

Supporting Information

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Dielectric Fitting Data

General: The obtained THz dielectric spectra of a series of binary liquid mixtures were fitted to either a two-component Debye model (D2), a constrained three-component Debye model (D3), or a Debye model containing a single Lorentz vibrational term (D1-V1).

$$\tilde{\varepsilon}_{Debye} = \varepsilon_{\infty} + \sum_{j=1}^2 \frac{\varepsilon_j}{1 + i\omega\tau_j} \quad (D2)$$

$$\tilde{\varepsilon}_{Debye} = \varepsilon_{\infty} + \sum_{j=1}^3 \frac{\varepsilon_j}{1 + i\omega\tau_j} \quad (D3)$$

$$\tilde{\varepsilon}_{Debye} = \varepsilon_{\infty} + \frac{\varepsilon_1}{1 + i\omega\tau_1} + \frac{A_v}{\omega_v^2 - \omega^2 + i\omega\gamma_v} \quad (D1 - V1)$$

The values ε_j and τ_j represent the relative relaxation strength and time of the j^{th} relaxation component, A_v is the amplitude of the vibrational mode, ω_v is the characteristic frequency of the vibrational mode and γ_v is the damping factor. For the D3 model, the values of τ_1 and τ_2 are fixed to that of the pure liquid components obtained from the D2 model.

Static dielectric constants (ε_s) of the mixtures are obtained from previously published results and fixed as

$$\varepsilon_s = \varepsilon_{\infty} + \sum_j \varepsilon_j$$

The measured THz dielectric spectra were fitted to these models by minimising both weighted least-squares, S_w

$$S_w = \sum_{\omega} \left[\frac{R(\tilde{\varepsilon}_{Debye}) - R(\tilde{\varepsilon}_{THz})}{R(\tilde{\varepsilon}_{THz})} \right]^2 + \sum_{\omega} \left[\frac{I(\tilde{\varepsilon}_{Debye}) - I(\tilde{\varepsilon}_{THz})}{I(\tilde{\varepsilon}_{THz})} \right]^2$$

where $\tilde{\varepsilon}_{THz}$ is the measured THz dielectric spectrum. Normalised residues were calculated for every fit as

$$\delta_w^2 = \frac{S^{min}}{2m}$$

Diol-water and diol-methanol mixtures: The obtained THz dielectric spectra of these mixtures were fitted to both the D2 and the constrained D3 models.

Dielectric relaxation parameters of diol-methanol mixtures evaluated using the D2 model. Values of the static dielectric constant (ϵ_s) were obtained from Ref. 1.

<i>mol%</i>	ϵ_s	ϵ_1	τ_1 (ps)	ϵ_2	τ_2 (fs)	$\delta_w^2 (10^{-4})$
Ethylene glycol-methanol						
0	32.88	29.3	13.5	1.47	158	2.68
10	33.55	30.1	12.0	1.40	142	4.67
20	34.59	31.1	14.1	1.44	126	6.19
30	35.47	31.9	16.5	1.49	111	3.70
40	36.34	32.7	17.9	1.47	125	4.34
50	37.14	32.7	16.5	1.58	115	4.74
60	37.84	33.4	16.2	1.56	113	4.95
70	38.48	34.8	19.7	1.47	108	4.00
80	39.02	35.2	20.3	1.50	111	3.04
90	39.78	35.7	24.9	1.60	112	3.08
100	40.68	36.9	33.6	1.55	104	2.00
1,2-propanediol-methanol						
0	32.88	29.4	14.5	1.48	180	5.88
10	31.97	28.5	16.6	1.39	152	7.16
20	31.53	28.2	20.8	1.35	140	8.60
30	31.07	27.8	21.9	1.20	117	3.87
50	30.15	26.9	24.2	1.12	104	1.63
60	29.74	26.4	24.9	1.13	103	1.92
70	29.38	26.1	34.9	1.12	104	2.07
100	28.30	25.1	40.8	1.01	96.4	2.32

Dielectric relaxation parameters of diol-methanol mixtures evaluated using the D3 model. Values of the static dielectric constant (ϵ_s) were obtained from Ref. 1.

