### Gas-Phase Salt Bridge Interactions between Glutamic Acid and Arginine

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Figure SI-1. Influence of diffuse functions on the calculated spectra of the systems investigated in this study. The black trace is without diffuse functions added to the basis set, the red trace is with diffuse functions added to the basis set.

	M05	-2X	B3L	YP	Interactions				
	ZPE	Gibbs	ZPE	Gibbs	SC - SC	Disp. Int.	BB - BB	BB - SC	
ER_Z1	0.00	0.00	4.06	5.79	A	$NH_2$	C10(O0-N3)	N1-O	
ER_Z2	3.76	3.24	5.89	5.01	С	N2	C10(O0-N3)	N1-O + O1-H <sub>2</sub> N	
ER_C1	3.81	2.00	5.41	5.03	OH - NH	NH <sub>2</sub>	C10(O0-N3)	N1-O	
ER_Z3	4.08	4.06	5.82	8.22	A	2 x NH <sub>2</sub>	C10(O0-N3)	$N1-O^{1} + N2-O^{2} + O2-H_{2}N$	
ER_Z4	4.28	0.54	0.00	0.00	A		C5(N1-O1) + C7(O1-N3)	N2-O + O2-H <sub>2</sub> N	
ER_Z5	4.37	3.60	6.60	7.20	В		C5(N1-O1)	$N2-O^{1} + N3-O^{2} + O0-(H\epsilon+H_{2}N)$	
ER_Z6	4.46	2.71	5.14	6.34	В	$H\epsilon + NH_2$	C7(O1-N3)	N1-O	
ER_Z7	4.56	2.57	6.55	6.89	A	NH <sub>2</sub>	C5(N1-O1)	N2-O + O0-H <sub>2</sub> N	
ER_Z8	4.85	4.27	11.21	11.97	С		C5(N1-O1) + C7(O1-N3)	N3-O + O0-H <sub>2</sub> N + O2-Hε	
ER_Z9	5.54	3.51	2.60	1.35	A		-		
ER_C2	6.59	4.41	5.88	3.74	OH - NH	NH <sub>2</sub>	C7(O1-N3)	N1-O	
ER_Z10	7.84	5.23	4.64	3.82	Α		C5(N1-O1)	N2-O <sup>1</sup> +N3-O <sup>2</sup>	
ER_Z11	8.80	5.28	5.44	5.75	A		C7(O1-N3)	N2-O + O2-N	
ER_Z12	9.08	8.07	10.02	9.55	A		C10(O0-N3)	N2-O	
ER_C3	9.75	6.06	8.37	7.56	OH - NH	N2	C10(O0-N3)	N1-O	
ER_T1	9.98	7.92	10.83	10.76	NH <sub>2</sub> -OH	NH <sub>2</sub>	C5(N1-O1)	Ν1-Ο + Ο3-ΗΟ + Ν2-Νε	
ER_Z13	10.13	5.68	7.00	6.18	С		-	N1-O + O1-H <sub>2</sub> N	
ER_Z14	10.13	6.30	8.30	9.48	В	$H\epsilon + NH_2$	-	Ν1-Ο + Ο2-Ηε	
ER_Z15	10.30	6.96	9.17	9.15	С		C5(N1-O1)	$N2-O^{1} + N3-O^{1} + O0-H_{2}N$	
ER_T2	11.12	8.16	7.66	7.29	O-(H <sub>2</sub> N) <sub>2</sub>		C7(O1-N3	Ν2-Νε + Ν3-Νε + Ο0-ΗΟ	
ER_C4	11.41	9.52	9.76	9.41			C5(N1-O1) + C7(O1-N3)	O0-HO	
ER_T3	12.80	10.21	11.96	10.69			C10(O0-N3)	Ν1-Ο + Ν3-Νε + Ο2-ΗΟ	
ER_C5	12.90	9.87	10.48	9.84	OH - NH		C5(N1-O1) + C7(O1-N3)	N2-O	
ER_T4	13.40	9.78	13.19	12.70	O-H <sub>2</sub> N	OH		Ν1-Νε + Ν2-Νε + Ο2-Η2Ν	
ER_C6	13.93	9.78	8.82	8.03	OH - NH		C7(O0-N2) + C7(O1-N3)		
ER_T5	14.53	12.09	9.98	8.32			C10(O0-N3)	N2-Nε + O0-H <sub>2</sub> N + O1-HO	
ER_C7	14.58	12.28	14.36	13.97	OH - NH	N2	C7(O0-N2) + C7(O1-N3)	O0-H <sub>2</sub> N	

Table SI-1. ZPE-corrected energies (ZPE), Gibbs free energies at 300K ( $\Delta G$ ) and intramolecular interactions for the optimized structures of Z-Glu-Arg-NHMe. The employed basis set for the M05-2X and B3LYP functionals is 6-311+G(d,p).

