## Mutual solubilities between water and non-aromatic sulfonium-, ammonium- and phosphonium- hydrophobic ionic liquids<sup>†</sup>

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## **Electronic Supporting Information**

Т/К —	$[S_{221}][NTf_2]$	$[S_{222}][NTf_2]$	$[N_{4111}][NTf_2]$	$[N_{4441}][NTf_2]$	$[N_{8881}][NTf_2]$	$[P_{8881}][NTf_2]$
			$x_{ m w}$ =	$\pm \sigma$		
288.15	$0.2284 \pm 0.0003$	$0.1841 \pm 0.0001$	$0.2104 \pm 0.0005$	$0.0979 \pm 0.0009$	$0.0169 \pm 0.0005$	$0.0132 \pm 0.0004$
293.15	$0.2420 \pm 0.0003$	$0.1936 \pm 0.0008$	$0.2243 \pm 0.0003$	$0.1023 \pm 0.0005$	$0.0215 \pm 0.0006$	$0.0163 \pm 0.0003$
298.15	$0.2561 \pm 0.0005$	$0.2065 \pm 0.0006$	$0.2396 \pm 0.0001$	$0.1089 \pm 0.0002$	$0.0272 \pm 0.0006$	$0.0195 \pm 0.0004$
303.15	$0.2705 \pm 0.0001$	$0.2177 \pm 0.0004$	$0.2513 \pm 0.0009$	$0.1179 \pm 0.0001$	$0.0354 \pm 0.0004$	$0.0240 \pm 0.0008$
308.15	$0.2854 \pm 0.0008$	$0.2324 \pm 0.0001$	$0.2723 \pm 0.0003$	$0.1252 \pm 0.0003$	$0.0426 \pm 0.0006$	$0.0316 \pm 0.0004$
313.15	$0.3007 \pm 0.0006$	$0.2455 \pm 0.0008$	$0.2857 \pm 0.0003$	$0.1329 \pm 0.0007$	$0.0508 \pm 0.0002$	$0.0416 \pm 0.0005$
318.15	$0.3163 \pm 0.0004$	$0.2559 \pm 0.0003$	$0.2995 \pm 0.0008$	$0.1399 \pm 0.0005$	$0.0597 \pm 0.0002$	$0.0474 \pm 0.0004$

**Table S1a.** Experimental mole fraction solubility of water,  $x_w$ , in the studied ILs as a function of temperature along with its standard deviation,  $\sigma$ .

**Table S1b**. Experimental mass fraction solubility of water,  $w_w$ , in the studied ILs as a function of temperature along with its standard deviation,  $\sigma$ .

	$[S_{221}][NTf_2]$	$[S_{222}][NTf_2]$	$[N_{4111}][NTf_2]$	$[N_{4441}][NTf_2]$	$[N_{1888}][NTf_2]$	$[P_{1888}][NTf_2]$
<i>1</i> /K			$\mathcal{W}_{\mathrm{W}}$ :	$\pm \sigma$		
288.15	$0.0137 \pm 0.00001$	$0.0101 \pm 0.00001$	$0.0120 \pm 0.00002$	$0.0041 \pm 0.00003$	$0.0005 \pm 0.00001$	$0.0004 \pm 0.00001$
293.15	$0.0147 \pm 0.00001$	$0.0107 \pm 0.00004$	$0.0130 \pm 0.00001$	$0.0043 \pm 0.00002$	$0.0006 \pm 0.00002$	$0.0004 \pm 0.00001$
298.15	$0.0158 \pm 0.00002$	$0.0116 \pm 0.00003$	$0.0141 \pm 0.00001$	$0.0046 \pm 0.00001$	$0.0008 \pm 0.00002$	$0.0005 \pm 0.00001$
303.15	$0.0170 \pm 0.00001$	$0.0124 \pm 0.00002$	$0.0150 \pm 0.00004$	$0.0050 \pm 0.00001$	$0.0010 \pm 0.00001$	$0.0007 \pm 0.00002$
308.15	$0.0183 \pm 0.00004$	$0.0135 \pm 0.00001$	$0.0167 \pm 0.00001$	$0.0053 \pm 0.00001$	$0.0012 \pm 0.00002$	$0.0009 \pm 0.00001$
313.15	$0.0197 \pm 0.00003$	$0.0145 \pm 0.00004$	$0.0179 \pm 0.00001$	$0.0057 \pm 0.00003$	$0.0015 \pm 0.00001$	$0.0012 \pm 0.00001$
318.15	$0.0212 \pm 0.00002$	$0.0153 \pm 0.00001$	$0.0191 \pm 0.00004$	$0.0061 \pm 0.00002$	$0.0018 \pm 0.00001$	$0.0013 \pm 0.00001$

