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Electronic Supplementary Information, Tables S1, S2, S3 and S4

Table S1.

Resulting interatomic distances from the calculated $Al(NO_3)_3 \cdot 9H_2O$ crystal structure compared with room-temperature neutron diffraction data from Ref. 15. All experimental numbers have standard deviations of 1 in the last digit for the Al–O distances and 1 or 2 for the N–O distances.

Distances (Å) in $Al(H_2O)_6^{3+}$ octahedra			Distances (Å) in NO_3^- ions			
	Calc.	Exp.		Calc.	Exp.	
Al(1)-O(1)	1.893	1.881	N(1) -O(11)	1.257	1.234	
-O(2)	1.896	1.873	-O(12)	1.263	1.242	
-O(3)	1.900	1.886	-O(13)	1.288	1.263	
Al(2)-O(4)	1.908	1.889	N(2)-O(21)	1.268	1.246	
-O(5)	1.884	1.868	-O(22)	1.268	1.238	
-O(6)	1.921	1.897	-O(23)	1.270	1.249	
			N(3)-O(31)	1.279	1.247	
			-O(32)	1.256	1.227	
			-O(33)	1.266	1.242	

Table S2.

Resulting O–H distances and H–O–H angles from the calculated Al(NO₃)₃·9H₂O crystal structure compared with room-temperature neutron diffraction data from Ref. 15. The experimental numbers have standard deviations of 0.002 Å for the bond distances and 0.002° for bond angles. Δ is the difference between calculated and experimental distances.

	O-I	H distances (Å	.)	H-O-H angles (°)		
	Calc.	Exp.	Δ	Calc	Exp	
O(1)-H(11)	0.999	0.975	+.024	112 4	1117	
-H(12)	1.001	0.979	+.022	112.4	111./	
O(2)-H(21)	1.009	0.980	+.029	109.1	107.9	
-H(22)	1.006	0.985	+.021	106.1	107.8	
O(3)-H(31)	0.995	0.972	+.027	1127	112.1	
-H(32)	1.014	0.985	+.029	112./	112.1	
O(4)-H(41)	1.003	0.981	+.022	110.7	110 4	
-H(42)	1.006	0.977	+.029	110.7	110.4	
O(5)-H(51)	1.014	0.981	+.033	109.6	107.5	
-H(52)	1.003	0.986	+.017	108.0	107.5	
O(6)-H(61)	0.995	0.966	+.029	109.2	107.9	
-H(62)	1.011	0.985	+.026	108.2	107.8	
O(7)-H(71)	1.002	0.974	+.028	106.2	105 7	
-H(72)	0.993	0.969	+024	100.5	103.7	
O(8)-H(81)	1.000	0.968	+.032	1077	106.0	
-H(82)	0.989	0.963	+.026	107.7	100.9	
O(9)-H(91)	0.980	0.938	+.042	105.0	105 7	
-H(92)	0.992	0.961	+.031	103.9	103.7	

Table S3.

Hydrogen bond distances and angles around the water molecules in the crystal structure of $Al(NO_3)_3$ ·9H₂O. Δ is the difference between calculated and experimental¹⁵ distances.

