

Insights into the Interactions of CO₂ with Amines: a DFT Benchmark Study

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Supporting Information

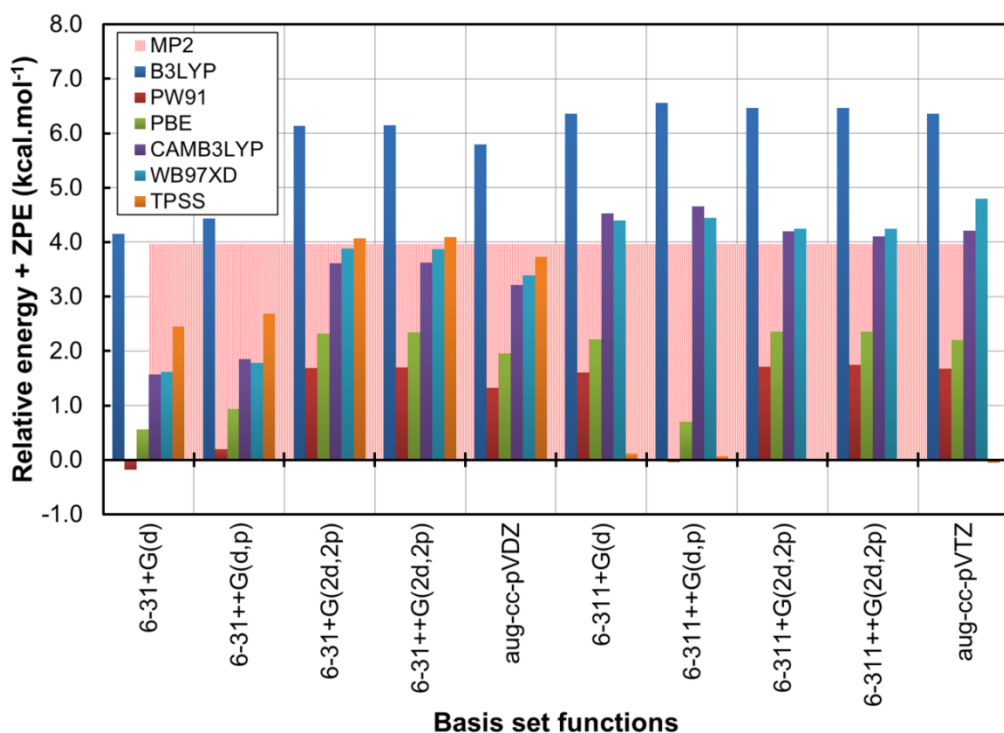


Figure S1. Zero-point corrected interaction energy between the 1,2-diaminoethane and CO₂ in kcal mol⁻¹ as function of the basis set for various DFT functionals. All calculations are compared with the MP2/aug-cc-pVTZ calculation equals to 3.9 kcal mol⁻¹, represented by the bottom continuous bar.

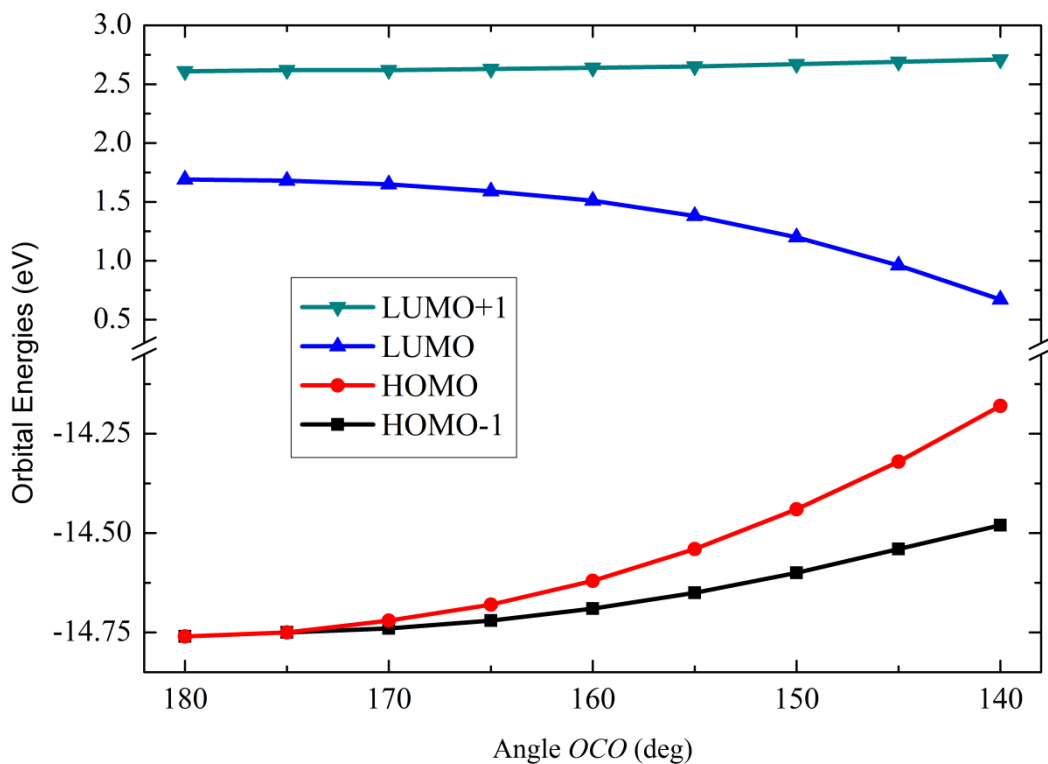
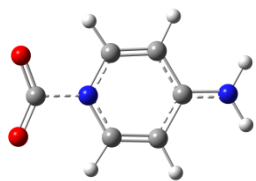


Figure S2. MP2(full)/6-311++G(2d,2p) orbital energies for CO₂ as a function of the COC angle.

