ESI (Electronic Supplementary Information)

Green Synthesis of Highly Fluorescent Au(I)@Ag₂/Ag₃-Thiolate Core-Shell Particles for Selective Detection of Cysteine and Pb(II)

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Figure S1: Elemental mapping of a single particle for the element silver, gold and sulfur in AuAgF solution.



Figure S2: (A) Fluorescence spectral profile and (B) bar diagram displaying the effect of interfering amino acids on 2.5 mL CyAuAgFL solution. Condition : [Amino acid] = 20×10^{-4} M, $\lambda_{ex} = 400$ nm.



Figure S3(a): MALDI mass spectrum of the drop-casted AuAgFL solution.



Figure S3(b): Liquid chromatography mass spectrum in ES+ mode of the AuAgFL solution.



Figure S4: Fluorescence decay profile of AuAgFL and aqueous suspension of WM.

Condition: $[Cys] = 30x10^{-6} M.$



Figure S5: (A) Degree of fluorescence quenching of AuAgF solution with [*l*-Cys] and [*d*-Cys].

(B) CD spectra of *l*-Cys, *d*-Cys, GSH, AuAgFL, (*l*-Cys + AuAgFL), (*d*-Cys + AuAgFL).

Condition : AuAgF = 2.5 mL, $[Cys/GSH] = 20x10^{-4} M$.



Figure S6: Elemental mapping for the element silver, sulfur, gold and lead from WM precipitate.



Figure S7: TEM image (A1, A2) from same copper grid and size distribution histogram (B) of AuAgFL.



Figure S8: The particle size distribution of (A) AuAgFL, (B) AuAgFL after heating at 70 $^{\circ}$ C, (C) CyAuAgFL, (D) aqueous suspension of WM and (E) aqueous suspension of WM + Na₂-EDTA measured by a DLS particle size analyzer.



Figure S9: FTIR spectra of AuAgFL, CyAuAgFL, WM, pure GSH and pure Cys.



Figure S10: Absorption spectral profile of (A) AuAgFL, CyAuAgFL, aqueous suspension of WM, aqueous suspension of WM + Na₂-EDTA, (B) aqueous GSH, aqueous GSH + aqueous solution of PbNO₃, (C) aqueous Cys, aqueous Cys + aqueous solution of PbNO₃, (D) aqueous solution of PbNO₃ + Na₂-EDTA.



Figure S11: Fluorescence spectral profile of AuAgFL at different excitation wavelength.



Figure S12: Excitation spectral profile of AuAgFL at the emission wavelength of 564 nm.