

Supporting Information for “Theoretical Prediction of New
Noble-Gas Molecules FNgBNR (Ng = Ar, Kr, and Xe; R = H,
CH₃, CCH, CHCH₂, F, and OH)”

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12 Tables, 3 Figures.

Submitted to *Physical Chemistry Chemical Physics*, March 2013

Table S1 The calculated geometries (in Å) of FArBNR at various methods.

	F–Ar	Ar–B	B–N	N–R
FArBNH				
MP2/aug-cc-pVDZ	2.041	1.794	1.244	1.004
MP2/aug-cc-pVTZ	2.009	1.775	1.236	0.995
B3LYP/aug-cc-pVDZ	2.024	1.831	1.229	1.000
B3LYP/aug-cc-pVTZ	2.018	1.816	1.225	0.994
MPW1PW91/aug-cc-pVDZ	2.000	1.807	1.228	0.998
MPW1PW91/aug-cc-pVTZ	1.991	1.796	1.223	0.992
CCSD(T)/aug-cc-pVTZ	2.019	1.791	1.234	0.996
FArBNCH ₃				
MP2/aug-cc-pVDZ	2.058	1.790	1.246	1.431
MP2/aug-cc-pVTZ	2.024	1.773	1.239	1.419
B3LYP/aug-cc-pVDZ	2.039	1.829	1.229	1.421
B3LYP/aug-cc-pVTZ	2.032	1.815	1.225	1.415
MPW1PW91/aug-cc-pVDZ	2.014	1.805	1.228	1.412
MPW1PW91/aug-cc-pVTZ	2.004	1.796	1.224	1.407
FArBNCCH				
MP2/aug-cc-pVDZ	2.023	1.800	1.254	1.320
MP2/aug-cc-pVTZ	1.991	1.780	1.247	1.308
B3LYP/aug-cc-pVDZ	2.012	1.833	1.242	1.306
B3LYP/aug-cc-pVTZ	2.006	1.817	1.237	1.300
MPW1PW91/aug-cc-pVDZ	1.987	1.808	1.240	1.303
MPW1PW91/aug-cc-pVTZ	1.977	1.797	1.236	1.297
FArBNCHCH ₂				
MP2/aug-cc-pVDZ	2.046	1.792	1.251	1.388
MP2/aug-cc-pVTZ	2.014	1.774	1.244	1.376
B3LYP/aug-cc-pVDZ	2.029	1.830	1.235	1.377
B3LYP/aug-cc-pVTZ	2.022	1.816	1.231	1.370
MPW1PW91/aug-cc-pVDZ	2.004	1.806	1.234	1.371
MPW1PW91/aug-cc-pVTZ	1.995	1.796	1.230	1.365
FArBNF				
MP2/aug-cc-pVDZ	2.021	1.806	1.239	1.312
MP2/aug-cc-pVTZ	1.992	1.785	1.230	1.299
B3LYP/aug-cc-pVDZ	2.014	1.843	1.225	1.307
B3LYP/aug-cc-pVTZ	2.007	1.827	1.219	1.303
MPW1PW91/aug-cc-pVDZ	1.989	1.816	1.222	1.294

MPW1PW91/aug-cc-pVTZ	1.982	1.805	1.217	1.290
		FArBNOH		
MP2/aug-cc-pVDZ	2.044	1.797	1.242	1.336
MP2/aug-cc-pVTZ	2.068	1.808	1.242	1.336
B3LYP/aug-cc-pVDZ	2.030	1.836	1.227	1.331
B3LYP/aug-cc-pVTZ	2.024	1.820	1.222	1.328
MPW1PW91/aug-cc-pVDZ	2.006	1.811	1.225	1.319
MPW1PW91/aug-cc-pVTZ	1.998	1.800	1.220	1.316

Table S2 The calculated geometries (in Å) of FKrBNR at various methods.