<i>mol%</i>	ϵ_s	ϵ_1	ϵ_2	ϵ_3	τ_3 (fs)	$\delta_w^2 (10^{-4})$
Ethylene glycol-methanol						
0	32.88	29.3	0.06	1.47	158	2.68
10	33.55	23.7	6.35	1.40	142	4.60
20	34.59	19.0	12.1	1.43	126	6.11
30	35.47	14.6	17.4	1.49	111	3.68
40	36.34	12.7	20.1	1.47	125	4.31
50	37.14	14.8	17.9	1.58	114	4.73
60	37.84	15.8	17.5	1.56	113	4.94
70	38.48	10.8	24.1	1.46	108	3.98
80	39.02	10.2	25.0	1.50	110	3.05
90	39.78	5.53	30.2	1.60	112	3.08
100	40.68	0.00	36.9	1.55	104	2.00
1,2-propanediol-methanol						
0	32.88	29.9	0.00	1.49	168	5.99
10	31.97	25.2	3.31	1.39	152	7.14
20	31.53	15.8	12.4	1.34	140	8.56
30	31.07	13.7	14.1	1.20	117	3.86
50	30.15	10.1	16.8	1.12	104	1.62
60	29.74	9.03	17.4	1.13	103	1.91

70	29.38	3.12	22.9	1.12	104	2.07
100	28.30	0.00	25.1	1.03	99	2.21

Dielectric relaxation parameters of diol-water mixtures evaluated using the D2 model. Values of the static dielectric constant (ϵ_s) were obtained from Ref. 2 and 3.

<i>mol%</i>	ϵ_s	ϵ_1	τ_1 (ps)	ϵ_2	τ_2 (fs)	$\delta_w^2 (10^{-4})$
Ethylene glycol-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	70.62	65.7	11.3	2.01	122	4.41
20	64.97	60.2	14.8	1.88	129	3.41
30	60.46	56.1	21.0	1.83	112	6.54
40	56.60	52.3	25.3	1.81	110	8.73
50	53.18	49.0	23.9	1.73	100	2.33
60	50.09	46.0	27.7	1.69	96	2.80
70	47.29	43.3	31.0	1.60	106	4.47
80	44.80	40.9	31.2	1.56	100	4.59
90	42.62	38.8	38.6	1.59	101	13.7
100	40.82	36.9	35.2	1.58	101	4.32
1,2-propanediol-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	64.81	59.9	12.5	1.93	110	7.80
20	56.44	52.0	18.5	1.65	105	4.92
30	50.42	46.4	25.7	1.53	94	5.73
40	45.74	42.0	21.7	1.33	78	2.42
50	41.91	38.1	30.6	1.32	93	1.92
60	38.35	34.8	23.2	1.16	90	6.20
70	35.36	32.0	24.9	1.08	85	1.03
80	32.67	29.4	24.4	1.06	80	1.01
90	30.34	27.1	38.1	1.01	101	0.82
100	28.36	25.1	29.7	0.98	89	1.86
1,3-propanediol-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	66.21	61.5	14.2	1.78	116	5.36
20	58.49	54.2	19.8	1.48	105	7.78
30	53.07	49.1	33.1	1.35	115	13.7
40	48.97	45.1	40.1	1.26	105	9.07
50	45.67	42.1	38.7	1.10	95	5.44
60	42.84	39.4	41.2	1.03	96	8.32
70	40.28	37.1	37.0	0.96	90	13.9
80	37.96	34.6	44.7	0.95	98	11.4
90	35.91	32.3	59.9	0.94	103	8.29
100	34.30	31.1	68.3	0.90	105	8.72
1,2-butanediol-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	60.57	55.7	15.1	1.85	122	13.5
20	50.40	45.8	20.4	1.65	119	13.0
30	43.65	39.9	23.3	1.13	109	16.3
40	38.72	35.2	27.9	1.06	102	4.71
50	34.88	31.5	39.5	0.98	109	5.63
60	31.73	28.5	37.7	0.89	108	1.78