ER_T6	14.77	12.39	12.11	9.49	OH-NH <sub>2</sub>	NH <sub>2</sub>	C7(O1-N3)	Ν1-Νε + Ν2-Νε
ER_C8	15.58	11.61	12.16	11.07			C5(N1-O1) + C5(N2-O2)	N2-O + N3-HNη + O2-HO
ER_T7	16.11	14.08	15.88	14.64		NH <sub>2</sub>		N2-Nε + O1-HO
ER_T8	17.32	14.15	14.77	13.23			C7(O1-N3)	N1-H2N + N2-Νε
ER_C9	18.69	14.61	15.69	14.08			C7(O1-N3)	N1-0
ER_T9	19.09	13.10	14.15	11.55		NH <sub>2</sub>	C7(O1-N3)	Ν1-Νε
ER_C10	19.12	12.11	11.78	7.36			C5(N1-O1) + C7(O1-N3)	N2-O
ER_Z16	20.33	18.60	16.20	16.03	A		C7(O0-N2) + C7(O1-N3)	
ER_T10	20.48	17.58	19.92	19.78			C7(O0-N2) + C7(O1-N3)	N1-O + O1-HO
ER_C11	21.04	16.09	16.51	13.81			C7(O0-N2) + C7(O1-N3)	O2-H <sub>2</sub> N
ER_C12	25.11	18.57	17.33	13.20			C7(O0-N2) + C7(O1-N3)	

Energies are given in kcal/mol.



ER\_Z3

ER\_Z4













ER\_Z9

ER\_Z7

ER\_Z10











Figure SI-2. Conformations of Z-Glu-Arg-NHMe, optimized with the M05-2x functional employed with the 6-311+G(d,p) basisset.





# IR wavenumbers (cm<sup>-1</sup>)

Figure SI-3. Comparison of the experimental IR-UV ion dip spectrum of Z-Glu-Arg-NHMe (black trace) with spectra calculated for various isomers. The black sticks represent the deconvoluted experimental spectrum. The red traces are the theoretical spectra of the optimized structures of Z-Glu-Arg-NHMe, where the red sticks represent the individual computed frequencies and intensities.







Figure SI-4. Experimental IR-UV ion-dip spectrum of Z-Glu-Arg-NHMe (black trace) and the theoretical spectra of the optimized structures (red trace).

