

Room-temperature silver-containing liquid metal salts with nitrate anions - Electronic supporting information

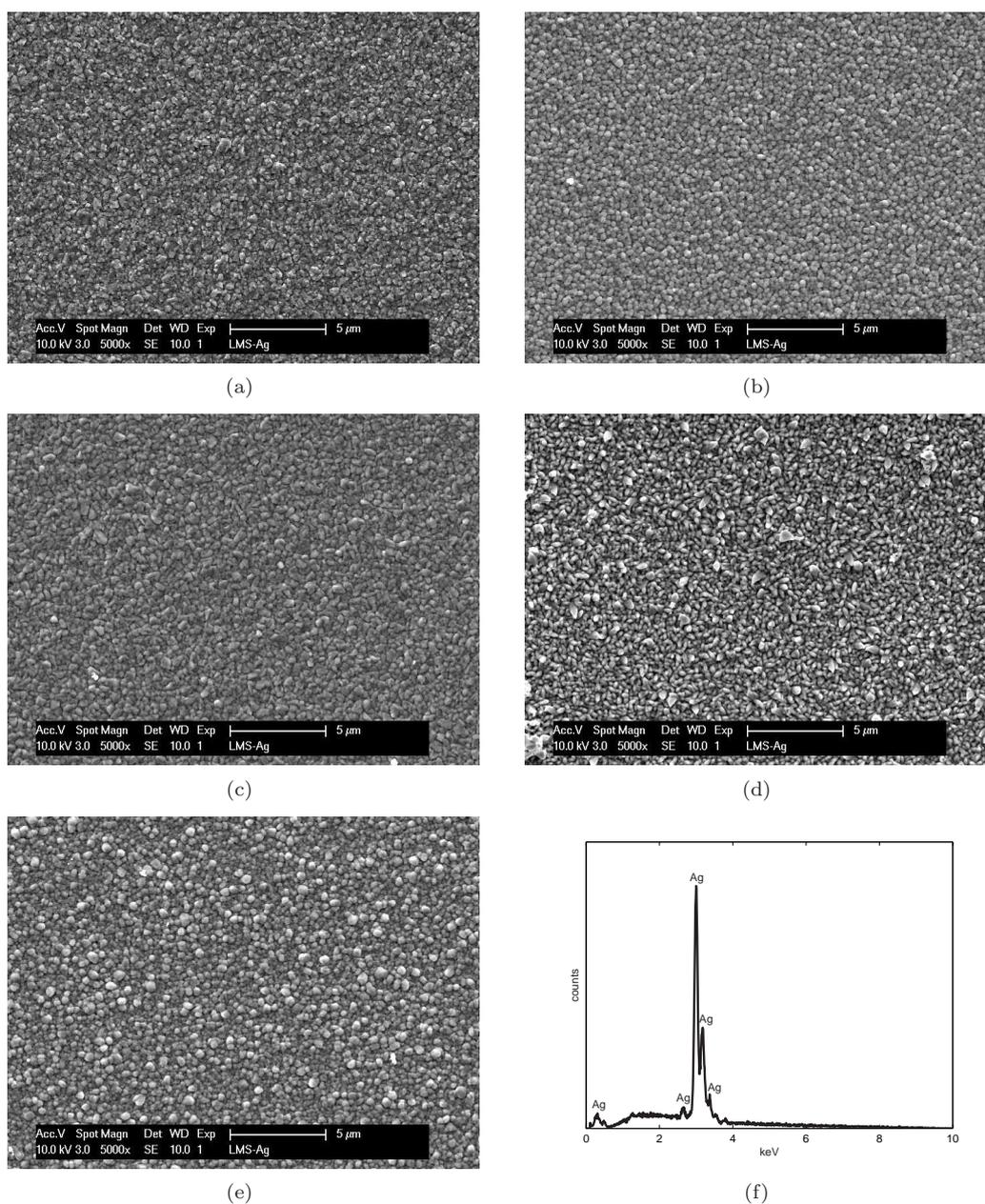


Figure 1: SEM micrographs of silver deposits from $[\text{Ag}(\text{EtIm})_2][\text{NO}_3]$ (a-b), $[\text{Ag}(\text{MeIm})(\text{EtIm})][\text{NO}_3]$ (c-d) and $[\text{Ag}(\text{BuIm})_2][\text{NO}_3]$ (e) on a Au working electrode at 90 °C obtained at -1 A dm^{-2} (a-c-e) and -5 A dm^{-2} (b-d). The theoretical thickness is $1 \mu\text{m}$. A typical EDX spectrum is shown in (f).