70	28.99	25.8	44.1	0.85	98	2.32
80	26.53	23/4	42.5	0.78	105	4.15
90	24.33	21.3	68.9	0.79	118	3.43
100	22.57	19.5	61.4	0.76	112	3.83
1,4-butanediol-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	63.71	59.3	15.8	1.59	107	3.87
20	55.26	50.9	25.4	1.48	155	8.79
30	49.61	45.7	34.4	1.19	118	5.76
40	45.48	41.8	38.7	1.16	77	3.98
50	42.25	38.7	53.8	0.96	113	9.88
60	39.59	36.2	59.2	0.92	95	4.22
70	37.27	34.1	80.3	0.80	111	3.88
80	35.18	32.1	59.1	0.76	107	6.43
90	33.32	30.2	68.7	0.73	111	6.99
100	31.82	28.8	96.7	0.78	68	9.23

Dielectric relaxation parameters of diol-water mixtures evaluated using the constrained D3 model.
Values of the static dielectric constant (ϵ_s) were obtained from Ref. 2 and 3.

<i>mol%</i>	ϵ_s	ϵ_1	ϵ_2	ϵ_3	τ_3 (fs)	$\delta_w^2 (10^{-4})$
Ethylene glycol-water						
0	80.40	75.3	0.00	2.02	102	5.83
10	70.62	38.1	27.7	1.99	119	4.34
20	64.97	22.5	37.8	1.87	127	3.64
30	60.46	10.1	46.0	1.82	111	6.35
40	56.60	5.29	47.0	1.81	109	8.62
50	53.18	6.05	42.9	1.73	99	2.38
60	50.09	3.14	42.9	1.69	96	2.81
70	47.29	1.35	42.0	1.60	106	4.45
80	44.80	1.17	39.7	1.56	100	4.58
90	42.62	0.00	38.8	1.58	94	17.0
100	40.82	0.00	37.0	1.58	101	6.34
1,2-propanediol-water						
0	80.40	75.3	0.00	2.02	102	5.83
10	64.81	30.1	29.8	1.92	108	7.41
20	56.44	13.3	38.7	1.65	104	4.83
30	50.42	5.10	41.3	1.53	94	5.64
40	45.74	7.44	34.6	1.33	78	2.41
50	41.91	1.88	36.3	1.32	93	1.93
60	38.35	5.17	29.6	1.16	90	0.62
70	35.36	3.84	28.1	1.08	85	1.03
80	32.67	3.78	25.6	1.06	80	1.01
90	30.34	0.00	27.1	1.00	99	0.94
100	28.36	0.00	25.1	0.98	90	2.12
1,3-propanediol-water						
0	80.40	75.3	0.00	2.02	102	5.83
10	66.21	29.3	32.3	1.77	114	4.90
20	58.49	16.6	37.6	1.47	103	7.78
30	53.07	6.56	42.5	1.35	114	13.4
40	48.97	3.96	41.1	1.26	105	9.01

50	45.67	3.99	38.1	1.10	94	5.45
60	42.84	3.23	36.2	1.03	96	8.29
70	40.28	3.90	33.2	0.96	90	13.9
80	37.96	2.27	32.3	0.95	98	11.4
90	35.91	0.57	32.0	0.94	103	8.29
100	34.30	0.00	31.1	0.90	105	8.71
1,2-butanediol-water						
0	80.40	75.3	0.00	2.02	102	5.83
10	60.57	24.2	31.5	1.83	119	13.3
20	50.40	13.0	32.8	1.64	117	12.5
30	43.65	9.19	30.8	1.12	108	16.0
40	38.72	5.92	29.2	1.06	102	4.70
50	34.88	2.44	29.0	0.98	109	5.64
60	31.73	2.50	26.0	0.89	108	1.78
70	28.99	1.42	24.4	0.85	98	2.32
80	26.53	1.46	22.0	0.78	105	4.14
90	24.33	0.00	21.3	0.78	115	3.72
100	22.57	0.00	19.5	0.76	112	3.64
1,4-butanediol-water						
0	80.40	75.3	0.00	2.02	102	5.83
10	63.71	25.1	34.3	1.59	100	3.32
20	55.26	12.0	38.9	1.45	148	9.74
30	49.61	6.98	38.8	1.18	113	4.38
40	45.48	5.20	36.7	1.17	74	3.55
50	42.25	2.61	36.1	0.96	111	8.72
60	39.59	1.95	34.2	0.92	94	4.22
70	37.27	0.61	33.4	0.80	110	3.98
80	35.18	1.74	30.4	0.75	105	6.41
90	33.32	1.04	29.2	0.73	109	6.49
100	31.82	0.00	28.7	0.78	68	8.74

