

Supporting Information for:

Clear microstructure-performance relationships in
Mn-containing perovskite and hexaaluminate compounds
prepared by Activated Reactive Synthesis

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Physical and textural characterizations

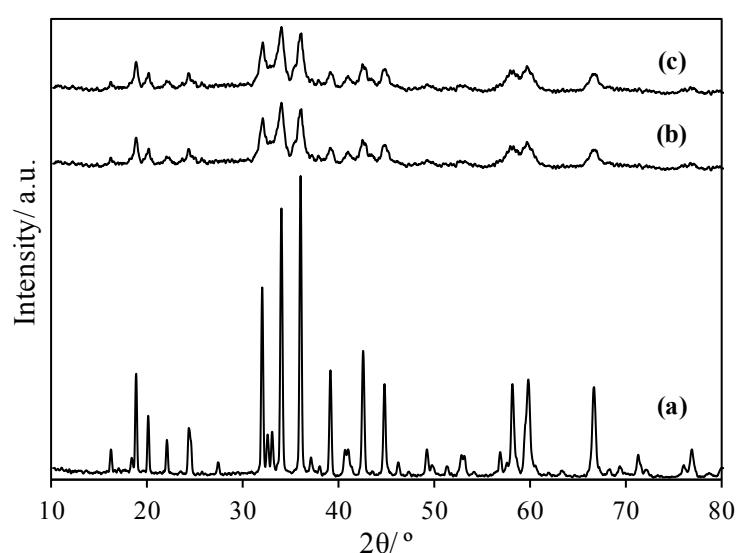


Fig. S1 Diffractograms recorded at each step of the synthesis for $\text{LaMnAl}_{11}\text{O}_{19}$. (a) SRR-LaMnAl, (b) HEBM-LaMn, and (c) HE-LE(RT)-LaMnAl.

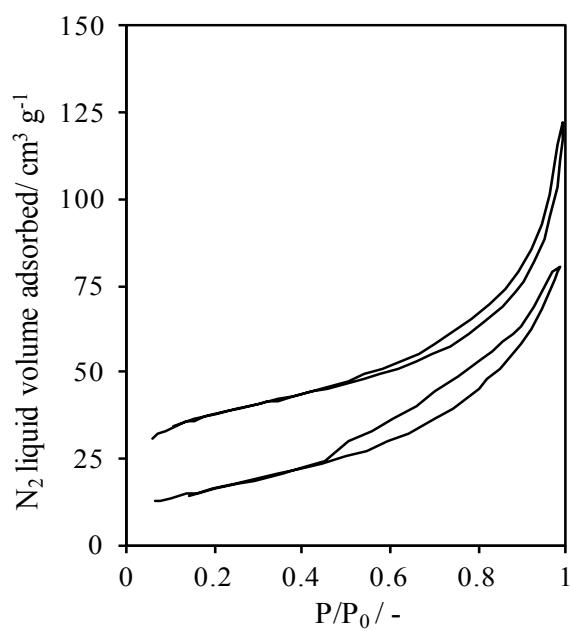


Fig. S2 Nitrogen adsorption/desorption isotherms recorded for selected mixed oxides (\blacktriangle) HE-LE(CR)-LaMn, (\bullet)HE-LE(RT)-LaMnAl.

Manganese reducibility

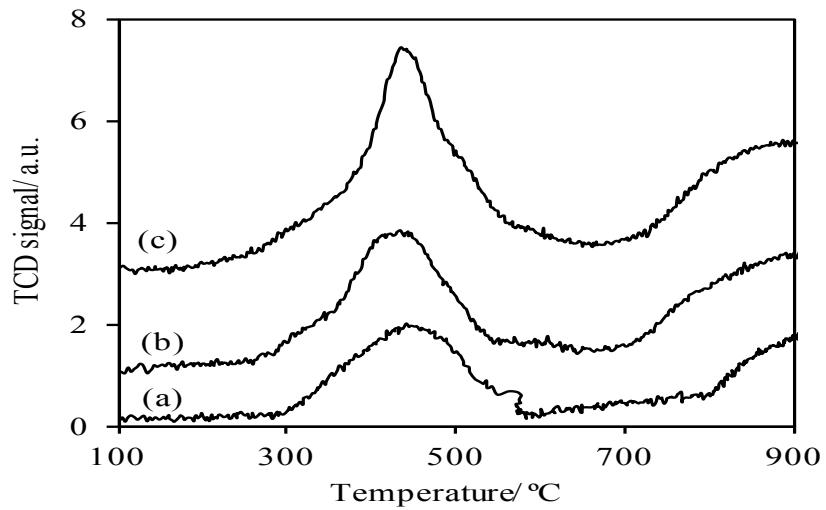


Fig S3 Temperature programmed reduction profiles recorded over $\text{LaMnO}_{3+\delta}$: (a) SSR-LaMn; (b) HEBM-LaMn; (c) HE-LE(CR)-LaMn.

