

ARTICLE

Decomposition of Nitrous Oxide on Fe-doped Boron Nitride Nanotubes: The Ligand Effect

Cite this: DOI: 10.1039/x0xx00000x

Natcha Injan,^a Jakkapan Sirijaraensre^b and Jumras Limtrakul^{*bc}

Table S1. Electronic energies with zero-point energy (ZPE) and computed $\langle S^2 \rangle$ values of various spin states of $\text{Fe}_{(B)}\text{-BNNT}$ and $\text{Fe}_{(N)}\text{-BNNT}$.

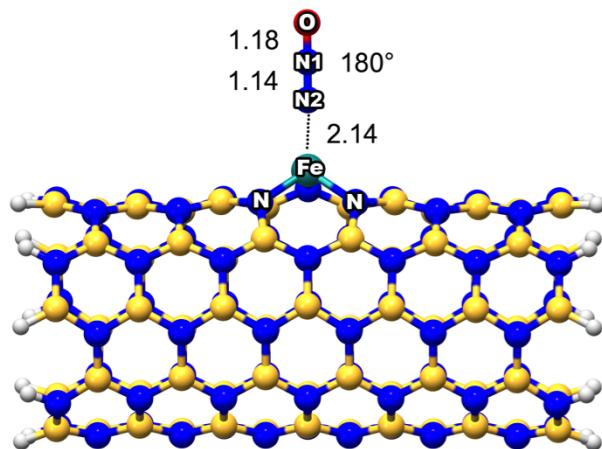
System	Spin multiplicity	Relative energy (kcal mol ⁻¹)	$\langle S^2 \rangle$	$\langle S^2 \rangle_{\text{cont}}^{\text{a}}$
$\text{Fe}_{(B)}\text{-BNNT}$	doublet	52.1	0.7890	5.20
	quartet	11.2	3.7523	0.06
	sextet	0.00	8.7501	0.00
$\text{Fe}_{(N)}\text{-BNNT}$	doublet	0.00	0.7861	4.81
	quartet	10.1	3.7591	0.24
	sextet	24.0	8.7503	0.00

^a The percent spin contamination $\langle S^2 \rangle_{\text{cont}} = [\langle S^2 \rangle - S(S+1)]100/[S(S+1)]$

Table S2. Partial natural charges of the adsorption complex (q_{Ads}) and charge-transfer of all species in the adsorption complexes (Δq_{Ads}) and transition states (Δq_{TS}).^a

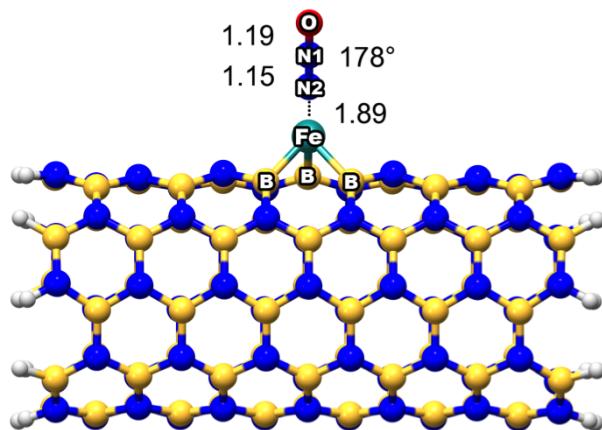
System	q_{Ads}			Δq_{Ads}			Δq_{TS}		
	tube	Fe	N_2O	tube	Fe	N_2O	tube	Fe	N_2O
sextet $\text{Fe}_{(B)}-\text{BNNT}:\text{N}_2\text{O}$	-1.333	1.241	0.091	-0.037	-0.055	0.091	0.548	-0.069	-0.479
doublet $\text{Fe}_{(N)}-\text{BNNT}:\text{N}_2\text{O}$	0.357	-0.039	-0.318	0.442	-0.083	-0.360	0.026	-0.007	-0.019

^a calculated from equations $\Delta q_{\text{Ads}} = q_{\text{Ads}} - q_{\text{iso}}$ and $\Delta q_{\text{TS}} = q_{\text{TS}} - q_{\text{Ads}}$ where q_{Ads} , q_{TS} and q_{iso} are charges of all species in the adsorption complexes, the transition states and the isolated N_2O and nanotubes, respectively.



Adsorption energy = $-15.0 \text{ kcal mol}^{-1}$

(a) sextet $\text{Fe}_{(B)}$ -BNNT



Adsorption energy = $-26.4 \text{ kcal mol}^{-1}$

(b) doublet $\text{Fe}_{(N)}$ -BNNT

Fig. S1 Geometrical structures and adsorption energy with ZPE correction of N-bound complexes over (a) sextet $\text{Fe}_{(B)}$ -BNNT and (b) doublet $\text{Fe}_{(N)}$ -BNNT. Distances and angles are given in Å and degrees, respectively.

Table S3. Relative energies with zero-point energy (ZPE) of N₂O decomposition on Fe_(B)-BNNT and Fe_(N)-BNNT.

	Relative energy (kcal mol ⁻¹)		
	Doublet state	Quartet state	Sextet state
Fe _(B) -BNNT + N ₂ O	52.1	11.2	0.0
Ads	5.2	2.1	-12.9
TS	15.5	9.6	12.3
Prod	-43.3	-39.2	-14.4
FeO _(B) -BNNT + N ₂	-40.9	-36.4	-15.4
Fe _(N) -BNNT + N ₂ O	0.0	10.1	24.0
Ads	-25.0	-3.1	1.9
TS	-25.2	-19.2	4.8
Prod	-51.3	-67.3	-32.8
FeO _(N) -BNNT + N ₂	-48.7	-64.7	-32.5