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Supplementary Information

Chemical bonding in electron-deficient boron oxide clusters: core boronyl group, dual 3c-4e hypervalent bond, and rhombic 4c-4e bond[†]

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- **Tables S1–S7**Natural charges, natural resonance theory (NRT) bond orders and atomic
valencies for the global-minimum structures 3-9 of $B_3O_2^+$, $B_3O_3^{-/0/+}$, and
 $B_3O_4^{-/0/+}$.
- Table S8Calculated vertical electron detachment energies (VDEs) at the TD-B3LYP level
based on the global minimum structures $D_{\infty h} B_3 O_2^- (\mathbf{1}, {}^{3}\Sigma_g), C_{2v} B_3 O_3^- (\mathbf{4}, {}^{1}A_1),$
and $C_{2v} B_3 O_4^- (\mathbf{7}, {}^{1}A_1).$
- Figs. S1–S7 Alternative optimized structures for B₃O₂⁺, B₃O₃^{-/0/+}, and B₃O₄^{-/0/+}, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.
- Figs. S8–S15 AdNDP bonding patterns and canonical molecular orbitals (CMOs) for the global-minimum structures 3–9 of $B_3O_2^+$, $B_3O_3^{-/0/+}$, and $B_3O_4^{-/0/+}$. A low-lying $D_{\infty h} B_3O_2^+$ ($^{1}\Sigma_g$) structure is also analyzed.

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- Gaussian 09, Revision A.2, M. J. Frisch, G. W. Trucks, H. B. Schlegel, G. E. Scuseria, M. A. Robb, J. R. Cheeseman, G. Scalmani, V. Barone, B. Mennucci, G. A. Petersson, H. Nakatsuji, M. Caricato, X. Li, H. P. Hratchian, A. F. Izmaylov, J. Bloino, G. Zheng, J. L. Sonnenberg, M. Hada, M. Ehara, K. Toyota, R. Fukuda, J. Hasegawa, M. Ishida, T. Nakajima, Y. Honda, O. Kitao, H. Nakai, T. Vreven, J. A. Montgomery, Jr., J. E. Peralta, F. Ogliaro, M. Bearpark, J. J. Heyd, E. Brothers, K. N. Kudin, V. N. Staroverov, R. Kobayashi, J. Normand, K. Raghavachari, A. Rendell, J. C. Burant, S. S. Iyengar, J. Tomasi, M. Cossi, N. Rega, J. M. Millam, M. Klene, J. E. Knox, J. B. Cross, V. Bakken, C. Adamo, J. Jaramillo, R. Gomperts, R. E. Stratmann, O. Yazyev, A. J. Austin, R. Cammi, C. Pomelli, J. W. Ochterski, R. L. Martin, K. Morokuma, V. G. Zakrzewski, G. A. Voth, P. Salvador, J. J. Dannenberg, S. Dapprich, A. D. Daniels, Ö. Farkas, J. B.
 - Foresman, J. V. Ortiz, J. Cioslowski, and D. J. Fox, Gaussian, Inc., Wallingford CT, 2009.

Table S1. Natural resonance theory (NRT) bond orders and atomic valencies of $C_{\infty v} B_3 O_2^+$ (3, $^1\Sigma$) at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in |e|) are also labeled.



 $C_{\text{ov}} B_3 O_2^+ \left({}^1\Sigma\right)$

			Natural B	ond Order		Natural Atomic Valency						
		B_1-O_4	B_2 - B_3	B_2-O_4	B ₃ -O ₅	O_5	B_3	\mathbf{B}_2	O_4	B_1		
NRT	t ^a	1.03	1.03	2.93	2.97	2.97	3.99	3.95	3.97	1.03		
	с	0.17	0.93	0.72	1.39	1.39	2.32	1.65	0.89	0.17		
	i	0.86	0.10	2.21	1.58	1.58	1.67	2.30	3.08	0.86		

^a t, the total bond orders; c, the covalent bond orders of NRT; i, the ionic bond orders of NRT.

Table S2. Natural resonance theory (NRT) bond orders and atomic valencies of $C_{2v} B_3 O_3^-(4, {}^1A_1)$ at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in lel) are also labeled.



 $C_{2v} B_3 O_3^{-} ({}^1A_1)$

			Nat	Natural Atomic Valencies						
		B_1 - B_2	B_2-B_3	B_1-O_4	B ₂ -O ₅	O ₃ -O ₆	O ₆	B ₃	B_2	O_5
NRT	t ^a	0.98	0.98	2.98	2.04	2.98	2.98	3.96	4.00	2.04
	с	0.90	0.90	1.08	0.95	1.08	1.08	1.98	2.75	0.95
i		0.08	0.08	1.90	1.09	1.90	1.90	1.98	1.25	1.09

^a t, the total bond orders; c, the covalent bond orders of NRT; i, the ionic bond orders of NRT.