Hydrogen bond distances (Å) and angles (°)										
	O…O dist.			H…O dist.			O-H…O angle		O…O…O angle	
	Calc.	Exp.	Δ	Calc.	Exp.	Δ	Calc.	Exp.	Calc.	Exp.
O(1)-H(11)···O(11)	2.702	2.750	05	1.706	1.777	07	174.3	174.8	110 1	117.0
-H(12)···O(12)	2.666	2.726	06	1.669	1.752	09	173.9	173.2	118.1	117.9
O(2)-H(21)···O(7)	2.628	2.656	03	1.633	1.692	06	168.1	166.9	101.0	100.0
-H(22)…O(23)	2.642	2.670	03	1.637	1.685	05	177.3	178.6	101.8	
O(3)-H(31)····O(33)	2.717	2.722	01	1.737	1.766	03	167.5	166.7	110.0	110.1
-H(32)···O(7)	2.621	2.693	07	1.608	1.711	11	175.2	174.6	118.9	118.1
O(4)-H(41)····O(21)	2.685	2.723	04	1.682	1.742	06	179.1	178.0	115.2	116.1
-H(42)…O(22)	2.648	2.699	05	1.647	1.731	09	172.5	170.7	115.2	
O(5)-H(51)···O(8)	2.592	2.650	06	1.581	1.675	09	174.1	171.8	107.0	105.3
-H(52)····O(13)	2.649	2.650	01	1.647	1.666	02	176.6	175.5	107.9	
O(6)-H(61)···O(31)	2.689	2.703	01	1.713	1.764	05	165.9	162.9	100 (99.8
-H(62)…O(8)	2.632	2.680	05	1.627	1.701	07	172.2	172.3	100.6	
O(7)-H(71)···O(9)	2.693	2.761	07	1.694	1.791	10	173.8	173.8	02.9	02.0
-H(72)····O(9')	2.778	2.851	07	1.805	1.901	10	165.4	166.1	93.8	93.0
O(8)-H(81)····O(32)	2.687	2.782	10	1.690	1.814	12	174.6	177.1	1147	112.2
-H(82)…O(31)	2.706	2.757	05	1.765	1.849	08	157.6	156.2	114./	115.5
O(9)-H(91)···O(23)	3.065	3.054	+.01	2.349	2.339	+.01	129.3	132.7		
-H(91)…O(33)	2.920	2.900	+.02	2.079	2.127	05	142.8	138.8	113.3 114 1	116.8 114 5
-H(92)···O(13)	2.771	2.854	07	1.796	1.918	12	167.1	164.0		111.0

Table S4.

Charges of various types for the constituents in the complexes **A**, **B**, **C**, **D** and **E** (see Fig. 9) calculated at the PBE/6-311G** level with the Gaussian program and charges of the same constituents in the $Al(NO_3)_3$ ·9H₂O crystal (only Bader charges) calculated at the PBE/(PAW+plane-wave basis) with the VASP program.

	Al1	W1	W1'	W2	W2'	W3	W3'	W7	NO ₃
Complex	Bader charges /e								
A	2.74	-	-	0.26	-	-	-	-	-
В	2.60	0.07	0.07	0.07	0.07	0.07	0.07	-	-
С	1.59	-	-	-0.03	-	-	-	0.15	0.30
D	2.60	0.05	0.05	-0.08	0.05	0.06	0.05	0.07	-0.85
Е	2.60	-0.03	-0.03	-0.07	0.05	0.05	0.04	0.06	-0.89 -0.89 -0.89
Crystal	2.52	-0.04	-0.04	-0.04	-0.04	-0.03	-0.03	0.03	-0.79 -0.79 -0.79
Complex	Mulliken charges / e								
A	2.42	-	-	0.58	-	-	-	-	-
В	1.58	0.24	0.24	0.23	0.23	0.24	0.24	_	-
С	1.29	-	-	0.20	-	_	_	0.19	0.33
D	1.51	0.22	0.21	0.05	0.21	0.23	0.21	0.11	-0.75
Е	1.48	0.09	0.10	0.06	0.19	0.21	0.18	0.09	-0.81 -0.79 -0.80
Complex					NBO cl	harges / e			
A	2.77	-	-	0.23	-	-	-	-	-
В	1.95	0.18	0.18	0.17	0.17	0.18	0.18	-	-
С	1.54	-	-	0.00	-	-	-	0.15	0.31
D	1.92	0.17	0.16	0.02	0.16	0.17	0.16	0.08	-0.84
Е	1.92	0.08	0.08	0.02	0.16	0.16	0.15	0.06	-0.88 -0.88 -0.88
Complex	Chelpg charges / e								
А	2.72	-	-	0.28	-	-	-	-	-
В	1.76	0.21	0.21	0.19	0.19	0.23	0.23	-	-
С	1.23	-	-	0.17	-	-	-	0.22	0.38
D	1.99	0.15	0.16	-0.06	0.13	0.16	0.09	0.13	-0.74
Е	2.08	0.01	0.01	-0.07	0.11	0.16	0.05	0.09	-0.81 -0.82 -0.80