T/V	[S <sub>221</sub> ][NTf <sub>2</sub> ]	$[S_{222}][NTf_2]$	$[N_{4111}][NTf_2]$	$[N_{4441}][NTf_2]$
<i>1</i> / <b>K</b>	$10^2 \cdot (x_{\text{IL}} \pm \sigma)$	$10^3 \cdot (x_{\text{IL}} \pm \sigma)$	$10^3 \cdot (x_{\text{IL}} \pm \sigma)$	$10^3 \cdot (x_{\text{IL}} \pm \sigma)$
288.15	$0.1194 \pm 0.0001$	$0.6436 \pm 0.0008$	$0.9250 \pm 0.0001$	$0.1190 \pm 0.0001$
293.15	$0.1228 \pm 0.0001$	$0.6596 \pm 0.0009$	$0.9353 \pm 0.0004$	$0.1286 \pm 0.0002$
298.15	$0.1266 \pm 0.0001$	$0.6778 \pm 0.0008$	$0.9491 \pm 0.0005$	$0.1395 \pm 0.0002$
303.15	$0.1308 \pm 0.0009$	$0.6983 \pm 0.0006$	$0.9664 \pm 0.0005$	$0.1518 \pm 0.0001$
308.15	$0.1354 \pm 0.0002$	$0.7211 \pm 0.0004$	$0.9872 \pm 0.0006$	$0.1656 \pm 0.0007$
313.15	$0.1405 \pm 0.0003$	$0.7466 \pm 0.0008$	$1.0116 \pm 0.0006$	$0.1810 \pm 0.0005$
318.15	$0.1461 \pm 0.0004$	$0.7747 \pm 0.0003$	$1.0399 \pm 0.0009$	$0.1981 \pm 0.0006$

**Table S2a**. Experimental mole fraction solubility of the studied ILs,  $x_{IL}$ , in water as a function of temperature along with its standard deviation,  $\sigma$ .

**Table S2b**. Experimental mass fraction solubility of the studied ILs,  $w_{IL}$ , in water as a function of temperature along with its standard deviation,  $\sigma$ .

T/V	$[S_{221}][NTf_2]$	$[S_{222}][NTf_2]$	$[N_{4111}][NTf_2]$	$[N_{4441}][NTf_2]$
1/ <b>K</b>			$w_{ m IL} \pm \sigma$	
288.15	$0.0249 \pm 0.00002$	$0.0141 \pm 0.00018$	$0.0200 \pm 0.00002$	$0.0032 \pm 0.00003$
293.15	$0.0256 \pm 0.00002$	$0.0144 \pm 0.00020$	$0.0202 \pm 0.00009$	$0.0034 \pm 0.00005$
298.15	$0.0264 \pm 0.00002$	$0.0148 \pm 0.00018$	$0.0205 \pm 0.00011$	$0.0037 \pm 0.00005$
303.15	$0.0272 \pm 0.00019$	$0.0153 \pm 0.00013$	$0.0208 \pm 0.00011$	$0.0040 \pm 0.00003$
308.15	$0.0282 \pm 0.00004$	$0.0157 \pm 0.00009$	$0.0213 \pm 0.00013$	$0.0044 \pm 0.00019$
313.15	$0.0292 \pm 0.00006$	$0.0163 \pm 0.00018$	$0.0218 \pm 0.00020$	$0.0048 \pm 0.00013$
318.15	$0.0304 \pm 0.00009$	$0.0169 \pm 0.00007$	$0.0224 \pm 0.00002$	$0.0053 \pm 0.00016$

**Table S3**. Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment calculated at the BP-TZVP level of theory for H<sub>2</sub>O.