Table S3. Natural resonance theory (NRT) bond orders and atomic valencies of C_{2v} B₃O₃ (**5**, ²A₁) at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in lel) are also labeled.



				Natural B	ond Order	•		Natural Atomic Valency						
		B ₁ - B ₃	B ₁ - O ₄	B ₁ -O ₅	B ₂ - O ₄	B ₂ -O ₅	B ₃ -O ₆	O_6	B ₃	\mathbf{B}_1	O_4	O ₅	B_2	
NRT ^a	t	0.99	1.45	1.45	1.46	1.46	3.00	3.00	3.98	3.88	2.91	2.91	2.92	
	с	0.95	0.49	0.49	0.42	0.42	1.23	1.23	2.18	1.93	0.91	0.91	0.84	
	i	0.04	0.96	0.96	1.04	1.04	1.77	1.77	1.80	1.95	2.00	2.00	2.08	

^a $C_{2v} B_3 O_3$ (²A₁) possesses two leading NRT reference structures: NRT₁ and NRT₂.



Table S4. Natural resonance theory (NRT) bond orders and atomic valencies of $C_{\infty v} B_3 O_3^+$ (6, ${}^{1}\Sigma$) at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in |e|) are also labeled.



			Natu	ıral Bond C	Order		Natural Atomic Valencies							
	$ B_1 - B_2 $							B_2	B_3	O_4	B_5	O ₆		
	t ^a	2.97	1.02	2.91	1.04	2.95	2.97	3.99	3.93	3.95	4.00	2.95		
INKI	с	1.40	0.93	0.72	0.34	1.31	1.40	2.32	1.65	1.06	1.65	1.31		
	i	1.57	0.09	2.19	0.70	1.64	1.57	1.67	2.28	2.89	2.35	1.64		

^a t, the total bond orders; c, the covalent bond orders of NRT; i, the ionic bond orders of NRT.

Table S5. Natural resonance theory (NRT) bond orders and atomic valencies of $C_{2v} B_3 O_4^-(7, {}^1A_1)$ at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in |e|) are also labeled in the structure.



		Natural Bond Order								Natural Atomic Valency						
		B ₁ - B ₂	B ₁ -O ₅	B ₁ -O ₇	B ₂ -O ₆	B ₃ -O ₄	B ₃ -O ₅	B ₃ -O ₇	O_6	B_2	\mathbf{B}_1	O_5	O ₇	B ₃	O_4	
NRT ^a	t ^b	0.99	1.47	1.47	2.99	2.04	1.38	1.38	2.99	3.98	3.93	2.86	2.86	4.81	2.04	
	с	0.95	0.53	0.53	1.13	0.82	0.34	0.34	1.13	2.08	2.01	0.88	0.88	1.51	0.82	
	i	0.04	0.94	0.94	1.86	1.22	1.04	1.04	1.86	1.90	1.92	1.98	1.98	3.30	1.22	

^a $C_{2v} B_3 O_4^{-}(^1A_1)$ possesses two significant NRT reference structures: NRT₁ and NRT₂.



^b t, the total bond orders; c, the covalent bond orders of NRT; i, the ionic bond orders of NRT.

Table S6. Natural resonance theory (NRT) bond orders and atomic valencies of C_s B₃O₄ (**8**, ²A') at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in |e|) are also labeled.



		Natural Bond Order								Natural Atomic Valencies					
	B_1-O_4 B_1-O_5 B_2-O_6 B_2-O_7 B_3-O_4 B_3-O_5 B_3-O_6							O_7	B_2	O_6	B ₃	O_4	O ₅	\mathbf{B}_1	
NRT ^a	t	1.47	1.47	1.08	2.91	1.42	1.42	1.04	2.91	3.99	2.12	3.88	2.89	2.89	2.93
	c	0.43	0.43	0.39	1.16	0.48	0.48	0.40	1.16	1.56	0.79	1.36	0.91	0.90	0.85
	i	1.04	1.04	0.69	1.75	0.94	0.94	0.64	1.75	2.43	1.33	2.52	1.98	1.98	2.08

^a $C_s B_3 O_4$ (²A') possesses two leading NRT reference structures: NRT₁ and NRT₂.



