

1    **Supporting Information**

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4    **Fig. S1.** Spectro-photoelectrochemical cell constructed from PEEK used for anodisation  
5    studies.

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7    *Gas chromatography measurements of H<sub>2</sub>*

8    H<sub>2</sub> was generated with solar simulator light source under the aforementioned described  
9    experimental conditions in a specifically designed gas-tight Teflon PEC cell for online H<sub>2</sub> gas  
10   detection and quantification with a gas chromatograph (GC-2014, Shimadzu). The generated  
11   H<sub>2</sub> is recirculated with Ar carrier gas by a membrane pump at 0.1 bar overpressure enabling  
12   us to monitor H<sub>2</sub> evolution online. For calibration, two reference gases containing 50 and 100  
13   ppm H<sub>2</sub> in Ar are used. 10 µL samples of the PEC cell headspace are injected to the Hayasep  
14   D column (10 Ft x 1/8 in SS). The H<sub>2</sub> amount exceeded our estimates and did not fit in the  
15   entire calibration range. The calibration curve was thus extrapolated to give an estimation of  
16   the evolved H<sub>2</sub>. The shown values should not be considered as absolute values.

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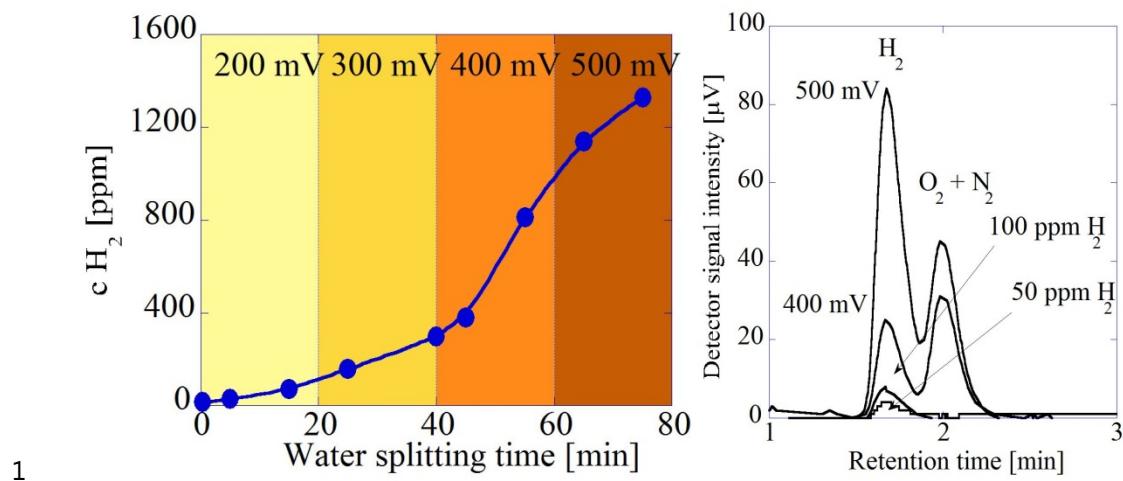


Fig. S2: Left - Evolution of hydrogen concentration with bias potential varied from 200 mV to 500 mV during 80 minutes water splitting time, as obtained from gas chromatograms. Right - Representative chromatograms for 5 minutes collection at 400 mV and 500 mV bias potential of hematite anodised for 10 minutes at 700 mV, plus two chromatograms from 50 ppm and 100 ppm  $H_2$  reference gases.

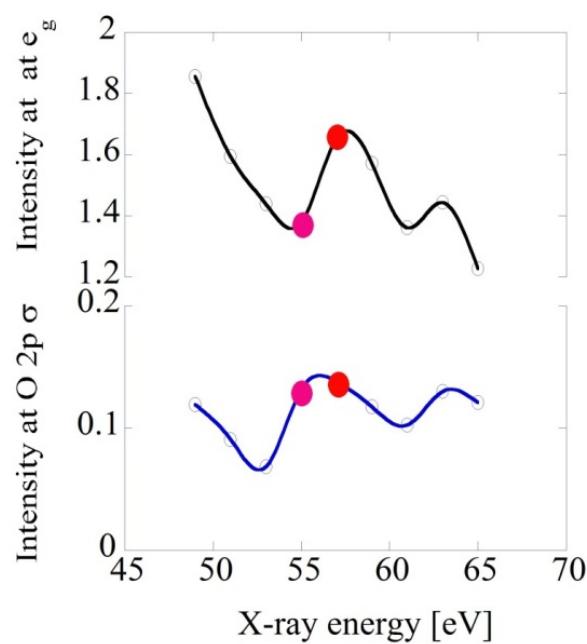
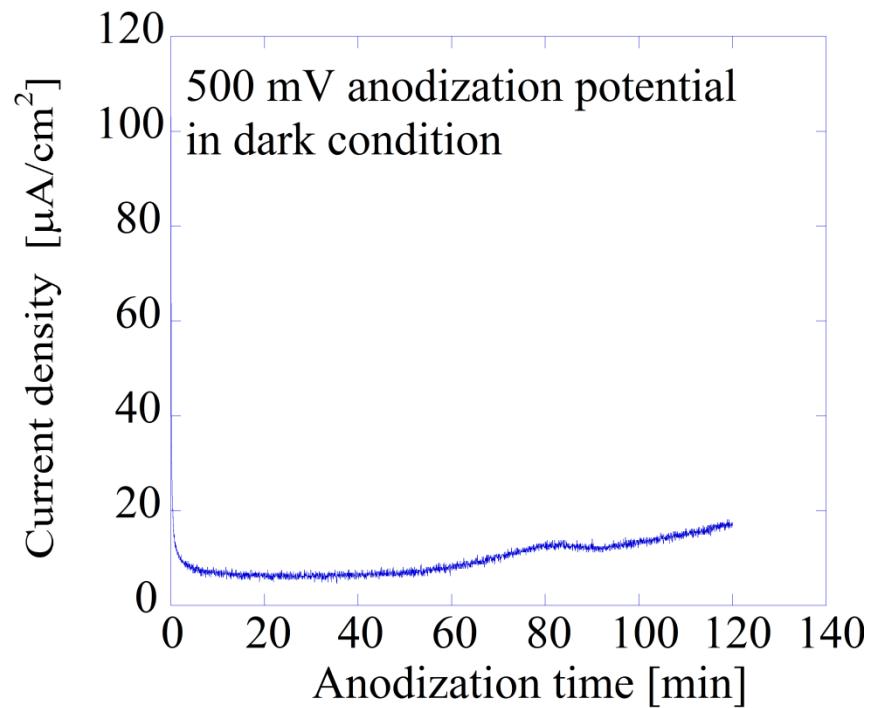


Fig. S3. Variation of the relative intensities of the O 2p  $\sigma$  peak and the  $e_g$  peak for the selected x-ray energies. Compare references [25, 26].

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4 **Fig. S4.** Evolution of the current density at 500 mV in dark condition.

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