	F–Kr	Kr–B	B–N	N–R
FKrBNH				
MP2/aug-cc-pVDZ	2.082	1.961	1.248	1.003
MP2/aug-cc-pVTZ	2.059	1.937	1.240	0.995
B3LYP/aug-cc-pVDZ	2.084	1.995	1.233	1.000
B3LYP/aug-cc-pVTZ	2.075	1.983	1.228	0.993
MPW1PW91/aug-cc-pVDZ	2.058	1.971	1.231	0.997
MPW1PW91/aug-cc-pVTZ	2.047	1.964	1.227	0.992
CCSD(T)/aug-cc-pVTZ	2.064	1.953	1.238	0.995
FKrBNCH ₃				
MP2/aug-cc-pVDZ	2.097	1.958	1.250	1.430
MP2/aug-cc-pVTZ	2.073	1.934	1.242	1.417
B3LYP/aug-cc-pVDZ	2.095	1.993	1.233	1.420
B3LYP/aug-cc-pVTZ	2.087	1.982	1.228	1.414
MPW1PW91/aug-cc-pVDZ	2.070	1.970	1.231	1.411
MPW1PW91/aug-cc-pVTZ	2.059	1.962	1.227	1.406
FKrBNCCH				
MP2/aug-cc-pVDZ	2.068	1.966	1.258	1.319
MP2/aug-cc-pVTZ	2.045	1.940	1.250	1.307
B3LYP/aug-cc-pVDZ	2.071	1.995	1.245	1.306
B3LYP/aug-cc-pVTZ	2.062	1.983	1.241	1.299
MPW1PW91/aug-cc-pVDZ	2.045	1.972	1.244	1.302
MPW1PW91/aug-cc-pVTZ	2.033	1.964	1.239	1.296
FKrBNCHCH ₂				
MP2/aug-cc-pVDZ	2.086	1.960	1.255	1.386
MP2/aug-cc-pVTZ	2.063	1.935	1.247	1.373
B3LYP/aug-cc-pVDZ	2.086	1.994	1.239	1.375
B3LYP/aug-cc-pVTZ	2.078	1.982	1.234	1.368
MPW1PW91/aug-cc-pVDZ	2.059	1.970	1.237	1.369
MPW1PW91/aug-cc-pVTZ	2.050	1.962	1.233	1.363
FKrBNF				
MP2/aug-cc-pVDZ	2.066	1.971	1.243	1.315
MP2/aug-cc-pVTZ	2.045	1.945	1.234	1.302
B3LYP/aug-cc-pVDZ	2.073	2.002	1.228	1.310
B3LYP/aug-cc-pVTZ	2.064	1.990	1.222	1.306
MPW1PW91/aug-cc-pVDZ	2.048	1.978	1.226	1.297

MPW1PW91/aug-cc-pVTZ	2.037	1.970	1.220	1.293
FKrBNOH				
MP2/aug-cc-pVDZ	2.085	1.963	1.247	1.339
MP2/aug-cc-pVTZ	2.063	1.938	1.238	1.328
B3LYP/aug-cc-pVDZ	2.088	1.998	1.230	1.333
B3LYP/aug-cc-pVTZ	2.079	1.985	1.225	1.330
MPW1PW91/aug-cc-pVDZ	2.062	1.973	1.229	1.321
MPW1PW91/aug-cc-pVTZ	2.052	1.965	1.224	1.318

Table S3 The calculated geometries (in Å) of FXeBNR at various methods.

	F–Xe	Xe–B	B–N	N–R
FXeBNH				
MP2/aug-cc-pVDZ	2.153	2.154	1.253	1.003
MP2/aug-cc-pVTZ	2.129	2.131	1.244	0.994
B3LYP/aug-cc-pVDZ	2.161	2.187	1.237	0.999
B3LYP/aug-cc-pVTZ	2.145	2.179	1.232	0.993
MPW1PW91/aug-cc-pVDZ	2.137	2.165	1.235	0.997
MPW1PW91/aug-cc-pVTZ	2.118	2.161	1.231	0.991
CCSD(T)/aug-cc-pVTZ	2.128	2.148	1.242	0.994
FXeBNCH ₃				
MP2/aug-cc-pVDZ	2.163	2.150	1.255	1.428
MP2/aug-cc-pVTZ	2.139	2.128	1.246	1.416
B3LYP/aug-cc-pVDZ	2.171	2.184	1.236	1.418
B3LYP/aug-cc-pVTZ	2.154	2.177	1.232	1.412
MPW1PW91/aug-cc-pVDZ	2.145	2.163	1.235	1.410
MPW1PW91/aug-cc-pVTZ	2.126	2.159	1.231	1.404
FXeBNCCH				
MP2/aug-cc-pVDZ	2.142	2.152	1.263	1.317
MP2/aug-cc-pVTZ	2.118	2.129	1.254	1.305
B3LYP/aug-cc-pVDZ	2.149	2.185	1.250	1.304
B3LYP/aug-cc-pVTZ	2.132	2.178	1.245	1.298
MPW1PW91/aug-cc-pVDZ	2.124	2.163	1.248	1.301
MPW1PW91/aug-cc-pVTZ	2.105	2.159	1.244	1.295
FXeBNCHCH ₂				
MP2/aug-cc-pVDZ	2.155	2.150	1.260	1.383
MP2/aug-cc-pVTZ	2.131	2.128	1.251	1.371
B3LYP/aug-cc-pVDZ	2.162	2.184	1.243	1.373
B3LYP/aug-cc-pVTZ	2.146	2.177	1.238	1.366
MPW1PW91/aug-cc-pVDZ	2.137	2.163	1.242	1.367
MPW1PW91/aug-cc-pVTZ	2.118	2.159	1.237	1.361
FXeBNF				
MP2/aug-cc-pVDZ	2.142	2.158	1.248	1.318
MP2/aug-cc-pVTZ	2.118	2.135	1.238	1.304
B3LYP/aug-cc-pVDZ	2.151	2.192	1.232	1.313
B3LYP/aug-cc-pVTZ	2.134	2.184	1.227	1.308
MPW1PW91/aug-cc-pVDZ	2.126	2.169	1.230	1.300