Table S1. CAM-B3LYP/6-311++G(2d,2p) optimized geometries, selected geometrical parameters, and interaction energy for the zwitterion formed by a set of different amines and CO₂. Only the interaction between CO₂ and the most basic nitrogen of each amine is provided.

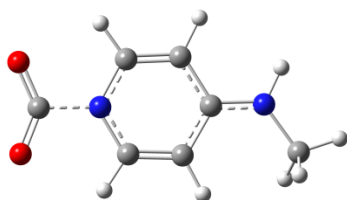
	OH---OCO distance	1.760 Å
	N-C distance	1.605 Å
	OCO angle	136.28°
	Int. Energy + BSSE	-2.50 kcal mol ⁻¹
	N-C distance	1.625 Å
	OCO angle	138.06°
	Int. Energy + BSSE	-0.01 kcal mol ⁻¹
	N-C distance	1.606 Å
	OCO angle	137.07°
	Int. Energy + BSSE	-2.03 kcal mol ⁻¹
	N-C distance	1.655 Å
	OCO angle	136.80°
	Int. Energy + BSSE	-0.53 kcal mol ⁻¹
	N-C distance	1.636 Å
	OCO angle	135.23°
	Int. Energy + BSSE	1.54 kcal mol ⁻¹
	N-C distance	1.637 Å
	OCO angle	137.65°
	Int. Energy + BSSE	0.64 kcal mol ⁻¹



N-C distance 1.544 Å

OCO angle 134.19°

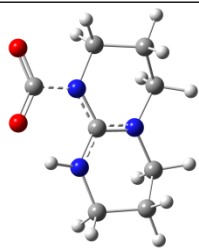
Int. Energy + BSSE -4.67 kcal mol⁻¹



N-C distance 1.540 Å

OCO angle 133.95°

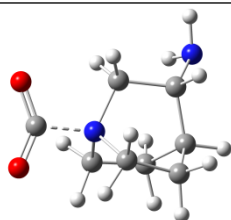
Int. Energy + BSSE -5.54 kcal mol⁻¹



N-C distance 1.470 Å

OCO angle 127.94°

Int. Energy + BSSE 1.26 kcal mol⁻¹



N-C distance 1.629 Å

OCO angle 136.45°

Int. Energy + BSSE -2.40 kcal mol⁻¹

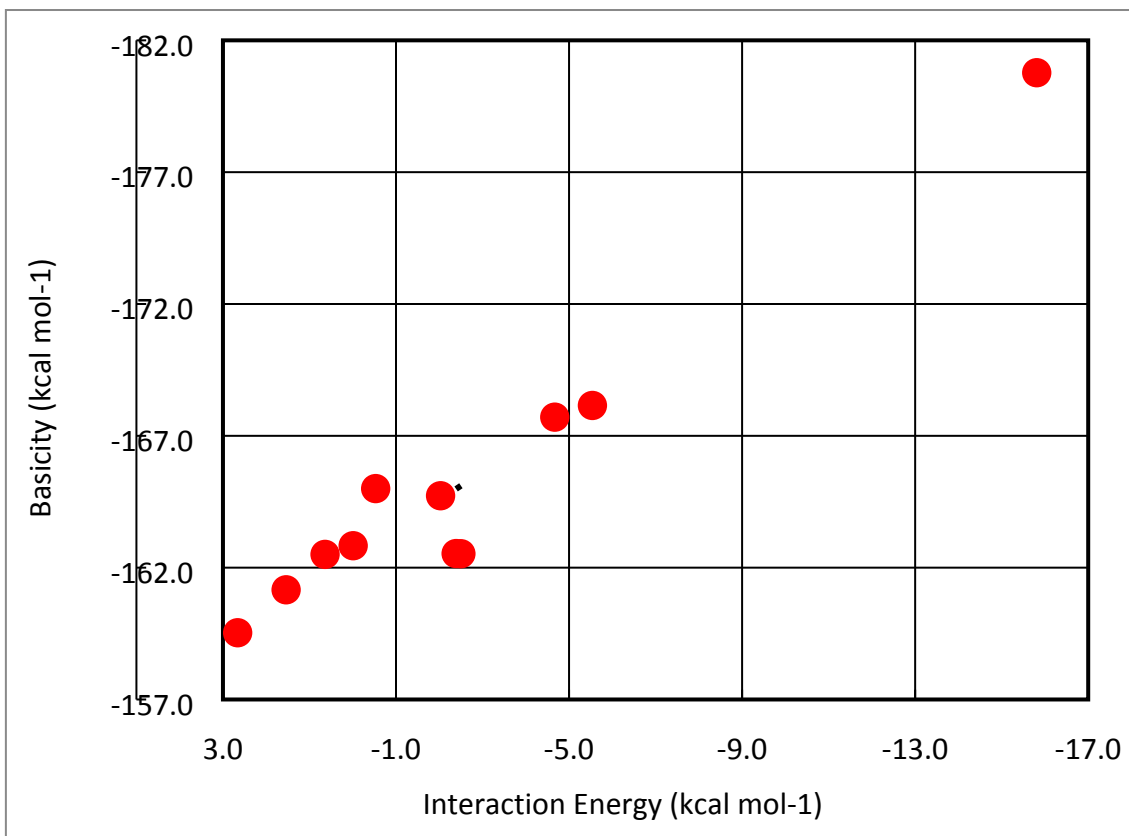


Figure S3. Correlation between the interaction energy of CO₂-amines and basicity of the amines.