

## Supporting information

### Molecular Reactivity Dynamics in a Confined Environment

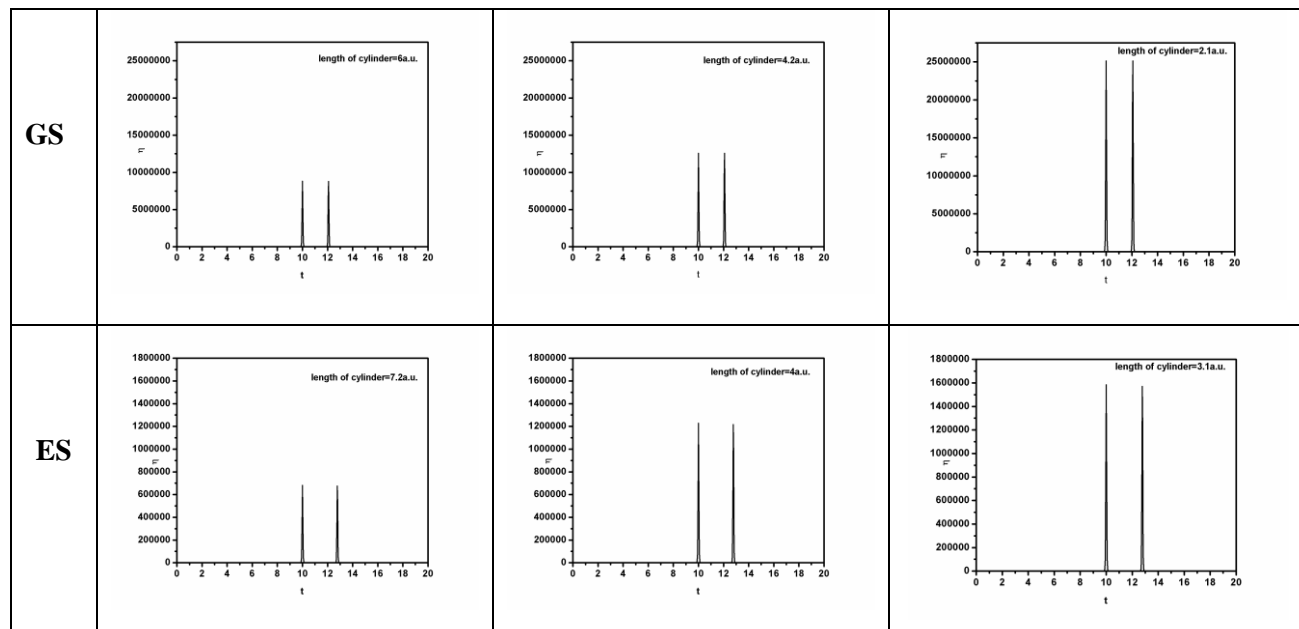
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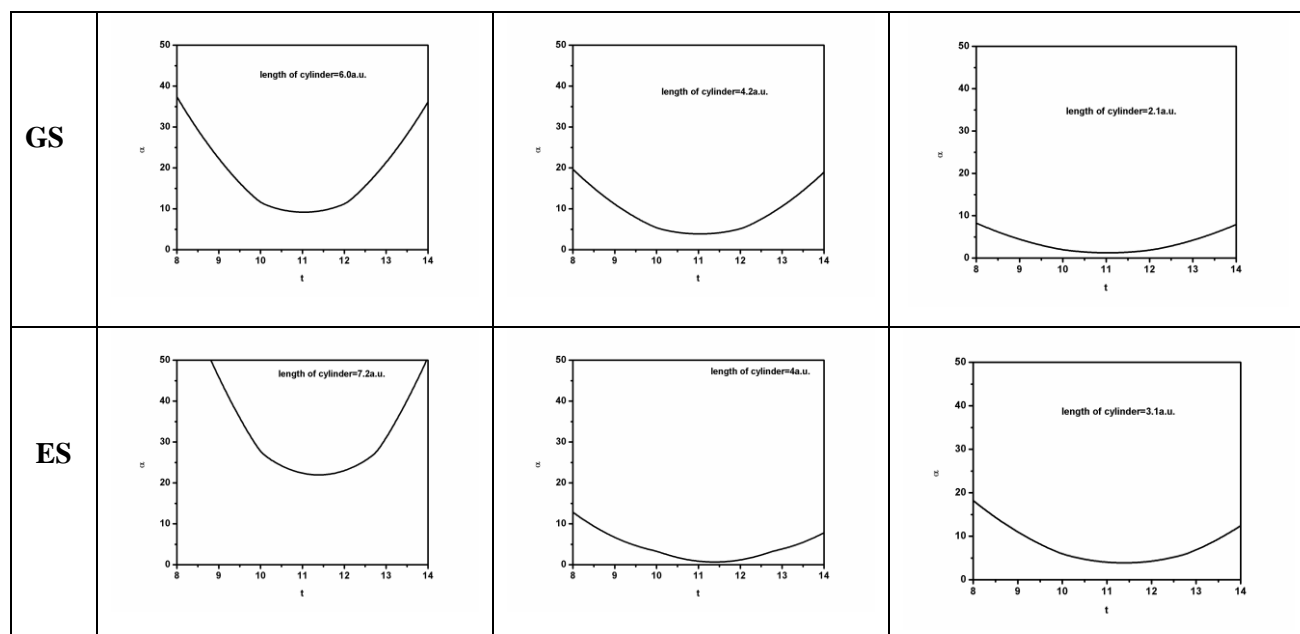
Indian Institute of Technology Kharagpur,

Kharagpur 721302, West Bengal, India.

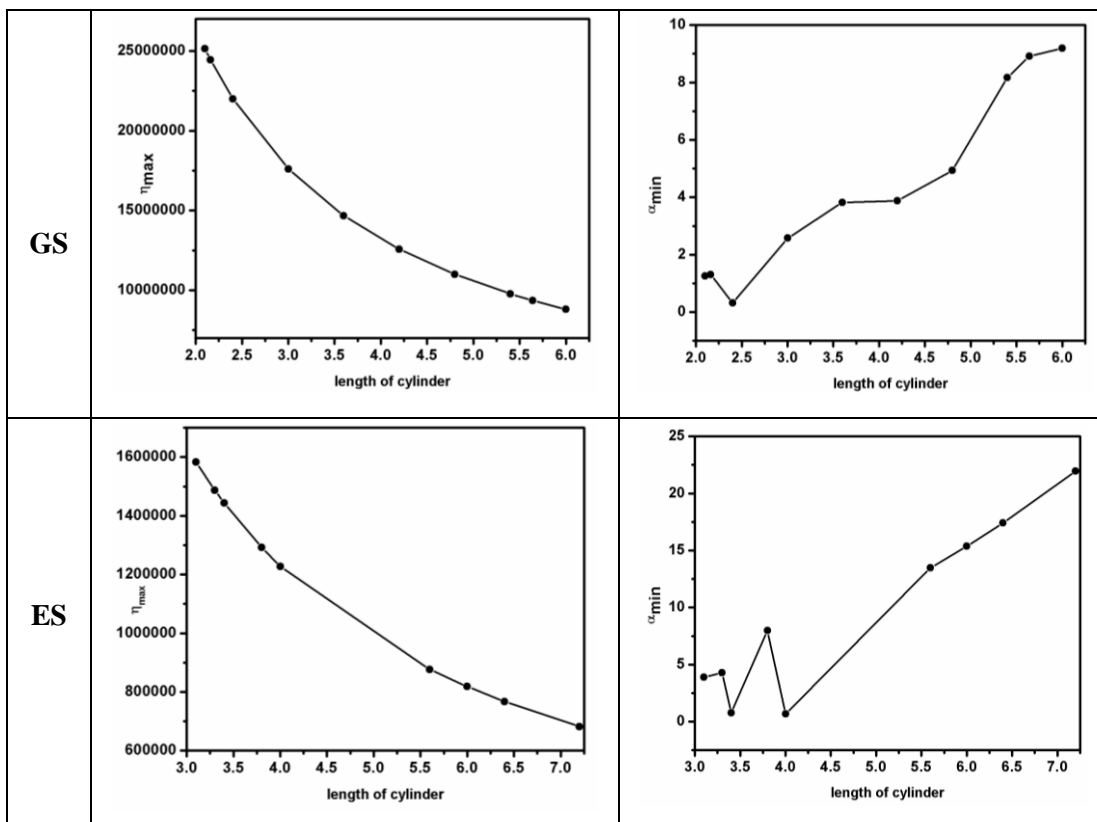
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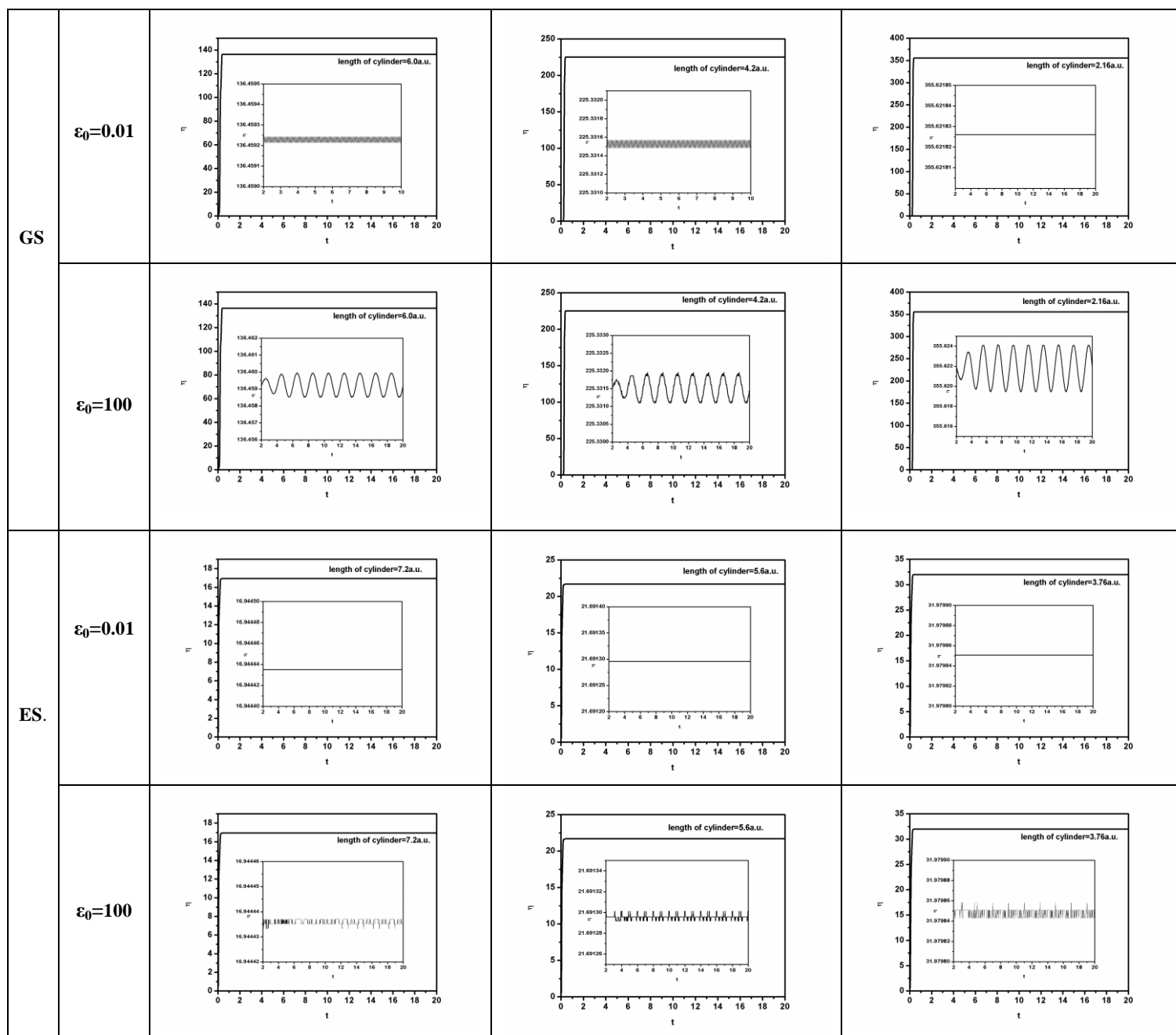
**Figure S1:** Time evolution of chemical hardness ( $\eta$ ) during a collision process between  $H^+$  and  $N_2$  molecule in ground state (GS) and excited state (ES) in a confined environment (length of cylinder = 6.0a.u. denotes unconfined system in GS and 7.2a.u. in ES).