MPW1PW91/aug-cc-pVTZ	2.108	2.165	1.225	1.295
FXeBNOH				
MP2/aug-cc-pVDZ	2.155	2.153	1.251	1.342
MP2/aug-cc-pVTZ	2.130	2.130	1.242	1.331
B3LYP/aug-cc-pVDZ	2.164	2.187	1.235	1.336
B3LYP/aug-cc-pVTZ	2.148	2.178	1.230	1.332
MPW1PW91/aug-cc-pVDZ	2.139	2.164	1.233	1.323
MPW1PW91/aug-cc-pVTZ	2.121	2.160	1.228	1.319

Table S4 The calculated three-body dissociation energies (in kcal/mol) of FNgBNR → F + Ng + BNR

	Ar	Kr	Xe
		R=H	
MP2/aug-cc-pVDZ	12.2	32.0	58.2
MP2/aug-cc-pVTZ	20.3	41.0	67.1
B3LYP/aug-cc-pVDZ	15.3	31.1	53.2
B3LYP/aug-cc-pVTZ	16.9	33.1	55.4
MPW1PW91/aug-cc-pVDZ	11.6	28.6	51.6
MPW1PW91/aug-cc-pVTZ	12.9	30.3	53.5
CCSD(T)/aug-cc-pVTZ	10.1	30.7	66.4
CCSD(T)/aug-cc-pVQZ ^a	11.5	31.8	59.6
		R=CH ₃	
MP2/aug-cc-pVDZ	19.3	35.1	61.0
MP2/aug-cc-pVTZ	23.6	44.0	69.8
B3LYP/aug-cc-pVDZ	17.2	32.5	54.0
B3LYP/aug-cc-pVTZ	18.7	34.4	56.2
MPW1PW91/aug-cc-pVDZ	11.6	29.9	52.5
MPW1PW91/aug-cc-pVTZ	14.6	31.5	54.4
CCSD(T)/aug-cc-pVTZ ^b	11.3	32.6	58.6
		R=CCH	
MP2/aug-cc-pVDZ	17.0	37.9	65.5
MP2/aug-cc-pVTZ	23.6	45.5	72.9
B3LYP/aug-cc-pVDZ	13.0	29.7	52.7
B3LYP/aug-cc-pVTZ	14.5	31.5	54.8
MPW1PW91/aug-cc-pVDZ	9.0	26.9	51.1
MPW1PW91/aug-cc-pVTZ	10.3	28.5	52.9
CCSD(T)/aug-cc-pVTZ ^b	8.5	30.3	58.0
		R=CHCH ₂	
MP2/aug-cc-pVDZ	17.0	39.3	65.8
MP2/aug-cc-pVTZ	26.4	47.4	73.9
B3LYP/aug-cc-pVDZ	15.5	31.3	53.3
B3LYP/aug-cc-pVTZ	17.0	33.2	55.5
MPW1PW91/aug-cc-pVDZ	9.0	28.6	51.7
MPW1PW91/aug-cc-pVTZ	12.9	30.3	53.6
CCSD(T)/aug-cc-pVTZ ^b	8.5	32.2	58.8
		R=F	

MP2/aug-cc-pVDZ	15.8	31.3	58.6
MP2/aug-cc-pVTZ	18.7	40.4	67.6
B3LYP/aug-cc-pVDZ	12.7	29.3	52.2
B3LYP/aug-cc-pVTZ	14.5	31.5	54.6
MPW1PW91/aug-cc-pVDZ	13.4	26.8	50.7
MPW1PW91/aug-cc-pVTZ	10.6	28.7	52.9
CCSD(T)/aug-cc-pVTZ ^b	12.2	28.8	56.2
R=OH			
MP2/aug-cc-pVDZ	14.3	34.3	60.8
MP2/aug-cc-pVTZ	22.4	43.4	69.9
B3LYP/aug-cc-pVDZ	15.4	31.3	53.4
B3LYP/aug-cc-pVTZ	17.2	33.5	55.8
MPW1PW91/aug-cc-pVDZ	11.7	28.8	51.9
MPW1PW91/aug-cc-pVTZ	13.2	30.6	54.1
CCSD(T)/aug-cc-pVTZ ^b	10.1	31.0	57.7

^aSingle-point energies using the CCSD(T)/aptz geometry.