Exp	V <sub>1</sub>	l <sub>1</sub>	V <sub>2</sub>	$I_2$	V <sub>3</sub>	$I_3$	V <sub>4</sub>	$I_4$	V <sub>5</sub>	$I_5$	V <sub>6</sub>	l <sub>6</sub>		
	1634.4	0.29	1657.3	0.18	1675.4	0.58	1698.7	0.78	1713.5	0.6	1736.0	0.68		
Calculations	$\Delta v_1$	$\Delta I_1$	$\Delta v_2$	$\Delta I_2$	$\Delta v_3$	Δl <sub>3</sub>	$\Delta v_4$	$\Delta I_4$	$\Delta v_5$	$\Delta I_5$	$\Delta v_6$	Δl <sub>6</sub>	<u>Σ( Δv )</u>	<u>Σ( Δ  )</u>
ER_Z1	7.0	0.07	15.1	0.12	21.2	-0.19	9.4	-0.06	2.8	-0.27	-6.4	-0.06	61.7	0.77
ER_Z2	-9.4	0.18	9.3	0.00	9.9	-0.52	-4.2	-0.40	-16.8	-0.13	-30.9	-0.35	80.5	1.59
ER_C1	-18.6	0.19	-10.2	0.59	18.5	-0.09	0.8	-0.23	0.5	-0.21	-17.0	-0.08	65.7	1.39
ER_Z3	22.1	0.00	8.7	-0.11	2.3	0.15	-4.5	-0.72	-4.9	0.40	21.9	0.08	64.3	1.47
ER_Z4	13.2	0.05	1.7	0.05	4.8	0.13	0.8	-0.74	5.4	0.31	-4.4	-0.21	30.3	1.48
ER_Z5	-2.5	0.00	26.9	0.33	12.6	-0.15	-2.3	-0.59	-10.6	-0.37	-26.7	-0.36	81.6	1.80
ER_Z6	-23.7	-0.04	17.9	0.03	25.4	-0.32	3.5	-0.36	-5.3	-0.18	16.9	-0.30	92.7	1.23
<u>ER_</u> Z7	11.9	0.05	19.6	-0.05	20.7	-0.07	2.7	-0.46	-5.9	-0.25	-12.4	-0.34	73.1	1.22
ER_Z8	0.3	0.71	6.8	0.48	5.2	-0.45	-9.4	-0.41	-16.0	-0.13	-21.1	-0.22	58.7	2.39
ER_Z9	0.9	-0.04	25.7	-0.05	27.7	-0.27	13.5	-0.30	-0.3	-0.20	0.8	-0.36	68.9	1.21
ER_C2	-33.6	-0.03	3.0	0.48	-4.4	-0.29	12.3	-0.33	2.2	0.01	5.1	-0.15	60.7	1.29
ER_Z10	8.5	-0.06	23.6	0.22	13.4	-0.16	5.9	-0.14	5.4	0.24	-4.1	-0.17	61.0	0.98
ER_Z11	11.6	0.05	9.3	-0.04	3.8	0.37	-7.3	-0.70	3.5	0.26	27.2	-0.17	62.6	1.60
ER_Z12	4.6	-0.12	20.2	-0.08	12.4	-0.13	11.5	-0.26	4.3	-0.03	-9.9	-0.48	62.9	1.10
ER_C3	-36.8	-0.01	-3.9	0.33	18.8	0.03	4.7	-0.40	-6.1	-0.23	13.4	-0.29	83.6	1.29
ERZ13	-10.4	0.45	16.3	0.14	4.2	-0.41	-1.3	-0.21	-15.2	-0.16	-5.1	-0.19	52.4	1.55
ER_Z14	-19.4	0.53	15.4	0.36	16.4	0.39	7.2	-0.59	3.9	-0.27	-3.7	0.05	66.0	2.19
ER_Z15	-2.3	0.58	17.6	0.82	14.2	-0.44	0.5	-0.37	-10.3	-0.44	-16.7	-0.19	61.6	2.84
ER_C4	-34.9	0.29	-6.9	0.57	-8.0	0.08	-12.0	-0.22	-10.8	0.37	22.2	0.22	94.8	1.75
ER_C5	-26.5	0.26	-3.2	0.73	-0.4	-0.21	-1.6	0.22	2.0	-0.08	6.5	0.05	40.3	1.55
ER_C6	-19.7	-0.16	-7.1	0.06	-2.8	-0.45	1.0	-0.71	-11.2	-0.09	-21.9	0.16	63.5	1.63
ER_C7	-10.8	0.22	8.7	0.45	2.1	-0.39	-9.6	-0.55	-9.4	0.40	-3.1	-0.21	43.7	2.21
ER_C8	-12.2	-0.03	-13.6	0.47	-4.2	0.35	-17.5	-0.60	14.8	-0.05	-4.7	-0.29	67.1	1.79
ER_C9	-21.6	-0.20	8.6	0.23	-4.9	-0.33	18.9	-0.25	29.1	-0.06	14.8	-0.04	97.8	1.11
ER_C10	-22.3	-0.06	1.5	0.73	-9.3	-0.44	13.5	-0.12	21.7	-0.09	46.0	0.14	114.3	1.58
ER_Z16	3.2	0.09	17.7	0.13	10.5	-0.39	-2.4	-0.45	0.9	0.26	-6.6	0.16	41.4	1.48
ER_C11	-16.9	0.27	13.5	0.51	0.4	-0.22	-1.1	-0.02	2.1	0.14	48.6	-0.05	82.6	1.21
ER_C12	-33.8	-0.14	16.7	0.09	5.7	-0.10	9.8	-0.51	2.2	0.22	49.4	-0.15	117.7	1.22

Table SI-2. Frequencies and intensities of the six deconvoluted Gaussians in the Amide I region in the experimental spectrum of Z-Glu-Arg-NHMe compared to the computed bands for the optimized structures, which are listed in order of increasing relative energy.

