

**Electronic Supplementary Information (ESI) for**

**Tuning the Oxygen Reduction Activity of the Pt-Ni Nanoparticles under Specific Anion Adsorption by Varying Heat Treatment Atmosphere**

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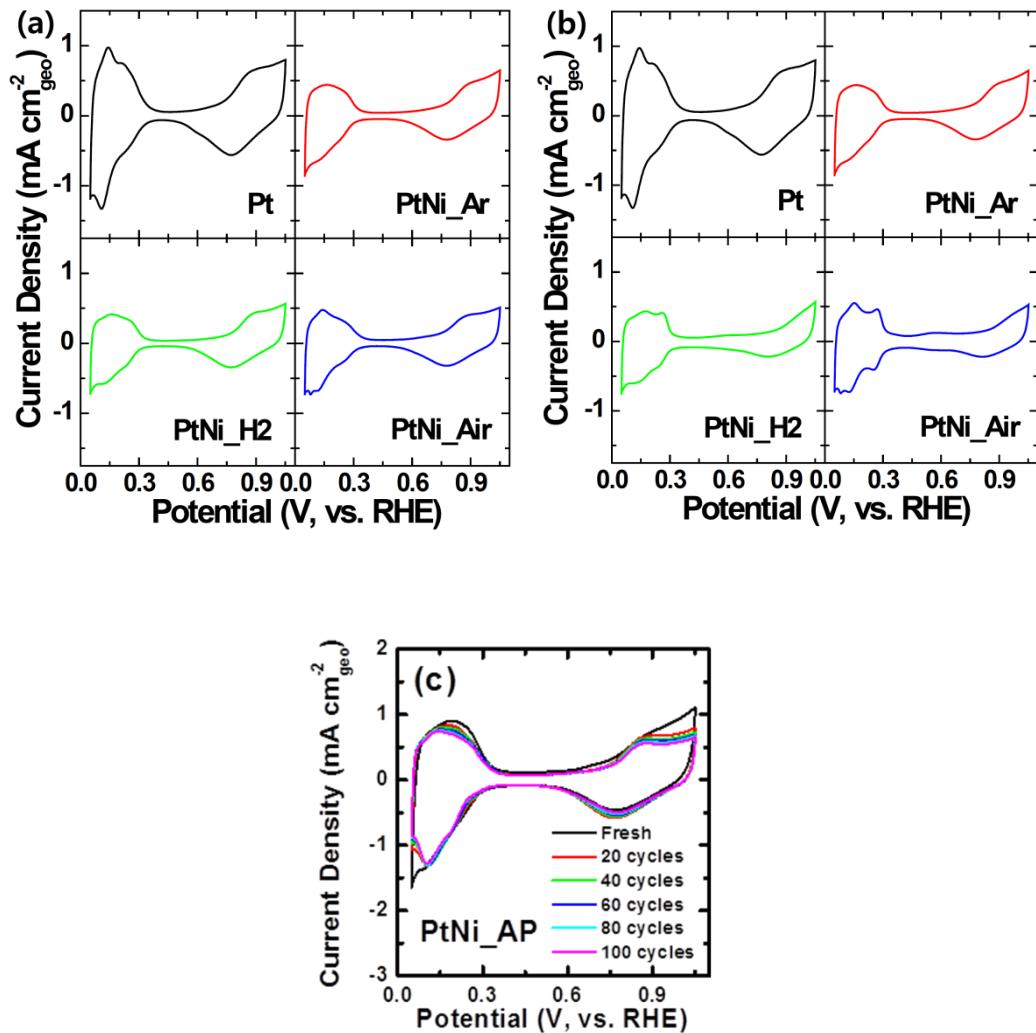
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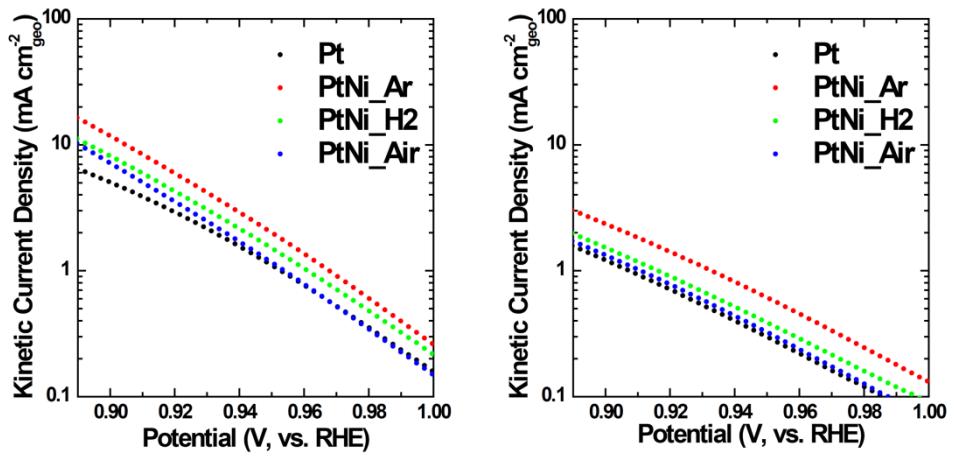
## S1. Supplementary Half-Cell Data

