

Supplementary Information

Electrochemical reduction induced self-doping of Ti^{3+} for efficient water splitting performance on TiO_2 based photoelectrodes

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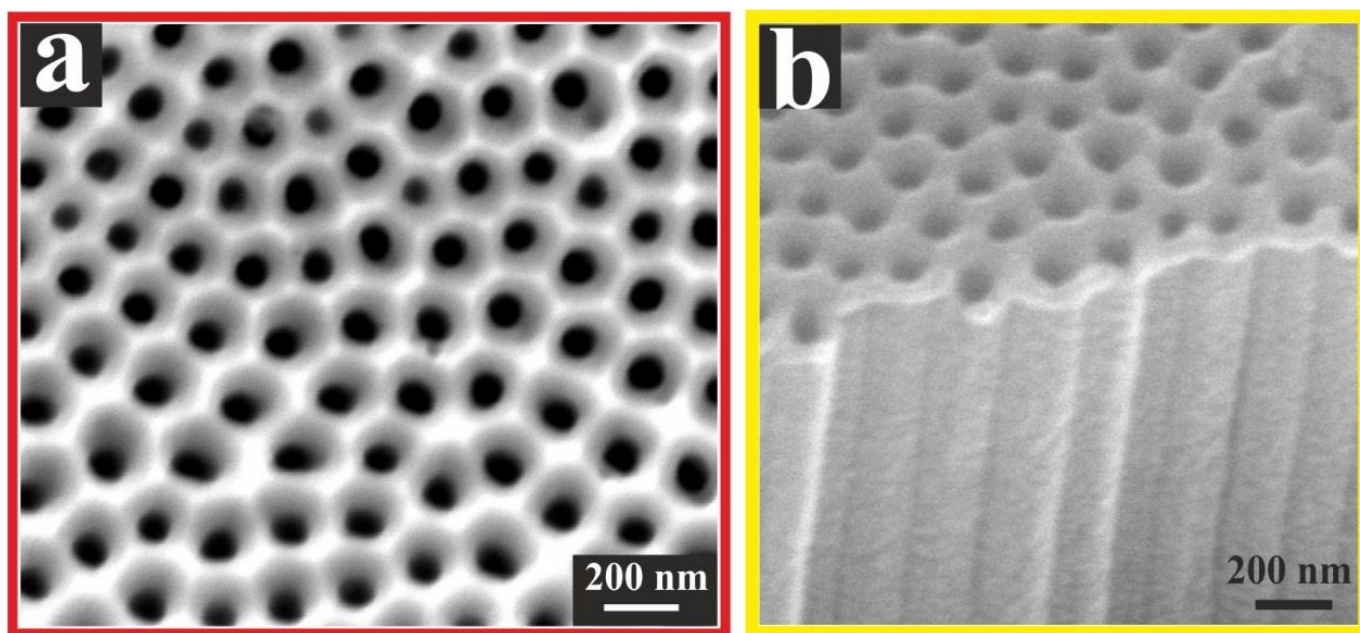


Fig. S1 SEM images of the TiO_2 NTs of (a) top view, and (b) cross-sectional view.