^bSingle-point energies using the MPW1PW91/apdz geometry.

CCSD(T)/CBS calculation of the H–C bond energies

Table S4 Calculated hemolytic bond dissociation energies ^a (in kcal/mol)

	H–CCH	H–CN
CCSD(T)/aug-cc-pVQZ	139.4	132.8
CCSD(T)/aug-cc-pV5Z	139.7	133.0
CCSD(T)/CBS	140.9	133.3
exp. ^b	136.8	129.4

^a Born-Oppenheimer energies, not including zero-point and thermal energies.

^b Obtained using the heats of formation from NIST Chemistry WebBook (Ref. S1) and the calculated zero-point and thermal energies at the CCSD(T)/aptz level.

Comparison of the bond energies of RCC–H, RNB–H and RNB–NgF

As seen in the following table, the bond energies are fairly constant regardless the identities of the R groups for all five types of molecules listed in the following table with the only exception of HCCOH. The bond energies of H–CCR are ~20 kcal/mol higher than those of H–BNR. The bond energies of F–Ng–BNR are significantly lower. It is interesting to note that the bond energies are predicted to be slightly more sensitive to the identities R groups for FNgBNR. It is difficult to say whether the trends in bond energies parallel or not among these five series of molecules since the bond energies in a series are all similar and the differences were probably within the calculation errors.

Table S5 The calculated bond dissociation energies (in kcal/mol) by CCSD(T)/aug-cc-pVQZ level.

	R=H ^a	R=CH ₃ ^b	R=CCH ^b	R=CHCH ₂ ^b	R=F ^b	R=OH ^b
H–BNR	116.3	116.3	117.8	117.3	116.6	116.4
H–CCR	139.5	140.1	141.1	138.5	140.8	132.9
F–Ar–BNR	11.5	13.6	9.8	12.6	8.5	11.5
F–Kr–BNR	31.8	33.5	31.1	33.0	29.7	31.9
F–Xe–BNR	59.8	61.1	60.3	61.3	58.0	60.0

^a Single-point energies using the CCSD(T)/aptz geometry.

^b Single-point energies using the MPW1PW91/apdz geometry.

Comparison of the bond lengths of RNBH, RCCH, and RNBNGF.

As seen in the following tables, the H–B bonds are ~ 0.1 Å longer than the H–C bonds. The H–B distance is almost independent of the R groups while the H–C distances show slightly more variation. The B–N distances are 0.02–0.03 Å longer than the C–C distances. Both bonds show only very small variations (< 0.02 Å) with different R groups. The C–R and N–R bonds show much stronger dependence on the identities of the R groups. Most importantly, the BN–CCH bond was predicted to be very short (1.30 Å), similar to a double bond, but the CC–CCH bond was predicted to be considerably longer (1.37 Å). Also the BN–CH₃ (1.41 Å) and BN–CHCH₂ (1.36 Å) bonds are significantly shorter than the CC–CH₃ (1.46 Å) and CC–CHCH₂ (1.42 Å) bonds. However, the BN–F (1.30 Å) and BN–OH (1.33 Å) bonds were predicted to be slightly longer (by 0.02 Å) than the CC–F (1.28 Å) and CC–OH (1.31 Å) bonds. It is obvious that although CC and BN are isoelectronic, *the nature of these two types of double bonds is different* and makes the trends in the relative bond lengths between the BN–R and CC–R not very uniform. On the other hand, the bond length trends of BN–R or CC–R within each series are quite similar.

If we compare the N–R bond lengths in HBNR and FNGBNR (see Figure 2), we found the *trends parallel*. The N–H bonds of course are both the shortest (~ 1.00 Å). For other R groups, the N–F and N–CCH bonds are the shortest N–R bonds in both HBNR and FNGBNR with bond length around 1.30 Å. The N–CH₃ are the longest N–C bonds in both HBNR and FNGBNR with bond length around 1.41 Å. In addition, for both sets of molecules, the N–CHCH₂ bond lengths (~ 1.37 Å) are between the N–CCH and N–CH₃. These trends indicated that in FNGBNR molecules the FNg– part does not seem to have significant effects on the bonding properties of the BN–R.

Table S6 The calculated geometries (in Å) of HBNR.

