

Synthesis and electrocatalytic activity of highly porous hollow palladium nanoshells for oxygen reduction in alkaline solution

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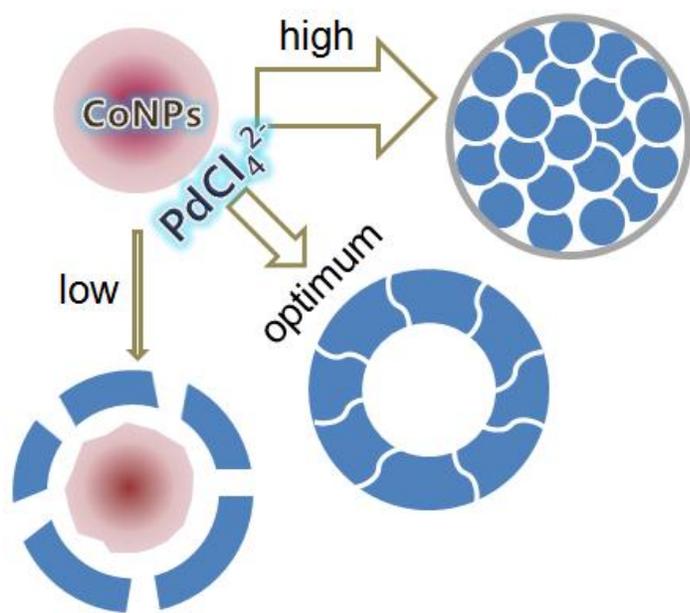


Fig. S1. Schematic illustration of the hPd nanoshells prepared at different concentrations of palladium precursor.