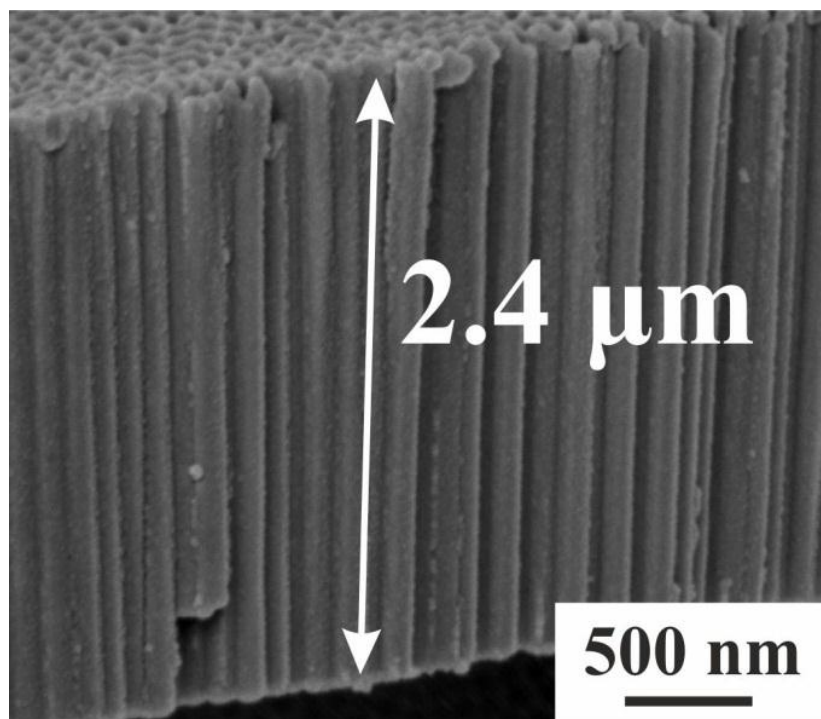


Fig. S2 Cross-sectional SEM image of the ECR-TiO₂ NTs.

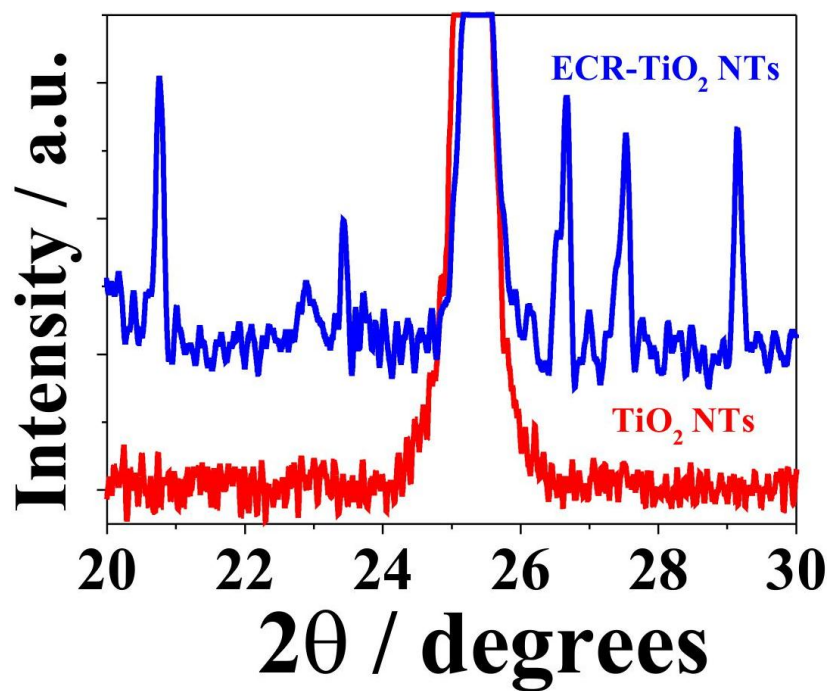


Fig. S3 XRD patterns of the TiO₂ NTs and ECR-TiO₂ NTs in magnification view in range of 20-30°.

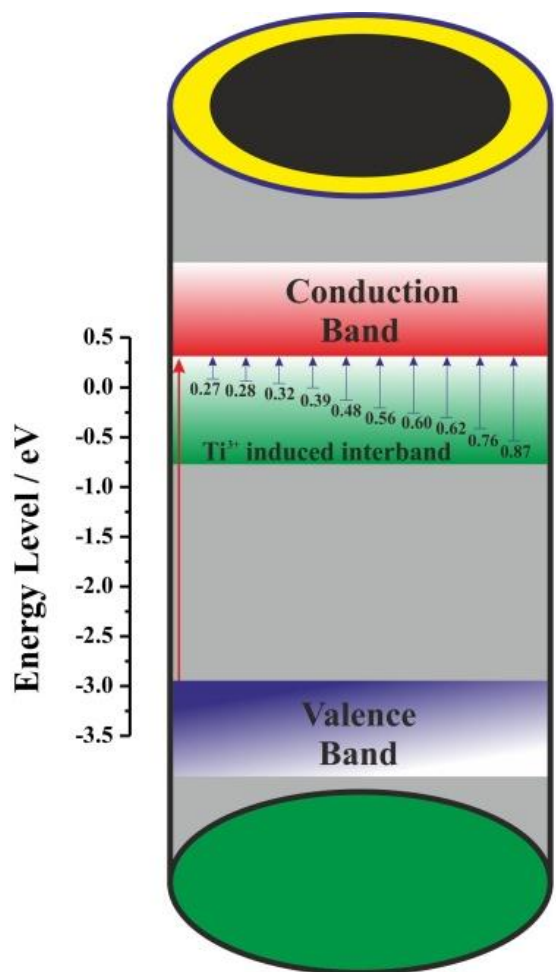


Fig. S4 Schematics of band energy level of Ti³⁺ self-doped ECR-TiO₂ NTs.

