

CARBON NANOFIBER MESOPOROUS FILMS: EFFICIENT PLATFORMS FOR BIO-HYDROGEN OXIDATION IN BIOFUEL CELLS

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Supporting Information

Figure SI 1: CV curves for H₂ oxidation on a 855 $\mu\text{g}\cdot\text{cm}^{-2}$ CNF-modified PG electrode with no MbH1 in 50 mM HEPES buffer, pH 6.8 at 60°C under H₂ atm. in quiescent conditions. Scan rate was 5 $\text{mV}\cdot\text{s}^{-1}$.

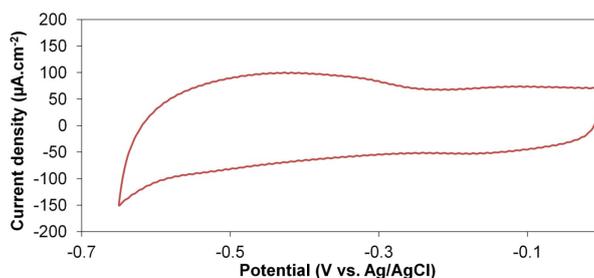


Figure SI 2: Adsorption kinetics (+) and stability of the catalytic signal (●) for H₂ oxidation by 2 μM MbH1 as a function of the incubation time on 855 $\mu\text{g}\cdot\text{cm}^{-2}$ CNF_H2-modified PG electrode. The currents are measured from CV curves for H₂ oxidation by MbH1 adsorbed on CNF_H2-modified PG electrodes in 50 mM HEPES buffer, pH 6.8 at 60°C under H₂ atm in quiescent conditions. The stability is reported as a percentage of the remaining current.

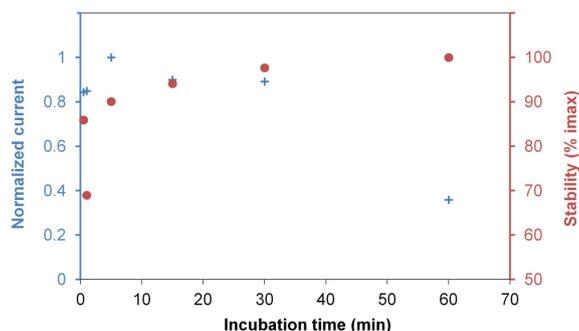


Figure SI 3: Impedance spectra recorded at a $36 \mu\text{g}\cdot\text{cm}^{-2}$ CNF_H2-modified electrode with $2\mu\text{M}$ MbH1 adsorbed for 30 min according to procedure (a) (+), and at a $36 \mu\text{g}\cdot\text{cm}^{-2}$ CNF_H2-modified electrode mixed with $2\mu\text{M}$ MbH1 according to procedure (b) (+).

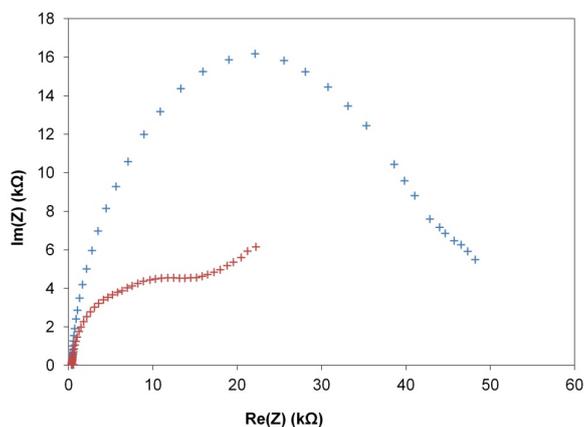


Figure SI 4: CV curves for H_2 oxidation by $2\mu\text{M}$ MbH1 adsorbed for 30 min on a bare PG electrode (A) or on a $855 \mu\text{g}\cdot\text{cm}^{-2}$ CNF_H2-modified PG electrode (B) in 50 mM HEPES buffer, pH 6.8 at 60°C under H_2 atm in H_2 OFF conditions (blue curves) and H_2 ON conditions at a H_2 flow rate $1 \text{ cm}^3\cdot\text{s}^{-1}$ (red curves). In (A) and (B) the blue and red CV curves are two consecutive cycles. Scan rate was $5 \text{ mV}\cdot\text{s}^{-1}$.