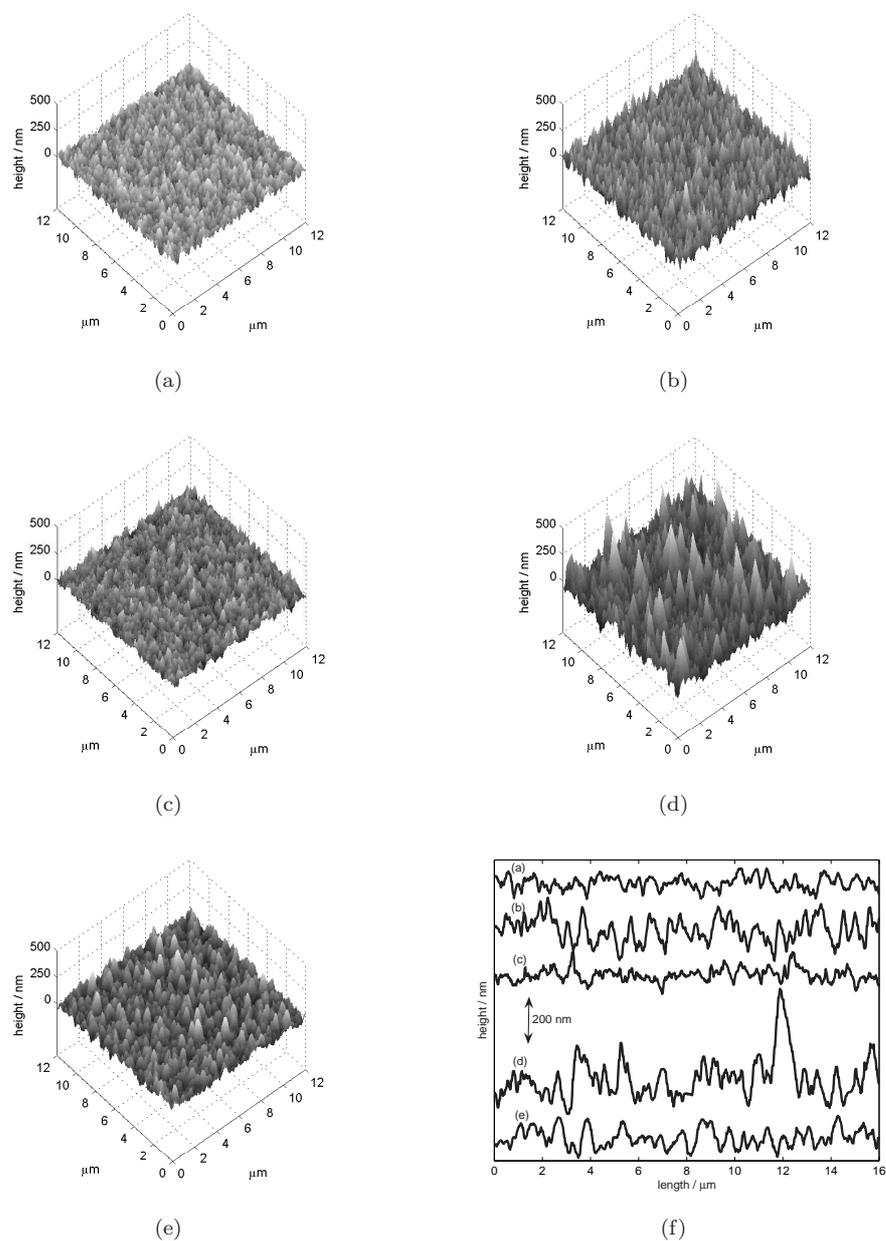


Figure 2: AFM images of silver deposits from [Ag(EtIm)₂][NO₃] (a-b), [Ag(MeIm)(EtIm)][NO₃] (c-d) and [Ag(BuIm)₂][NO₃] (e) on a Au working electrode at 90 °C obtained at -1 A dm⁻² (a-c-e) and -5 A dm⁻² (b-d). (f): roughness profiles.