

Electronic Supplementary Information

Ternary Pt/SnO_x/TiO₂ photocatalysts for hydrogen production: Consequence of Pt sites for synergy of dual co-catalysts

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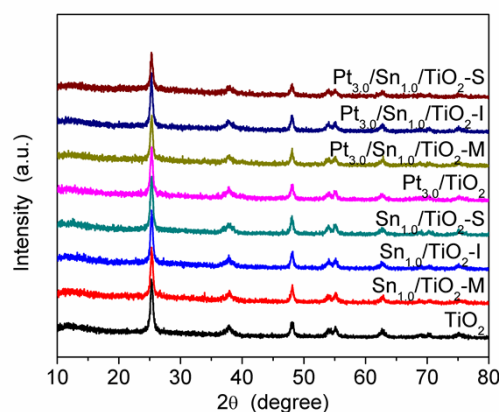


Fig. S1 The XRD patterns of TiO_2 , $\text{Pt}_{3.0}/\text{TiO}_2$, $\text{Sn}_{1.0}/\text{TiO}_2$ samples prepared by different methods, and $\text{Pt}_{3.0}/\text{Sn}_{1.0}-\text{TiO}_2$ samples prepared by different methods.

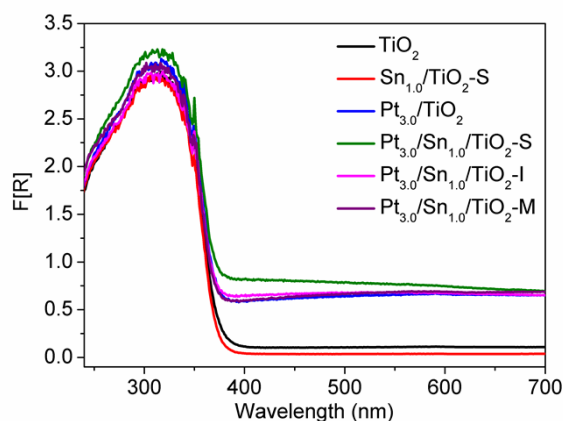


Fig. S2 UV-vis diffuse reflectance spectra of samples

Figure S2 shows UV-Vis DRS spectra of $\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$, $\text{Pt}_{3.0}/\text{TiO}_2$, $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$, $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-I}$ and $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-M}$ samples. Clearly, UV-Vis DRS spectrum of $\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$ is coincide with that of TiO_2 , indicating that Sn modification results in no alterations in the optical absorption properties of TiO_2 . The spectra of the all Pt loaded samples ($\text{Pt}_{3.0}/\text{TiO}_2$, $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$, $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-I}$ and $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-M}$) show an enhanced absorption in the visible-light region but the shape and optical absorption threshold of the spectra is similar with that of parent TiO_2 . According to the reference (J. Yu, L. Qi, M. Jaroniec, *J. Phys. Chem. C* 2010, **114**, 13118.), absorption in visible light region is due to the fact that highly dispersed platinum nanoparticles can absorb nearly all the incident light. Therefore, this result can further confirms that the Pt nanoparticles are highly dispersed onto surface of samples and closely contacted with TiO_2 .