# Deconvolution and assignment of the Z-Glu-Arg-NHMe spectrum in the Amide I region

The Amide I region of the experimental spectrum of Z-Glu-Arg-NHMe is deconvoluted into 6 Gaussian functions, each Gaussian representing a vibrational mode of Z-Glu-Arg-NHMe. The obtained frequencies and relative intensities are compared with the theoretically obtained frequencies and relative intensities of the DFT calculations. The sum of the deviations in frequencies and relative intensities with respect to the computed ones are given in the last two columns of Table SI-2.

The closest agreement between experimental and calculated frequencies is found for structure ER\_Z4, with the sum of the absolute frequency deviations between theory and experiment being 30 cm<sup>-1</sup>. However, there is a large mismatch between the relative intensities, resulting in a completely different shape of the overall feature (see Supporting Information: Fig SI-3).

The next closest match between computed and experimental frequencies is found for ER\_Z16 with a summed absolute deviation of 41.4 cm<sup>-1</sup>. However, the energy of this structure is calculated to be 20.3 kcal/mol higher than the lowest energy structure. Moreover, this structure also shows a large mismatch in relative intensities. The calculated spectrum of ER\_Z16 is shown together with the experimental spectrum in the bottom panel of Figure 7, showing a strong absorption at 1585 cm<sup>-1</sup> which is not observed in the experimental spectrum.

The experimental spectrum of ER\_C7 in the Amide I region is shown in Figure SI-3. At first glance, there is a fairly good agreement between theory and experiment and only the lowest frequency vibration is shifted to the red by 11 cm<sup>-1</sup>. However, upon comparing the individual vibrations (Table SI-2), one observes a large mismatch in relative intensities. In addition, the experimental spectrum shows a moderately strong band at 1713 cm<sup>-1</sup>, which shifts to 1689 cm<sup>-1</sup>. In the calculated spectrum of ER\_C7 leading to the conclusion that this structure is unlikely to be responsible for the observed spectrum.

The relative intensities are thus important for assigning the experimental spectrum to one of the theoretical structures. The structure with the lowest deviation in relative intensity is ER\_Z1. In addition, the deviation in frequency is also fairly low. The experimental spectrum together with computed spectrum of ER\_Z1 is shown in the top trace of Figure 7 in the 1000 – 1850 cm<sup>-1</sup> range. Besides the good match in the Amide I region, the remainder of the spectrum is also in good agreement with the experimental spectrum. To conclude, the experimental spectrum of Z-Glu-Arg-NHMe is assigned to ER\_Z1.













Figure SI-5. Conformations of Z-Glu-Ala-Arg-NHMe, optimized with the M05-2x functional employed with the 6-311+G(d,p) basis set.









theoretical spectra of the optimized structures for Z-Glu-Ala-Arg-NHMe (red trace).