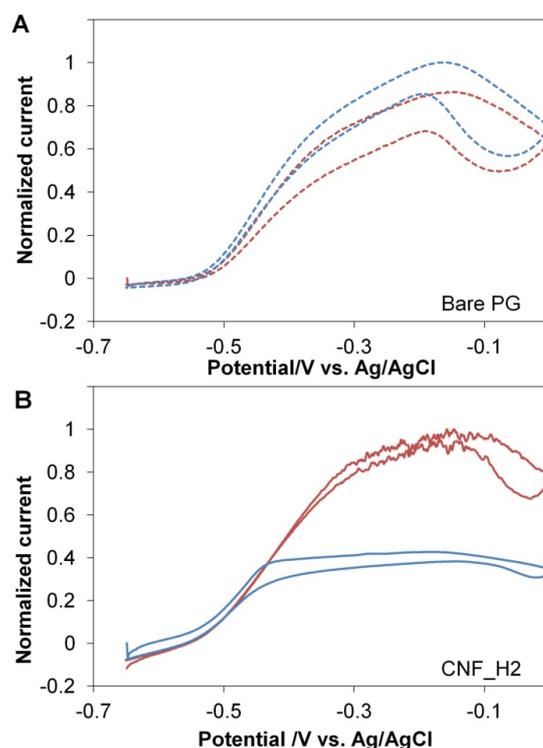


Figure SI 5: Adsorption isotherms (A) and kinetics (B) of MbH1 for two CNF deposits on a PG electrode. The currents are measured from CV curves for H₂ oxidation by MbH1 adsorbed on CNF_H2-modified PG electrodes in 50 mM HEPES buffer, pH 6.8 at 60°C, under H₂ atm. in quiescent conditions (blue marks), and H₂ ON conditions at a H₂ flow rate 1 cm³.s⁻¹ (red marks). In (A) the circles stand for a 285 μg.cm⁻² CNF_H2-modified PG electrode and the crosses for a 855 μg.cm⁻² CNF_H2-modified PG electrode.

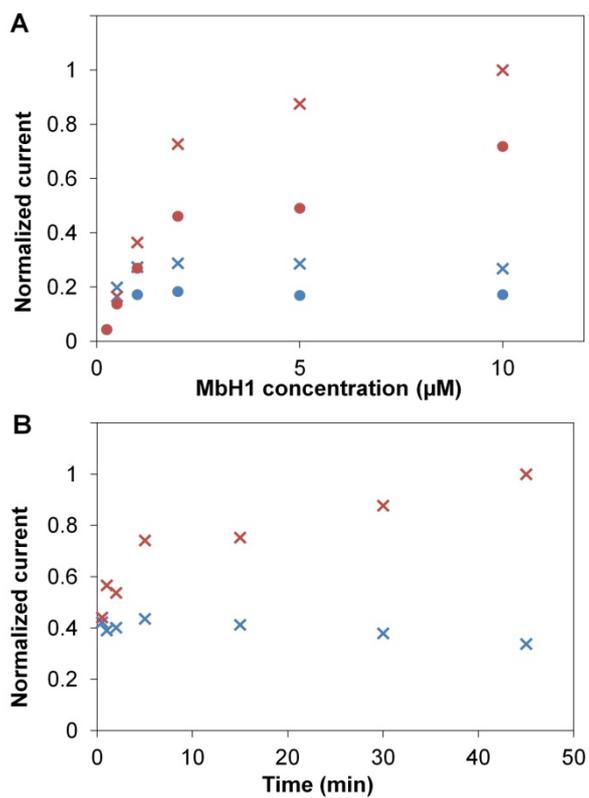


Figure SI 6: CV curves for H_2 oxidation by $2\mu\text{M}$ MbH1 adsorbed at a PG electrode modified by $1140\ \mu\text{g}\cdot\text{cm}^{-2}$ CNF under a H_2 flow rate of $5.2\ \text{cm}^3\cdot\text{s}^{-1}$. $V = 5\ \text{mV}\cdot\text{s}^{-1}$, 50 mM HEPES buffer, pH 6.8, $60\ ^\circ\text{C}$.