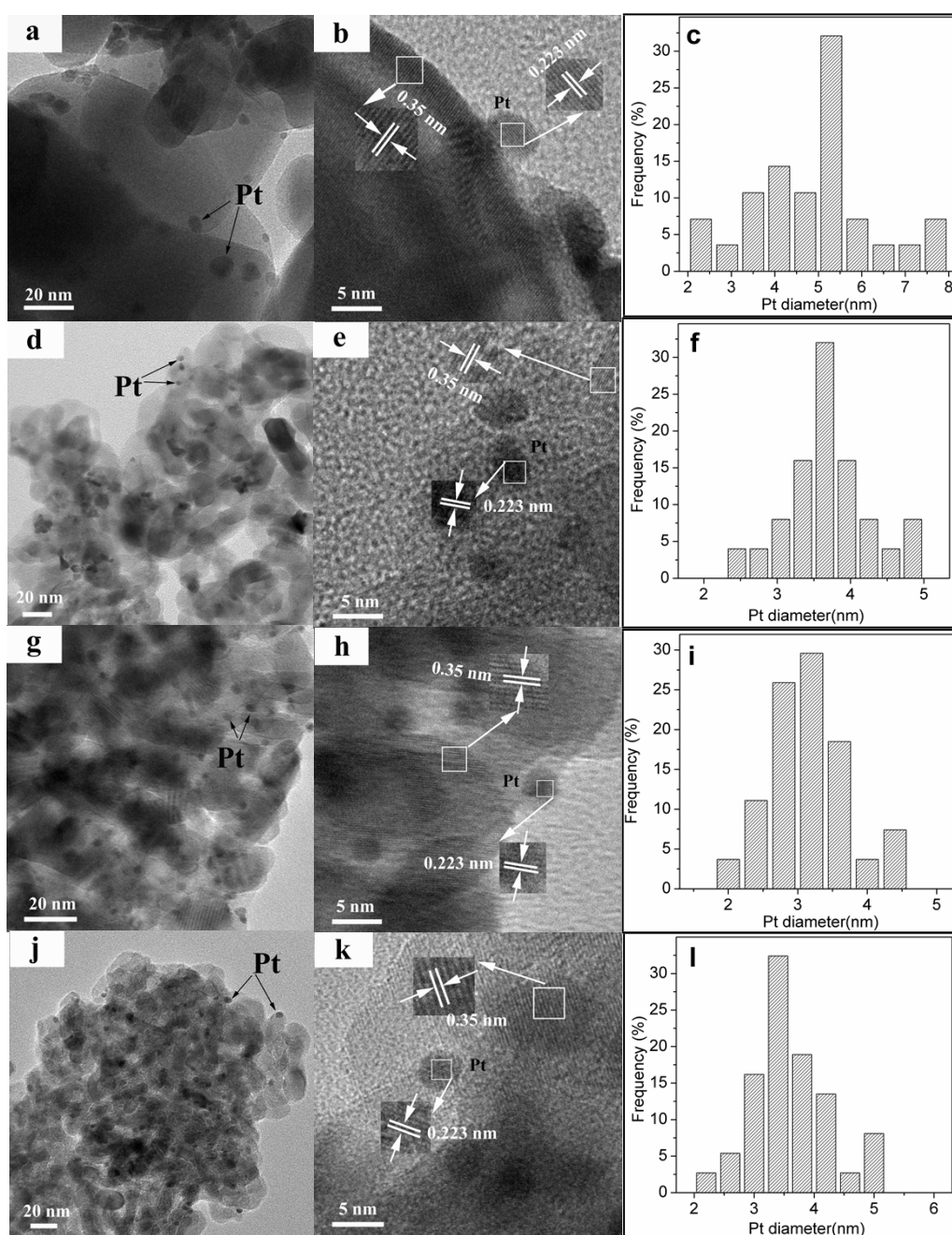


Fig. S3 TEM and HRTEM images of Pt_{3.0}/SnO₂ (a and b), Sn_{1.0}/Pt_{3.0}/TiO₂-M (d and e), Sn_{1.0}/Pt_{3.0}/TiO₂-I (g and h), and Sn_{1.0}/Pt_{3.0}/TiO₂-S (j and k) samples. The statistical size distributions of Pt particles on Pt_{3.0}/SnO₂ (c), Sn_{1.0}/Pt_{3.0}/TiO₂-M (f), Sn_{1.0}/Pt_{3.0}/TiO₂-I (i), and Sn_{1.0}/Pt_{3.0}/TiO₂-S (l).