0	MOS	5-2X	B3LYP		Interactions				
	ZPE	Gibbs	ZPE	Gibbs	SC - SC	Disp. Int.	BB - BB	BB - SC	
EAR_Z1	0.00	0.00	2.56	5.86	А	$2 \times NH_2$	C10(O0-N3)	$N1-O^{1} + N2-O^{2}$	
EAR_Z2	0.73	0.47	3.47	6.28	В	NH <sub>2</sub> + Νε	C10(O0-N3)	$N1-O^{1} + N2-O^{2}$	
EAR_Z3	1.11	2.54	4.07	7.73	В	NH <sub>2</sub>	C10(O0-N3) + C10(O1-N4)	O3-( $H_2N + H\epsilon$ )	
EAR_Z4	1.85	1.31	2.67	5.61	А	$2 \times NH_2$	C7(O2-N4)	$N1-O^{1} + N2-O^{2}$	
								$N1-O^{1} + N2-O^{1} + N3-O^{1} + O4-$	
EAR_Z5	1.90	2.09	5.16	8.20	A	2 x NH <sub>2</sub>	C10(O1-N4)	H <sub>2</sub> N	
EAR_Z6	2.13	1.96	4.65	7.88	A	$2 \times NH_2$	C10(O1-N4)	N3-O + O3-H <sub>2</sub> N	
EAR_Z7	2.85	2.82	3.46	7.12	A	$2 \times NH_2$	C7(O2-N4)	$N1-O^{1} + N3-O^{2} + O3-H_{2}N$	
EAR_Z8	3.03	3.42	5.67	8.88	В	NH <sub>2</sub> + Νε	C10(O0-N3)	$N1-O^1 + N2-O^2$	
EAR_Z9	3.91	5.82	5.93	8.89	В	NH <sub>2</sub>	C10(O0-N3) + C10(O1-N4)	N2-0	
								$N1-O^{1} + N_{2}O^{1} + N_{4}O^{1} + O^{1}$	
EAR_Z10	4.40	5.12	6.33	8.62	C*	N2		$(H_2N)' + O2 - (H_2N)^2$	
EAR_Z11	4.82	4.67	5.61	7.42	С		C7(O0-N2) + C10(O1-N4)	$N3-O + O3-H_2N$	
EAR_Z12	5.16	5.08	7.41	8.72	B	NH <sub>2</sub> + Νε	C10(O0-N3) + C10(O1-N4)	N1-O <sup>+</sup> + N2-O <sup>2</sup>	
EAR_Z13	5.52	4.30	4.25	5.01	A		C7(O0-N2) + C10(O1-N4)	N3-O + O3-H <sub>2</sub> N	
EAR_Z14	5.52	5.73	6.67	9.20	В	HE	C7(O1-N3)	$N1-O^{1} + N2-O^{2} + O2-H_{2}N$	
EAR_Z15	6.07	5.97	7.52	11.60	В		C5(N1-O1) + C7(O1-N3)	$N2-O + O0-(H_2N)^1 + O3-(H_2N)^2$	
EAR_Z16	6.10	7.21	9.13	11.29	C*	2 x NH <sub>2</sub>	C10(O1-N4)	$N1-O^{1} + N2-O^{1} + N3-O^{1}$	
EAR_Z17	6.74	6.85	7.40	10.39	В	NH <sub>2</sub> + Νε	C7(O2-N4)	$N1-O^{1} + N3-O^{2}$	
								N2-O <sup>1</sup> + N3-O <sup>1</sup> + N4-O <sup>1</sup> + O0-	
EAR_Z18	6.79	5.86	7.45	8.46	A	NH <sub>2</sub>	C5(N1-O1)	H <sub>2</sub> N	
EAR_Z19	6.87	6.45	7.24	10.47	С	HE	C5(N1-O1) + C7(O2-N4)	$N2-O + O0-(H_2N)^1 + O3-(H_2N)^2$	
EAR_Z20	7.17	6.87	4.34	6.29	A		C10(O1-N4)	$N1-O^{1} + N3-O^{2} + O3-NH2$	
EAR_Z21	7.17	4.26	3.58	3.95	A		C5(N1-O1) + C10(O1-N4)	$N2-O^{1} + N3-O^{2}$	
EAR_Z22	7.43	8.35	9.09	12.12	В	NH <sub>2</sub> + Νε	C13(O0-N4) + C7(O1-N3)	$N1-O^{1} + N2-O^{2} + O2-H_{2}N$	
EAR_Z23	7.46	3.02	0.00	0.00	А		C5(N1-O1) + C7(O2-N4)	$N2-O^{1} + N3-O^{1}$	
EAR_Z24	7.74	6.30	4.28	4.71	A		C7(O0-N2) + C10(O1-N4)	N3-O + O3- H <sub>2</sub> N	
EAR_Z25	8.01	7.10	4.50	6.40	В	NH <sub>2</sub>	C7(O1-N3) + C7(O2-N4)	00- H <sub>2</sub> N	
EAR_Z26	9.07	8.11	8.17	10.40	В		C10(O0-N3) + C7(N2-O4)	O1- H <sub>2</sub> N	
EAR_Z27	9.49	7.20	4.96	6.16	A		C5(N1-O1)	$N2-O^{1} + N3-O^{1} + O3-H_{2}N$	

Table SI-3. ZPE-corrected energies (ZPE), Gibbs free energies at 300 K ( $\Delta$ G) and intramolecular interactions for the optimized structures of Z-Glu-Ala-Arg-NHMe. The employed basis set for the M05-2X and B3LYP functionals is 6-311+G(d,p).

