**Characterization details:** Otsuka ELS Z particle analyzer was used to measure the hydrodynamic diameters of micelles. The morphology of the samples was observed under a field emission scanning electron microscope (FE-SEM, HITACHI SU-8000) and a transmission electron microscope (TEM, JEOL JEM-1210). The crystalline phases and crystalline degrees were measured by X-ray powder diffraction (XRD, SHIMADZU XRD-7000) analysis. X-ray photoelectron spectroscopy (XPS) spectra were taken using PHI Quantera SXM system equipped with an Al anode X-ray source. Thermogravimetric analysis (TGA) was carried out using a TG/DTA instrument (SEIKO-6300) at a heating rate of 10 °C·min<sup>-1</sup> in air. The surface area of samples was measured by a  $N_2$  adsorption/desorption isotherm.

**Electrochemical measurement:** Mesoporous TiO<sub>2</sub>-RuO<sub>2</sub> composite was homogeneously mixed with a polyvinylidine difluoride (PVDF, 20 wt%) in N-methylpyrolidinone solvent. The slurry was coated onto a graphite substrate. The electrodes were dried at 80 °C for 2 h in a vacuum oven. For comparison, mesoporous TiO<sub>2</sub> was also used. Each electrode contained 0.3 mg·cm<sup>-2</sup> of electroactive material. The electrochemical measurements were conducted in a three-electrode electrochemical cell with a Pt counter electrode and Ag/AgCl as a reference electrode in a 0.5 M H<sub>2</sub>SO<sub>4</sub> solution. The graphite substrate coated with mesoporous TiO<sub>2</sub>-RuO<sub>2</sub> composite was used as the working electrode. Cyclic voltammetry measurements were obtained using an electrochemical workstation (CHI 660E CH Instruments, USA) in the scan range of 0.1 to 0.9 V.





**Fig. S1** Photographs of (a) PS-*b*-PVP-*b*-PEO solution dissolved in THF and (b) PS-*b*-PVP-*b*-PEO micellar solution after addition of HCl solution (Tyndall scattering is clearly confirmed).

## Fig. S2



**Fig. S2** (a) SEM and (b) TEM image of Ti/Ru/PS-*b*-PVP-*b*-PEO composite micelles. Yellow-colored circles indicate the shell region of inorganic precursors.

Fig. S3



**Fig. S3** Thermogravimetry/differential thermal analysis (TG/DTA) of Ti/Ru/PS-*b*-PVP-*b*-PEO composite micelles.

## Fig. S4



Fig. S4 HRTEM image of mesoporous  $TiO_2$ -RuO<sub>2</sub> composites.





Fig. S5  $N_2$  adsorption-desorption isotherm of mesoporous TiO<sub>2</sub>-RuO<sub>2</sub> composite. Pore size distribution is also shown in an inset figure.

Fig. S6



Fig. S6 XPS spectra (a) Ti 2p and (b) Ru 3d of mesoporous TiO<sub>2</sub>-RuO<sub>2</sub> composite.





**Fig. S7** (a) Cyclic voltammograms (CVs) of mesoporous TiO<sub>2</sub>-RuO<sub>2</sub> composite at various scan rates and (b) effect of the scan rates on the capacitance performance.

Fig. S8



Fig. S8 Cyclic voltammogram (CV) of mesoporous TiO<sub>2</sub>.