### S1.1. Cyclic voltamograms of Pt and PtNi nanoparticles



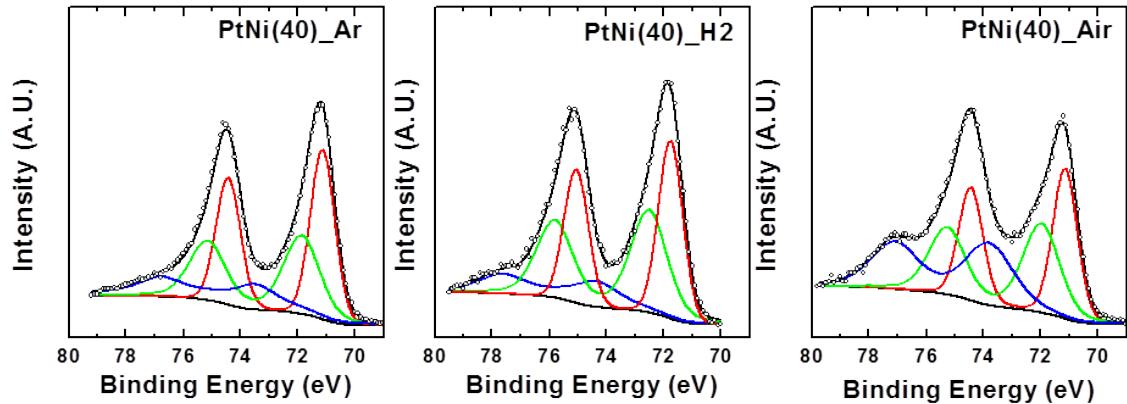
**Fig. S1.** Cyclic voltamograms (CVs) of Pt, PtNi\_Ar, PtNi\_H2, and PtNi\_Air nanoparticles in (a) 0.1 M HClO<sub>4</sub> and (b) 0.1 M HClO<sub>4</sub> + 0.1 M H<sub>3</sub>PO<sub>4</sub>; and (c) CVs of PtNi\_AP in 0.1 M HClO<sub>4</sub> after potential cycling. All measurements were carried out at 20 °C in saturated Ar condition (99.999%).