Aprotic liquid-water mixtures: The measured THz dielectric spectra were fitted to the D2, D1-V1 and constrained D3 models.

Dielectric relaxation parameters of aprotic liquid-water mixtures evaluated using the D2 model.

Values of the static dielectric constant (ϵ_s) were obtained from Ref. 4, 5 and 6.

$mol\%$	ε_s	ε_1	τ_1 (ps)	ε_2	τ_2 (fs)	$\delta_w^2(10^{-4})$
Acetonitrile-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	72.35	66.9	6.64	2.47	112	5.61
20	63.90	58.4	6.74	2.85	132	4.88
30	57.04	52.0	5.83	2.73	137	9.03
40	51.97	47.1	4.95	2.71	151	8.72
50	48.11	43.6	4.94	2.76	144	9.47
60	44.93	40.7	4.33	2.66	146	10.4
70	42.27	38.3	3.61	2.55	135	19.1
80	40.22	36.3	3.15	2.44	150	21.0
90	38.46	35.3	2.63	1.89	108	15.3
100	36.74	33.5	3.03	2.41	141	14.7
Acetone-water						

0	80.40	75.3	7.45	2.02	102	4.29
10	61.89	57.4	10.4	1.81	113	5.80
20	51.36	47.2	10.6	1.79	119	6.04
30	42.87	39.0	9.19	1.67	148	3.52
40	37.65	34.0	8.45	1.65	152	5.22
50	30.85	27.4	5.96	1.58	166	3.08
60	27.75	24.4	4.83	1.51	189	2.93
70	25.57	22.4	4.80	1.47	185	3.57
80	23.64	20.4	3.85	1.53	254	3.51
90	22.00	19.2	3.26	1.24	216	2.27
100	20.80	18.1	3.01	1.15	222	3.19
Tetrahydrofuran-water						
0	80.40	75.3	7.45	2.02	102	4.29
10	57.76	53.1	12.0	1.72	115	7.96
20	41.20	37.2	10.1	1.34	115	6.78
30	30.57	26.8	10.0	1.23	137	7.97
40	23.55	20.1	8.26	1.07	152	6.44
50	18.75	15.5	5.94	0.95	159	5.54
60	15.32	12.6	5.17	0.71	149	1.94
70	12.72	9.75	3.31	0.77	158	3.01
80	10.60	7.75	3.41	0.81	236	1.92
90	8.82	6.31	2.23	0.54	204	2.60
100	7.58	5.22	2.26	0.47	231	6.18

Dielectric relaxation parameters of aprotic liquid-water mixtures evaluated using the D1-V1 model.

