# Tri-metallic deltahedral Zintl ions: A theoretical survey of the series $[Sn_{9-m-n}Ge_mBi_n]^{(4-n)-}$ for n = 1 - 4 and m = 0 - (9-n). Prediction and rationalization of their possible structures.

Alvaro Muñoz-Castro<sup>a</sup> and Slavi S. Sevov<sup>b</sup>

<sup>a</sup>Departamento de Ciencias Quimicas, Universidad Andres Bello, Av. Republica 275, Santiago, Chile.

<sup>b</sup>Departament of Chemistry and Biochemistry, University of Notre Dame, Notre Dame, Indiana 46556, USA.



**Figure S1.** Natural Charge distribution in both  $[Ge_9]^{4-}$  and  $[Sn_9]^{4-}$  displaying  $C_{4\nu}$  and  $D_{3h}$  structures.



**Figure S2.** Global minimum structure for  $[Sn_6Ge_3]^{4-}$ .



**Figure S3.** Low-energy isomers of the series  $[Sn_{8-m}Ge_mBi_1]^{3-}$  for m = 0 - 8. The difference in energy (kcal/mol) from the global minimum is given in parenthesis.



**Figure S4.** Low-energy isomers of the series  $[Sn_{7-m}Ge_mBi_2]^{2-}$  for m = 0 - 7. The difference in energy (kcal/mol) from the global minimum is given in parenthesis.



**Figure S5.** Low-energy isomers of the series  $[Sn_{6-m}Ge_mBi_3]^{1-}$  for m = 0 - 6. The difference in energy (kcal/mol) from the global minimum is given in parenthesis.



**Figure S6.** Low-energy isomers of the series  $[Sn_{5-m}Ge_mBi_4]$  for m = 0 - 5. The difference in energy (kcal/mol) from the global minimum is given in parenthesis.

Electronic Supplementary Material (ESI) for Physical Chemistry Chemical Physics This journal is The Owner Societies 2013

## SUPPORTING INFORMATION

**Table S1.**Total bonding energies (TBE) for the global minima of the Sn-Ge-Bitrimetallic deltahedral Zintl ions with various stoichiometries.

Composition	TBE (eV) <sup>a</sup>		
composition			
[Ge <sub>8</sub> Bi] <sup>3-</sup>	-24 5671		
[SnGe <sub>7</sub> Bi] <sup>3-</sup>	-24.0916		
$[Sn_2Ge_6Bi]^3$	-23.6624		
[Sn <sub>3</sub> Ge <sub>5</sub> Bi] <sup>3-</sup>	-23.1884		
$[Sn_4Ge_4Bi]^{3-1}$	-22.7652		
[Sn <sub>5</sub> Ge <sub>3</sub> Bi] <sup>3-</sup>	-22.2569		
$[Sn_6Ge_2Bi]^{3-}$	-21.8156		
[Sn <sub>7</sub> GeBi] <sup>3-</sup>	-21.3653		
[Sn <sub>8</sub> Bi] <sup>3-</sup>	-20.8604		
$[Ge_7Bi_2]^{2-}$	-27.9900		
[SnGe <sub>6</sub> Bi <sub>2</sub> ] <sup>2-</sup>	-27.4557		
$[Sn_2Ge_5Bi_2]^{2}$	-26.9466		
$[Sn_3Ge_4Bi_2]^{2}$	-26.4894		
$[Sn_4Ge_3Bi_2]^{2-}$	-25.9587		
$[Sn_5Ge_2Bi_2]^{2-}$	-25.4537		
$[Sn_6Ge_1Bi_2]^{2-}$	-25.0252		
$[Sn_7Bi_2]^{2-}$	-24.4361		
$[Ge_6Bi_3]^-$	-27.8418		
[SnGe <sub>5</sub> Bi <sub>3</sub> ] <sup>-</sup>	-27.3292		
$[Sn_2Ge_4Bi_3]^-$	-26.8592		
$[Sn_3Ge_3Bi_3]^-$	-26.2902		
$[Sn_4Ge_2Bi_3]^-$	-25.5336		
[Sn <sub>5</sub> GeBi <sub>3</sub> ] <sup>-</sup>	-25.2701		
$[Sn_6Bi_3]^-$	-24.7807		
[Ge <sub>5</sub> Bi <sub>4</sub> ]	-23.9934		
[SnGe <sub>4</sub> Bi <sub>4</sub> ]	-23.7368		
[Sn <sub>2</sub> Ge <sub>3</sub> Bi <sub>4</sub> ]	-23.0077		
[Sn <sub>3</sub> Ge <sub>2</sub> Bi <sub>4</sub> ]	-22.5404		
[Sn <sub>4</sub> GeBi <sub>4</sub> ]	-22.0813		
[Sn <sub>5</sub> Bi <sub>4</sub> ]	-21.4782		

<sup>a</sup>Corrected values according to V. Branchadell, M. Sodupe, *Chem.Phys.Lett.* **1997**, 265, 481. These values indicate that the inclusion of Sn nuclei decrease the bonding energy in each series.