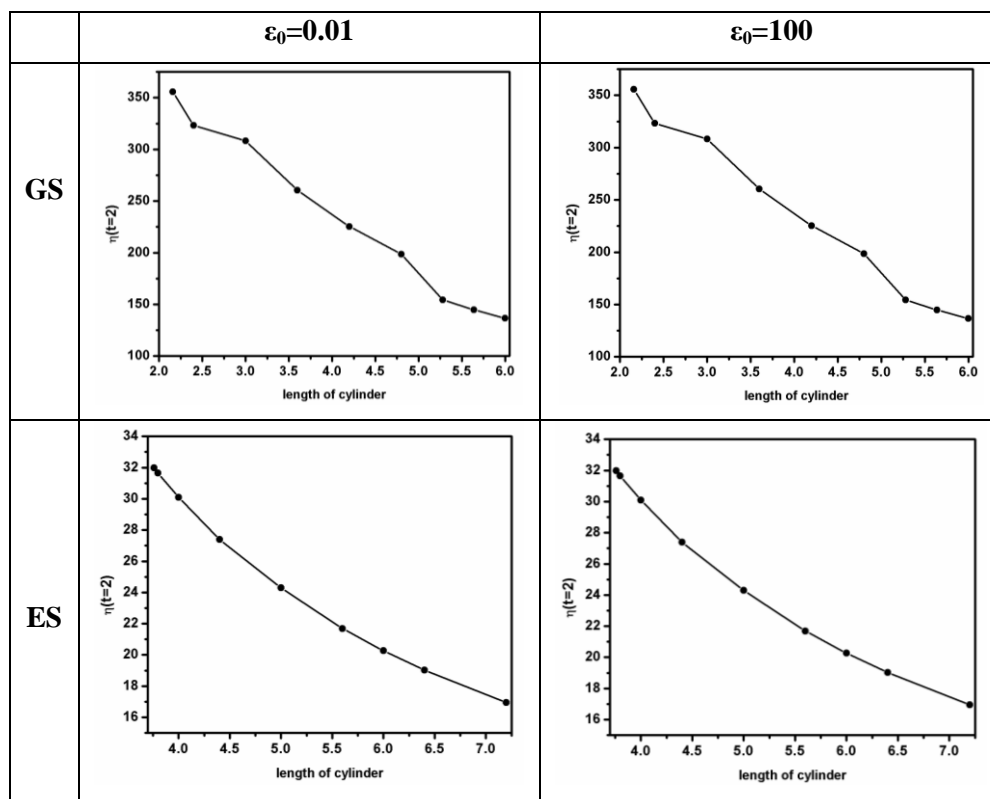
**Figure S2:** Time evolution of polarizability ( $\alpha$ ) during a collision process between  $H^+$  and  $N_2$  molecule in ground state (GS) and excited state (ES) in a confined environment (length of cylinder = 6.0a.u. denotes unconfined system in GS and 7.2a.u. in ES).



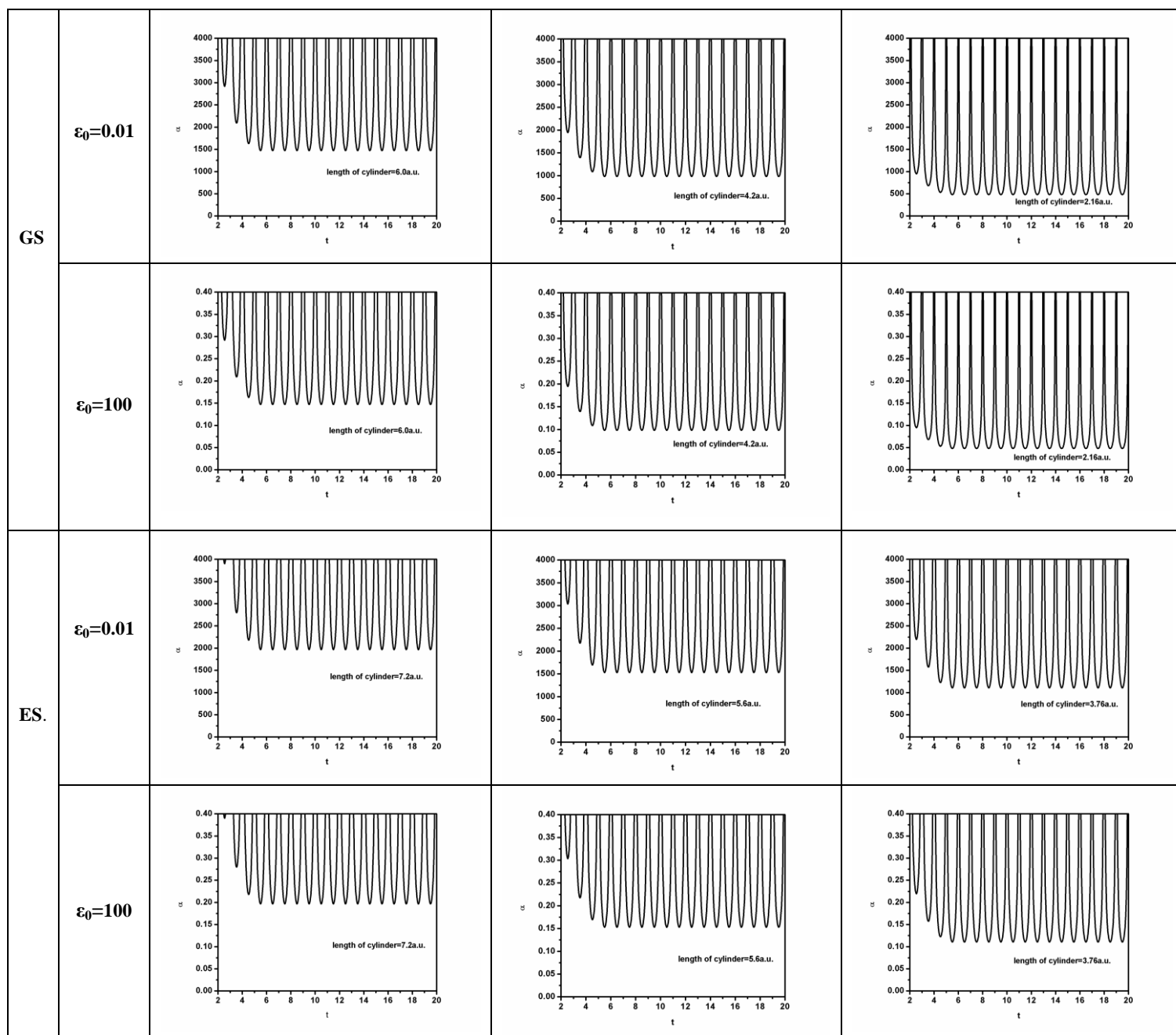
**Figure S3:** Plots of  $\eta_{\max}$  vs. length of cylinder and  $\alpha_{\min}$  vs. length of cylinder during a collision process between  $\text{H}^+$  and  $\text{N}_2$  molecule in GS and ES.