### S1.2. Kinetic current density

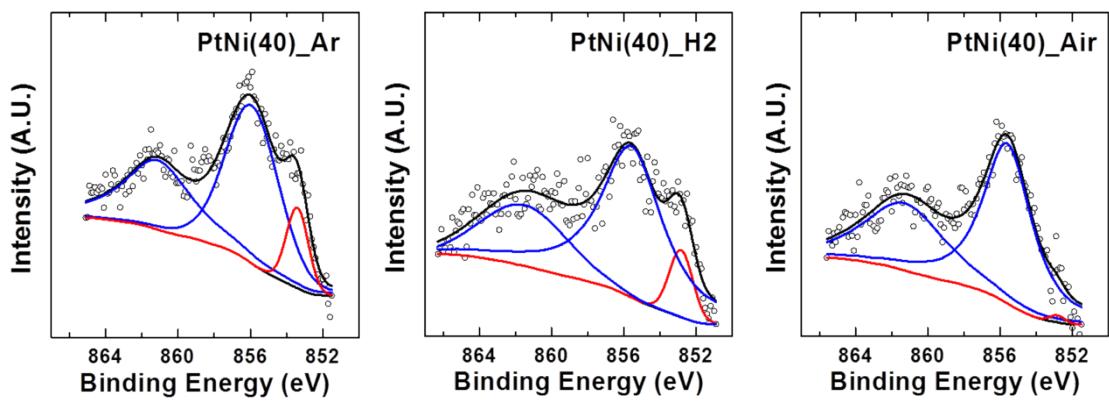


**Fig. S2.** Kinetic current density of Pt, PtNi\_Ar, PtNi\_H2, and PtNi\_Air nanoparticles in 0.1 M  $\text{HClO}_4$  (left) and 0.1 M  $\text{HClO}_4 + 0.1 \text{ M H}_3\text{PO}_4$ .(Right)

## S2. XPS Spectra of Pt 4f and Ni 2p



**Fig. S3.** XPS spectra of PtNi\_Ar (left), PtNi\_H2 (middle), and PtNi\_Air (right) nanoparticles for Pt 4f.



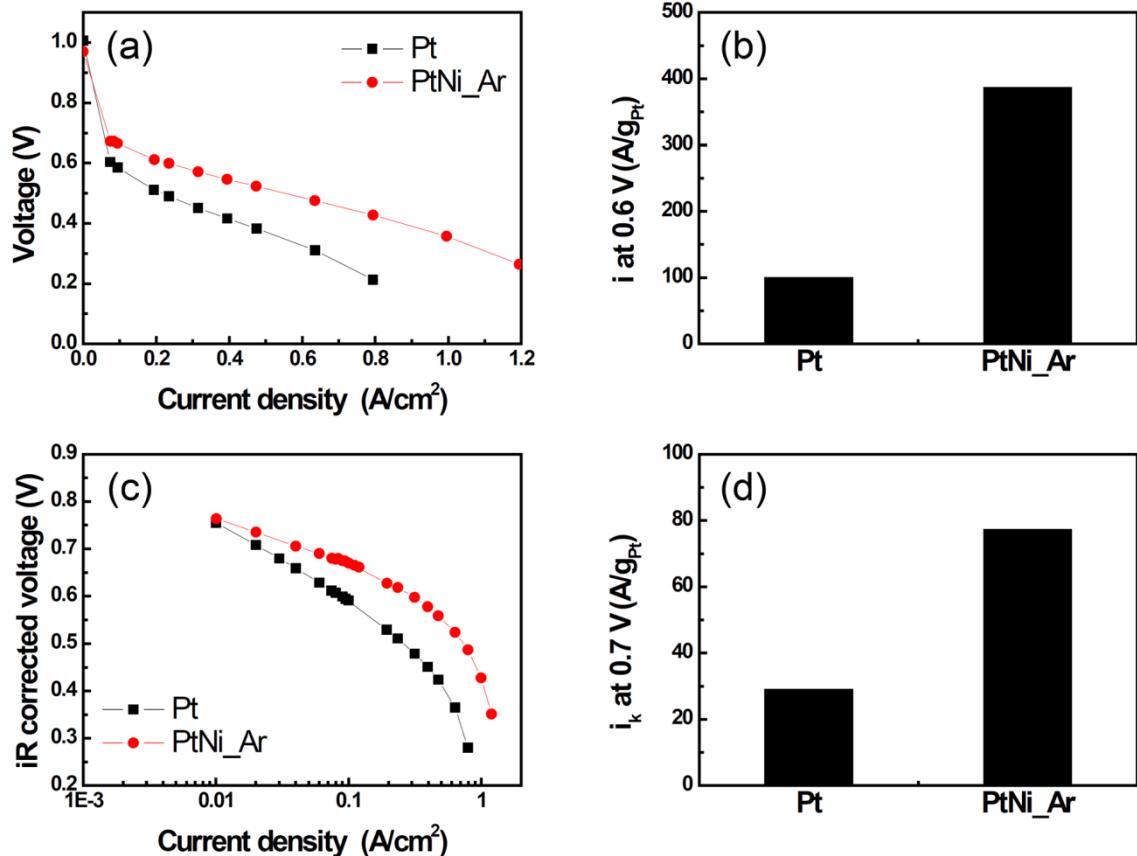
**Fig. S4.** XPS spectra of PtNi\_Ar (left), PtNi\_H2 (middle), and PtNi\_Air (right) nanoparticles for Ni 2p.