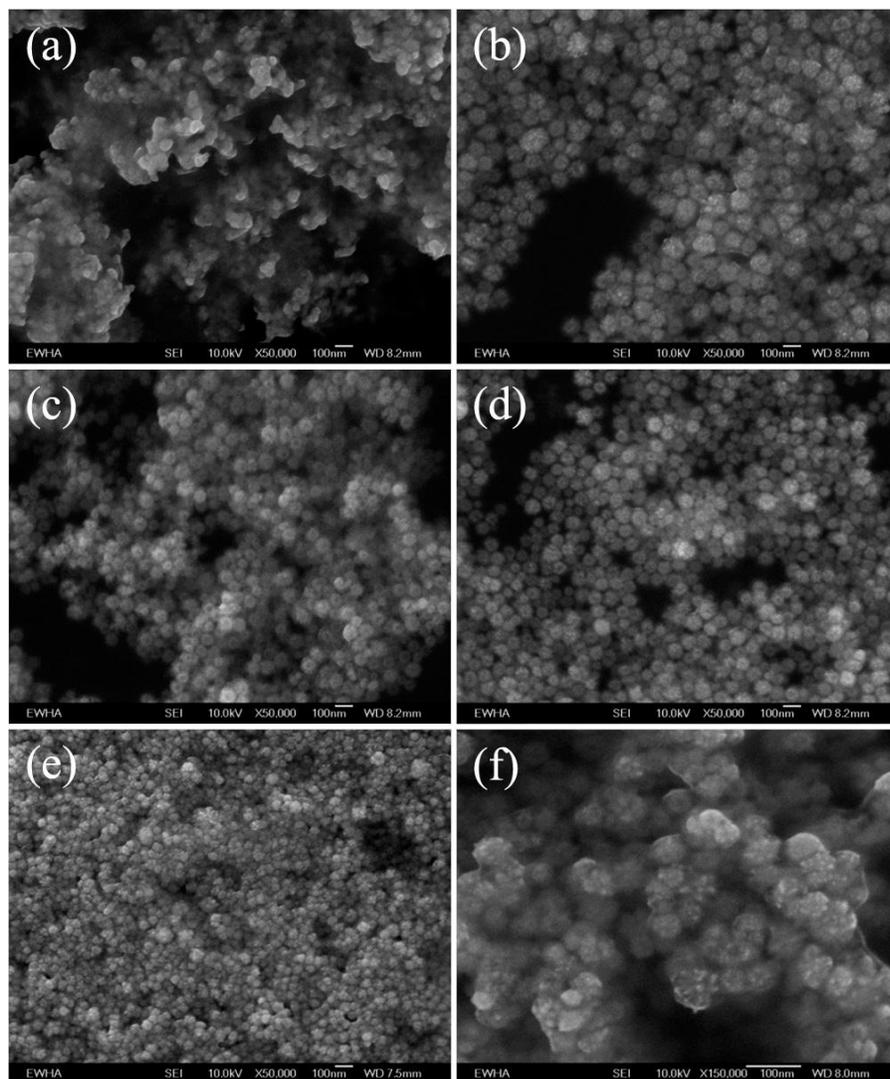


Fig. S2. Representative SEM images of palladium nanoshells obtained by reacting cobalt nanoparticles with an aqueous K_2PdCl_4 solution ((a) 1 mM, (b) 6 mM, (c) 12 mM, (d) 20 mM, and (e and f) 40 mM). Panel (f) is a higher-magnified view of a sample (e).

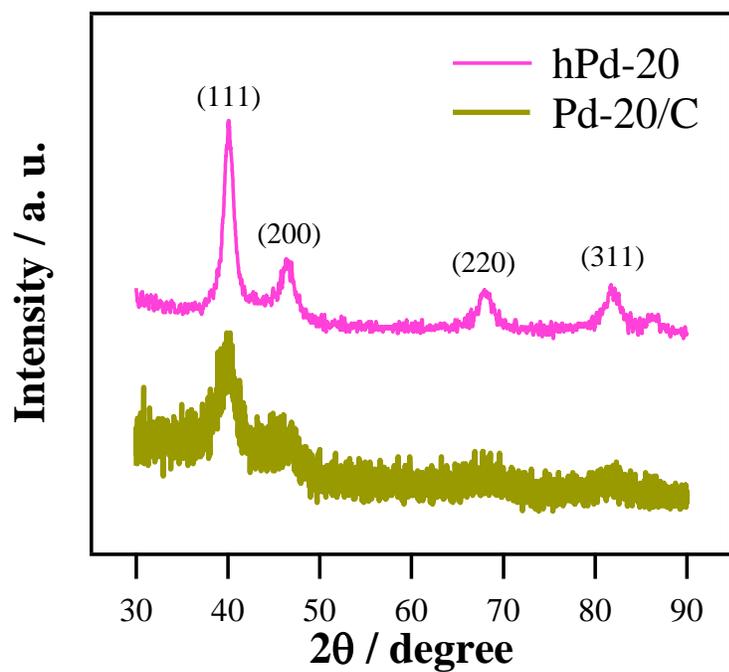


Fig. S3. XRD patterns of hPd-20 and commercial Pd-20/C.