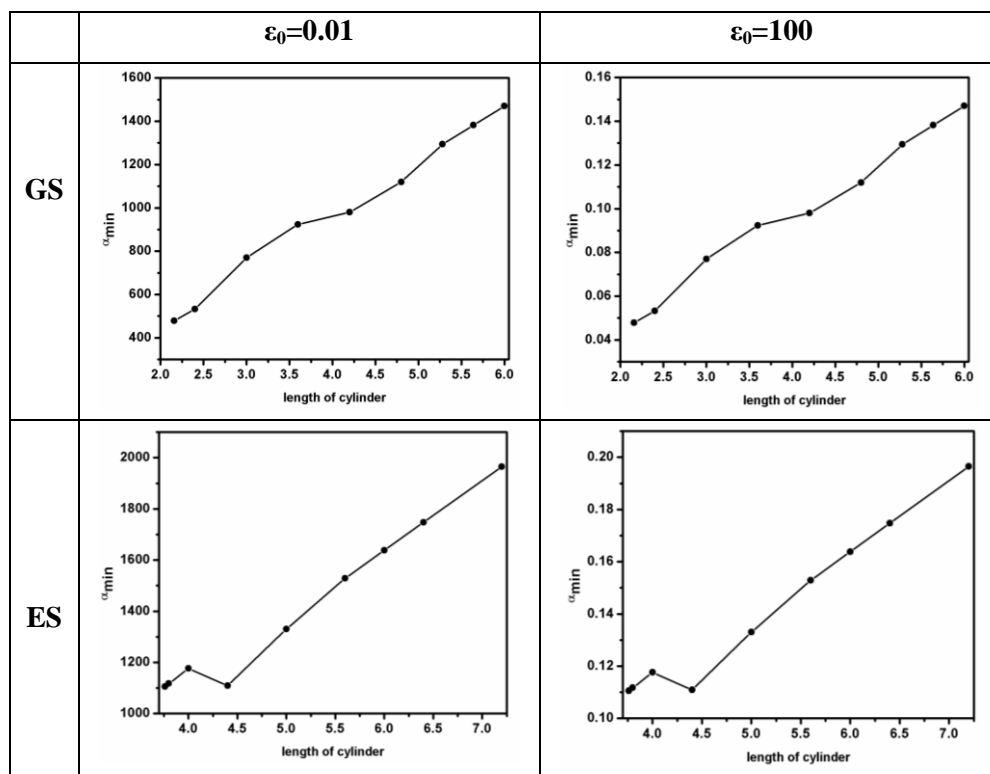
**Figure S4:** Time evolution of chemical hardness ( $\eta$ ) when  $N_2$  molecule in ground state (GS) and first excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01a.u.$  &  $100a.u.$  and frequency  $\omega_0 = \pi$ .



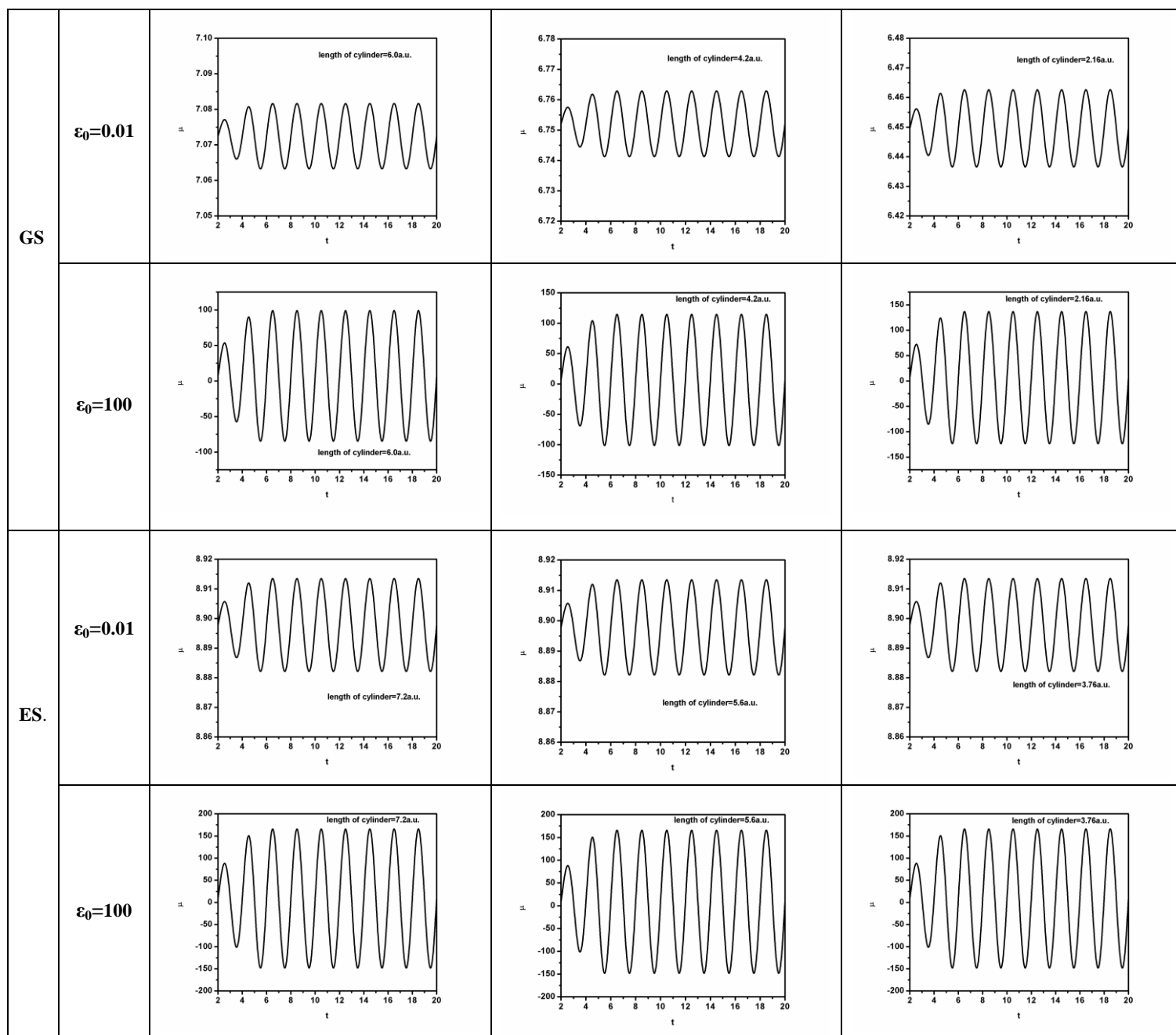
**Figure S5:** Plot of  $\eta(t = 2)$  vs. length of cylinder when  $N_2$  molecule in GS and in first ES is placed in an external field ( $\epsilon_0 = 0.01$ a.u. &  $100$ a.u. and  $\omega_0 = \pi$ ).



**Figure S6:** Time evolution of polarizability ( $\alpha$ ) when  $N_2$  molecule in ground state (GS) and first excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01a.u.$  &  $100a.u.$  and frequency  $\omega_0 = \pi$ .

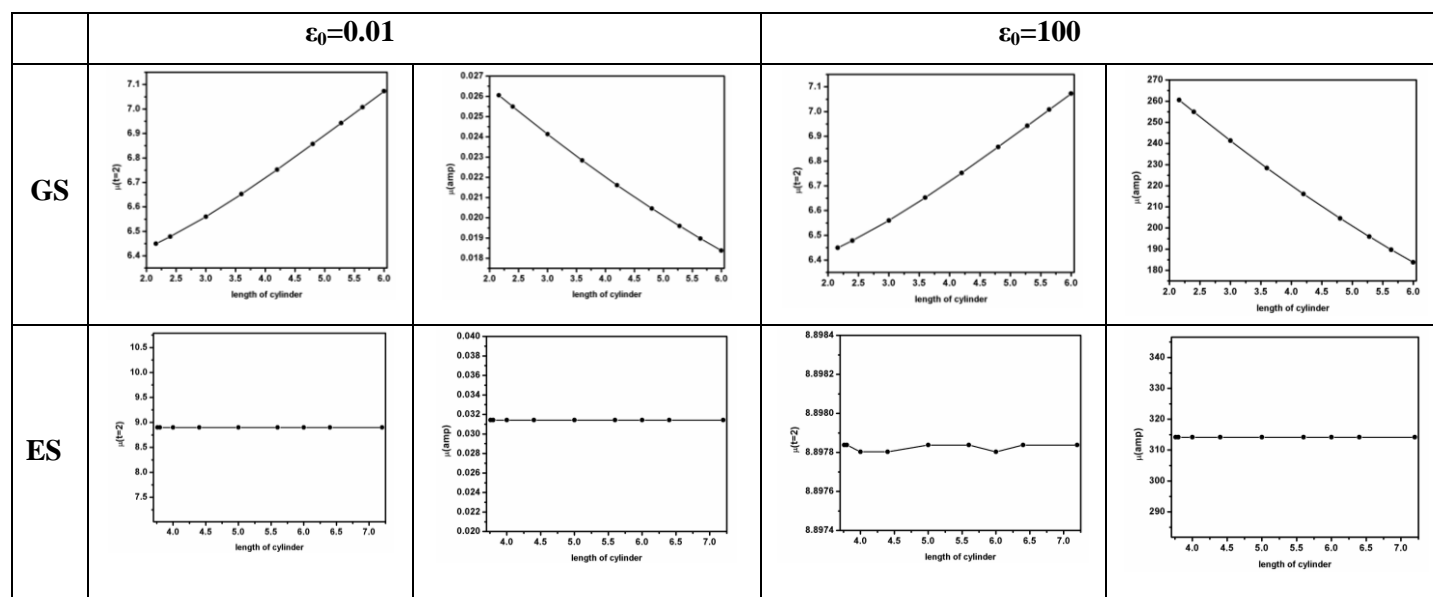


**Figure S7:** Plot of  $\alpha_{\min}$  vs. length of cylinder when  $N_2$  molecule in GS and in first ES is placed in an external field ( $\epsilon_0 = 0.01$ a.u. &  $100$ a.u. and  $\omega_0 = \pi$ ).