Centre	Atomic	Atom	С	Coordinates/Å			σ-m	σ-moment	
number	Number	Atom	Х	у	Z	INFA	Hb_acc3	Hb_don3	
1	8	0	0.6391	-0.0279	0.4947	-0.948	5.6933	0	
2	1	Н	0.1598	0.0324	-0.3698	0.474	0	1.9255	
3	1	Н	-0.7990	-0.0045	-0.1249	0.474	0	1.9252	
Total							5.6933	3.8506	

**Table S4**. Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for [NTf2]- anion.



Centre	Atomic	Atom	(	Coordinates	/Å	NDA	σ-ma	oment
number	Number	Atom	Х	у	Z	NFA	Hb_acc3	Hb_don3
1	6	С	-0.3638	-0.0445	2.6363	0.867	0	0
2	9	F	-1.4434	-0.7669	2.2879	-0.353	0	0
3	16	S	0.8690	0.0948	1.1568	2.195	0	0
4	8	0	1.8978	1.0554	1.6514	-0.944	0.5086	0
5	7	Ν	-0.0011	0.8858	0.0003	-1.219	0.6511	0
6	16	S	-0.8696	0.0948	-1.1573	2.195	0	0
7	8	0	-1.2738	-1.3118	-0.8670	-0.937	0.2948	0
8	8	0	1.2734	-1.3116	0.8663	-0.937	0.3072	0
9	9	F	0.2600	-0.6546	3.6631	-0.361	0	0
10	9	F	-0.7620	1.1790	3.0295	-0.356	0	0
11	12	С	0.3643	-0.0445	-2.6359	0.867	0	0
12	9	F	1.4455	-0.7635	-2.2855	-0.353	0	0
13	8	0	-1.8983	1.0550	-1.6525	-0.944	0.5305	0
14	9	F	-0.2575	-0.6581	-3.6618	-0.361	0	0
15	9	F	0.7596	1.1792	-3.0314	-0.356	0	0
Total						1.000	2.2922	0

**Table S5.** Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for  $[S_{221}]^+$  cation.



Centre	Atomic	Atom	Co	oordinates	ordinates/Å		σ-moment	
number	Number	Atom	Х	У	Z	ΝrΑ	Hb_acc3	Hb_don3
1	6	С	-0.0342	0.9937	2.4073	-0.711	0	0
2	6	С	0.4013	0.9776	0.9491	-0.587	0	0
3	16	S	-0.4653	-0.3527	-0.0014	0.881	0	0
4	6	С	0.2704	-1.8716	0.6878	-0.834	0	0
5	1	Н	-0.0668	-2.6985	0.0534	0.290	0	0.1681
6	6	С	0.3670	-0.2673	-1.6530	-0.586	0	0
7	6	С	-0.0494	0.9736	-2.4284	-0.712	0	0
8	1	Η	1.4783	0.8085	0.8223	0.268	0	0.0692
9	1	Η	0.1122	1.9013	0.4322	0.284	0	0.0641
10	1	Η	-0.1262	-1.9990	1.7001	0.291	0	0.0723
11	1	Η	1.3622	-1.7878	0.6901	0.271	0	0.0873
12	1	Н	1.4474	-0.3187	-1.4677	0.269	0	0.0452
13	1	Η	0.0389	-1.1899	-2.1490	0.285	0	0.1363
14	1	Н	0.4452	1.8549	2.8929	0.285	0	0
15	1	Н	-1.1216	1.1112	2.5069	0.262	0	0
16	1	Н	0.2803	0.0909	2.9463	0.249	0	0
17	1	Н	0.4043	0.9146	-3.4274	0.286	0	0
18	1	Н	-1.1388	1.0296	-2.5545	0.262	0	0
19	1	Н	0.3060	1.8994	-1.9580	0.249	0	0
Total						1.000	0	0.6425