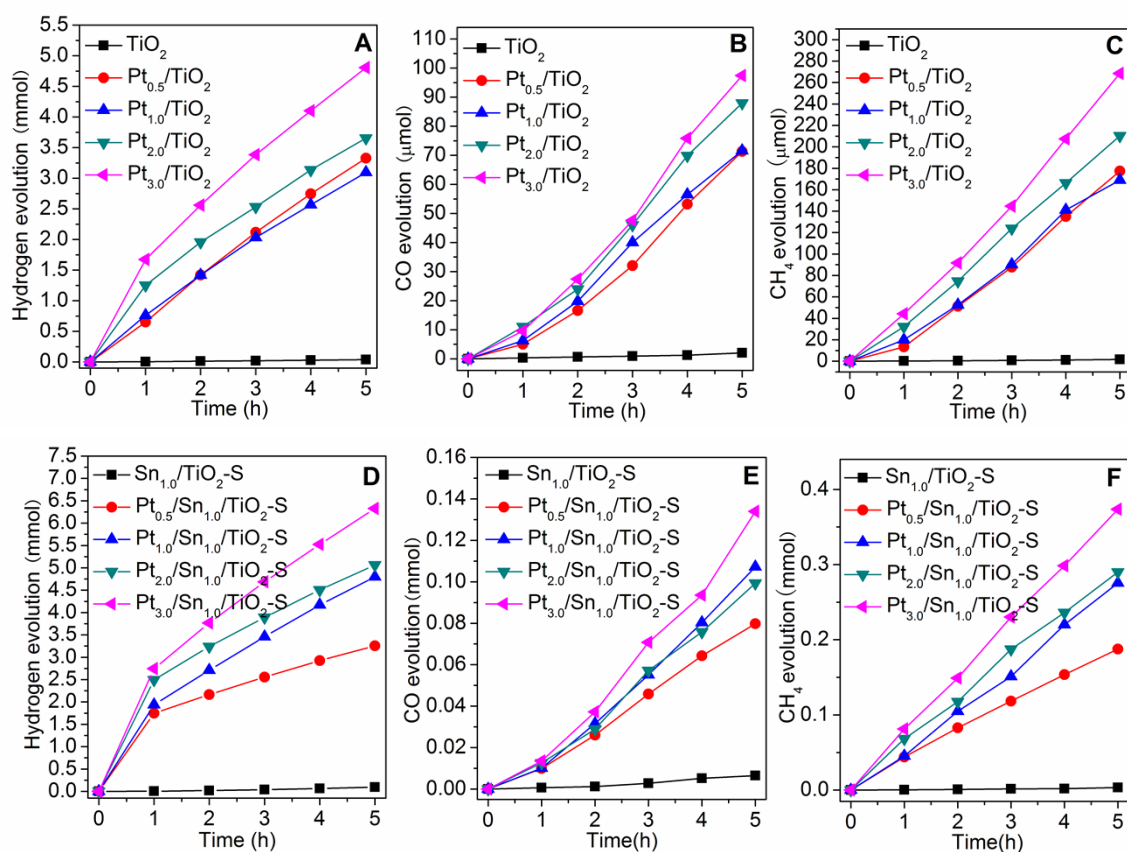


Fig. S4 The generation of H₂ (A and D), CO (B and E), and CH₄ (C and F) from the ethanol solution over the Pt_x/TiO₂ and Pt_x/Sn_{1.0}/TiO₂-S with different Pt contents, respectively.

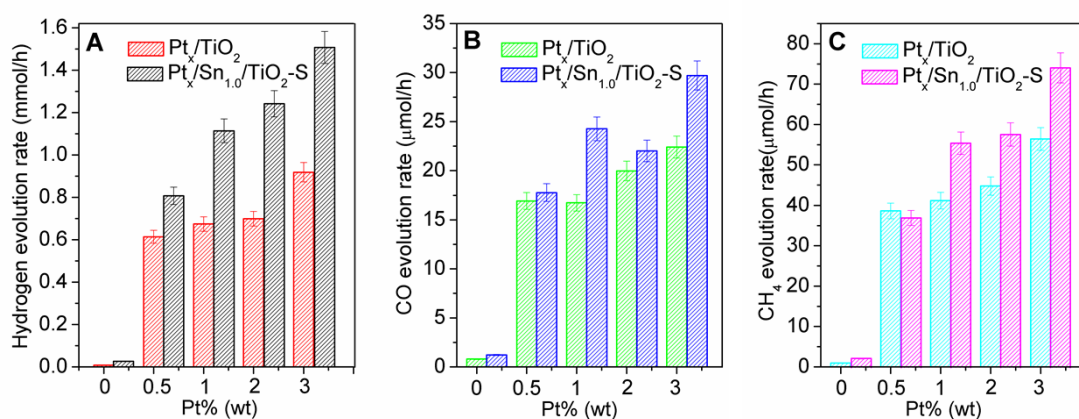


Fig. S5. The production rates of H₂ (A), CO (B), and CH₄ (C) from ethanol solution over the Pt_x/TiO₂ and Pt_x/Sn_{1.0}/TiO₂-S with different Pt contents

