# Ab initio Molecular Dynamics Simulations of Aqueous Triflic Acid Confined in Carbon Nanotubes 

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## Supplementary Information

Table S1. Hydrogen bond data for the optimized 14N3 system using different methods.

|  | Avg. $\mathrm{O} \cdots \mathrm{O}$ Distance $(\AA)$ |  | Avg. $\mathrm{H} \cdots \mathrm{O}$ Distance $(\AA)$ |  | Avg. $\|\delta\|(\AA)^{+}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Functional | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ |
| PBE | 2.65 | 2.56 | 1.65 | 1.53 | 0.64 | 0.49 |
| PBE0 | 2.64 | 2.55 | 1.66 | 1.54 | 0.67 | 0.51 |
| HSEsol | 2.64 | 2.55 | 1.66 | 1.54 | 0.67 | 0.51 |
| PBE-D3 | 2.65 | 2.57 | 1.66 | 1.53 | 0.65 | 0.49 |
| optB86 | 2.67 | 2.58 | 1.67 | 1.54 | 0.66 | 0.50 |

${ }^{\dagger}$ The value of $\delta$ always has water molecules/solvated protons as the hydrogen bond donor.

Table S2. Hydrogen bond data for the optimized 17F3 system using different methods.

|  | Avg. $\mathrm{O} \cdots \mathrm{O}$ Distance ( $\AA$ 足 |  | Avg. $\mathrm{H} \cdots \mathrm{O}$ Distance $(\AA)$ |  | Avg. $\|\delta\|(\AA)^{+}$ |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Functional | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{SO}_{3} \mathrm{H}$ | $\mathrm{H}_{2} \mathrm{O} / \mathrm{H}_{2} \mathrm{O}$ |
| PBE | 2.72 | 2.51 | 1.75 | 1.42 | 0.75 | 0.32 |
| PBE-D3 | 2.73 | 2.51 | 1.76 | 1.42 | 0.77 | 0.32 |
| optB86 | 2.75 | 2.52 | 1.77 | 1.43 | 0.78 | 0.33 |

${ }^{\dagger}$ The value of $\delta$ always has water molecules/solvated protons as the hydrogen bond donor.

Table S3. 14N3 dissociation data from sample MD runs using PBE and PBE-D2. ${ }^{\dagger}$

| Functional | Bound | Shared: <br> $\mathrm{SO}_{3} \mathrm{H} / \mathrm{SO}_{3} \mathrm{H}$ | Shared: <br> $\mathrm{SO}_{3} \mathrm{H} / \mathrm{H}_{2} \mathrm{O}$ | Dissociated: <br> Zundel | Dissociated: <br> $\mathrm{H}_{3} \mathrm{O}^{+}$ |
| :---: | :---: | :---: | :---: | :---: | :---: |
| PBE | - | - | 0.025 | 0.276 | 0.699 |
| PBE-D2 | - | - | 0.026 | 0.286 | 0.688 |

[^0]
(b) 17F2


Figure S1. Snapshots showing ring formation in (a) 17N2 and (b) 17F2.
(a) 14 N 3

(b) 17 N 3


(c) 17F3



Figure S2. Snapshots showing ring formation in (a) 14N3, (b) 17N3 showing a 5membered ring (top) and a 9-membered ring, and (c) 17F3 showing a 4-membered ring (top) and an 8-membered ring (bottom).
(a) 14 N 1

(b) 17 N 1


(d) 17 N 3


Figure S3. Snapshots of the bare CNT systems that contained OH $\cdots \mathrm{F}$ hydrogen bond between water molecules and a fluorine atom of triflic acid over 15\% of the time: (a) 14N1,
(b) 17 N 1, (c) 17 N 2 , and (d) 17 N 3 .
(a) 14 F 1

(b) 14 F 2

(c) 14 F 3


Figure S4. Snapshots of hydrogen bonding to the fluorinated wall in the smaller CNT systems: (a) 14F1, (b) 14F2, and (c) 14F3.


Figure S5. Snapshots of hydrogen bonding to the fluorinated wall in the larger CNT systems: (a) 17F1, (b) 17F2, and (c) 17F3.


[^0]:    ${ }^{+}$Data represents fraction of time spent in the different states

