Supplemental materials

Rapid Synthesis of Nitrogen-Doped Graphene for Lithium

ion Battery Anode with Excellent Rate Performance and

Superlong Cyclic Stability

Tao Hu,^{a,b} Xiang Sun, ^bHongtao Sun, ^bGuoqing Xin,^b Dali Shao,^c Changsheng Liu,^aand Jie Lian^b*

 ^{a.} Key Laboratory for Anisotropy and Texture of Materials of Ministry of Education Northeastern University, Shenyang, Liaoning 110004, China
^{b.} Department of Mechanical, Aerospace & Nuclear Engineering, Rensselaer Polytechnic Institute, Troy, NY 12180, USA
^{c.} Department of Electrical, Computer and Systems Engineering, Rensselaer Polytechnic Institute, Troy, NY 12180, USA

[*] E-mail: lianj@rpi.edu

Nitrogen Source	Experimental method	Reaction time	Reference				
~99% NH ₃ +Ar (1:2V/V)	Heat treatment	2 h	14				
Acetonitrile vapors	CVD	3-15 min	15				
Lithium nitride	Solvothermal	6 h or 10 h	17				
Ammonia gas	Electrothermal	unknown	18				
Nitrogen plasma	Plasma assisted	20 min	19				
Urea	Hydrothermal	3 h	20				

Table S1 Common nitrogen-doped graphene by different method using different

dopants and experimental conditions



Figure S1. SEM images of the NGr show the porous morphologies (a) and the edge appearance (b)

Typical SEM images of the pGr at different magnifications(c and d).



Figure S2. XRD patterns of pGr, NGr and GO



Figure S3. N_2 adsorption-desorption isotherm (a) and the pore size distribution plot (b) for the NGr powder

Current density (Ag ⁻¹)	Charge (mAhg ⁻¹)	1 st cycle Discharge (mAhg ⁻¹)	Efficiency (%)	Charge (mAhg ⁻¹)	2000 th cycle Discharge (mAhg ⁻¹)	Efficiency (%)	Capacity retention (%)
10	163	190	85.8	179	180	99.4	94
20	90	119	75.6	106	106	101.9	87
30	47	76	61.8	65	65	100	85

Table S2 Rate performance of the NGr electrode under 10, 20 and 30Ag⁻¹



Figure S4. N species content of NGr electrodes by XPS after lithiation and delithiation



Figure S5. Nyquist plots of the pGr and NGr electrodes (inset, Modeled equivalent

circuit of EIS)