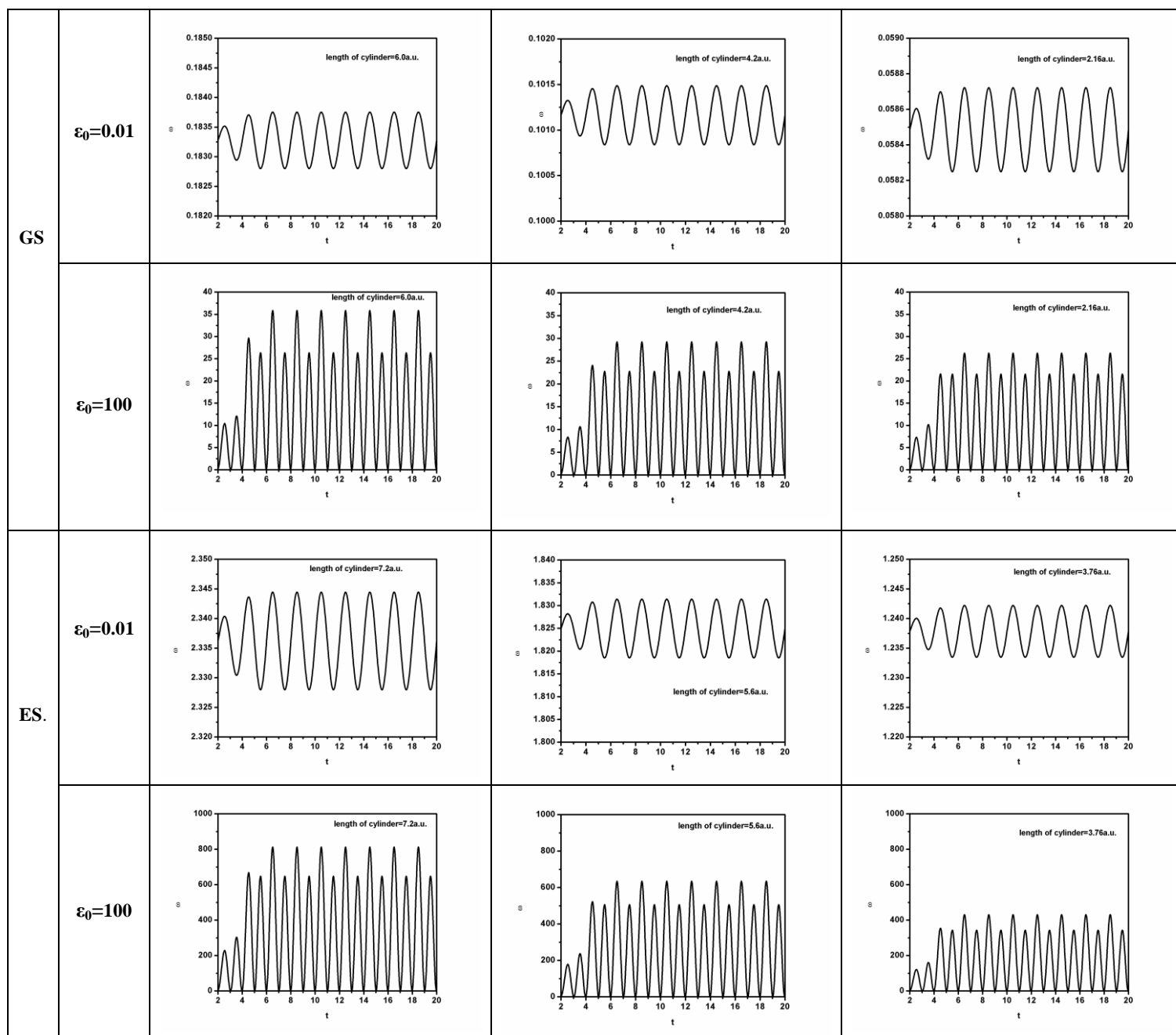


**Figure S8:** Time evolution of chemical potential ( $\mu$ ) when  $N_2$  molecule in ground state (GS) and first excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01a.u.$  &  $100a.u.$  and frequency  $\omega_0 = \pi$ .

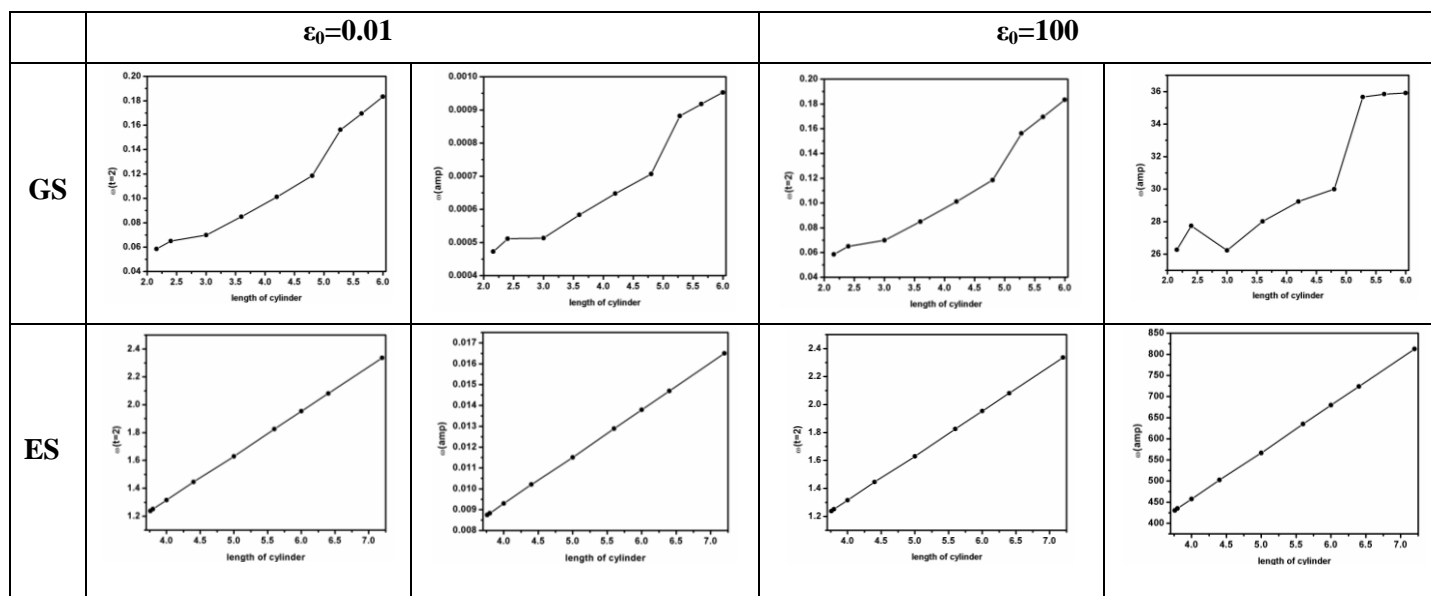




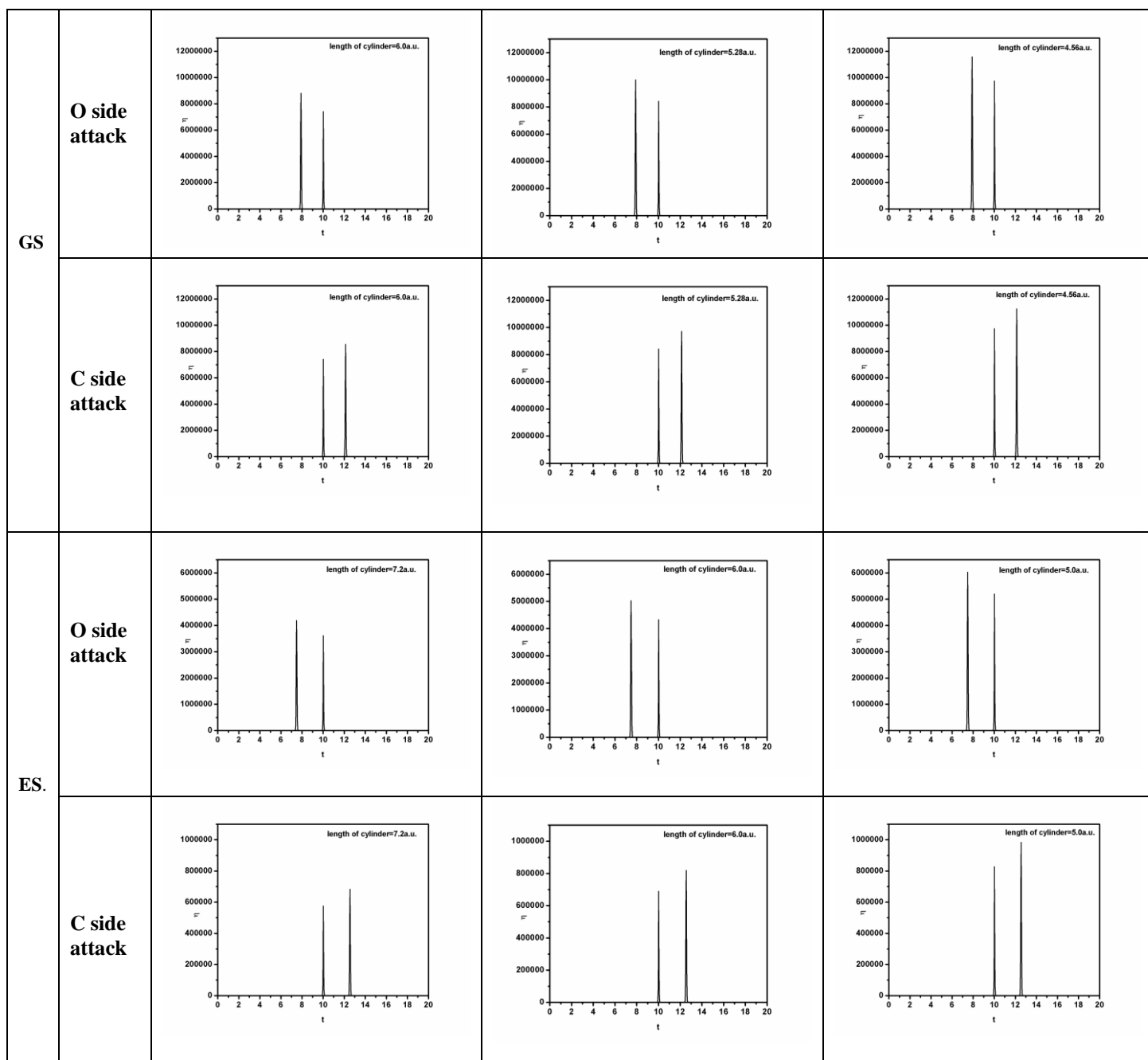
**Figure S9:** Plot of  $\mu(t = 2)$  vs. length of cylinder and  $\mu(\text{amp})$  vs. length of cylinder when  $\text{N}_2$  molecule in GS and in first ES is placed in an external field ( $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and  $\omega_0 = \pi$ ).