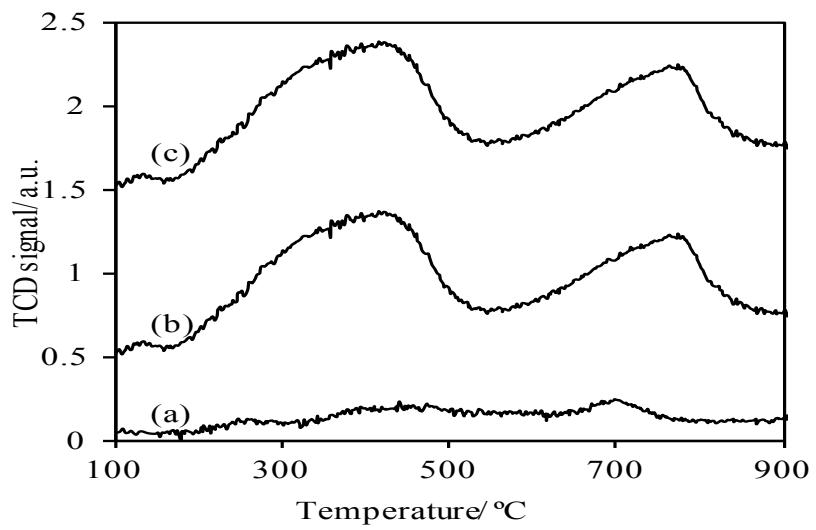


Fig S4 Temperature programmed reduction profiles recorded over $\text{LaMnAl}_{11}\text{O}_{19-\delta}$: (a) SSR-LaMnAl; (b) HEBM-LaMnAl; (c) HE-LE(RT)-LaMnAl.

Temperature-programmed isotopic exchange

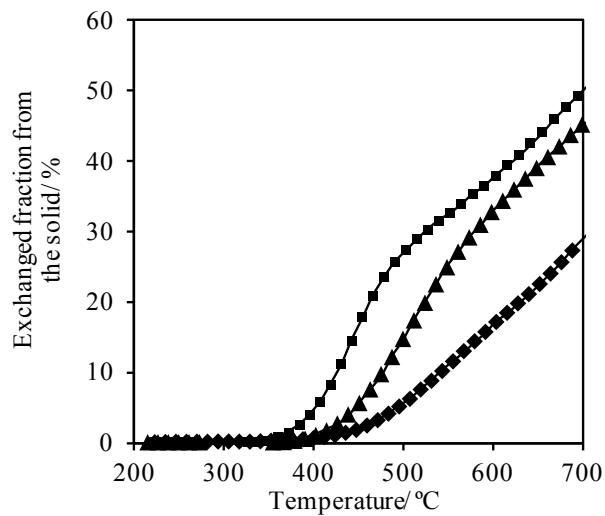


Fig S5 Evolution of the fraction of oxygen exchanged from the solid with the reaction temperature.
(◆) SSR-LaMn; (▲) HEBM-LaMn; (■) HE-LE (CR)-LaMn.

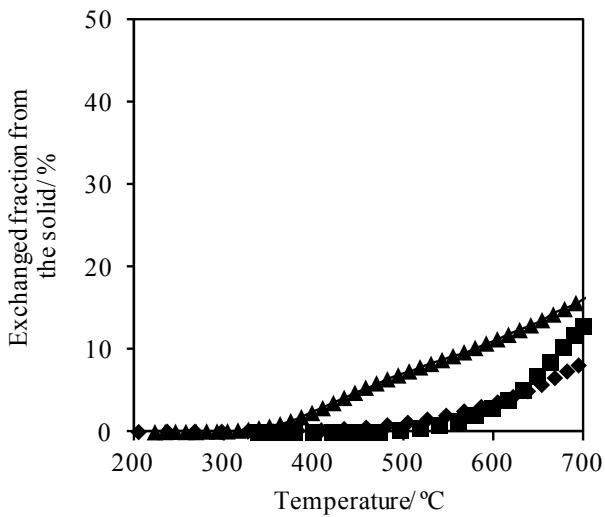


Fig S6 Evolution of the fraction of oxygen exchanged from the solid with the reaction temperature.
(◆) SSR-LaMnAl, (■) HEBM-LaMnAl, (▲) HE-LE(RT)-LaMnAl.