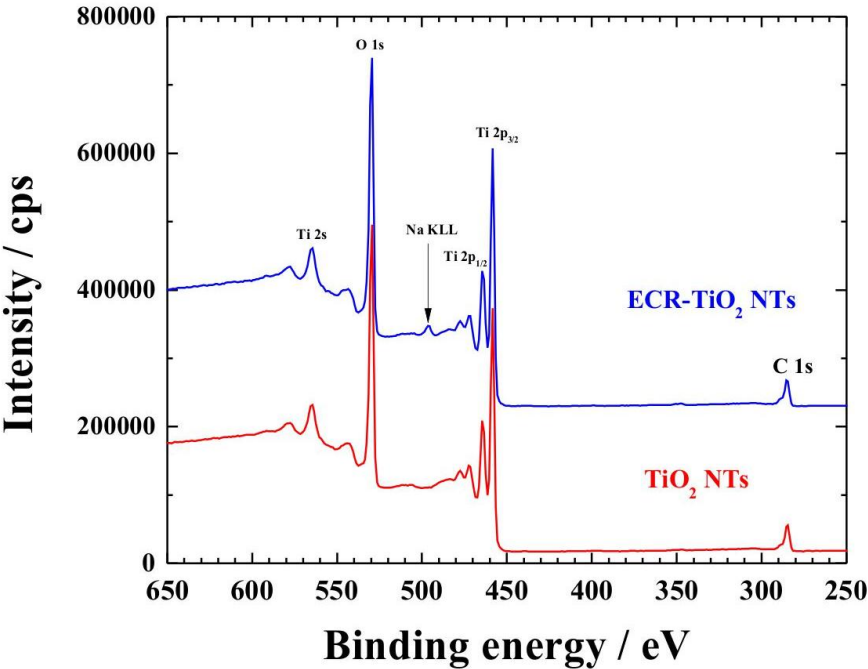


Fig. S5 XPS survey of the TiO₂ NTs and ECR-TiO₂ NTs.

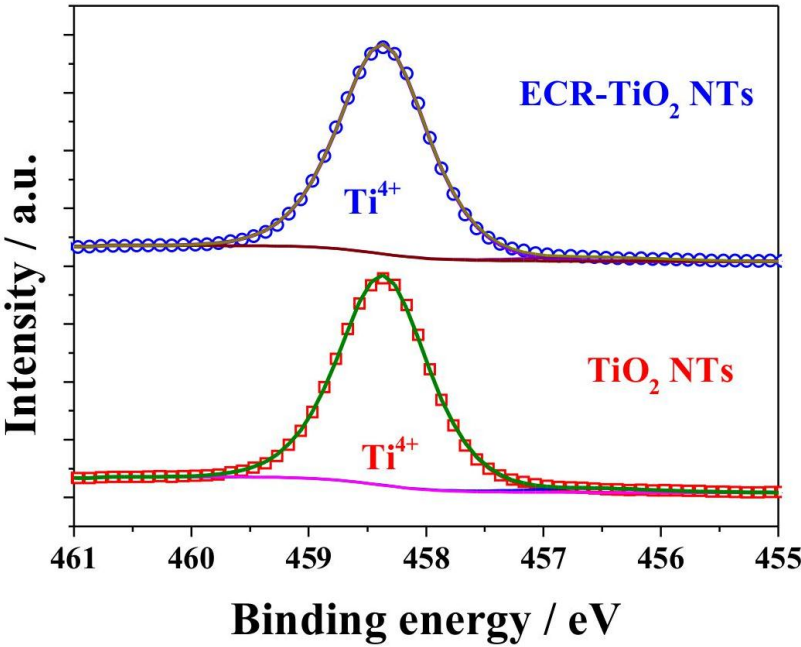


Fig. S6 XPS core level of Ti 2p_{3/2} of the TiO₂ NTs and ECR-TiO₂ NTs in the absence of argon sputtering.