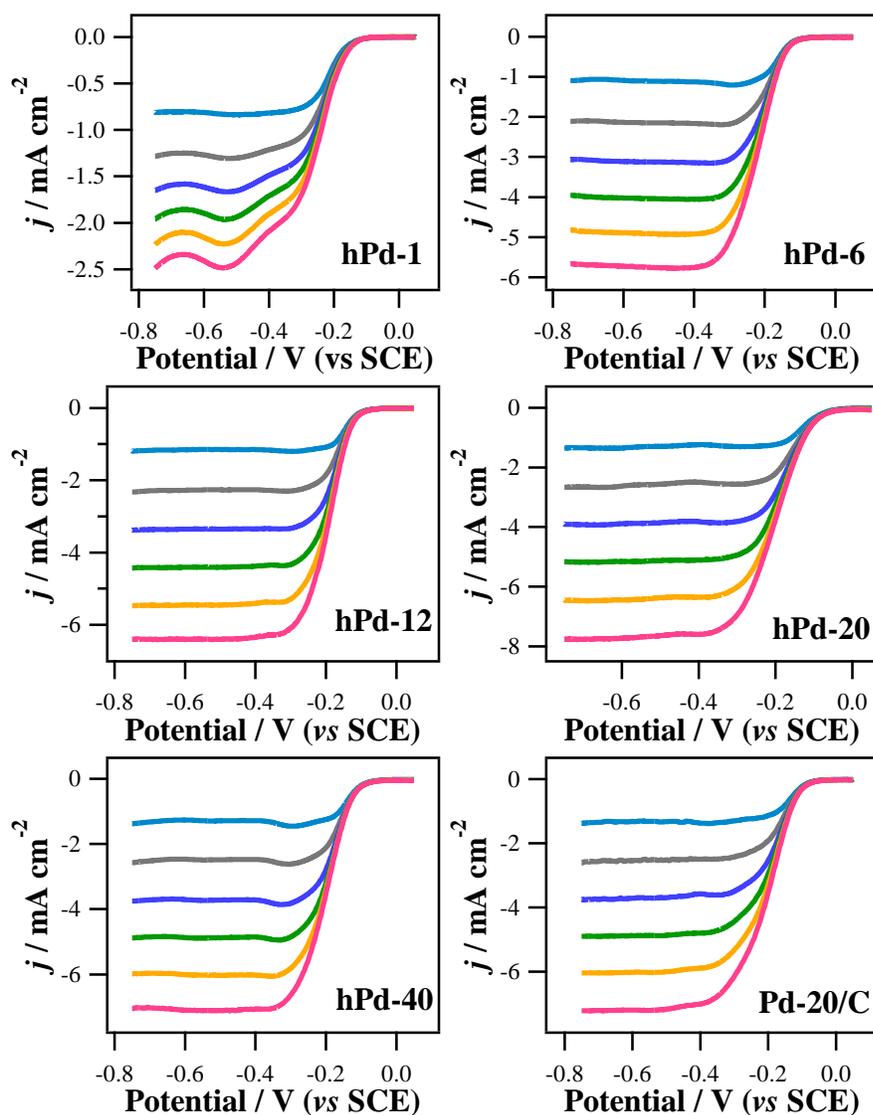


Fig. S4. RDE voltammograms for oxygen reduction in an O_2 -saturated 0.1 M NaOH at the hPd-1, hPd-6, hPd-12, hPd-20, hPd-40, and Pd-20/C loaded GC electrodes at a scan rate of 10 mV s^{-1} with a rotation speed of 100, 400, 900, 1600, 2500 and 3600 rpm.