	R(H–B)	R(B–N)	R(N–R)
R=H ^a	1.170	1.244	0.993
R=CH ₃ ^b	1.173	1.238	1.408
R=CCH ^b	1.171	1.249	1.301
R=CHCH ₂ ^b	1.173	1.243	1.364
R=F ^b	1.173	1.231	1.304
R=OH ^b	1.173	1.234	1.328

^aThe calculated geometries were obtained at the CCSD(T)/aptz level.

^bThe calculated geometries were obtained at theMPW1PW91/apdz level.

Table S7 The calculated geometries (in Å) of HCCR.

	R(H–C)	R(C–C)	R(C–R)
R=H ^a	1.064	1.210	1.064
R=CH ₃ ^b	1.075	1.222	1.460
R=CCH ^b	1.069	1.215	1.370
R=CHCH ₂ ^b	1.069	1.214	1.424
R=F ^b	1.062	1.203	1.283
R=OH ^b	1.068	1.208	1.307

^aThe calculated geometries were obtained at the CCSD(T)/aptz level.

^bThe calculated geometries were obtained at theMPW1PW91/apdz level.

Table S8 Calculated reaction energies (in kcal/mol) of FNgBNR → Ng + FBNR

	Ar	Kr	Xe
R=H			
MPW1PW91/apdz	−142.3	−125.3	−102.2
MP2/apdz	−146.1	−126.3	−100.0
CCSD(T)/aptz ^a	−148.9	−128.2	−101.8
R=CH ₃			
MPW1PW91/apdz	−139.1	−122.6	−100.0
MP2/apdz	−142.4	−123.1	−97.2
CCSD(T)/aptz ^a	−145.4	−125.1	−99.1
R=CCH			
MPW1PW91/apdz	−144.1	−126.2	−102.0
MP2/apdz	−147.6	−126.6	−99.1
CCSD(T)/aptz ^a	−150.7	−128.9	−101.2
R=CHCH ₂			
MPW1PW91/apdz	−141.3	−124.3	−101.2
MP2/apdz	−144.6	−124.6	−98.1
CCSD(T)/aptz ^a	−147.6	−126.7	−100.0
R=F			
MPW1PW91/apdz	−140.4	−122.6	−98.7
MP2/apdz	−144.7	−123.8	−96.5
CCSD(T)/aptz ^a	−147.4	−125.8	−98.3
R=OH			
MPW1PW91/apdz	−138.2	−121.1	−98.0
MP2/apdz	−142.2	−122.1	−95.6
CCSD(T)/aptz ^a	−145.0	−124.0	−97.3

^aSingle-point energies using the MPW1PW91/apdz geometry. The level of theory is the same as the CCSD(T)/aptz in Table 2 and approximately the same as the CCSD(T)/aptz in Table 1 in which the geometry was also calculated at the CCSD(T)/aptz level.

Table S9 Calculated barrier heights (in kcal/mol) of FNgBNR → Ng + FBNR

	Ar	Kr	Xe
		R=H	
MP2/apdz	15.0	23.7	33.3
MP2/aptz	15.4	23.3	31.5
B3LYP/apdz	20.4	27.9	35.8
B3LYP/aptz	19.5	26.8	33.8
MPW1PW91/apdz	20.7	28.5	36.6
MPW1PW91/aptz	19.9	27.5	34.7
CCSD(T)/aptz	15.9	23.9	32.5
CCSD(T)/apqz ^a	16.0	24.1	32.3
CCSD(T)/ap5z ^a	16.1	24.1	32.7
		R=CH ₃	
MP2/apdz	13.9	22.4	32.0
MP2/aptz	14.5	22.2	30.3
B3LYP/apdz	19.1	26.6	34.5
B3LYP/aptz	18.4	25.6	32.7
MPW1PW91/apdz	19.5	27.2	35.4
MPW1PW91/aptz	18.9	26.4	33.7
CCSD(T)/aptz ^b	14.8	22.7	31.2
		R=CCH	
MP2/apdz	16.0	25.0	34.7
MP2/aptz	16.5	24.6	32.6
B3LYP/apdz	21.0	28.9	36.9
B3LYP/aptz	20.2	27.7	34.9
MPW1PW91/apdz	21.4	29.5	37.8
MPW1PW91/aptz	20.7	28.5	35.9
CCSD(T)/aptz ^b	16.0	25.1	33.5
		R=CHCH ₂	
MP2/apdz	14.4	23.0	32.8
MP2/aptz	14.9	22.8	30.9
B3LYP/apdz	19.6	27.3	35.2
B3LYP/aptz	18.9	26.2	33.4
MPW1PW91/apdz	20.2	27.9	36.2
MPW1PW91/aptz	19.4	27.0	34.4
CCSD(T)/aptz ^b	15.3	23.4	41.3
		R=F	