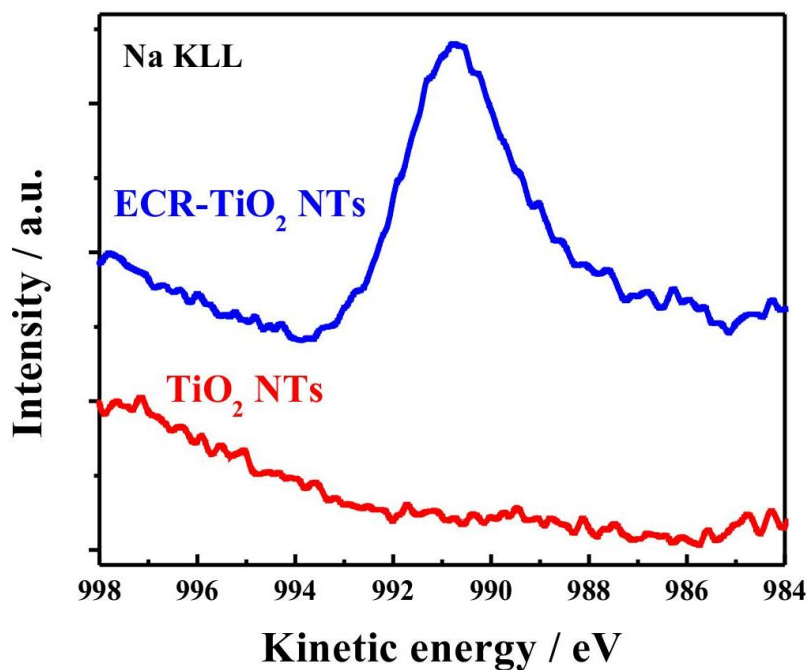


Fig. S7 Na KLL Auger spectra from TiO₂ NTs and ECR-TiO₂ NTs.