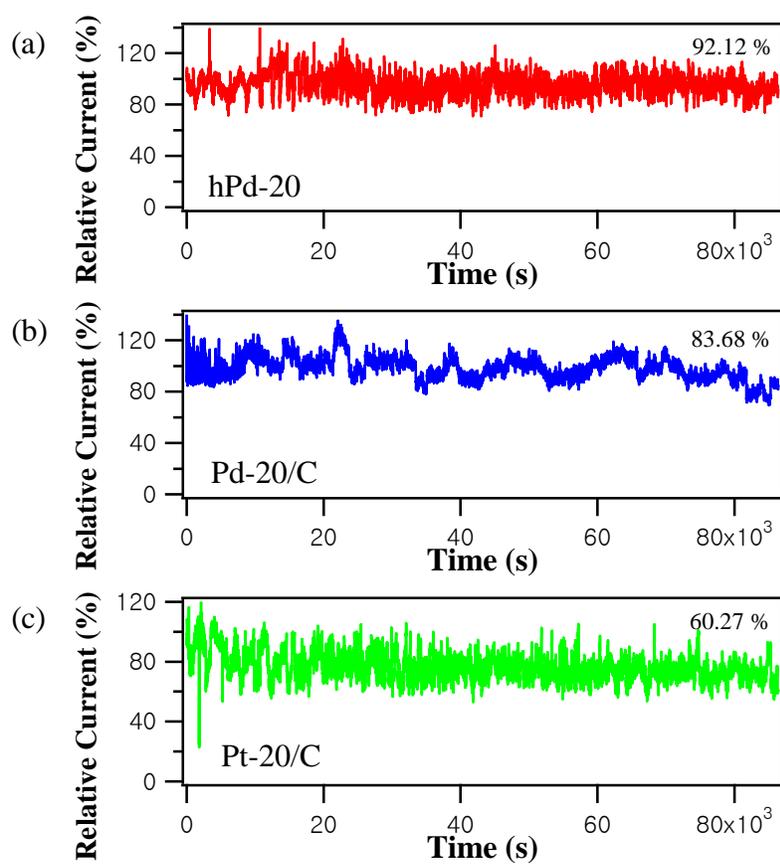


Fig. S5. The steady-state currents measured continuously for hPd-20, Pd-20/C, and Pt-20/C-loaded GC electrodes at -0.5 V vs. SCE in an O_2 -saturated 0.1 M NaOH solution, magnetically being stirred.

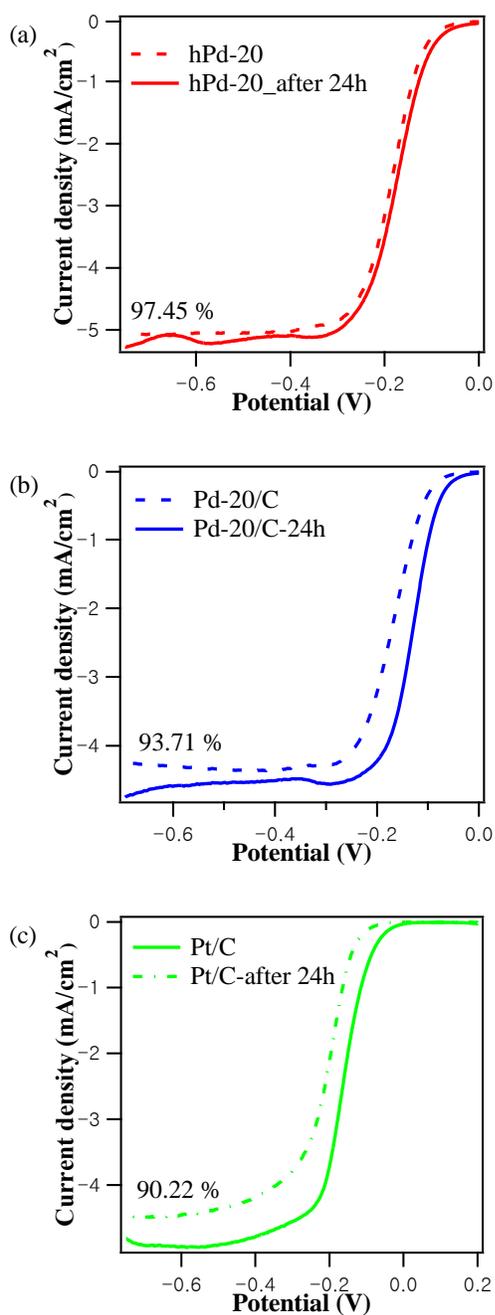


Fig. S6. RDE voltammograms before and after 24-h continuous chronoamperometric measurements (Fig. S5) obtained with hPd-20 (a), Pd-20/C (b), and Pt-20/C (c) -modified GC electrode in an O₂-saturated 0.1 M NaOH solution at a rotation rate of 1600 rpm and a scan rate of 10 mV s⁻¹.

Table S1. Mass activities of hPd-20 with and without carbon support, commercial Pd and Pt catalysts.

	Current (mA)^a	Relative current (%)	Mass activity (mA $\mu\text{g}_{\text{metal}}^{-1}$)
hPd-20	-0.580	100	0.0290
hPd/C-20wt%	-0.494	85.2	0.124
hPd/C-15wt%	-0.385	66.4	0.128
Pd-20/C	-0.422	72.8	0.106
Pt-20/C	-0.388	66.9	0.0970

^aThe values are the limiting currents measured at -0.7 V vs. SCE in the RDE curve for ORR obtained with a rotation speed of 1600 rpm.