**Table S6**. Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for  $[S_{222}]^+$  cation.



Centre	Atomic	Atom	С	oordinates	/Å	NDA	σ-m	oment
number	Number	Atom	X	у	Z	NPA	Hb_acc3	Hb_don3
1	6	С	-1.5907	-0.4510	0.4745	-0.585	0	0
2	16	S	0.0001	0.3713	0.0000	0.861	0	0
3	6	С	0.3848	-0.4512	-1.6148	-0.585	0	0
4	6	С	1.2071	-0.4501	1.1407	-0.585	0	0
5	6	С	1.5855	0.1897	-2.2946	-0.709	0	0
6	1	Н	0.5281	-1.5172	-1.3963	0.268	0	0.0459
7	1	Н	-0.5324	-0.3233	-2.2040	0.281	0	0.0191
8	1	Н	2.1760	-0.3197	0.6422	0.281	0	0.0140
9	6	С	1.1927	0.1893	2.5210	-0.709	0	0
10	1	Н	0.9489	-1.5168	1.1543	0.268	0	0.0456
11	6	С	-2.7796	0.1897	-0.2268	-0.709	0	0
12	1	Н	-1.4731	-1.5170	0.2414	0.268	0	0.0450
13	1	Н	-1.6430	-0.3231	1.5633	0.281	0	0.0172
14	1	Н	1.7428	-0.3196	-3.2557	0.283	0	0
15	1	Н	1.4156	1.2551	-2.4976	0.262	0	0
16	1	Н	2.5065	0.0802	-1.7068	0.247	0	0
17	1	Н	1.9473	-0.3189	3.1376	0.283	0	0
18	1	Н	1.4507	1.2554	2.4766	0.262	0	0
19	1	Н	0.2232	0.0766	3.0238	0.247	0	0
20	1	Н	-3.6908	-0.3197	0.1171	0.283	0	0
21	1	Н	-2.8706	1.2551	0.0215	0.262	0	0
22	1	Н	-2.7305	0.0795	-1.3183	0.247	0	0
Total						1.000	0.0000	0.1868

**Table S7**. Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment calculated at the BP-TZVP level of theory for  $[N_{4111}]^+$  cation.



Centre	Atomic	Atom	C	oordinates	/Å	NDA	<u>σ-m</u>	oment
number	Number	Atom	Х	у	Z	INF A	Hb_acc3	Hb_don3
1	6	С	-0.6065	-0.0152	-1.4247	-0.474	0	0
2	7	Ν	-0.0856	0.0649	-0.0141	-0.295	0	0
3	6	С	1.1129	-0.8678	0.1811	-0.252	0	0
4	6	С	0.3819	1.4735	0.2511	-0.472	0	0
5	6	С	-1.1910	-0.2679	0.9523	-0.474	0	0
6	1	Н	-0.7874	-0.2339	1.9695	0.263	0	0.0269
7	1	Н	-1.9852	0.4765	0.8336	0.264	0	0.0273
8	1	Н	-1.5804	-1.2645	0.7275	0.265	0	0.0033
9	1	Н	-1.0062	-1.0166	-1.6060	0.265	0	0.0036
10	1	Н	-1.4015	0.7287	-1.5391	0.264	0	0.0309
11	1	Н	0.2168	0.2000	-2.1137	0.263	0	0.0247
12	1	Н	0.7532	1.5307	1.2795	0.264	0	0.0300
13	1	Н	1.1809	1.7159	-0.4569	0.264	0	0.0285
14	1	Н	-0.4658	2.1529	0.1146	0.263	0	0.0258
15	1	Н	1.4573	-0.6844	1.2078	0.260	0	0.0152
16	6	С	0.8471	-2.3510	-0.0439	-0.489	0	0
17	1	Н	1.8813	-0.4993	-0.5118	0.260	0	0.0223
18	6	С	2.1353	-3.1596	0.1927	-0.457	0	0
19	1	Н	0.0674	-2.7155	0.6420	0.245	0	0
20	1	Н	0.4967	-2.5321	-1.0714	0.245	0	0
21	6	С	1.9277	-4.6624	-0.0213	-0.681	0	0
22	1	Н	2.9234	-2.7938	-0.4857	0.241	0	0
23	1	Н	2.4950	-2.9774	1.2187	0.241	0	0
24	1	Н	2.8606	-5.2173	0.1540	0.257	0	0
25	1	Н	1.1654	-5.0602	0.6661	0.235	0	0
26	1	Н	1.5979	-4.8755	-1.0495	0.235	0	0