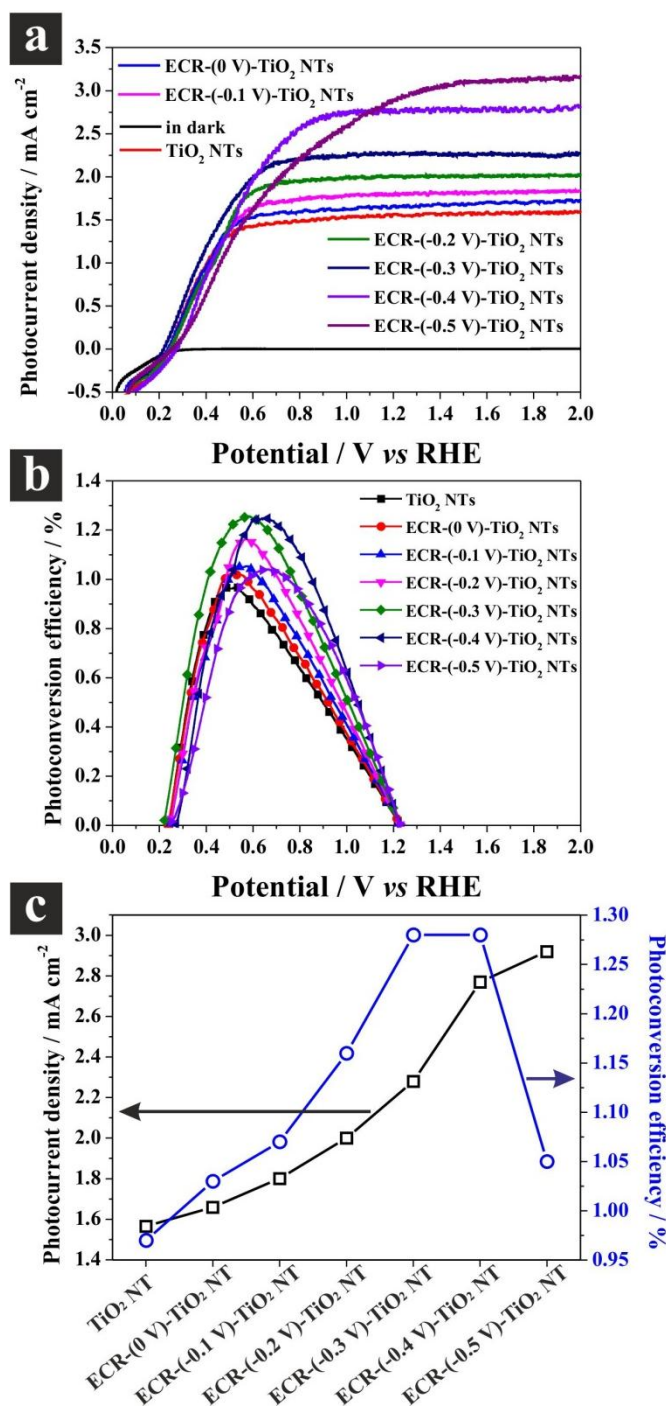


Fig. S8 PEC performance of the ECR-TiO₂ NT photoelectrodes reduced with different reduction potential, where x in ECR-(x)-TiO₂ NTs was the value of the applied potential (V). (a) linear-sweep voltammograms, collected with a scan rate of 5 mV s⁻¹ under simulated solar light (AM 1.5G); (b) photoconversion efficiency as a function of applied potential (calculated using equation 1); (c) summarized photocurrent density and photoconversion efficiency data.

