Supporting information available for:

Decreasing Operating Potential for Water Electrolysis to Hydrogen via Local Confinement of Irion-Based Soft Coordination Suprapolymers

Yawei Liang, Limin Xu, Yinglin Zhou^{*}, Xinxiang Zhang, Jianbin Huang^{*}, Yun Yan^{*},

Beijing National Laboratory for Molecular Sciences (BNLMS), State Key Laboratory for Structural Chemistry of Unstable and Stable Species, Key Laboratory of Biochemistry and Molecular Engineering, College of Chemistry and Molecular Engineering, Peking University, Beijing, 100871, PR China

^{*} Correspondence: yunyan@pku.edu.cn; zhouyl@pku.edu.cn; jbhuang@pku.edu.cn

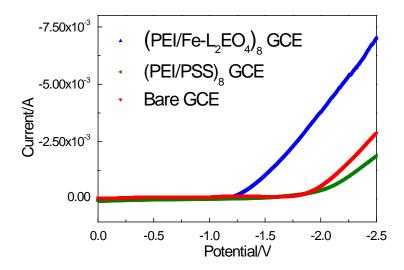
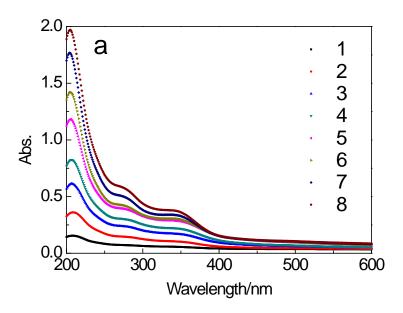


Figure S1. Experimental voltammetry of H⁺ reduction at the different GCE surface. Replacing the Fe-SCSP (the blue line) with another negatively charged electrolytespolystyrene sodium sulphite (PSS), only decreases of the proton reduction current without a potential movement was observed(the green line). This control experiment demonstrated that the Fe-SCSP is very crucial in the catalytic hydrogen evolution.



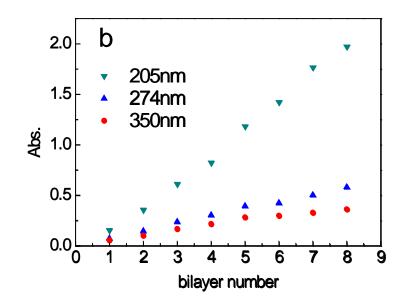


Figure S2. The linear growth of the mass of the layer by layer assembly film: (a) The UV-Vis spectrumscopy of the $(PEI/Fe^{3+}-L_2EO_4)_n$ layer by layer assembly film on the quartz glass substrate, bilayer number n = 1, 2, 3, 4, 5, 6, 7, 8; (b) And the intensity of this UV-Vis spectrumscopy growths linearly at certain wavelengths with an increasing of the bilayer number n. The absorptions at 274 nm and 350 nm wavelength characterize the coordination bonds in the Fe³⁺-L₂EO₄ structure.