## **Supporting Information**

# Graphene-based 3D Composite Hydrogel by Anchoring Co<sub>3</sub>O<sub>4</sub> Nanoparticles with Enhanced Electrochemical Properties

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#### <sup>10</sup> SI-1. Electrochemical Measurement with a two-electrode system

The electrochemical measurements of as-obtained products were conducted using a two-electrode system for comparison.

The detail procedure of experiment is as following: First, the water of prepared Co<sub>3</sub>O<sub>4</sub>/graphene hydrogel (CGH) composites was exchanged with 6 M KOH electrolyte for 2 h before tests. Then two <sup>15</sup> pieces of CGH were attached on platinum foils, the symmetry parts were assembled with a neutral paper sandwiched between the hydrogel. A platinum wire was clipped onto the foil by a clip at the end of each platinum foil, which was connected with CHI 660B for CV and galvanostatic charge-discharge measurements between 0 V and 1 V. The specific capacitance, energy density and maximum peak power density of the electrodes were calculated according to the equation reported previously<sup>[1]</sup>.

### 20 SI-2. Additional electrochemical data of CGH and GH with a two-electrode system



Figure S1 (a) CV measurements of CGH and GH at 5 mV $\cdot$ s<sup>-1</sup>; (b) Galvanostatic charge/discharge tests

of CGH and GH at 0.5  $A \cdot g^{-1}$ .



<sup>5</sup> Figure S2 (a) Gravimetric capacitances of CGH and GH measured at various current densities; (b)

Ragone plots of CGH and GH.

SI-3. Nyquist plots of CGH, Co<sub>3</sub>O<sub>4</sub> and GH with a three-electrode system



Figure S3 the Nyquist plots of CGH, Co<sub>3</sub>O<sub>4</sub> and GH.

#### References

[1] X. W. Yang, J. W. Zhu, L. Qiu, D. Li. Adv. Mater., 23 (2011) 2833–2838.