**Table S1.** Deconvolution results from XPS spectra of Pt 4f and Ni 2p.

Area Ratio (%)	Pt 4f			Ni 2p		$x^*$ ( $\text{Pt}_1\text{Ni}_x$ )
	Pt(0)	Pt(II)	Pt(IV)	Ni(0)	Ni(II)	
PtNi_Ar	52.4	33.4	14.2	10.0	90.0	0.44
PtNi_H2	56.6	23.9	19.5	11.6	88.4	0.45
PtNi_Air	33.0	36.1	30.9	0.6	99.4	0.50

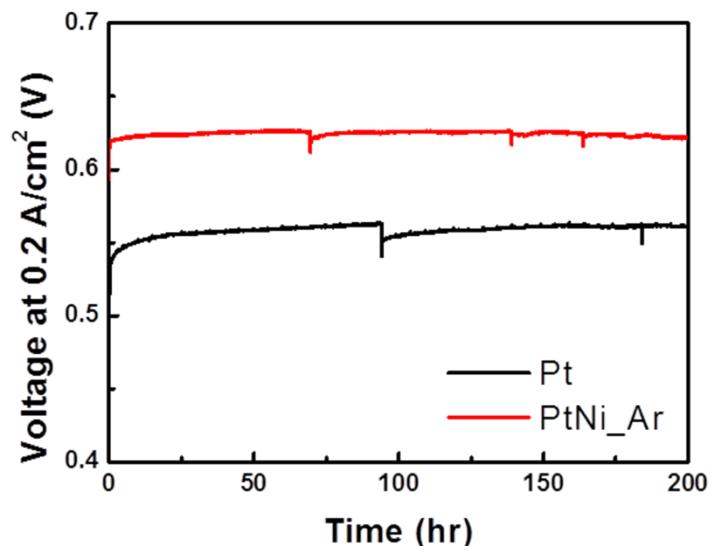
### S3. Supplementary Single-Cell Data

#### S3.1. Polarization curves of Pt and PtNi\_Ar under air feed



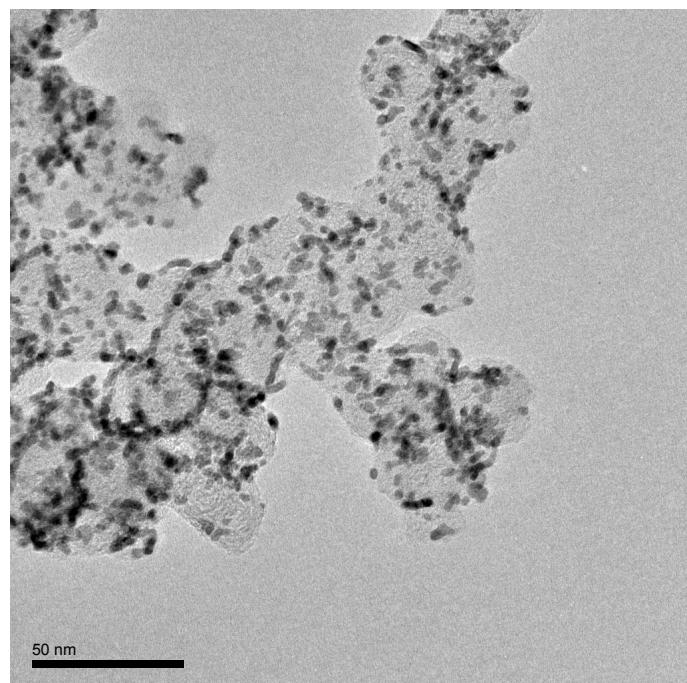
**Fig. S5.** (a) Polarization curve for Pt(HiSpec 4000, 40 wt%,  $0.78 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ) and PtNi\_Ar(40 wt%,  $0.6 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ) as cathode catalysts (anode catalyst  $1.0 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ). The feed gas to cathode is non-humidified air (stoichiometry 2) and cell temperature is  $160^\circ\text{C}$ . (b) Comparison of current density normalized with respect to platinum weight at 0.6V. (c) Tafel plot and (d) kinetic current density normalized with respect to platinum weight at 0.7 V