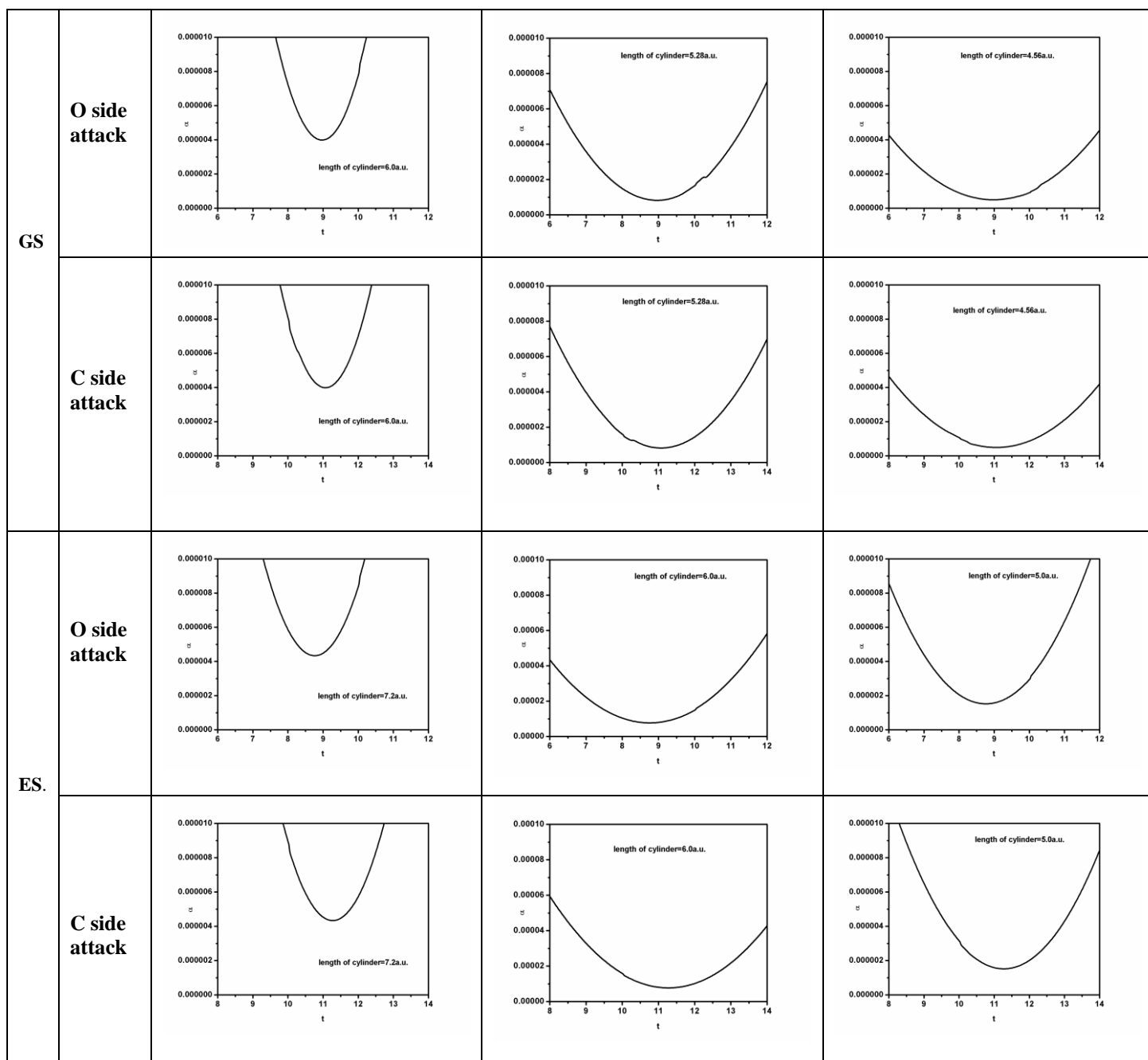
**Figure S10:** Time evolution of electrophilicity index ( $\omega$ ) when  $N_2$  molecule in ground state (GS) and first excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01$ a.u. & 100a.u. and frequency  $\omega_0 = \pi$ .



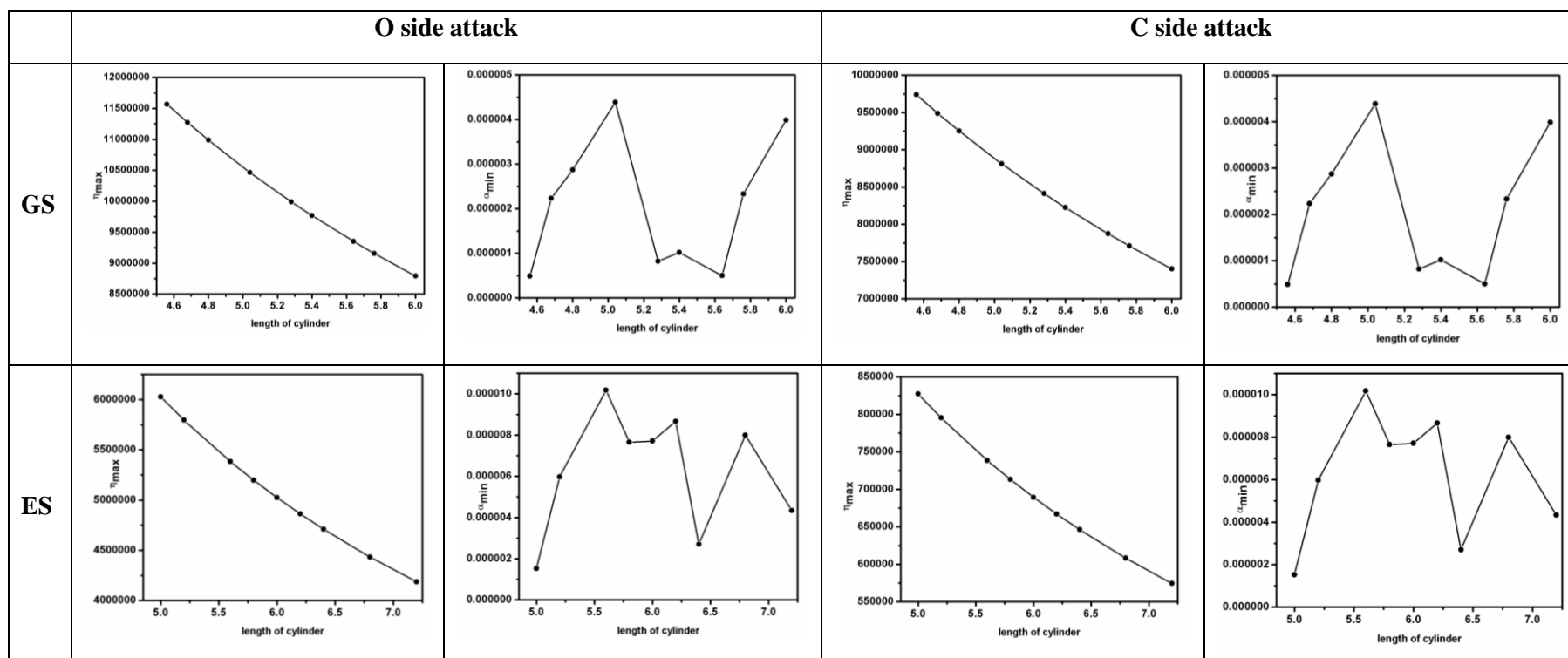
**Figure S11:** Plot of  $\omega(t=2)$  vs. length of cylinder and  $\omega(\text{amp})$  vs. length of cylinder when  $\text{N}_2$  molecule in GS and in first ES is placed in an external field ( $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and  $\omega_0 = \pi$ ).