Table S7. Natural resonance theory (NRT) bond orders and atomic valencies of $D_{\infty h} B_3 O_4^+$ (9, ${}^1\Sigma_g$) at B3LYP/aug-cc- pVTZ. The bond lengths (in Å) and natural atomic charges (in |e|) are also labeled.



Natural Bond Order								Natural Atomic Valencies						
		O_1 - B_2	B ₂ -O ₇	B ₃ -O ₇	B ₃ -O ₄	O ₄ -B ₅	B5-O6	O_1	B_2	O 7	B ₃			
	T ^b	2.96	1.04	1.97	1.97	1.04	2.95	2.96	3.99	3.00	3.93			
NRT ^a	с	1.30	0.34	0.55	0.55	0.34	1.30	1.30	1.64	0.89	1.10			
	i	1.66	0.70	1.42	1.42	0.70	1.65	1.66	2.35	2.11	2.83			

^a $D_{\infty h} B_3 O_4^+$ (¹ Σ_g) possesses three significant NRT reference structures: NRT₁, NRT₂ and NRT₃.



^b t, the total bond orders; c, the covalent bond orders of NRT; i, the ionic bond orders of NRT.

Species	Feature	Final State	VDE (eV)
$B_3O_2^- D_{\infty h} (1, {}^3\Sigma_g)$	X	$^{2}\Pi_{u}$	3.07
	А	$^{2}\Pi_{u}$	6.43
$B_{3}O_{3}^{-}C_{2v}(4, {}^{1}A_{1})$	X	$^{2}\mathbf{B}_{2}$	4.06
	А	${}^{2}\mathbf{B}_{2}$	6.40
$B_{3}O_{4}^{-}C_{2v}(7, {}^{1}A_{1})$	X	$^{2}B_{2}$	5.10
	А	$^{2}A_{2}$	6.10

Table S8.Calculated vertical electron detachment energies (VDEs) at the TD-B3LYP level based on the
global minimum structures $D_{ooh} B_3 O_2^- (\mathbf{1}, {}^3\Sigma_g), C_{2v} B_3 O_3^- (\mathbf{4}, {}^1A_1), and C_{2v} B_3 O_4^- (\mathbf{7}, {}^1A_1).$

Figure S1. Alternative optimized structures for B₃O₂⁺, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S2. Alternative optimized structures for B₃O₃⁻, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S3. Alternative optimized structures for B₃O₃, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S4. Alternative optimized structures for B₃O₃⁺, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S5. Alternative optimized structures for B₃O₄⁻, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S6. Alternative optimized structures for B₃O₄, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



Figure S7. Alternative optimized structures for B₃O₄⁺, along with their point group symmetries, electronic states, and minimum vibrational frequencies. Relative energies are given at B3LYP/aug-cc-pVTZ and CCSD(T)//B3LYP/aug-cc-pVTZ (in curly brackets). The structures were obtained initially using the Coalescence Kick and Basin Hopping global-minimum search programs at B3LYP/3-21G.



+150.03 kcal/mol

Figure S8. (a) AdNDP bonding pattern for the $C_{\infty v} B_3 O_2^+$ (**3**, ${}^1\Sigma$) global-minimum structure and (b) its canonical molecular orbitals (CMOs).



Figure S9. (a) AdNDP bonding pattern for $D_{\infty h} B_3 O_2^+ ({}^1\Sigma_g)$ and (b) its canonical molecular orbitals (CMOs).



Figure S10. (a) AdNDP bonding pattern for the $C_{2v} B_3 O_3^- (4, {}^1A_1)$ global-minimum structure and (b) its canonical molecular orbitals (CMOs).



Figure S11. (a) AdNDP bonding pattern for the C_{2v} B₃O₃ (**5**, ²A₁) global-minimum structure and (b) its canonical molecular orbitals (CMOs).



Figure S12. (a) AdNDP bonding pattern for the $C_{\infty v} B_3 O_3^+$ (**6**, ${}^1\Sigma$) global-minimum structure and (b) its canonical molecular orbitals (CMOs).



Figure S13. Canonical molecular orbitals (CMOs) for the $C_{2v} B_3 O_4^-$ (7, ¹A₁) global-minimum structure. For its AdNDP bonding pattern, see Figure 4 in the text.



Figure S14. (a) AdNDP bonding pattern for the C_s B₃O₄ (**8**, ²A') global-minimum structure and (b) its canonical molecular orbitals (CMOs).



Figure S15. (a) AdNDP bonding pattern for the $D_{\infty h} B_3 O_4^+ (\mathbf{9}, {}^1\Sigma_g)$ global-minimum structure and (b) its canonical molecular orbitals (CMOs).