$mol\%$	ε_s	ε_1	τ_1 (ps)	$\frac{A_v}{4\pi^2}$ (THz ²)	$\frac{\gamma_v}{2\pi}$ (THz)	δ_w^2 (10 ⁻⁴)
Acetonitrile-water ($\omega_v = 2.79$ THz)						
0	80.40	75.3	7.42	4.48	2.70	15.2
10	72.35	66.9	6.54	15.4	6.12	8.10
20	63.90	58.4	6.70	19.0	7.25	9.42
30	57.04	52.1	5.78	18.3	7.43	11.0
40	51.97	47.1	4.93	18.6	8.17	9.63
50	48.11	43.6	4.38	17.4	6.57	17.9
60	44.93	40.8	4.29	18.0	7.78	11.6
70	42.27	38.4	3.58	16.9	7.34	21.8
80	40.22	36.3	3.13	16.6	8.09	23.1
90	38.46	35.4	2.60	11.5	6.02	17.2
100	36.74	33.6	2.97	15.7	7.34	14.1
Acetone-water ($\omega_v = 1.83$ THz)						
0	80.40	75.3	7.42	4.48	2.70	15.2
10	61.89	57.0	9.77	3.91	2.65	28.3
20	51.36	46.9	10.0	4.01	2.80	26.2
30	42.87	38.5	8.56	4.05	3.26	15.9
40	37.65	33.0	7.79	4.06	3.37	15.2
50	30.85	27.0	5.60	3.95	3.58	10.0
60	27.75	24.5	4.65	3.92	4.00	7.26
70	25.57	22.2	4.54	3.73	3.88	5.86
80	23.64	20.7	3.72	4.05	4.98	2.25
90	22.00	19.5	3.28	3.51	4.86	2.49

100	20.80	18.2	2.97	3.19	4.81	3.18
Tetrahydrofuran-water ($\omega_v = 1.78$ THz)						
0	80.40	75.3	7.42	4.48	2.70	15.2
10	57.76	53.2	11.5	3.59	2.62	31.2
20	41.20	37.2	9.61	2.78	2.57	32.1
30	30.57	26.8	9.59	2.78	2.98	25.9
40	23.55	20.2	7.80	2.45	3.14	21.1
50	18.75	15.6	5.60	2.16	3.19	16.5
60	15.32	12.6	4.89	1.55	2.93	8.58
70	12.72	9.83	3.15	1.69	3.06	8.58
80	10.60	7.94	3.01	1.68	3.96	3.22
90	8.82	6.43	2.10	1.10	3.45	1.50
100	7.58	5.31	2.15	1.02	3.96	3.48

Dielectric relaxation parameters of aprotic liquid-water mixtures evaluated using the constrained D3 model. Values of the static dielectric constant (ϵ_s) were obtained from Ref. 4, 5 and 6.

<i>mol%</i>	ϵ_s	ϵ_1	ϵ_2	ϵ_3	τ_3 (fs)	δ_w^2 (10^{-4})
Acetonitrile-water						
10	72.35	61.3	5.71	2.44	105	5.52
20	63.90	53.8	4.69	2.81	119	9.13
30	57.04	42.9	9.25	2.67	130	10.2
40	51.97	33.4	13.8	2.66	143	10.2
50	48.11	29.5	14.3	2.66	129	7.56
60	44.93	22.5	18.4	2.55	133	7.71
70	42.27	13.8	24.7	2.47	123	19.1
80	40.22	6.79	29.6	2.36	139	20.6
90	38.46	0	35.1	2.02	121	16.7
Acetone-water						
10	61.89	57.4	0.00	1.47	62	70.3
20	51.36	47.3	0.00	1.44	69	90.9
30	42.87	38.8	0.00	1.39	115	17.6
40	37.65	33.1	0.00	1.52	137	8.10
50	30.85	21.2	5.70	1.54	161	2.47
60	27.75	14.5	9.96	1.45	181	2.43
70	25.57	12.7	9.48	1.41	177	2.55
80	23.64	6.77	13.9	1.45	245	3.48
90	22.00	2.29	17.2	1.21	212	2.25
Tetrahydrofuran-water						
10	57.76	53.7	0.00	1.77	27	341
20	41.20	37.6	0.00	1.14	49	112
30	30.57	27.1	0.00	0.94	73	93.4
40	23.55	20.3	0.00	0.89	119	16.6
50	18.75	14.2	1.36	0.93	156	5.27
60	15.32	10.4	2.22	0.69	144	1.53
70	12.72	4.35	5.44	0.74	151	2.40
80	10.60	3.69	4.10	0.78	230	1.92
90	8.82	0.00	6.29	0.56	210	2.63