

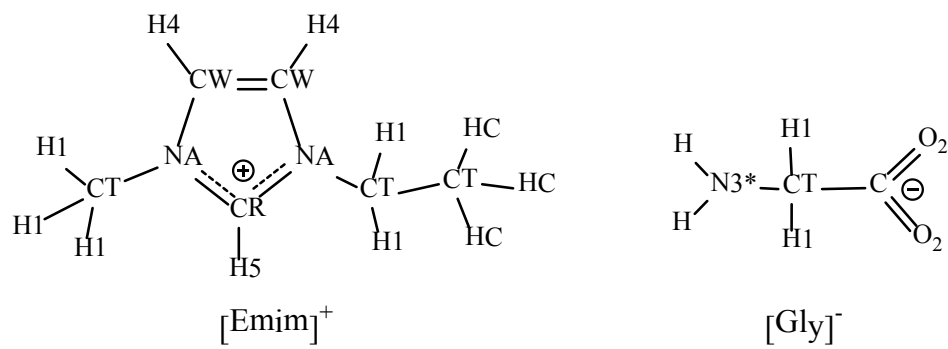


Fig. S1 Schematic structure and atom type notations of the 1-ethyl-3-methylimidazolium cation ($[\text{Emim}]^+$) and glycine anion ($[\text{Gly}]^-$) in the AMBER force field.