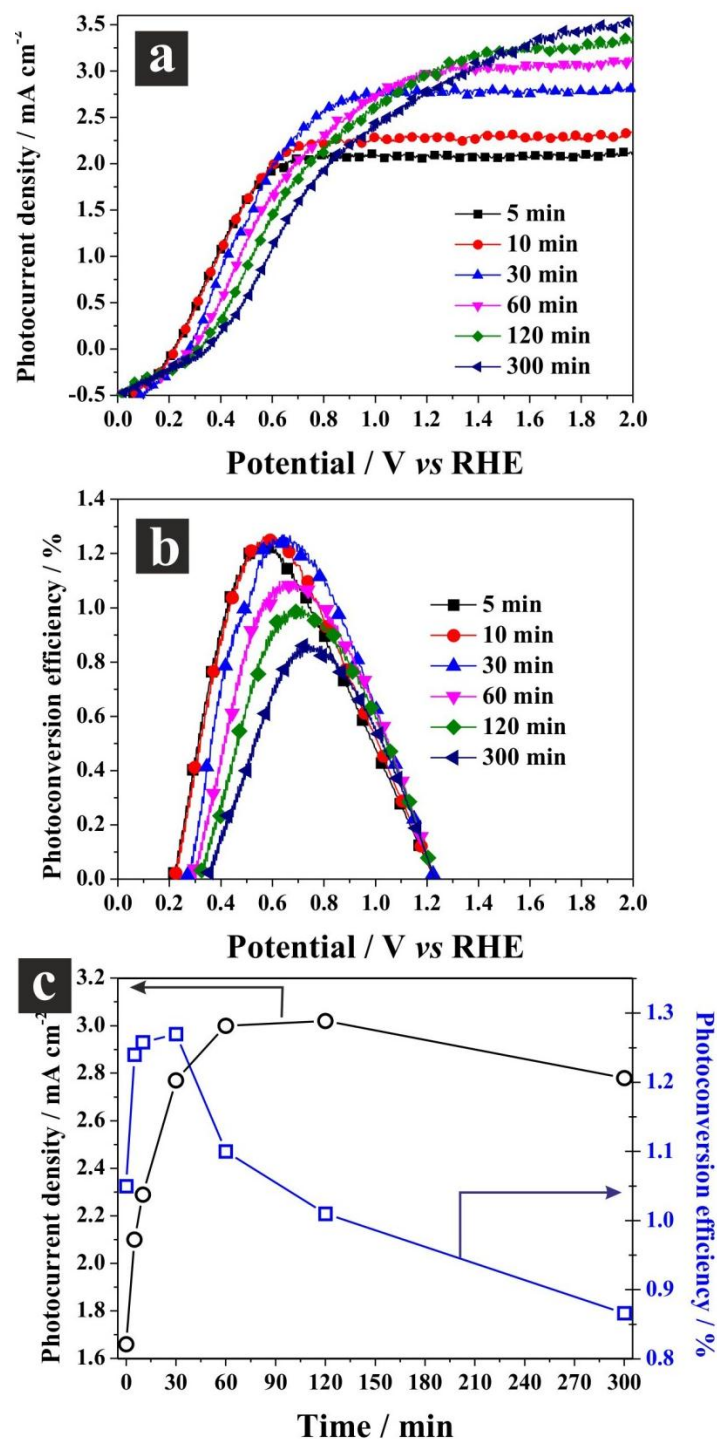


Fig. S9 PEC performance of the ECR-TiO₂ NTs photoelectrodes reduced with different duration length. (a) linear-sweep voltammograms, collected with a scan rate of 5 mV s⁻¹ under simulated solar light (AM 1.5G); (b) photoconversion efficiency as a function of applied potential (calculated using equation 1); (c) summarized photocurrent density and photoconversion efficiency data.

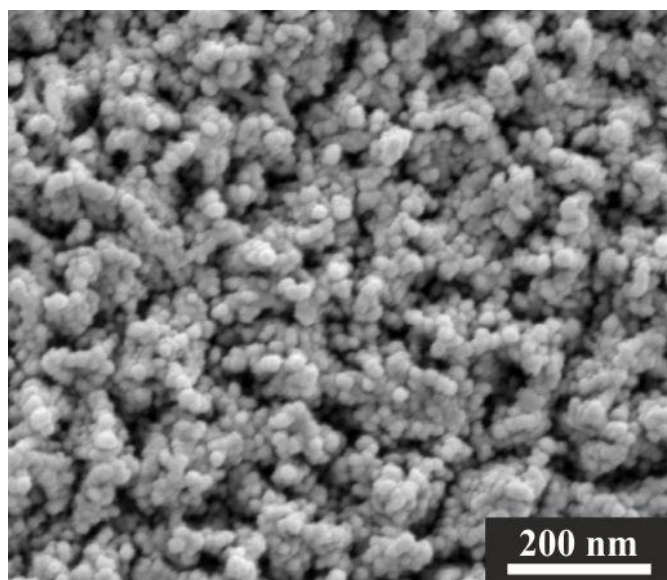


Fig. S10 SEM image of the TiO_2 film on indium tin oxide (ITO) prepared from commercial P25 TiO_2 nanoparticles.

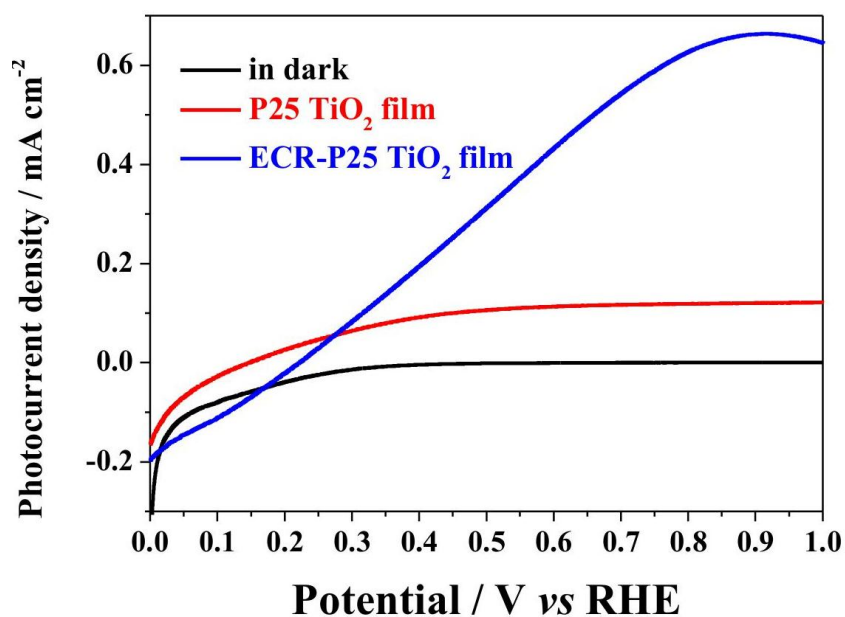


Fig. S11 PEC performance of the P25 TiO_2 film and ECR-P25 TiO_2 film photoelectrodes, prepared on indium tin oxide (ITO) with commercial P25 TiO_2 nanoparticles.