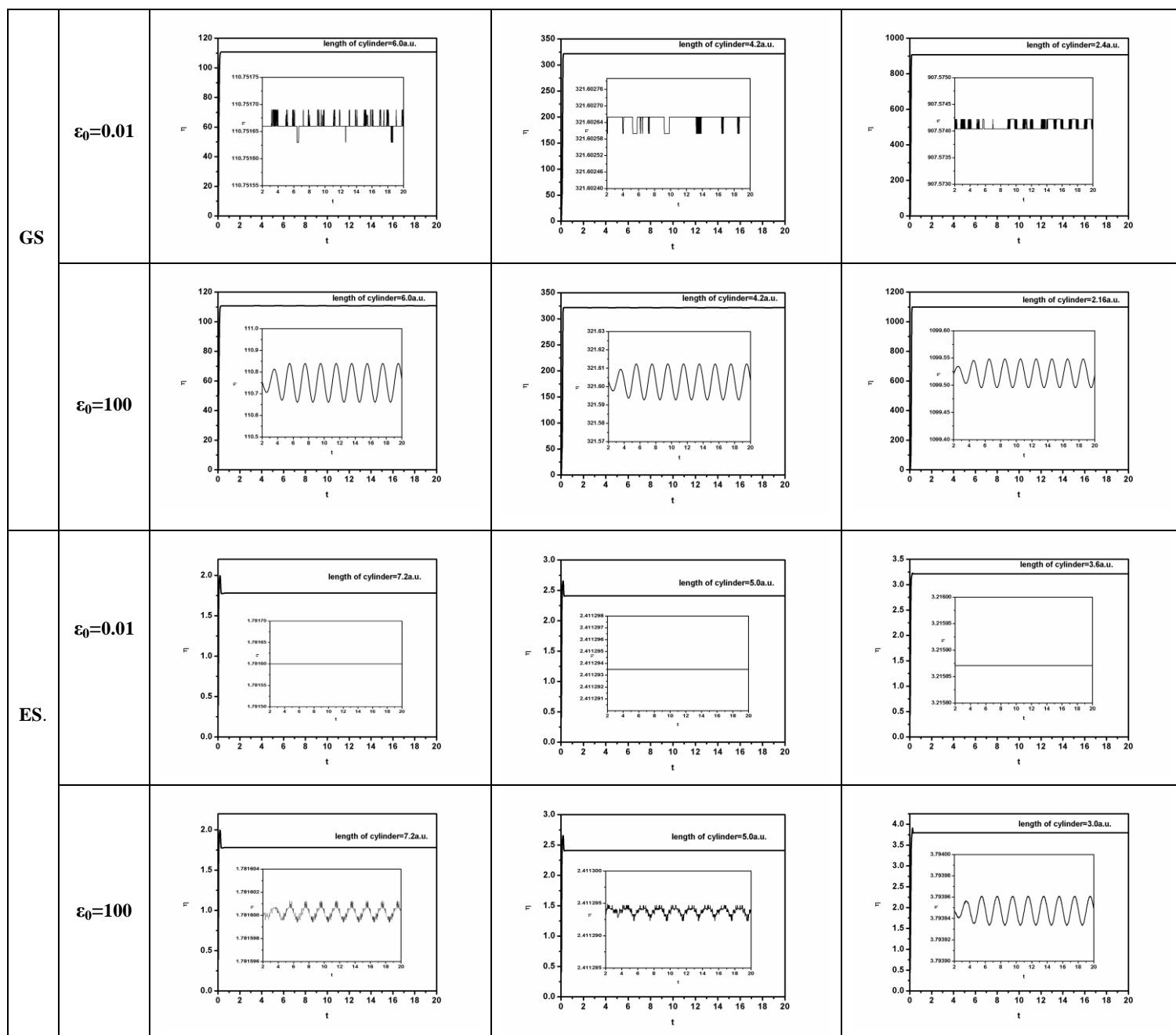
**Figure S12:** Time evolution of chemical hardness ( $\eta$ ) during a collision process between  $H^+$  and CO molecule in ground state (GS) and excited state (ES) in a confined environment (length of cylinder = 6.0a.u. denotes unconfined system in GS and 7.2a.u. in ES).



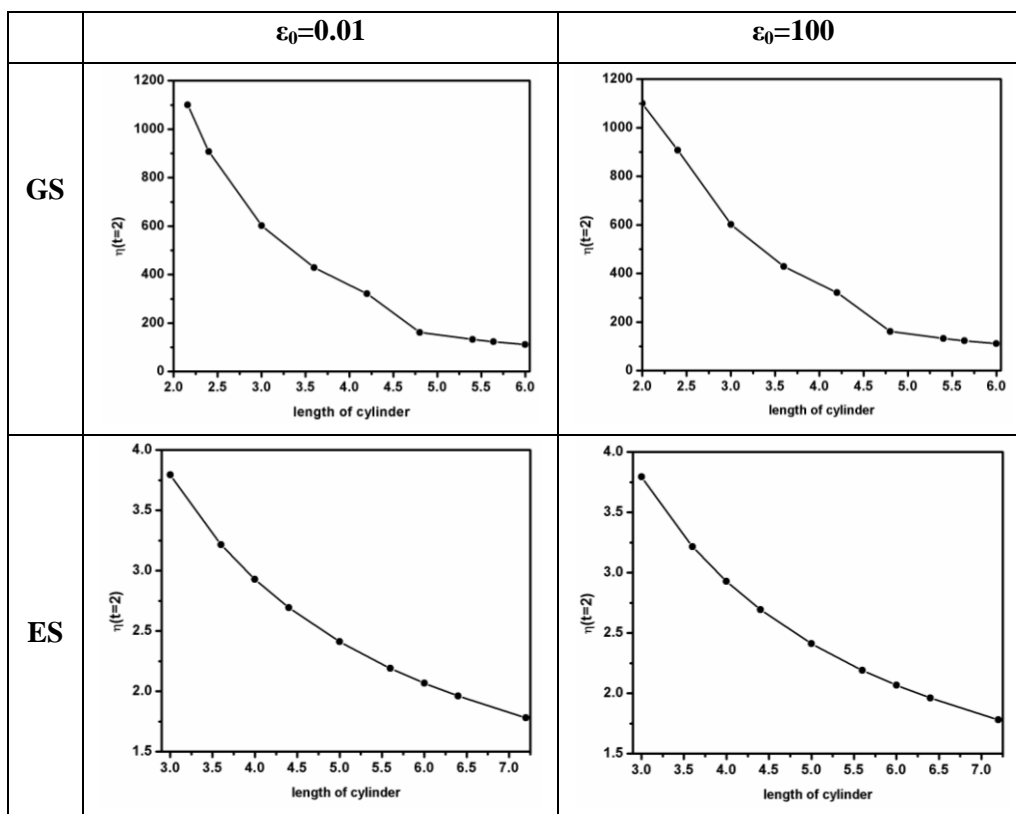
**Figure S13:** Time evolution of polarizability ( $\alpha$ ) during a collision process between  $H^+$  and CO molecule in ground state (GS) and excited state (ES) in a confined environment (length of cylinder = 6.0a.u. denotes unconfined system in GS and 7.2a.u. in ES).



**Figure S14:** Plots of  $\eta_{\max}$  vs. length of cylinder and  $\alpha_{\min}$  vs. length of cylinder during a collision process between  $\text{H}^+$  and CO molecule in GS and ES.