EAR_Z28	9.56	7.33	4.07	5.81	O-H <sub>2</sub> N		C5(N1-O1) + C7(O2-N4)	O0- H <sub>2</sub> N + N2-O <sup>1</sup> + N3-O <sup>1</sup>
							C7(O0-N2) + C7(O1-N3) +	
EAR_Z29	9.88	9.65	8.06	10.93	A	$2 \times NH_2$	C7(O2-N4)	N1-O + O3-H₂N
EAR_Z30	10.03	10.34	11.36	14.50	С	NH₂ + Νε	C7(O2-N4)	N2-O <sup>1</sup> +N3-O <sup>1</sup> + O0-H <sub>2</sub> N
								$N2-O'+N3-O'+O_{2}-(H_{2}N)'+O_{3}-$
EAR_Z31	10.11	10.07	8.33	10.31	С		C5(N1-O1)	$(H_2N)^2$
	44.00	44.04	10.17	44.04	0 I I I I			$OO-(NH_2)_2 + N2-H_2N + N3-N\epsilon +$
EAR_11	11.96	11.01	12.17	14.04	$O-H_2N$		C7(02-N4)	<u>U3-HO</u>
EAR_Z32	15.02	11.53	7.85	8.36	C*		C5(N1-O1) + C7(O2-N4)	$N2-O' + N3-O' + O3-H_2N$
	10.00							N1-O+ O1- $(H_2N)'$ + O3- $(H_2N)^2$ +
EAR_12	18.03	15.54	14.53	14.59			C7(O0-N2) + C7(O2-N4)	Ν3-Νε
	10.04	45.04	40.00	40.74				$N3-O + O3-HO + N4-N\epsilon + O2-$
EAR_13	18.04	15.21	12.26	12.74			C7(00-N2) + C5(N3-O3)	
	10 20	17 21	15 72	17.00			CE(N1 01)	$N3 - N1 H + N4 - N1 H + O1 - H_2 N + O2 HO$
		1/31						
	10.00	17.01	13.73	17.00				
EAR_T4	18.52	17.87	21.37	23.85		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3)	02-110 03-HO + N4-Νε
EAR_T4	18.52	17.87	21.37	23.85		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3)	O3-HO + N4-Nε N2-O + N3-OH + N4-Nε + O2-
EAR_T4 EAR_T5	18.52 20.00	17.87 17.87 15.59	21.37 15.61	23.85 15.59		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) C5(N1-O1) + C5(N3-O3)	O3-HO + N4-Nε N2-O + N3-OH + N4-Nε + O2- H <sub>2</sub> N + O3-HO
EAR_T4 EAR_T5 EAR_C2	18.52 20.00 19.72	17.87 15.59 16.17	21.37 15.61 14.36	23.85 15.59 13.66		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) C5(N1-O1) + C5(N3-O3) C7(O1-N3) + C7(O2-N4)	$03-HO + N4-N\epsilon$ $N2-O + N3-OH + N4-N\epsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\epsilon + O3-HO$ $020(100+100) + N2(200) + N2(200)$
EAR_T4 EAR_T5 EAR_C2	18.52 20.00 19.72	17.87 15.59 16.17	21.37 15.61 14.36	23.85 15.59 13.66		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) C5(N1-O1) + C5(N3-O3) C7(O1-N3) + C7(O2-N4)	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$
EAR_T4 EAR_T5 EAR_C2 EAR_T6	18.52 20.00 19.72 20.38	17.87 15.59 16.17 18.42	21.37 15.61 14.36 <u>16.66</u>	23.85 15.59 13.66 17.97		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-$ $H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-$ $N\varepsilon + O4-HO$