### S3.2. Durability test of PtNi\_Ar nanoparticles

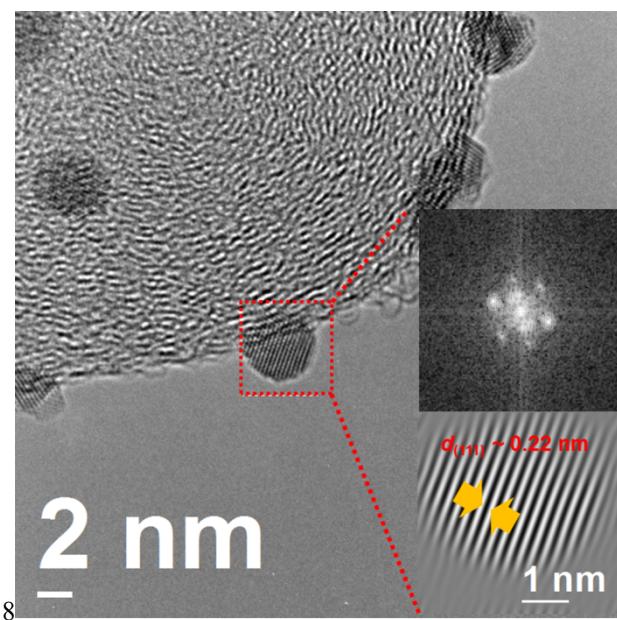


**Fig. S6.** Voltage at  $0.2 \text{ A/cm}^2$  for Pt (40 wt.%,  $0.78 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ) and PtNi\_Ar (40 wt%,  $0.6 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ) as cathode catalysts (anode catalyst  $1.0 \text{ mg}_{\text{Pt}} \text{ cm}^{-2}$ ). The feed gas to cathode is non-humidified air(stoichiometry 2) and cell temperature is  $160^\circ\text{C}$ .

#### S4. Supplementary TEM Images

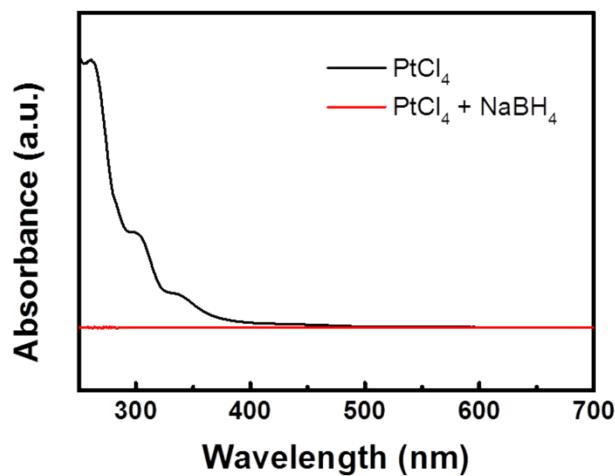


**Fig. S7.** TEM image of as-prepared  $\text{Pt}_1\text{Ni}_2 / \text{C}$ .



**Fig. S8.** High resolution-TEM image of the  $\text{PtNi}_\text{Ar}$ .

## S5. UV-vis Spectra



**Fig S9.** UV-vis spectra of  $\text{PtCl}_4$  solution in the ethanol before (black)/after (red) the addition of the reducing agent ( $\text{NaBH}_4$ ).