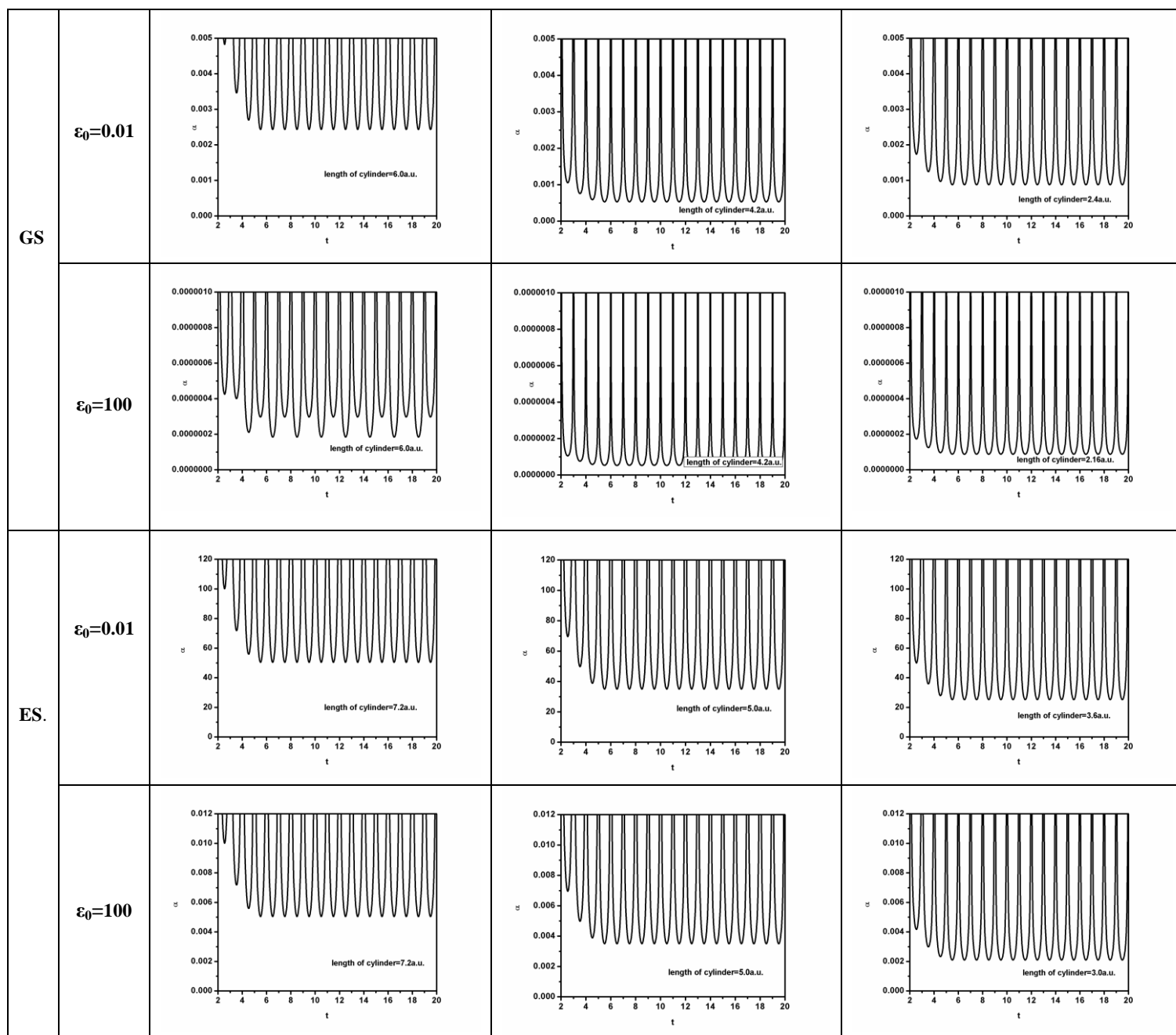


**Figure S15:** Time evolution of chemical hardness ( $\eta$ ) when CO molecule in ground state (GS) and excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and frequency  $\omega_0 = \pi$ .

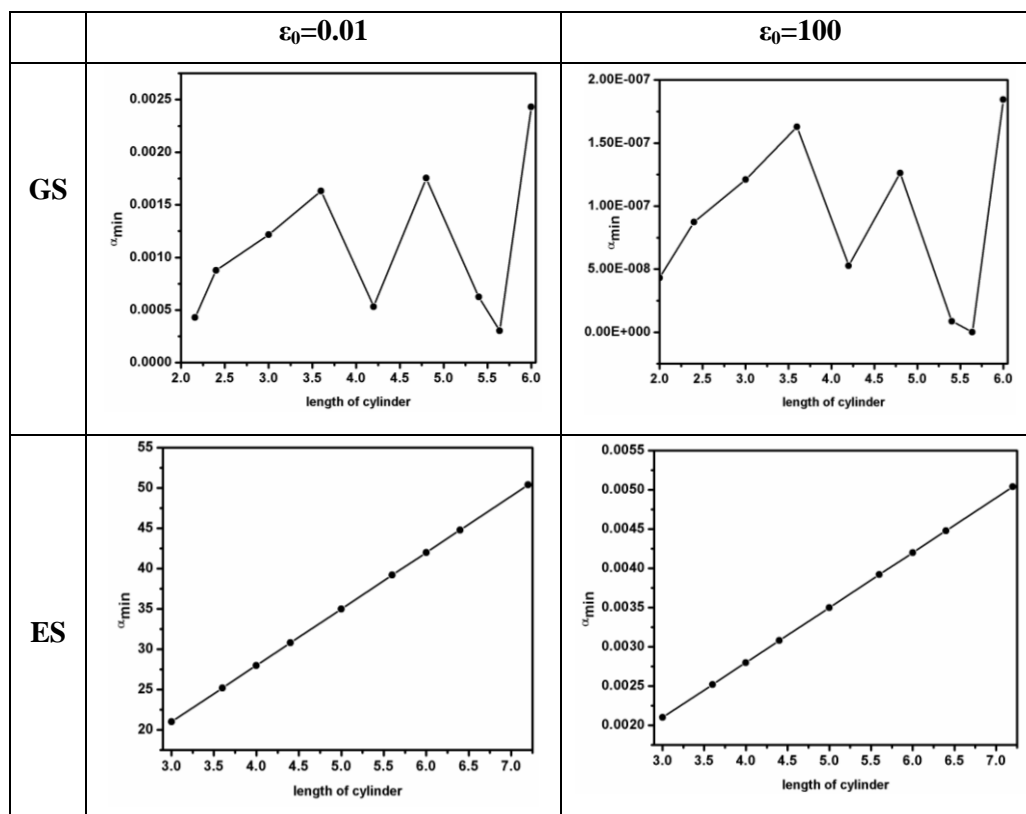


**Figure S16:** Plot of  $\eta(t = 2)$  vs. length of cylinder when CO molecule in GS and ES is placed in an external field ( $\epsilon_0 = 0.01$ a.u. &  $100$ a.u. and  $\omega_0 = \pi$ ).

