

Electronic Supplementary Information (ESI)

Volume shrinkage of metal organic framework host induced by the dispersive attraction of guest gas molecules

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A comparison of the DFT-*ulg* method (PBE-*ulg*) with a post Hartree-Fock calculation such as Møller-Plesset perturbation (MP2)

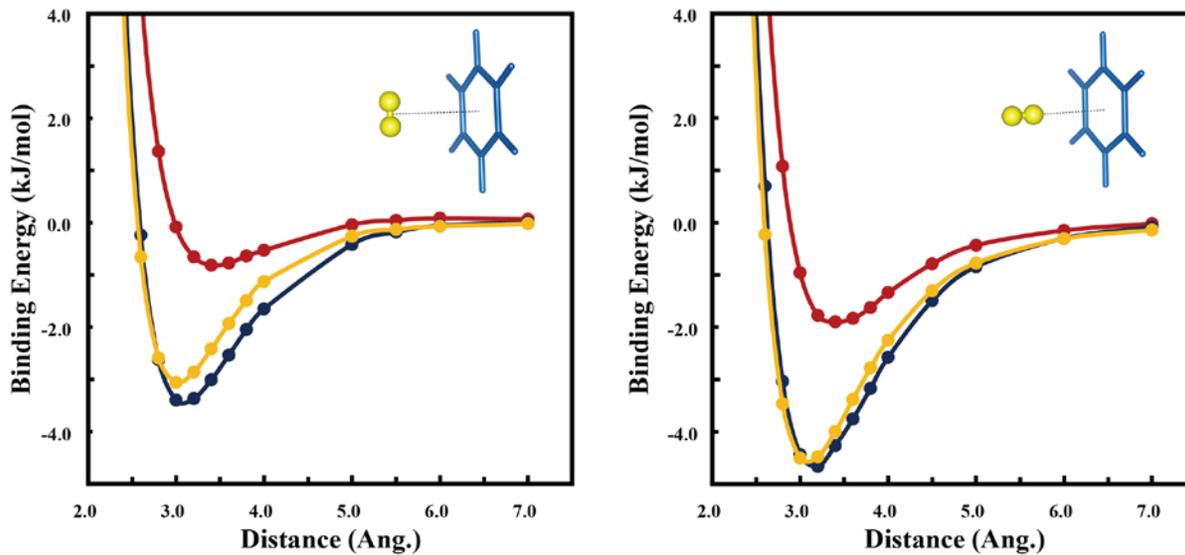


Fig. S1. Calculated H₂-C₆H₆ binding energies for the parallel (left) and vertical (right) configurations by PBE (red), PBE-*ulg* (blue), and MP2 (yellow) with a basis set of cc-pVQZ. The MP2 calculations were performed with the Q-CHEM software.

Variations of bonds and angles in MOF-5 after H₂ adsorption

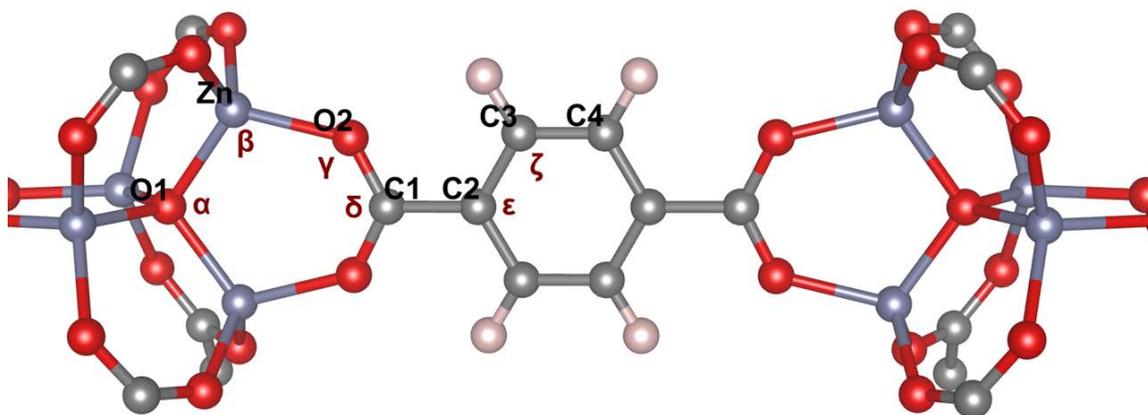


Fig. S2. Bonds and angles in MOF-5

Table S1. Variations of bonds in MOF-5 after H₂ adsorption, where the codes are in Fig. S2.

H ₂ adsorption	O1-Zn	Zn-O2	O2-C1	C1-C2	C2-C3	C3-C4
No adsorption (Å)	1.967	1.967	1.277	1.495	1.403	1.391
Site 1 (%)	0.0	-0.1	0.0	-0.1	0.0	0.0
Site 1+2 (%)	0.0	-0.1	0.0	-0.1	0.0	0.0
Site 1+2+3 (%)	-0.4	-0.3	0.0	-0.3	0.0	-0.1
Site 1+2+3+4 (%)	-0.4	-0.5	0.0	-0.4	-0.1	-0.1

Table S2. Variations of angles in MOF-5 after H₂ adsorption, where the codes are in Fig. S2.

H ₂ adsorption	α	β	γ	δ	ϵ	ζ
No adsorption (°)	109.48	111.52	130.61	126.25	119.95	120.02
Site 1 (%)	0.0	0.0	0.0	0.0	0.1	0.0
Site 1+2 (%)	0.0	0.0	0.0	0.0	0.1	0.0
Site 1+2+3 (%)	0.0	0.1	0.0	-0.1	0.2	-0.1
Site 1+2+3+4 (%)	0.0	0.1	-0.2	0.1	0.3	-0.1

Table S3. Variations of Bader charges in MOF-5 after H₂ adsorption, where the codes are in Fig. S2.

H ₂ adsorption	O1	Zn	O2	C1	C2	C3
No adsorption (e)	-0.1500	1.2909	-1.1523	1.5379	-0.0265	0.0488
Site 1 (e)	-1.1491	1.2884	-1.1520	1.5387	-0.0245	0.1103