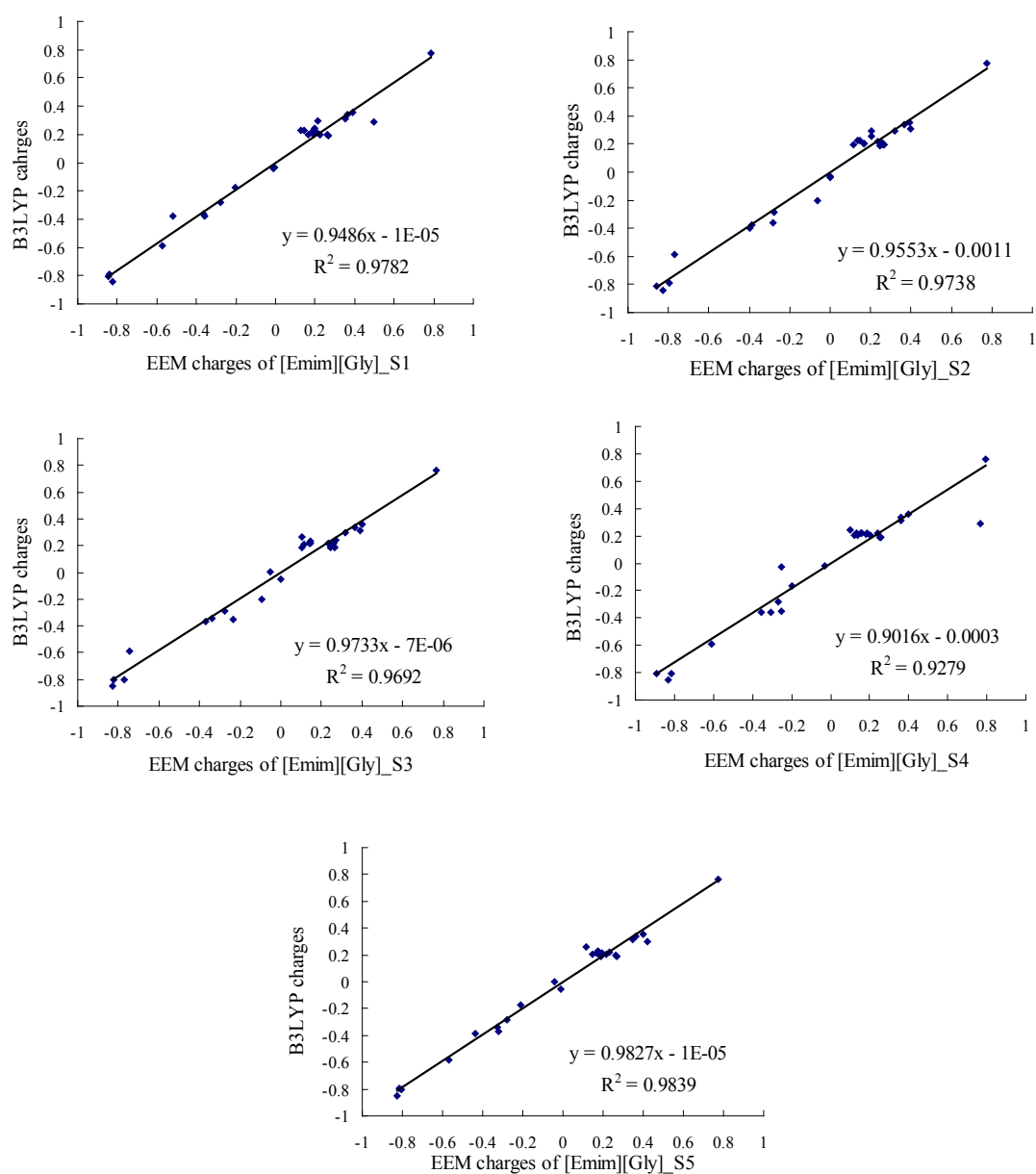


Fig. S2 The EEM/MM charges vs the corresponding B3LYP/6-311+G(d,p) charges (e) of five [Emim][Gly] conformers.

Table S1 Force field parameters for the [Emim][Gly] ILs used in this work.^a

Bonds							
	K_r (kJ·mol ⁻¹ ·Å ⁻²)	r_0 (Å)		K_r (kJ·mol ⁻¹ ·Å ⁻²)	r_0 (Å)		
CT-CT	1297	1.52	CW-CW	1715	1.34		
CT-H1	1423	1.08	CT-HC	1423	1.09		
CW-H4	1611	1.07	CT-H1	1359.8	1.0864		
CR-H5	1590	1.07	CT-C	1326.3	1.5499		
CR-NA	1674	1.32	CT-N3*	1535.5	1.4608		
CW-NA	1506	1.38	C-O2	2092	1.2374		
CT-NA	1172	1.47	N3*-H	1882.8	1.0036		
Angles							
	K_θ (kJ·mol ⁻¹ ·rad ⁻²)	θ_θ (deg)		K_θ (kJ·mol ⁻¹ ·rad ⁻²)	θ_θ (deg)		
CT-CT-H1	159	109.5	H4-CW-NA	126	122.1		
CT-CT-HC	155	109.5	H5-CR-NA	126	125.7		
CT-CT-NA	293	112.2	CW-CW-H4	126	130.7		
H1-CT-H1	146	109.5	C-CT-N3*	376.6	116.5		
HC-CT-HC	142	109.5	C-CT-H1	242.7	108.1		
H1-CT-NA	230	109.5	N3*-CT-H1	251	108.8		
CW-NA-CT	209	125.7	H1-CT-H1	161.1	106.1		
CR-NA-CT	209	126.3	CT-C-O2	334.7	115.4		
CW-CW-NA	502	107.1	O2-C-O2	355.6	129.1		
CR-NA-CW	502	108	CT-N3*-H	209.2	107.3		
NA-CR-NA	502	109.9	H-N3*-H	209.2	103.4		
Proper Torsion							
	K_ϕ (kJ mol ⁻¹ rad ⁻²)	γ (deg)	n		K_ϕ (kJ mol ⁻¹ rad ⁻²)	γ (deg)	n
NA-CR-NA-CW	50.21	180	2	NA-CT-CT-HC	0.669	0	3
NA-CR-NA-CT	8.368	180	2	H1-CT-CT-HC	0.628	0	3
H5-CR-NA-CW	6.276	180	2	HC-CT-CT-HC	0.628	0	3
H5-CR-NA-CT	6.276	180	2	H1-CT-NA-CW	1.021	0	3
CW-CW-NA-CR	50.21	180	2	H1-CT-NA-CR	0.686	0	3
CW-CW-NA-CT	8.368	180	2	CT-CT-NA-CW	-0.745	0	1
H4-CW-NA-CR	8.368	180	2	CT-CT-NA-CR	-0.987	0	1
H4-CW-NA-CT	6.276	180	2	N3-CT-C-O2	0.962	0	2
NA-CW-CW-H4	6.276	180	2	H1-CT-C-O2	0	0	2
NA-CW-CW-NA	50.21	180	2	H1-CT-N3*-H	0.92	0	3
H4-CW-CW-H4	6.276	180	2				
Improper Torsion							
	K_ϕ (kJ mol ⁻¹ rad ⁻²)	γ (deg)	n		K_ϕ (kJ mol ⁻¹ rad ⁻²)	γ (deg)	n
NA-NA-CR-H5	4.602	180	2	CR-CW-NA-CT	8.368	180	2
CW-NA-CW-H4	4.602	180	2	CT-O2-C-O2	43.93	180	2