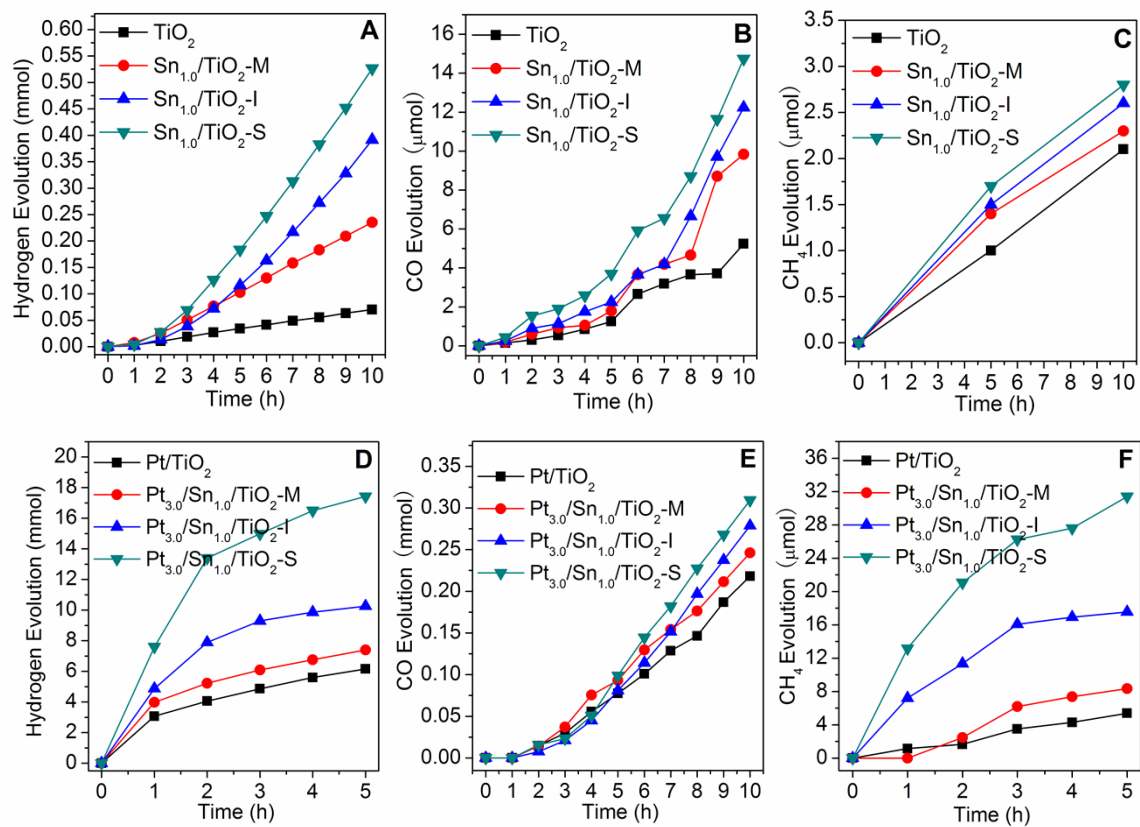


Fig. S6 The generation of H₂ (A and D), CO (B and E), and CH₄ (C and F) from the methanol solution over the Sn_{1.0}/TiO₂ and Pt_{3.0}/Sn_{1.0}/TiO₂ prepared by different methods, respectively