**Table S2**. Results from the *Continuos-Shape-Measure* (CShM) calculations. Each number represents the deviation of a given nine-atom deltahedral Zintl ion from a perfect *msa* or *ttp* polyhedron. Smaller deviation (denoted in bold) indicates better fit with the corresponding shape. Deviations of around 0.5 and 1.5 indicate intermediate situation.

Composition	CShM	CShM	φ	Symmetry
_	msa	ttp		
[Ge <sub>8</sub> Bi] <sup>3-</sup>	1.192	0.832	0.10	$\sim D_{3h}$
[SnGe7Bi]3-	0.704	1.030	1.00	$C_{4v}$
$[Sn_2Ge_6Bi]^{3-}$	0.478	1.455	0.50	$\sim C_{2v}$
[Sn <sub>3</sub> Ge <sub>5</sub> Bi] <sup>3-</sup>	1.205	0.870	0.05	$D_{3h}$
[Sn <sub>4</sub> Ge <sub>4</sub> Bi] <sup>3-</sup>	1.306	0.904	0.06	$D_{3h}$
[Sn <sub>5</sub> Ge <sub>3</sub> Bi] <sup>3-</sup>	1.089	0.706	0.01	$D_{3h}$
$[Sn_6Ge_2Bi]^{3-}$	0.840	0.594	0.06	$D_{3h}$
[Sn <sub>7</sub> GeBi] <sup>3-</sup>	0.332	0.902	0.97	$C_{4v}$
[Sn <sub>8</sub> Bi] <sup>3-</sup>	0.924	0.581	0.23	$\sim D_{3h}$
$[Ge_7Bi_2]^{2-}$	0.564	1.597	0.40	$\sim C_{2v}$
$[SnGe_6Bi_2]^{2}$	0.618	1.627	0.71	$\sim C_{2v}$
$[Sn_2Ge_5Bi_2]^{2-}$	0.458	1.409	0.89	$\sim C_{4v}$
$[Sn_3Ge_4Bi_2]^{2-}$	0.458	1.409	0.49	$\sim C_{2v}$
$[Sn_4Ge_3Bi_2]^{2-}$	0.569	1.388	1.00	$C_{4v}$
$[Sn_5Ge_2Bi_2]^{2-}$	0.394	1.281	0.96	$C_{4v}$
$[Sn_6Ge_1Bi_2]^{2-}$	0.234	1.148	0.98	$C_{4v}$
$[Sn_7Bi_2]^{2-}$	0.247	1.042	0.99	$C_{4v}$
[Ge <sub>6</sub> Bi <sub>3</sub> ] <sup>-</sup>	0.722	1.691	0.96	$C_{4v}$
[SnGe <sub>5</sub> Bi <sub>3</sub> ] <sup>-</sup>	0.677	1.674	0.47	$\sim C_{2\nu}$
[Sn <sub>2</sub> Ge <sub>4</sub> Bi <sub>3</sub> ] <sup>-</sup>	0.626	1.677	0.59	$\sim C_{2\nu}$
[Sn <sub>3</sub> Ge <sub>3</sub> Bi <sub>3</sub> ] <sup>-</sup>	0.640	1.657	0.68	$\sim C_{2v}$
$[Sn_4Ge_2Bi_3]^-$	0.219	0.657	0.81	$\sim C_{4v}$
[Sn <sub>5</sub> GeBi <sub>3</sub> ] <sup>-</sup>	0.312	0.422	0.96	$C_{4v}$
$[Sn_6Bi_3]^-$	1.062	0.064	0.00	$D_{3h}$
[Ge <sub>5</sub> Bi <sub>4</sub> ]	0.691	1.799	0.93	$C_{4v}$
[SnGe <sub>4</sub> Bi <sub>4</sub> ]	0.700	1.808	0.96	$C_{4v}$
[Sn <sub>2</sub> Ge <sub>3</sub> Bi <sub>4</sub> ]	0.516	1.656	0.49	$\sim C_{2v}$
[Sn <sub>3</sub> Ge <sub>2</sub> Bi <sub>4</sub> ]	0.382	1.554	0.97	$C_{4v}$
[Sn <sub>4</sub> GeBi <sub>4</sub> ]	0.247	1.435	1.00	$C_{4v}$
[Sn <sub>5</sub> Bi <sub>4</sub> ]	0.174	0.830	0.58	~ <i>C</i> <sub>2v</sub>