

Electronic Supplementary Information

Elucidating a twin-dependent chemical activity of hierarchical copper sulfide nanocages

Shaodong Sun, Dongchu Deng, Xiaoping Song and Zhimao Yang*

School of Science, State Key Laboratory for Mechanical Behavior of Materials, MOE Key Laboratory for Non-Equilibrium Synthesis and Modulation of Condensed Matter, Xi'an Jiaotong University, Xi'an 710049, ShaanXi, People's Republic of China

* Corresponding author. E-mail: zmyang@mail.xjtu.edu.cn

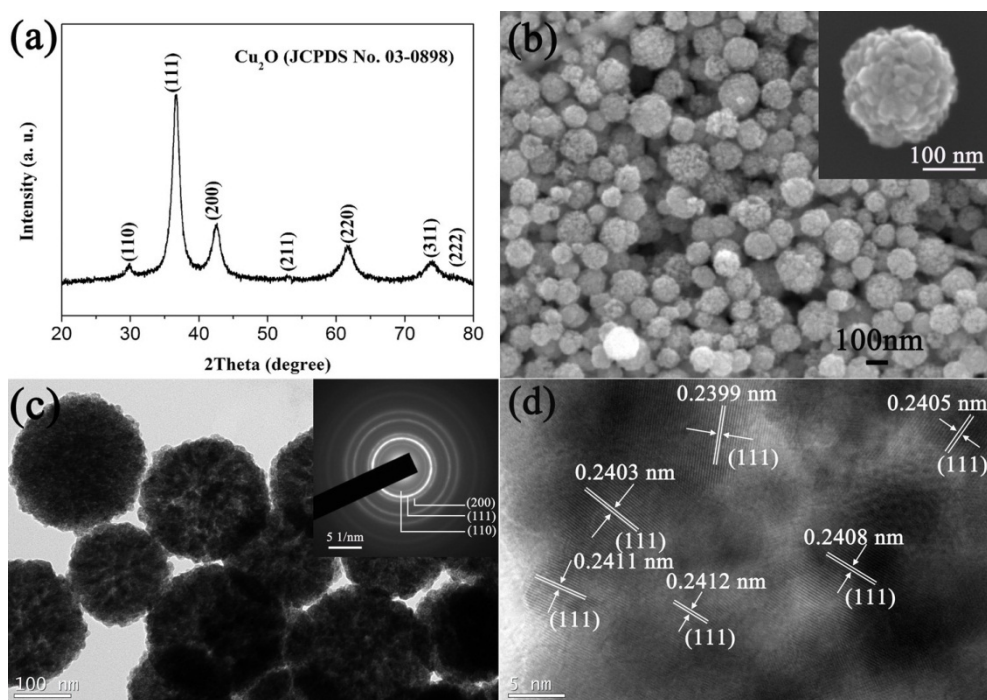


Fig. S1 (a) XRD pattern of the hierarchical Cu_2O templates; (b) FESEM image of the hierarchical Cu_2O nanospheres, and the inset is a typical FESEM image of an individual particle; (c) TEM image of the hierarchical Cu_2O nanospheres, and the inset is the corresponding SAED pattern; (d) HETEM image of the hierarchical Cu_2O nanosphere.

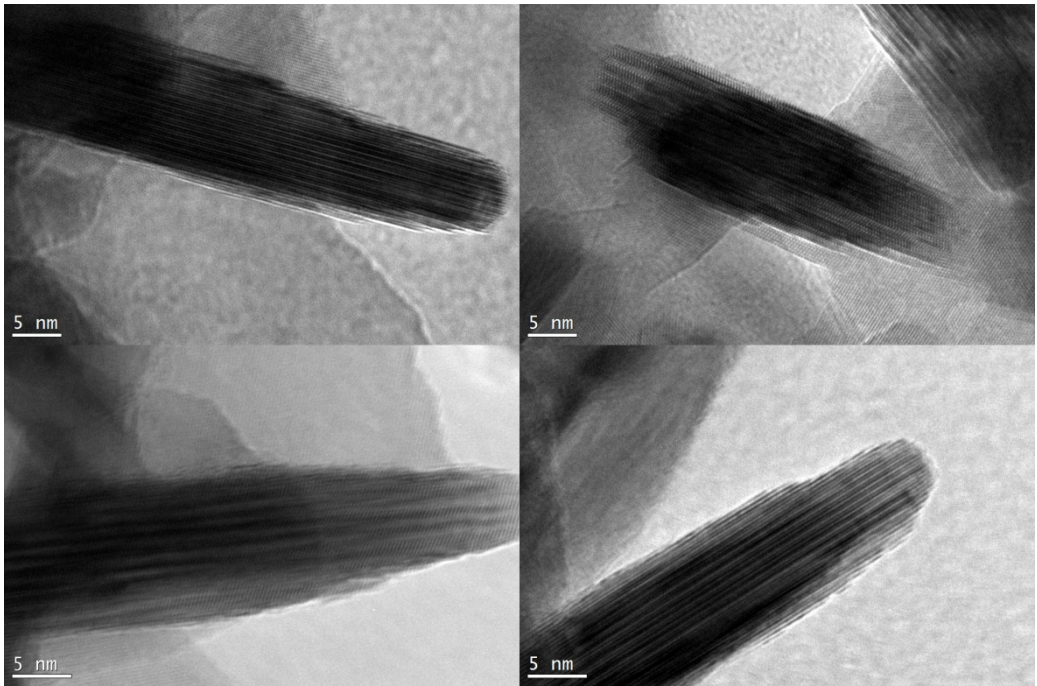


Fig. S2 HRTEM images of other four nanoplates of the as-prepared hierarchical Cu₇S₄ hollow nanocage assembly of twinned nanoplate building blocks.