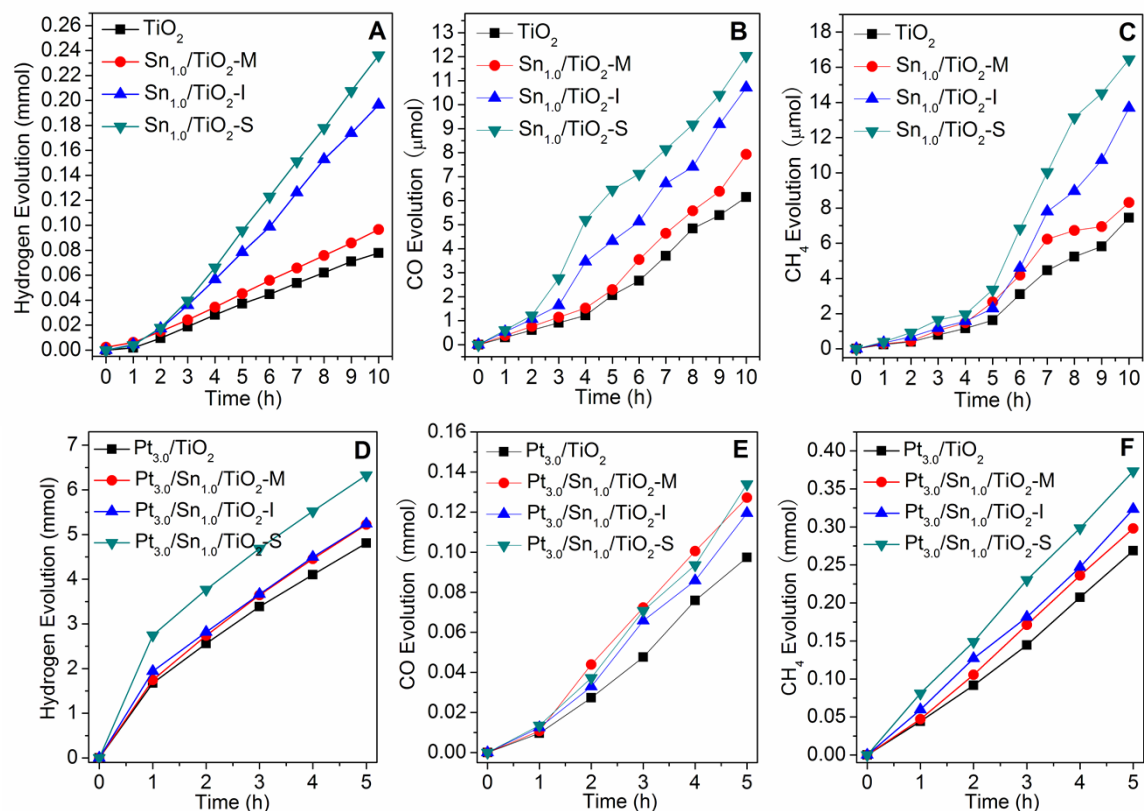
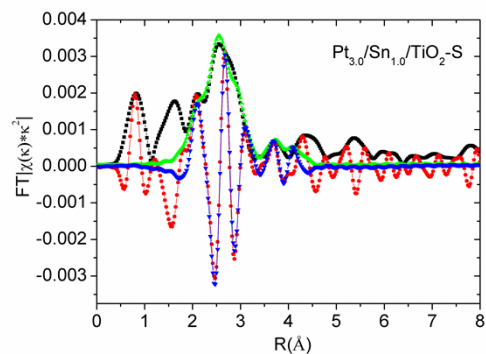
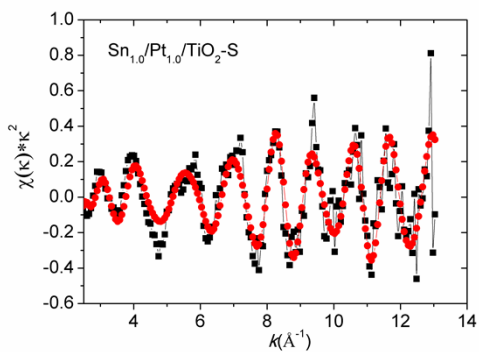
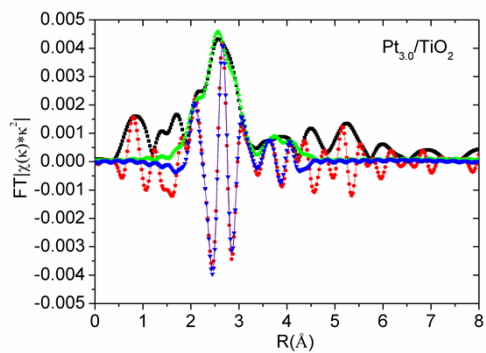
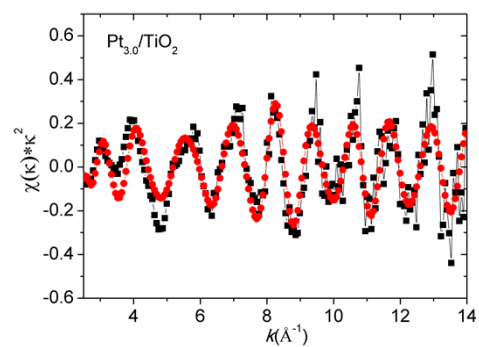
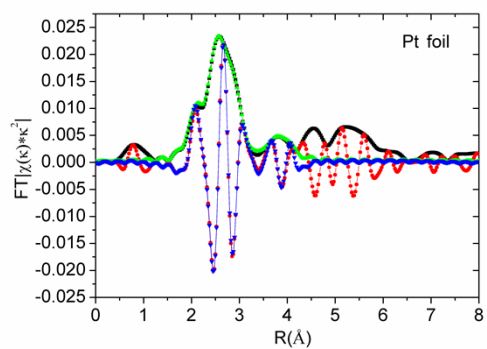
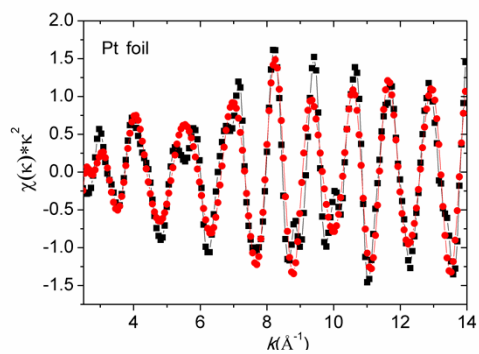
