnH	1.595	InO	1.875
ZnF	1.799		

Table 3 Internuclear distances d of the diatomic spin-1/2 molecules as reported by P. Belanzoni, E. van Lenthe and E. J. Baerends, J. Chem. Phys., 2001, **114**, 4421-4433

S.2 Doublet Molecules

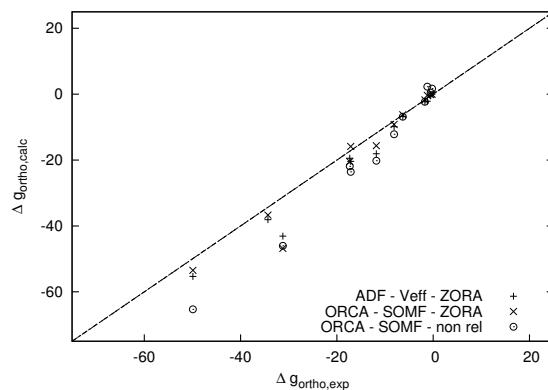


Figure 1 Correlation diagram between $\Delta g_{\text{ortho,exp}}$ and $\Delta g_{\text{ortho,calc}}$ (y-axis) for several doublet molecules. Results of SOMF (with and without ZORA) and ADF (two-component method) are presented. For all calculations a triple- ζ quality basis set with polarization functions (Ahrlrichs 2df if possible) was used.

S.3 High Spin Molecules

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Molecule	d (Å)	Molecule	d (Å)
<i>B</i> ₂	1.618	<i>CrN</i>	1.556
<i>NH</i>	1.053	<i>Ge</i> ₂ ⁺	2.461
<i>NF</i>	1.343	<i>GaAs</i> ⁺	2.744
<i>NCl</i>	1.643	<i>YB</i> ⁺	2.263
<i>NBr</i>	1.808	<i>YAl</i> ⁺	2.816
<i>NI</i>	2.007	<i>ZrV</i>	2.076
<i>O</i> ₂	1.235	<i>MoN</i>	1.659
<i>PH</i>	1.453	<i>NbO</i>	1.714
<i>SeO</i>	1.677	<i>TiNb</i>	2.046
<i>SO</i>	1.518	<i>ZrNb</i>	2.245
<i>S</i> ₂	1.931	<i>CrF</i>	1.784
<i>BC</i>	1.496	<i>CrAg</i>	2.554
<i>C</i> ₂ ⁺	1.418	<i>MnO</i>	1.629
<i>AlC</i>	1.986	<i>MnS</i>	2.043
<i>SiB</i>	1.935	<i>MnH</i>	1.710
<i>SiAl</i>	2.459	<i>MnF</i>	1.833
<i>Si</i> ₂ ⁺	2.327	<i>MnCl</i>	2.241
<i>VO</i>	1.597	<i>MnBr</i>	2.395
<i>TiV</i>	1.835	<i>MnAg</i>	2.539
<i>V</i> ₂ ⁺	1.741		

Table 4 Internuclear distances d of the diatomic high spin molecules as used by S. Pathkovskii and T. Ziegler, J. Phys. Chem. A, 2001, **105**, 5490-5497