MP2/apdz	16.3	25.4	35.2
MP2/aptz	16.7	24.8	33.2
B3LYP/apdz	21.8	29.6	37.5
B3LYP/aptz	20.9	28.4	35.6
MPW1PW91/apdz	22.1	30.1	38.3
MPW1PW91/aptz	21.3	29.0	36.4
CCSD(T)/aptz ^b	17.0	25.5	34.2
R=OH			
MP2/apdz	19.4	27.2	35.5
MP2/aptz	14.5	22.3	30.7
B3LYP/apdz	19.1	26.7	34.7
B3LYP/aptz	18.3	25.6	32.9
MPW1PW91/apdz	14.0	22.6	32.4
MPW1PW91/aptz	18.7	26.3	33.8
CCSD(T)/aptz ^b	14.8	22.9	31.6

^aSingle-point energies using the CCSD(T)/aptz geometry.

^bSingle-point energies using the MPW1PW91/apdz geometry.

Stability comparison of the FNgBNH and FNgNBH

Table S10 Calculated Relative Energies (kcal/mol) by CCSD(T)/aug-cc-pVQZ level.^a

	FNgBNH	FNgNBH
Ng= Ar	0.0	41.4
Ng= Kr	0.0	31.0
Ng= Xe	0.0	19.1

^aSingle-point energies using the CCSD(T)/aptz geometry for FNgBNH and using MPW1PW91/apdz geometry for FNgNBH.

Table S11 Calculated bond dissociation energies and barrier heights (in kcal/mol) of FNgBNH and FNgNBH isomers

	FNgBNH		FNgNBH	
	three-body dissociation energy ^a	Barrier height ^b	three-body dissociation energy ^b	Barrier height ^b
Ng=Ar	11.5	15.7	−9.9	NA ^c
Ng=Kr	31.8	23.8	20.7	48.6
Ng=Xe	59.6	32.4	60.5	60.5

^aCCSD(T)/apqz single-point energies using the CCSD(T)/aptz geometry.

^bCCSD(T)/aptz single-point energies using the MPW1PW91/apdz geometry.

^cIn CCSD(T)/aptz single-point energy calculations, the energy of transitions state was lower than that of the reactant.

Table S12 Calculated atomic charges^a of FNgCCH by ChelpG and NBO (in parentheses) methods, in atomic unit, *e*.

	F	Ng	C	C	H
FArCCH	−0.493 (−0.629)	0.542 (0.722)	−0.057 (−0.122)	−0.253 (−0.212)	0.261 (0.241)
FKrCCH	−0.484 (−0.667)	0.594 (0.884)	−0.104 (−0.239)	−0.273 (−0.217)	0.268 (0.239)
FXeCCH	−0.451 (−0.710)	0.609 (1.071)	−0.151 (−0.380)	−0.259 (−0.216)	0.253 (0.234)