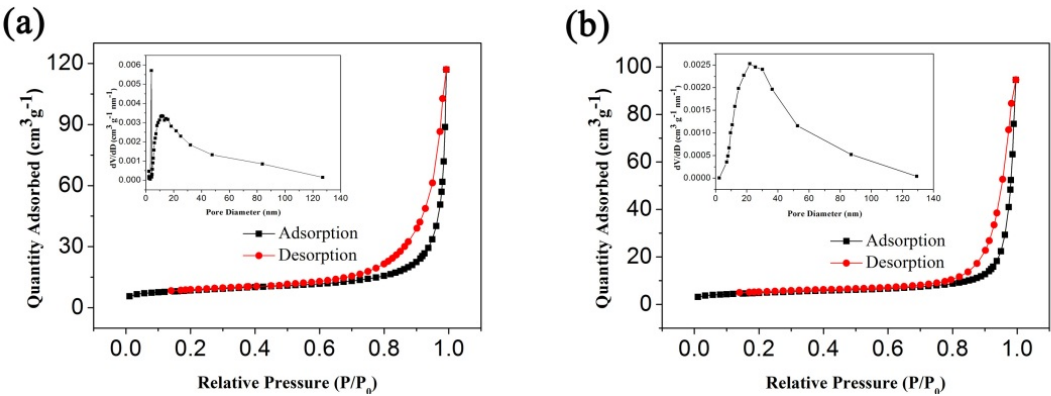


Fig. S3 Typical nitrogen gas adsorption-desorption isotherms and pore size distribution curves (insets) of the hierarchical Cu₇S₄ nanocages: (a) the hierarchical Cu₇S₄ hollow nanocages with twinned nanoplate building blocks; (b) the hierarchical Cu₇S₄ hollow nanocages with nanoparticle building blocks.

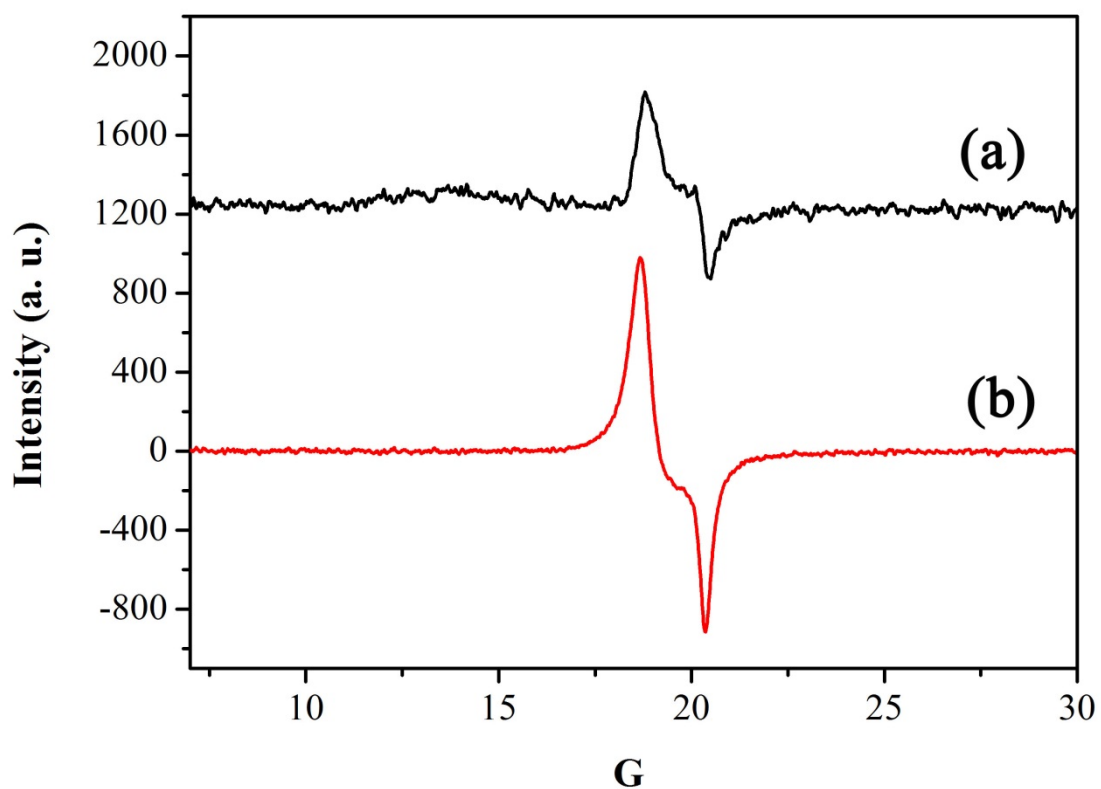


Fig. S4 Experimental ESR spectrum of the as-prepared two Cu_7S_4 nanocages. (a) the hierarchical Cu_7S_4 hollow nanocages with nanoparticle building blocks; (b) the hierarchical Cu_7S_4 hollow nanocages with twinned nanoplate building blocks.