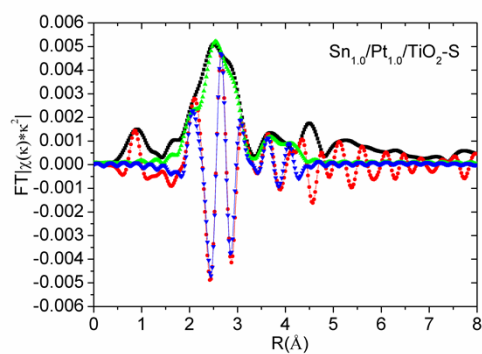
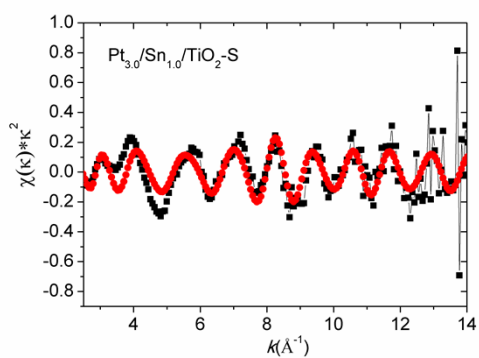
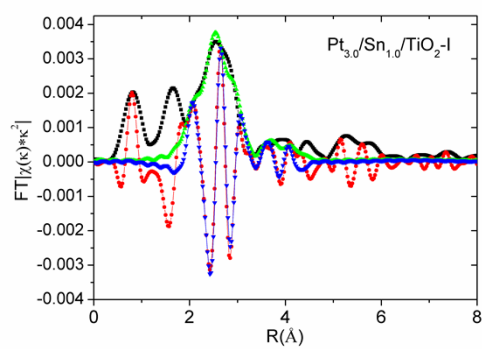
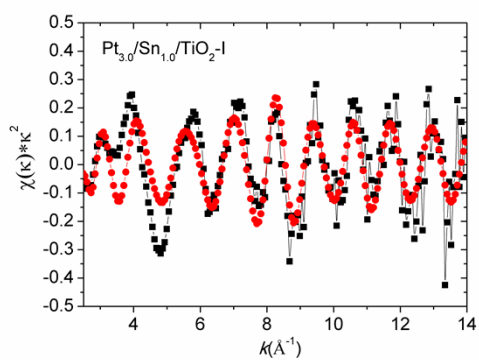
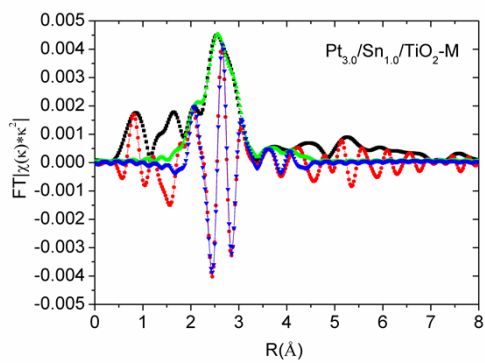
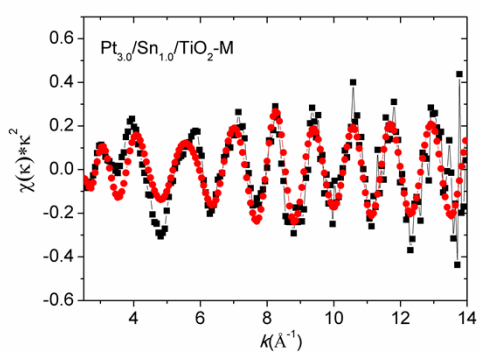