Total

**Table S8.** Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for  $[N_{4441}]^+$  cation.



Centre	Atomic	Atom	С	oordinates	/Å	NDA	σ-m	oment
number	Number	Atom	X	у	Z	NPA	Hb_acc3	Hb_don3
1	6	С	3.5856	-0.5121	-1.2998	-0.486	0	0
2	6	С	2.9721	-1.0713	-0.0204	-0.255	0	0
3	7	Ν	1.4459	-1.1490	0.0062	-0.288	0	0
4	6	С	0.8949	-1.9236	-1.1964	-0.254	0	0
5	6	С	1.4033	-3.3501	-1.3688	-0.486	0	0
6	6	С	0.8507	0.2344	-0.0467	-0.475	0	0
7	1	Н	-0.2292	0.1525	0.1162	0.260	0	0.0158
8	6	С	0.9936	-1.8550	1.2894	-0.252	0	0
9	6	С	1.3072	-1.1410	2.6005	-0.485	0	0
10	1	Н	1.0431	0.6678	-1.0323	0.261	0	0.0080
11	1	Н	1.3075	0.8550	0.7288	0.262	0	0.0040
12	1	Н	3.2512	-0.4466	0.8362	0.258	0	0.0004
13	1	Н	3.3314	-2.0899	0.1738	0.259	0	0.0017
14	1	Н	-0.0904	-1.9917	1.1756	0.256	0	0.0163
15	1	Н	1.4702	-2.8417	1.2728	0.258	0	0.0005
16	1	Н	-0.1960	-1.9091	-1.0702	0.258	0	0.0166
17	1	Н	1.1363	-1.3175	-2.0781	0.259	0	0.0010
18	6	С	5.1155	-0.4257	-1.1595	-0.456	0	0
19	1	Н	3.1898	0.4923	-1.5151	0.243	0	0
20	1	Н	3.3428	-1.1514	-2.1620	0.242	0	0

21	6	С	0.8087	-1.9846	3.7874	-0.456	0	0
22	1	Н	2.3900	-0.9790	2.7111	0.243	0	0
23	1	Н	0.8195	-0.1560	2.6353	0.243	0	0
24	6	С	0.7923	-3.9788	-2.6337	-0.456	0	0
25	1	Н	1.1382	-3.9703	-0.5001	0.244	0	0
26	1	Η	2.5007	-3.3646	-1.4583	0.242	0	0
27	6	С	5.7871	0.1226	-2.4228	-0.680	0	0
28	1	Η	5.3652	0.2154	-0.2983	0.239	0	0
29	1	Η	5.5169	-1.4269	-0.9312	0.239	0	0
30	6	С	1.0938	-1.3177	5.1366	-0.680	0	0
31	1	Η	-0.2750	-2.1566	3.6808	0.239	0	0
32	1	Н	1.2882	-2.9764	3.7556	0.239	0	0
33	6	С	1.2672	-5.4182	-2.8564	-0.680	0	0
34	1	Н	-0.3070	-3.9588	-2.5542	0.240	0	0
35	1	Н	1.0537	-3.3621	-3.5093	0.238	0	0
36	1	Н	0.7268	-1.9371	5.9676	0.255	0	0
37	1	Н	2.1739	-1.1640	5.2836	0.234	0	0
38	1	Н	0.6017	-0.3356	5.2062	0.234	0	0
39	1	Н	0.8199	-5.8441	-3.7660	0.255	0	0
40	1	Н	2.3616	-5.4628	-2.9669	0.233	0	0
41	1	Н	0.9880	-6.0642	-2.0099	0.234	0	0
42	1	Н	6.8784	0.1729	-2.2983	0.255	0	0
43	1	Н	5.5769	-0.5167	-3.2937	0.234	0	0
44	1	Н	5.4278	1.1368	-2.6554	0.234	0	0
Total						1.000	0.0000	0.0643