^aUsing electron density calculated at the MP2/aug-cc-pVTZ level

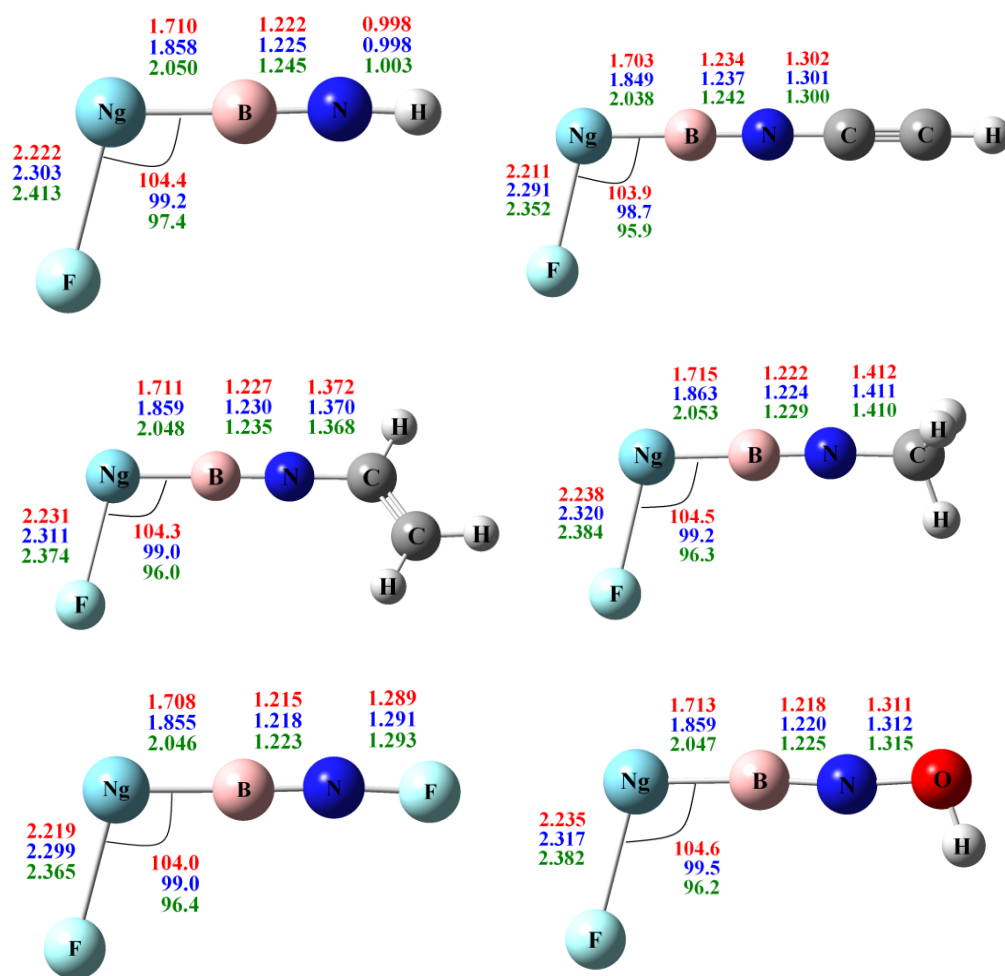


Figure S1 Calculated geometry of the transition states of the two-body dissociation of FNgBNR at the MPW1PW91/aug-cc-pVDZ level. The values in red, blue, and green colors are for Ng = Ar, Kr, and Xe, respectively. The bond lengths are in angstroms, and the bond angles in degrees.