Fig. S7 The generation of H₂ (A and D), CO (B and E), and CH₄ (C and F) from the ethanol solution over the Sn_{1.0}/TiO₂ and Pt_{3.0}/Sn_{1.0}/TiO₂ prepared by different methods, respectively





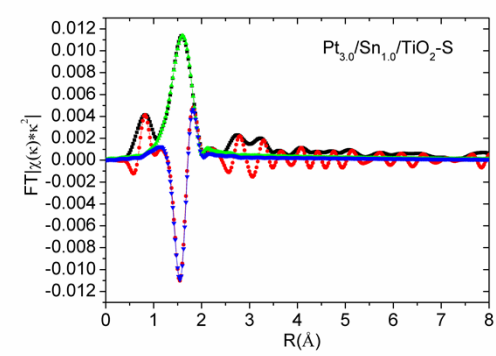
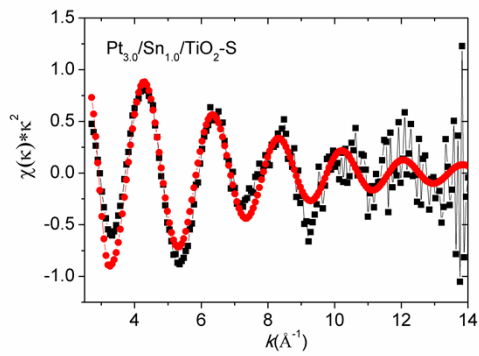
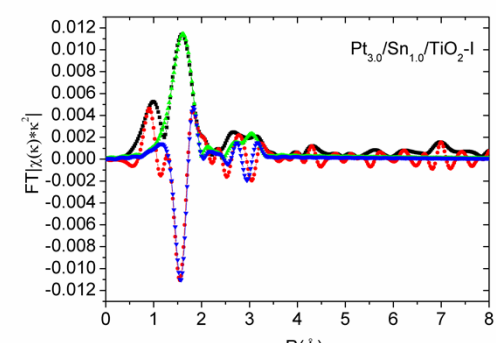
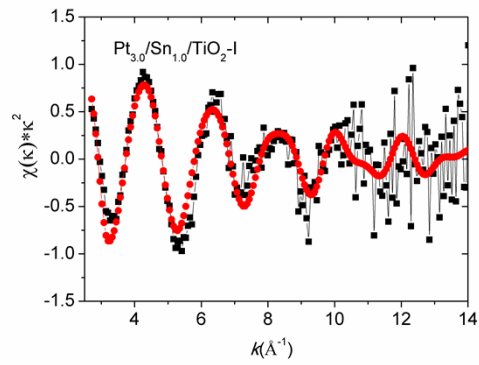
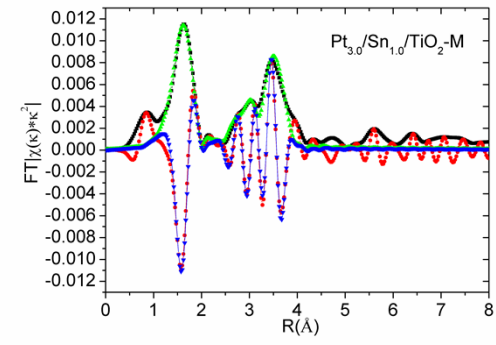
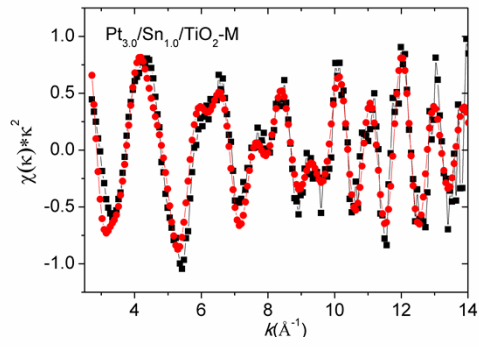
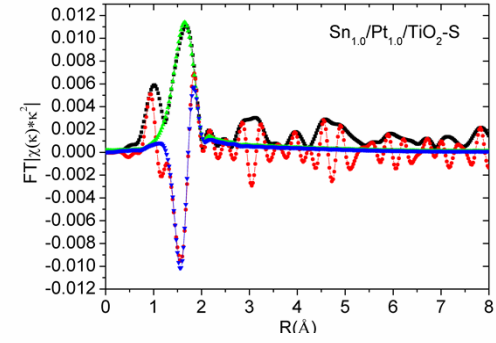
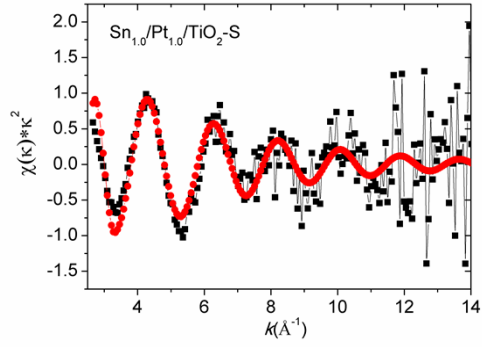
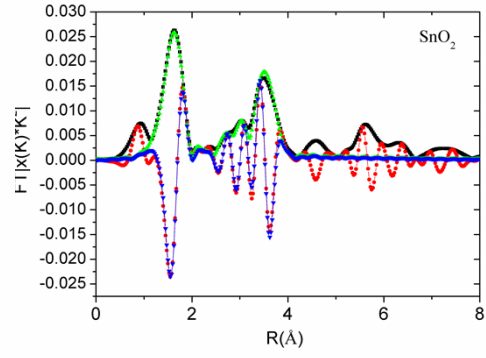
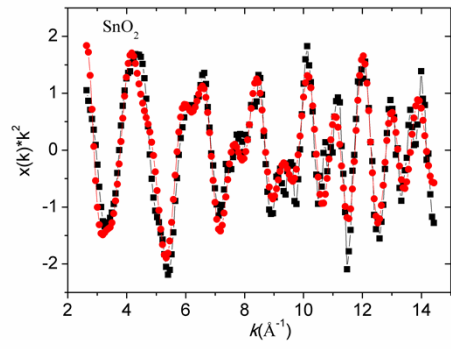


