

Supporting Information

Phosphine and Phosphine Oxide Groups in Metal-Organic Frameworks detected by P K-edge XAS

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1. Infrared spectroscopy

Infrared spectra were recorded in the range of 4000-400 cm^{-1} on a PerkinElmer 2000 FTIR spectrometer.

1.1. $\text{PPh}_2\text{-bdc}$ and $\text{POPh}_2\text{-bdc}$

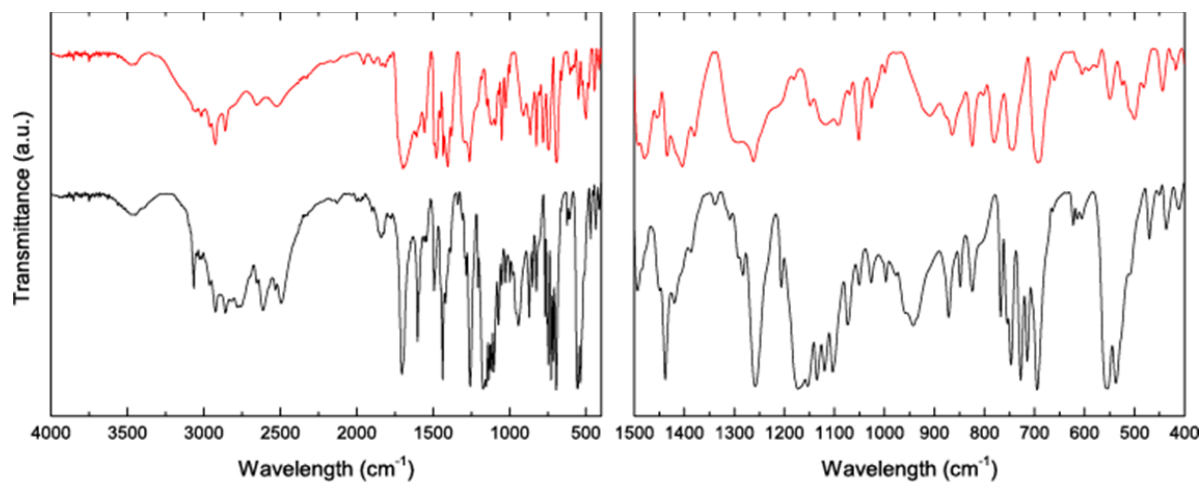


Figure S1. Infrared spectroscopy of $\text{PPh}_2\text{-bdc}$ (red) and $\text{POPh}_2\text{-bdc}$ (black). Full spectrum (left) and region of interest (right).

1.2. LSK-12 and LSK-15

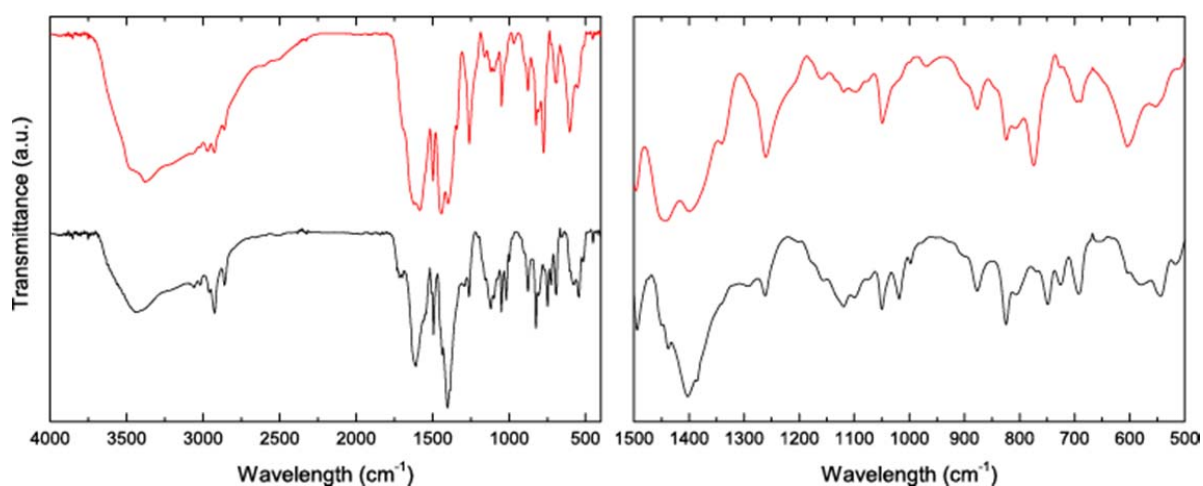


Figure S2. Infrared spectroscopy of LSK-15 (red) and LSK-12 (black). Full spectrum (left) and region of interest (right).