EAR_T4 EAR_T5 EAR_C2 EAR_T6	18.52 20.00 19.72 20.38	17.87 17.87 15.59 16.17 18.42	10.73 21.37 15.61 14.36 16.66	23.85 15.59 13.66 17.97		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) + C7(O1-N3)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-$ $H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-$ $N\varepsilon + O4-HO$
EAR_T4 EAR_T5 EAR_C2 EAR_T6 EAR_C3	18.52 20.00 19.72 20.38 21.54	17.87 17.87 15.59 16.17 18.42 18.44	21.37 15.61 14.36 16.66 19.60	23.85 15.59 13.66 17.97 19.95		NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) +$ $C7(O1-N3)$ $C7(O1-N3) + C7(O1-N3) + C$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$ $O2-H\varepsilon$
EAR_T4 EAR_T5 EAR_C2 EAR_T6 EAR_C3 EAR_C4	18.52 20.00 19.72 20.38 21.54 21.56	17.87 17.87 15.59 16.17 18.42 18.44 18.02	13.73 21.37 15.61 14.36 16.66 19.60	23.85 15.59 13.66 17.97 19.95	OH - NH	NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) +$ $C7(O1-N3)$ $C7(O0-N2) + C7(O1-N3) +$ $C7(O2-N4)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$ $O2-H\varepsilon$
EAR_T4 EAR_T5 EAR_C2 EAR_T6 EAR_C3 EAR_C4	18.52 20.00 19.72 20.38 21.54 21.56	17.87 17.87 15.59 16.17 18.42 18.44 18.02	13.73 21.37 15.61 14.36 16.66 19.60 13.17	23.85 15.59 13.66 17.97 19.95 12.02	OH - NH	NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) + C7(O1-N3)$ $C7(O0-N2) + C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C7(O2-N4)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$ $O2-H\varepsilon$
EAR_T4 EAR_T5 EAR_C2 EAR_T6 EAR_C3 EAR_C4 EAR_T7	18.52 20.00 19.72 20.38 21.54 21.56 24.06	17.87 17.87 15.59 16.17 18.42 18.44 18.02 21.24	21.37 15.61 14.36 16.66 19.60 13.17 19.33	23.85 15.59 13.66 17.97 19.95 12.02 19.18	OH - NH	NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) +$ $C7(O1-N3)$ $C7(O0-N2) + C7(O1-N3) +$ $C7(O2-N4)$ $C7(O0-N2) + C7(O1-N3) +$ $C5(N3-O3)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$ $O2-H\varepsilon$ $O3-HO + N4-(NH_2)^1 + O2-(H_2N)^2$
EAR_T4 EAR_T5 EAR_C2 EAR_T6 EAR_C3 EAR_C4 EAR_T7	18.52 20.00 19.72 20.38 21.54 21.56 24.06	17.87 15.59 16.17 18.42 18.44 18.02 21.24	21.37 15.61 14.36 16.66 19.60 13.17 19.33	23.85 15.59 13.66 17.97 19.95 12.02 19.18	OH - NH	NH <sub>2</sub>	C7(O1-N3) + C5(N3-O3) $C5(N1-O1) + C5(N3-O3)$ $C7(O1-N3) + C7(O2-N4)$ $C7(O2-N4)$ $C11(N1-O3) + C7(O0-N2) + C7(O1-N3)$ $C7(O0-N2) + C7(O1-N3) + C7(O2-N4)$ $C7(O0-N2) + C7(O1-N3) + C7(O2-N4)$ $C7(O0-N2) + C7(O1-N3) + C5(N3-O3)$	$03-HO + N4-N\varepsilon$ $N2-O + N3-OH + N4-N\varepsilon + O2-H_2N + O3-HO$ $N1-O + O1-H\varepsilon + O3-HO$ $O0-(H_2N + H_2N) + N2-NH_2 + N3-N\varepsilon + O4-HO$ $O2-H\varepsilon$ $O3-HO + N4-(NH_2)^1 + O2-(H_2N)^2$