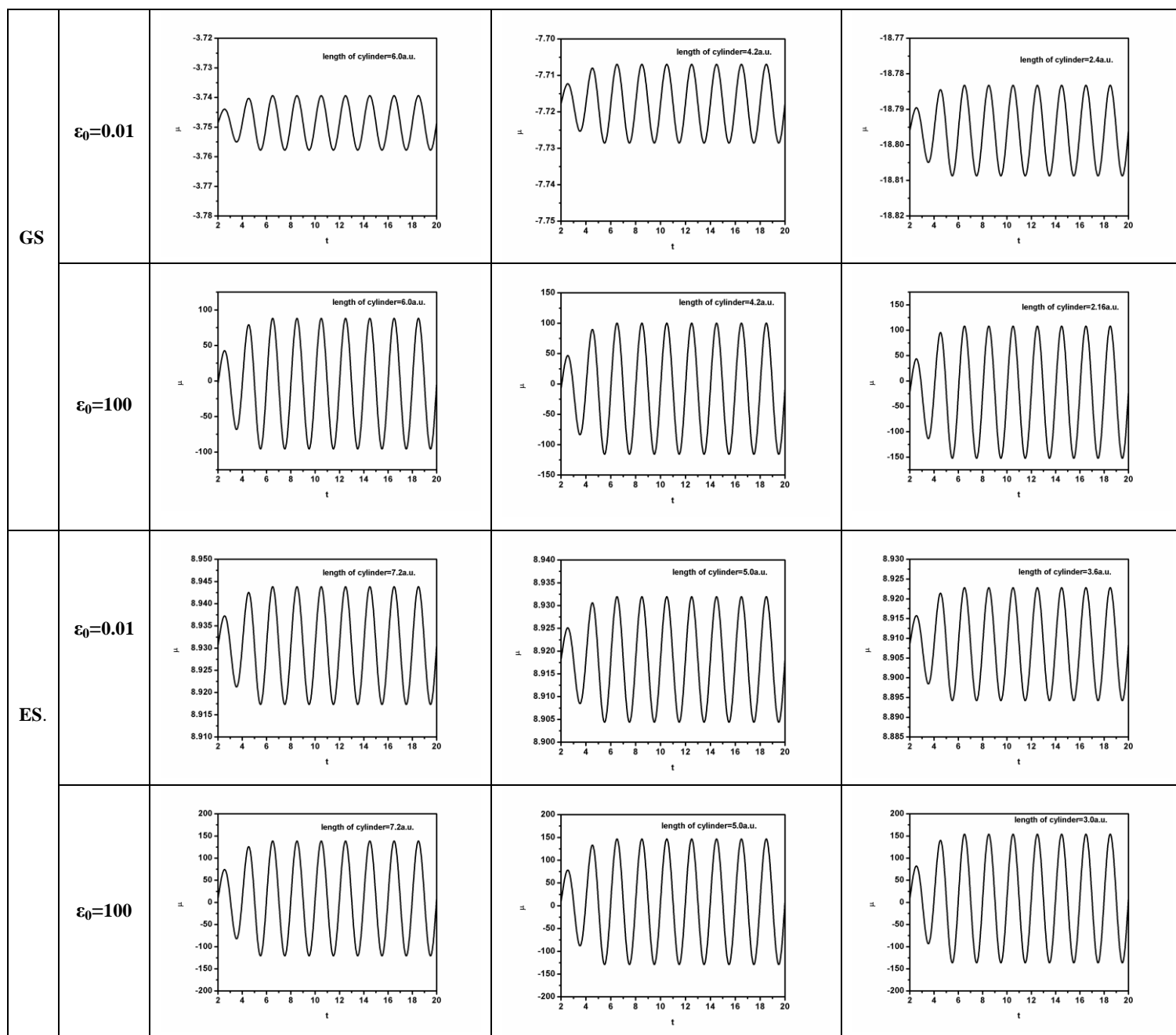




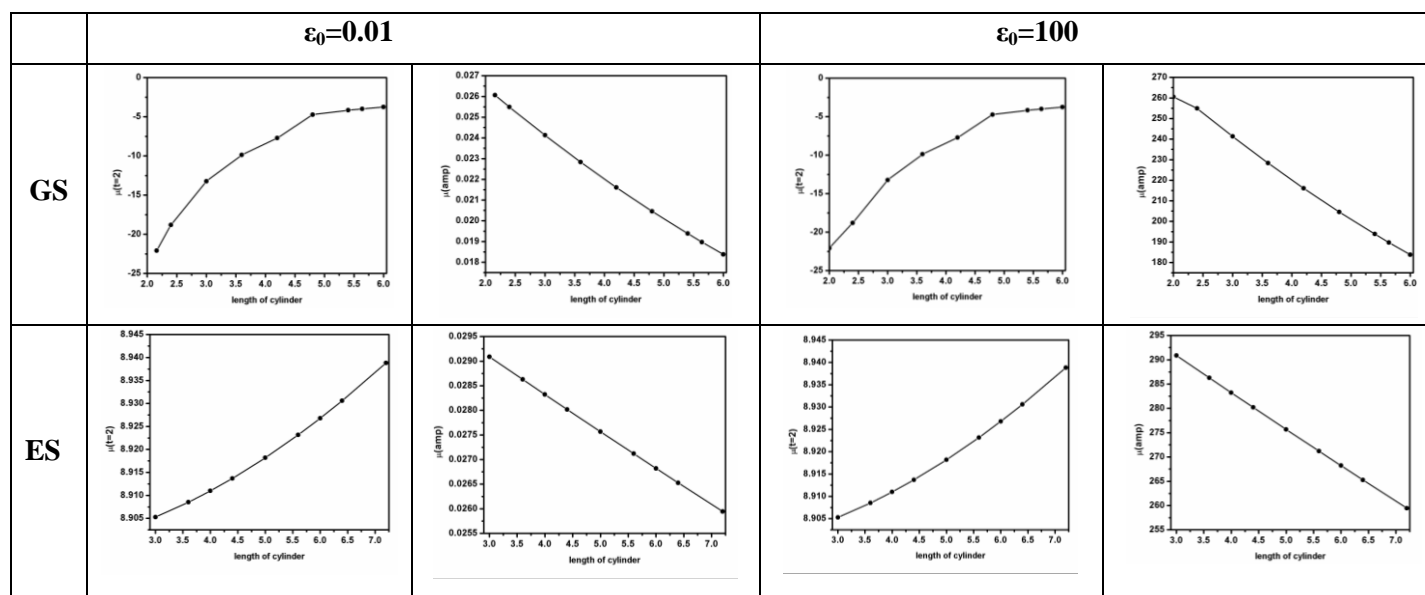
**Figure S17:** Time evolution of polarizability ( $\alpha$ ) when CO molecule in ground state (GS) and excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and frequency  $\omega_0 = \pi$ .



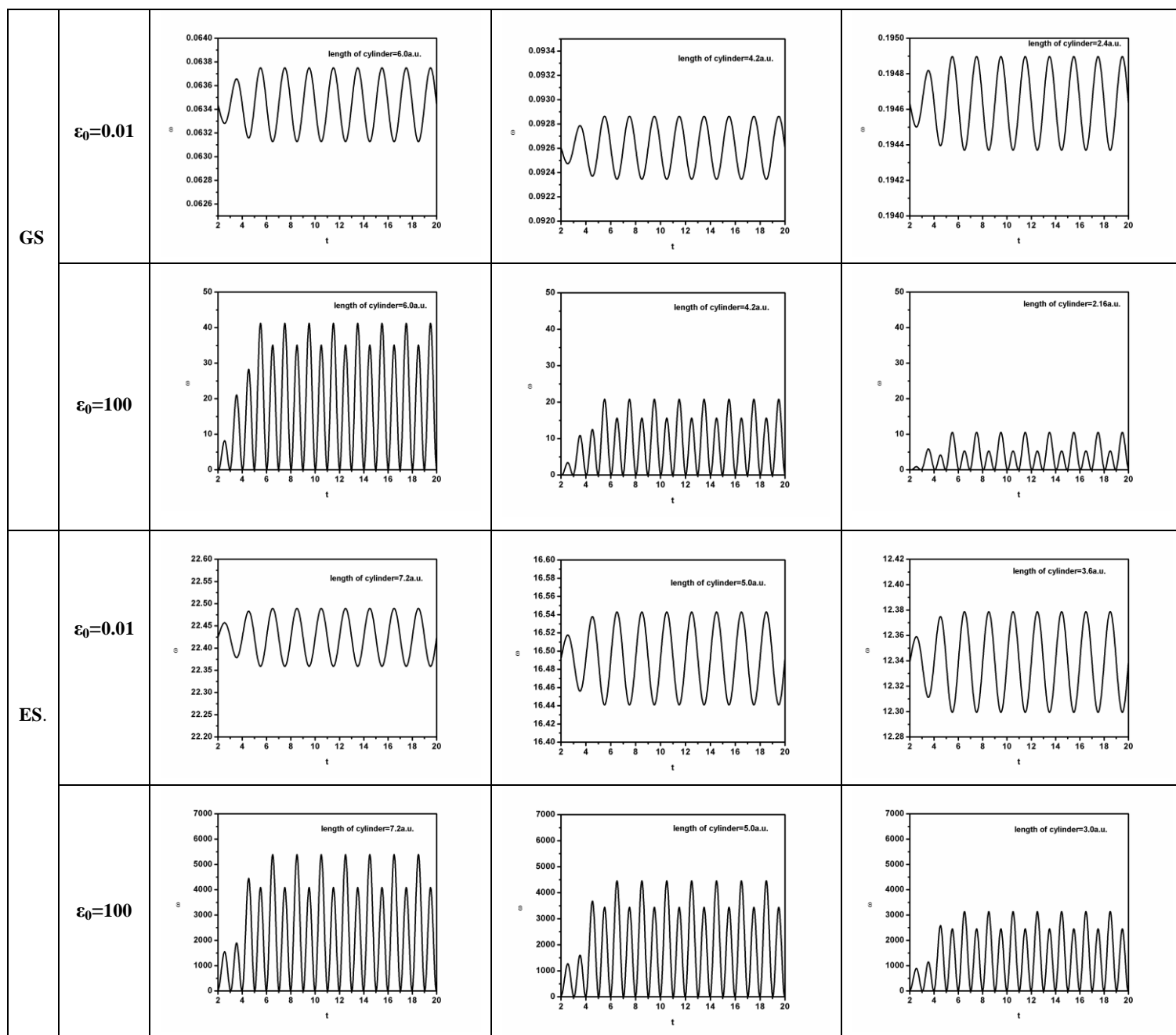
**Figure S18:** Plot of  $\alpha_{\min}$  vs. length of cylinder when CO molecule in GS and ES is placed in an external field ( $\epsilon_0 = 0.01$ a.u. & 100a.u. and  $\omega_0 = \pi$ ).



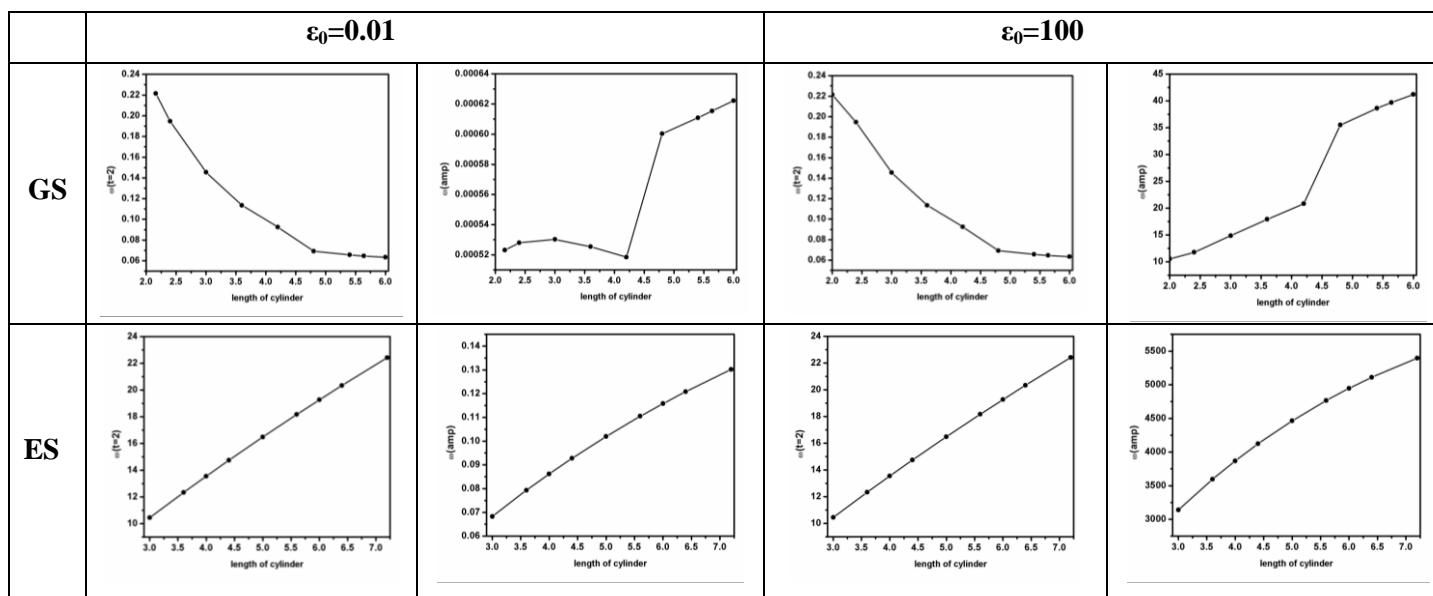
**Figure S19:** Time evolution of chemical potential ( $\mu$ ) when CO molecule in ground state (GS) and excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and frequency  $\omega_0 = \pi$ .



**Figure S20:** Plot of  $\mu(t = 2)$  vs. length of cylinder and  $\mu(\text{amp})$  vs. length of cylinder when CO molecule in GS and ES is placed in an external field ( $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and  $\omega_0 = \pi$ ).



**Figure S21:** Time evolution of electrophilicity index ( $\omega$ ) when CO molecule in ground state (GS) and excited state (ES) is placed in an external field of amplitude  $\epsilon_0 = 0.01$ a.u. & 100a.u. and frequency  $\omega_0 = \pi$ .



**Figure S22:** Plot of  $\omega(t = 2)$  vs. length of cylinder and  $\omega(\text{amp})$  vs. length of cylinder when CO molecule in GS and ES is placed in an external field ( $\epsilon_0 = 0.01\text{a.u.}$  &  $100\text{a.u.}$  and  $\omega_0 = \pi$ ).