^a to be continued.

Table S1 (continued) Force field parameters for the [Emim][Gly] ILs used in this work.^a

	Van der Waals			
	σ_i (Å)	ε_i (kJ/mol)		ε_i (kJ/mol)
CT	3.3997	0.4577	O2	0.8786
C	3.3997	0.3598	H	0.0657
CW	3.3997	0.3598	HC	0.0657
CR	3.3997	0.3598	H1	0.0657
NA	3.2500	0.7113	H4	0.0628
N3*	3.3409	0.7113	H5	0.0628

^aThe labels of atoms refer to those used in Fig. S1.

Table S2 Comparison of EEM/MM and B3LYP/6-311+G(d,p) vibrational frequencies (cm⁻¹) for gas-phase [Emim]⁺ and [Gly]⁻.

[Emim] ⁺							
EEM/MM	B3LYP	EEM/MM	B3LYP	EEM/MM	B3LYP	EEM/MM	B3LYP
52.2	46.4	58.3	70.5	138.6	135.1	206.6	208.3
245.7	232.8	295.7	291.9	370.2	378.1	453.9	425.3
547.7	591.7	596.3	632.6	626.6	660.2	662.1	698.2
717.9	752.9	818.6	802.5	885.5	834.0	896.4	881.0
914.1	962.0	937.8	1036.3	956.4	1042.8	982.1	1099.0
985.0	1102.3	1051.7	1124.6	1073.7	1135.8	1104.3	1149.8
1110.5	1175.5	1174.6	1270.3	1193.3	1311.4	1263.4	1342.8
1362.9	1384.9	1366.0	1411.0	1368.6	1428.0	1396.8	1437.3
1417.0	1461.0	1438.5	1484.1	1468.2	1491.0	1494.6	1494.3
1536.4	1508.9	1556.0	1510.4	1632.1	1594.4	1641.2	1603.5
2870.4	3046.5	2871.7	3069.4	2914.5	3079.2	2976.6	3110.2
2978.8	3122.3	2979.4	3134.1	2980.0	3150.0	2980.8	3165.0
3120.1	3273.7	3139.6	3278.0	3143.9	3291.2		
[Gly] ⁻							
EEM/MM	B3LYP	EEM/MM	B3LYP	EEM/MM	B3LYP	EEM/MM	B3LYP
40.6	95.7	295.4	190.0	332.4	294.8	464.9	481.3
566.6	573.5	675.2	672.6	824.8	854.2	891.7	935.1
1040.8	983.7	1151.3	1102.0	1207.1	1145.1	1361.4	1299.1
1418.2	1354.3	1494.1	1368.9	1619.9	1470.7	1678.3	1656.0
1752.6	1676.8	2918.8	2998.5	2977.2	3055.8	3345.8	3419.5
3386.8	3504.3						