Energies are given in kcal/mol.



Figure SI- 7. Seven Gaussian functions, each representing a vibrational mode in the Amide I region of Z-Glu-Arg-NHMe (red), the sum of the six Gaussian functions (black), the experimental IR-UV spectrum (blue) and the computed spectrum of the proposed structure ( EAR\_Z1) of Z-Glu-Arg-NHMe (green).

	M05-2X B3LYP			Interactions						
					SC -	Disp.				
	ZPE	Gibbs	ZPE	Gibbs	SC	Int.	BB - BB	BB - SC		
						2 x	C10(O0-N3) + C7(O2-N4) + C5(N5-			
EA3R_Z1	0.00	0.00	0.00	0.00	А	$NH_2$	O5)	$O4-H_2N + N1-O^1 + N6-O^2$		
						$NH_2 +$	C7(O1-N3) + C10(O2-N5) +			
EA3R_Z2	0.07	0.89	3.71	2.32	В	Νε	C10(O3-N6)	N1-O <sup>1</sup> + N4-O <sup>2</sup> + O4- H <sub>2</sub> N		
							C5(N1-O1) + C7(O1-N3) + C10(O1-			
EA3R_Z3	0.36	0.49	3.93	1.15	А	$NH_2$	N4) + C10(O2-N5) + C10(O3-N6)	$N2-O + O0-(H_2N)^1 + O4-(H_2N)^2$		
							C5(N1-O1) + C7(O1-N3) + C10(O2-			
EA3R_Z4	3.18	2.03	3.69	-0.75	А		N5) + C10(O3-N6)	$N2-O + O0-(H_2N)^1 + O4-(H_2N)^2$		
						$NH_2 +$	C7(O1-N3) + C10(O2-N5) +			
EA3R_Z5	3.35	3.52	5.89	3.93	В	Νε	C13(O2-N6)	N1-O <sup>1</sup> + N4-O <sup>2</sup> + O4- H <sub>2</sub> N		
							C10(O0-N3) + C7(O2-N4) + C7(O4-			
EA3R_Z6	3.81	3.64	6.44	4.92	A		N6) + C11(N2-O4)	N1-O		
							C5(N1-O1) + C11(N2-O4) + C7(O2-	4		
EA3R_Z7	4.86	3.37	7.42	5.26	В		N4) + C7(O4-N6)	$N3-O + O0-(H_2N)^{T} + O3-(H_2N)^{2}$		
EA3R_Z8	8.64	5.41	8.61	4.97	C		C7(O0-N2) + C10(O2-N5)	$N3-O + O1-(H_2N)' + O3-(H_2N)^2$		
					_		C10(O0-N3) + C13(O1-N5) +			
EA3R_Z9	9.07	6.04	7.15	3.23	В		C7(O4-N6)	O5-(N $\epsilon$ +H <sub>2</sub> N)		
					-			$N3-O' + N4-O' + O1-(H_2N)' + O4-$		
EA3R_Z10	9.27	6.27	4.39	-0.34	С		C7(O0-N2) + C7(O3-N5)	$(H_2N)^2$		
			- · -					N2-O' + N4-O' + N5-O' + O5-N + O0 –		
EA3R_Z11	11.05	7.70	6.17	1.35	O-NH2		C5(N1-O1) + C7(O4-N6)	$(H_2N + H_2N)$		
/							C10(O0-N3) + C11(N2-O4) +			
EA3R_T1	14.84	13.95	18.09	15.00			C7(O2-N4)	Ν5-Νε + Ν6-Νε + Ν1-Ο + Ο5-ΗΟ		
								N3-O + N6-NηH + O4-(Hε + H <sub>2</sub> N) +		
EA3R_C1	16.69	15.59	16.41	11.82			C7(O0-N2) + C10(O1-N4)	O5-OH		
							C10(O0-N3) + C14(N2-O5) +			
EA3R_12	17.73	14.62	18.28	13.54			C/(O2-N4) + C5(N5-O5)	Ν1-Ο + Ν6-Νε		
	47.05	15.00	47.50	10.05			C5(N1-O1) + C7(O1-N3) + C7(O2-			
EA3R_13	17.95	15.98	17.58	13.35			N4)	00-H <sub>2</sub> N + N2-Nε + O3-HO + N5-O		

Table SI-4. ZPE-corrected energies (ZPE), Gibbs free energies at 300 K ( $\Delta$ G) and intramolecular interactions for the optimized structures of Z-Glu-Ala-Ala-Ala-Arg-NHMe. The employed basis set for the M05-2X and B3LYP functionals is 6-311G(d,p).

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					$NH_2$	C10(O0-N3) + C10(O1-N4) +	
EA3R_T4	21.31	17.84	22.20	15.98		C10(O2-N5)	Ν1-Νε
_						C7(O0-N2) + C7(N3-O5) + C7(N4-	
EA3R_C2	33.79	28.46	24.48	16.24		O6)	Ο1- H₂N + N3- NηH
_						C7(00-N2) + C7(01-N3) + C7(02-	
EA3R_C3	40.35	30.33	28.03	18.42		N4) + C7(O3-N5) + C7(O4-N6)	Ο5-Ηε
Energies a	re given	in kcal	/mol.	•	"		











Figure SI-8. Conformations of Z-Glu-Ala-Ala-Ala-Arg-NHMe, optimized with the M05-2x functional employed with the 6-311G(d,p) basis set.





Figure SI-9. Experimental IR-UV ion-dip spectrum of Z-Glu-Ala-Ala-Ala-Arg-NHMe (black trace) and the theoretical spectra of the optimized structures for Z-Glu-Ala-Ala-Ala-Ala-Arg-NHMe (red trace).