Fig. S8 Pt L_{III}-edge and Sn K-edge EXAFS fit results.

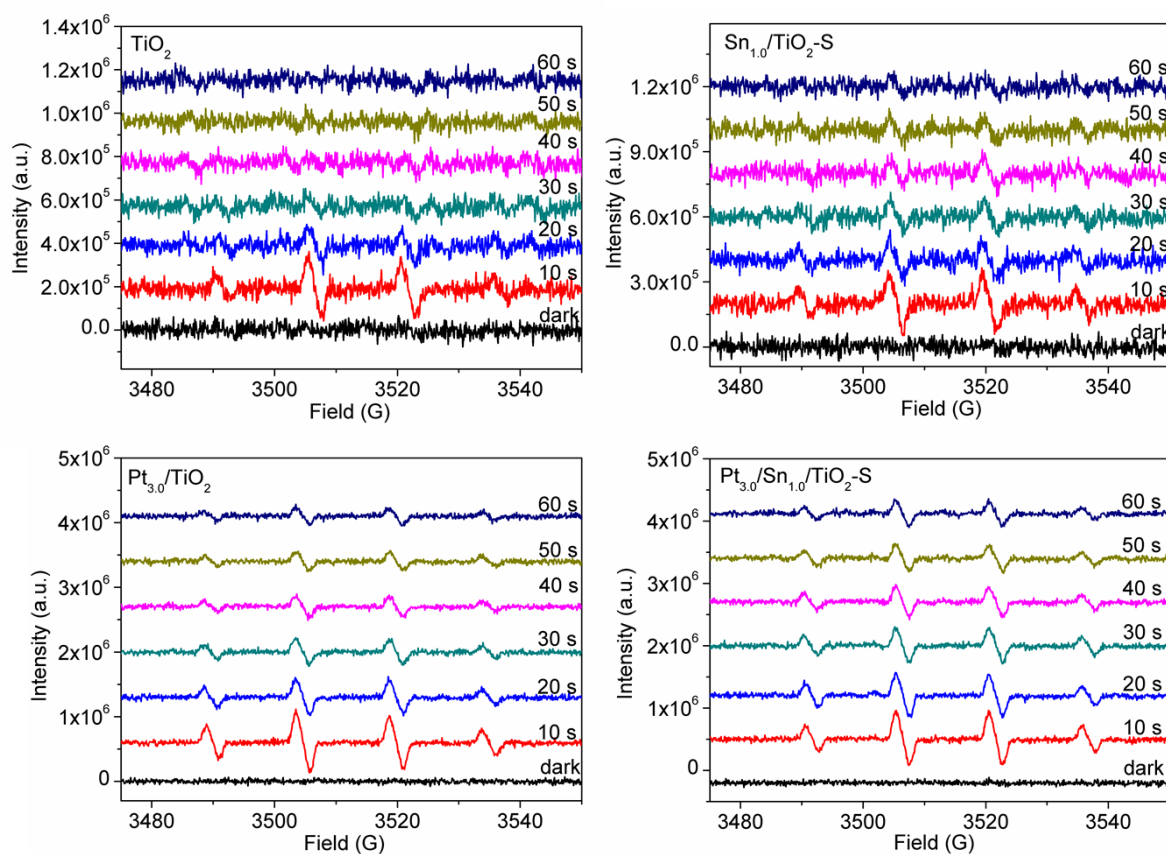


Fig. S9 The changes of EPR signal of the DMPO-•OH spin adducts in the suspension of TiO_2 , $\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$, $\text{Pt}_{3.0}/\text{TiO}_2$, and $\text{Pt}_{3.0}/\text{Sn}_{1.0}/\text{TiO}_2\text{-S}$ with 365 nm light irradiation time, respectively.

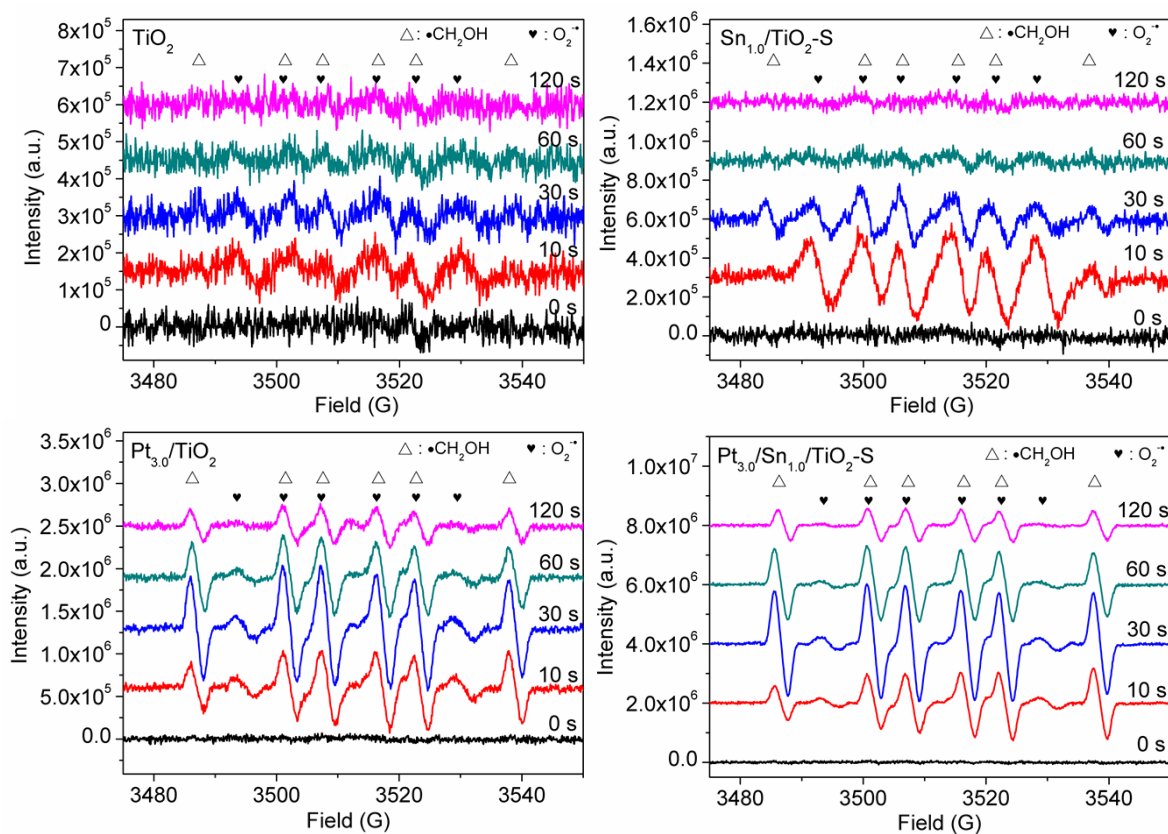


Fig. S10 EPR spectra of TiO₂, Sn_{1.0}/TiO₂-S, Pt_{3.0}/TiO₂, and Pt_{3.0}/Sn_{1.0}/TiO₂-S dispersed in DMPO/methanol with 365 nm light irradiation.