

Supplementary Infos

S1 Calculation of the different contributions to free energy of a molecule in the gas phase

Fvib_T, Frot_T, Ftrans_T evaluations The different contributions to free energy Fvib_T, Ftrans_T, Frot_T of DAE and water are calculated considering a perfect gas.⁸⁹

Fvib is defined as Fvib = ZPE + Fvib_T

Where

$$\text{ZPE} = \mathcal{N} \left(\frac{1}{2} \sum_i h \nu_i \right)$$

$$\text{Fvib}_T = \mathcal{N} \sum_i \left(\frac{h \nu_i}{\exp\left(\frac{h \nu_i}{kT}\right) - 1} \right) - \mathcal{N} kT \left[\sum_i \left(\frac{\left(\frac{h \nu_i}{kT}\right)}{\exp\left(\frac{h \nu_i}{kT}\right) - 1} \right) - \ln \left(1 - \exp\left(-\frac{h \nu_i}{kT}\right) \right) \right]$$

where \mathcal{N} is the Avogadro number, k the Boltzmann constant, T the temperature, h the Planck constant, and ν_i the vibrational modes.

The rotational free energy contribution of each reactant/product in the gas phase (water and DAE) was calculated with the formula:

$$\text{Frot}_T = \mathcal{N} kT \left\{ \frac{3}{2} - \left[\ln \left(\frac{\sqrt{\pi}}{\sigma} \left(\frac{8\pi^2 kT}{h^2} \right)^{3/2} \sqrt{A_e \times B_e \times C_e} \right) + \frac{3}{2} \right] \right\}$$

where σ is a symmetry number, and A_e , B_e , C_e are inertial moments with respect to the molecule axis.

The translational free energy of each reactant/product in the gas phase (water and DAE) was calculated with the Sackur-Tetrode formula.⁹⁰⁻⁹²

$$\text{Ftrans}_T = \mathcal{N} kT \left[\frac{3}{2} - (\ln q_t + 1 + 3/2) \right]$$

where

$$q_t = \left(\frac{2\pi m k_B T}{h^2} \right)^{3/2} \frac{k_B T}{P}$$

where m is the molecule (DAE or water) mass.

We calculated, in eV, in standard conditions (T = 298K and P = 1 atm)

	TiO ₂ -4H ₂ O	DAE ^{gas}	H2O ^{gas}
ZPE	2.96	2.91	0.55
Fvib _T	0.32	0.08	0.04
Frot _T		-0.28	-0.11
Ftrans _T		-0.46	-0.41

S2 Thermal Contributions to Free Energy of the adsorbed DAE molecule

When the ‘surface’ includes an adsorbate, $G(\text{surf})$ can be decomposed in two components, $G(\text{mol-ads})$ and $G(\text{surf-noads})$. $G(\text{surf-noads})$ is the contribution of the surface without adsorbate and $G(\text{mol-ads})$ the contribution of the adsorbed molecules. To evaluate free energy balance before and after adsorption, we consider only variations of the component $G(\text{surf}) = G(\text{mol-ads})$, which means that the properties of the adsorbed molecule (vibrational, rotational and translational) are taken into account but not variations of free energy of the bulk phase (TiO_2). In this context, for each adsorbed molecule, we consider that $\text{Frot}(\text{mol-ads})_T = 2/3 \text{Frot}(\text{mol})_{T,\text{gas}}$ when this molecule has no nearest neighbor in the same [010] row and otherwise $\text{Frot}(\text{mol-ads})_T = 0$; thus $\text{Frot}(\text{mol-ads})_T = 2/3 \text{Frot}(\text{mol})_{T,\text{gas}}$, $\text{Frot}(\text{mol-ads})_T = 4/3 \text{Frot}(\text{mol})_{T,\text{gas}}$, $\text{Frot}(\text{mol-ads})_T = 2/3 \text{Frot}(\text{mol})_{T,\text{gas}}$ and $\text{Frot}(\text{mol-ads})_T = 0$ for one, two, three and four adsorbed molecules per cell, respectively. Then we write $\Delta \text{Frot}_{\text{ads}} = \text{Frot}(\text{mol-ads})_T - \text{Frot}(\text{mol})_{T,\text{gas}}$

Similar considerations are valid for the $\text{Ftrans}(\text{surf})$ term. In this case, the translational free energy of adsorbed DAE molecules are calculated considering that $\text{Ftrans}(\text{mol-ads})_T = 2/3 \text{Ftrans}(\text{mol})_{T,\text{gas}}$ for DAE molecules adsorbed at low coverages (1.28 and 2.56 molecule/nm²), and zero in all other cases. Then we write $\Delta \text{Ftrans}_{\text{ads}} = \text{Ftrans}(\text{mol-ads})_T - \text{Ftrans}(\text{mol})_{T,\text{gas}}$. For DAE substituting water molecules, the translational energy is zero.

Table SI: Calculated values (in eV) of the vibrational, translation al and rotational free energies of the different configurations considered. ‘Ads’ refers to DAE adsorbed above a water layer. “Subst” refers to DAE adsorbed in substitution of water molecules.

System and coverage (molecule/m ²)	Fvib(T) (ads)	Frot(ads)	Ftrans(ads)
DAE Ads $\theta=1.28$	0.46	-0.18	-0.30
2 DAE Ads $\theta=2.56$	0.65	-0.37	-0.61
3 DAE ads $\theta=3.84$	0.82	-0.18	0
4 DAE Ads $\theta=5.12$	1.09	0	0
DAE subst $\theta=1.28$	0.28	0	0
2 DAE subst $\theta=2.56$	0.32	0	0
3 DAE subst $\theta=3.84$	0.34	0	0
4 DAE on TiO ₂ $\theta=5.12$	0.54	0	0

S3 Configurational Entropy

Configurational entropy was also considered for each DAE coverage. For each system, the – TS_{config} term to the free energy estimation:

$S_{\text{conf}} = k \ln[(N - n)!/(N!n!)]$ for n sites occupied by DAE over N surface sites.

Applying the Stirling formula yields

$$TS_{\text{config}} = \mathcal{N}kT \ln \left(\left(1 + \frac{n}{N}\right) + \frac{n}{N} \ln \left(1 + \frac{n}{N}\right) \right)$$

TS_{config} can thus be estimated per surface area. This value was found to be negligible in the present system.