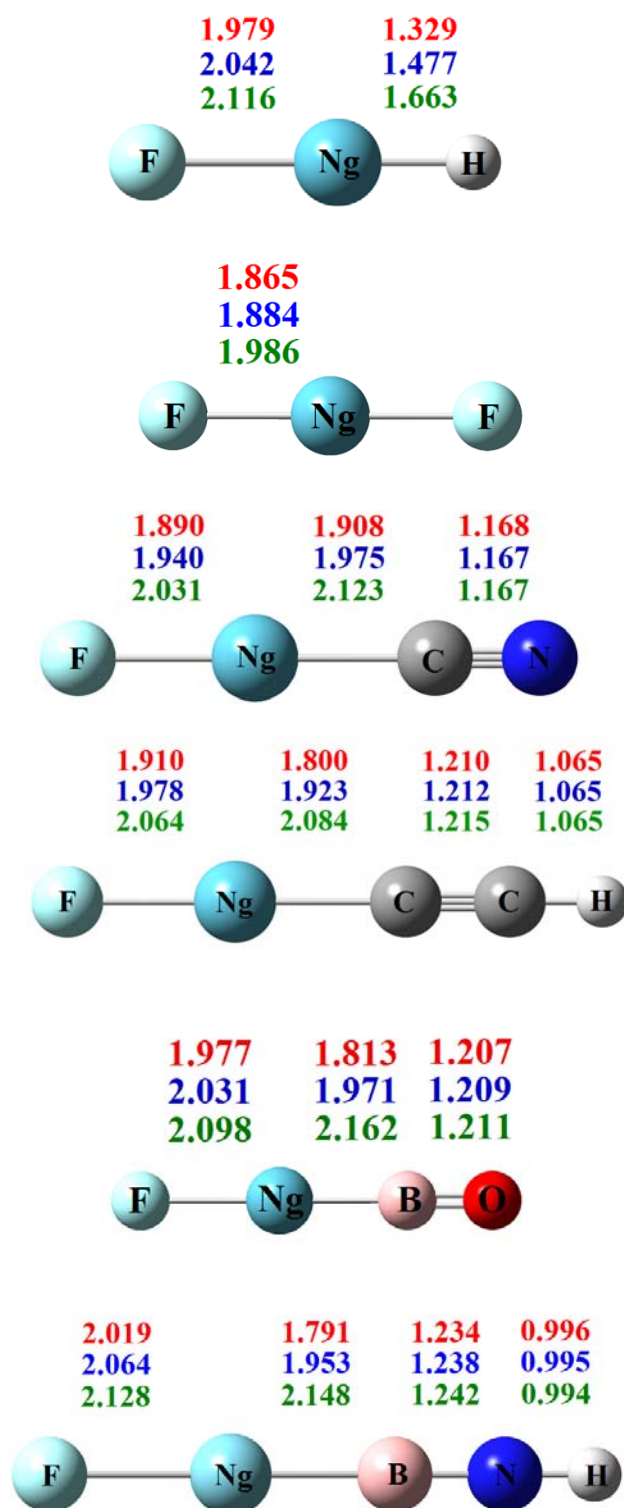


Figure S2 Calculated geometry of the FNgY at the CCSD(T)/aug-cc-pVTZ level. From top Y= BNH, H, CN, and CCH. The values in red, blue, and green colors are for Ng = Ar, Kr, and Xe, respectively. The bond lengths are in angstroms. All of molecules are linear. The experimental bond length of KrF₂ and XeF₂ were 1.875 and 1.977 Å, respectively.

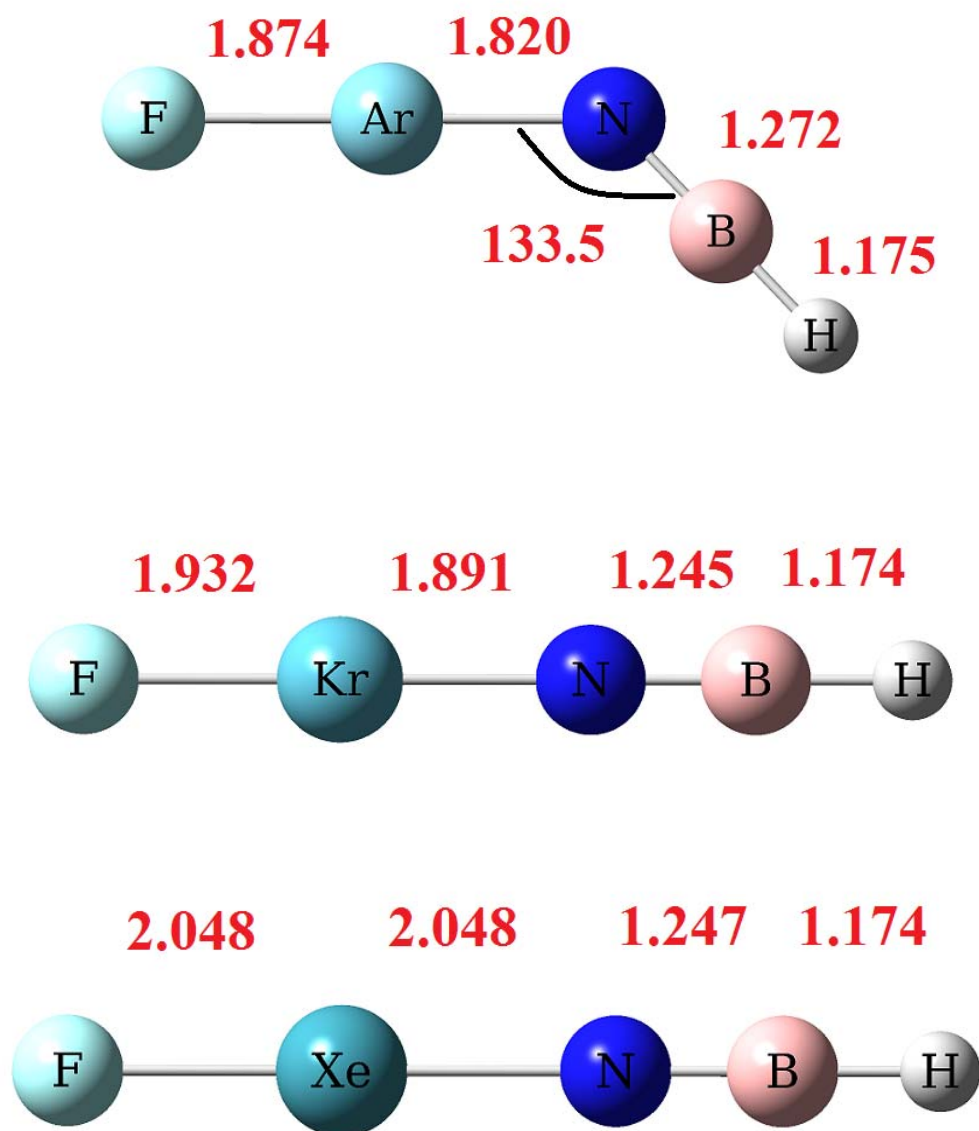


Figure S3 Calculated geometry of the FNgNBH at the MPW1PW91/aug-cc-pVDZ level. The bond lengths are in angstroms, and the bond angles in degrees.

References

- S1. The NIST Chemistry WebBook, <http://webbook.nist.gov/chemistry/>