2. Summary of nitrogen physisorption experiments

Table S1. Nitrogen physisorption properties of LSK-12, LSK-15 and related materials.^{1,2}

Materials	S_{BET} (m^2/g)	V_{pore} (cm^3/g)
LSK-12	3020	2.04
LSK-15	2692	1.32
MIL-101(Cr)	4100	1.90
MIL-101(Al)-NH ₂	3099	1.53

3. Additional characterization of LSK-12

3.1. Powder X-ray diffraction pattern of LSK-12

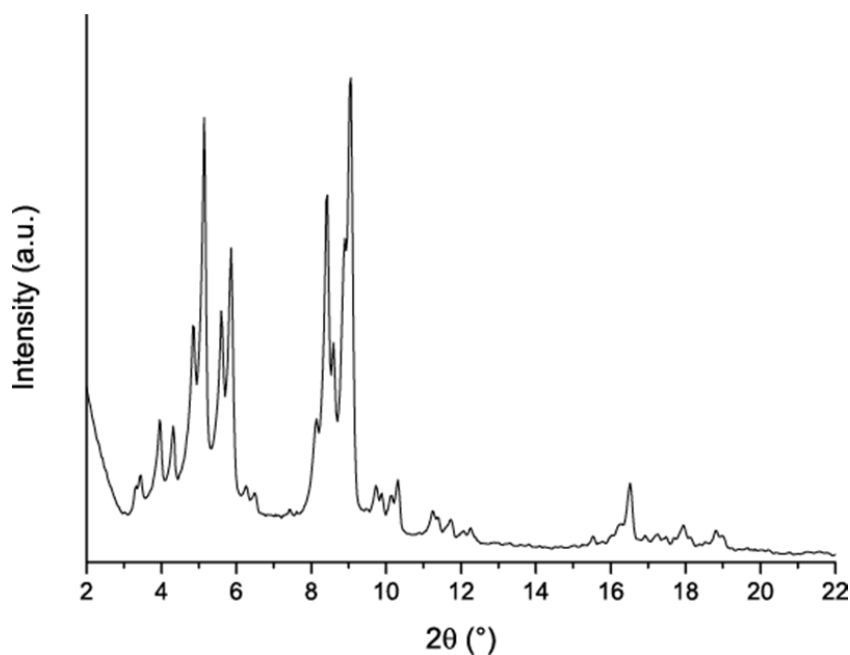


Figure S3. Powder X-ray diffraction pattern of LSK-12.

3.2. Liquid chromatography after digestion

LSK-12 (1 mg) was dissolved in a NaOH solution (400 μL 1 M) at 80 °C. Ethanol (400 μL) and phosphoric acid (60 μL , 85 % w/w) were added to the mixture, which was then filtered and analyzed in a Waters Acquity UPLC H-Class system equipped with a photodiodes array detector and a Waters BEH C18 (1.7 μm , 1.0 $\times$ 150 mm) column. The relative ratio bdc : POPh₂-bdc : PPh₂-bdc was equal to 20 : 1 : 0.

4. References

- (1) Férey, G.; Mellot-Draznieks, C.; Serre, C.; Millange, F.; Dutour, J.; Surblé, S.; Margiolaki, I. A chromium terephthalate-based solid with unusually large pore volumes and surface. *Science* **309**, 2040–2042 (2005).
- 2) Hartmann, M.; Fischer, M. Amino-functionalized basic catalysts with MIL-101 structure. *Microporous Mesoporous Mater.* **164**, 38–43 (2012).