**Table S9.** Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for  $[N_{1888}]^+$  cation.



Centre	Atomic	Atom	Coo	ordinates/Å	L	ΝΙΟΛ	σ-moment	
number	Number	Atom	X	у	Z	NPA	Hb_acc3	Hb_don3
1	7	Ν	-0.0022	-1.5530	0.8441	-0.288	0	0
2	6	С	-1.2289	-1.8449	-0.0297	-0.253	0	0
3	6	С	-0.0017	-0.1158	1.3610	-0.256	0	0
4	6	С	-0.0030	-2.4774	2.0336	-0.471	0	0
5	6	С	1.2242	-1.8454	-0.0296	-0.253	0	0
6	1	Н	0.8819	-0.0322	2.0057	0.260	0	0.0008
7	6	С	-0.0017	0.9701	0.2932	-0.482	0	0
8	1	Н	-0.8847	-0.0318	2.0063	0.260	0	0.0007
9	1	Н	0.8870	-2.2827	2.6378	0.260	0	0.0036
10	1	Н	-0.8970	-2.2866	2.6332	0.260	0	0.0030
11	1	Н	0.0001	-3.5108	1.6716	0.258	0	0.0208
12	1	Н	-1.1433	-2.9079	-0.2917	0.257	0	0.0159
13	6	С	-2.5875	-1.5505	0.5948	-0.482	0	0
14	1	Н	-1.0908	-1.2609	-0.9482	0.259	0	0.0011
15	6	С	2.5831	-1.5590	0.5980	-0.482	0	0
16	1	Н	1.1350	-2.9071	-0.2959	0.257	0	0.0169
17	1	Н	1.0893	-1.2574	-0.9459	0.259	0	0.0010
18	6	С	3.7056	-1.9629	-0.3747	-0.450	0	0
19	1	Н	2.6850	-0.4887	0.8337	0.243	0	0
20	1	Н	2.7079	-2.1171	1.5380	0.242	0	0

21	6	С	-3.7103	-1.9548	-0.3772	-0.450	0	0
22	1	Н	-2.7155	-2.1036	1.5373	0.242	0	0
23	1	Н	-2.6860	-0.4789	0.8260	0.243	0	0
24	6	С	-5.1097	-1.6865	0.1910	-0.451	0	0
25	1	Н	-3.5887	-1.4041	-1.3256	0.236	0	0
26	1	Н	-3 6129	-3 0262	-0 6216	0 2 3 7	0	0
27	6	C	5 1052	-1 6924	0 1916	-0.451	Ő	0
28	1	н	3 6089	-3 0346	-0.6181	0.237	Ő	0
29	1	Н	3 5827	-1 4131	-1 3235	0.237	0	0
30	6	C II	6 2332	-2 0738	-0 7763	-0.452	0	0
31	1	н	5 1932	-0.6224	0.4496	0.132	0	0
32	1	н	5 2294	-2 2512	1 1358	0.231	0	0
32	6	П С	-6 2384	-2.2312	-0 7766	-0.452	0	0
24	1	с u	-0.2384	-2.0008	-0.7700	-0.452	0	0
25	1	11 11	-5.2323	-2.2473	0.4500	0.231	0	0
55 26	1	П	-3.1965	-0.01/1	0.4309	0.251	0	0
30 27	0		/.0303	-1.8201	-0.2111	-0.454	0	0
3/	1	Н	6.1369	-3.1409	-1.0446	0.230	0	0
38	l	Н	6.1120	-1.506/	-1./163	0.229	0	0
39	6	C	8.7666	-2.1854	-1.1832	-0.454	0	0
40	l	H	7.7299	-0.7553	0.0687	0.227	0	0
41	1	Н	7.7617	-2.3958	0.7235	0.227	0	0
42	6	С	10.1651	-1.9382	-0.6065	-0.676	0	0
43	1	Н	8.6686	-3.2472	-1.4684	0.229	0	0
44	1	Н	8.6443	-1.6047	-2.1138	0.229	0	0
45	1	Н	10.9523	-2.2055	-1.3273	0.240	0	0
46	1	Н	10.3028	-0.8786	-0.3391	0.227	0	0
47	1	Н	10.3293	-2.5346	0.3048	0.227	0	0
48	6	С	-7.6411	-1.8109	-0.2112	-0.454	0	0
49	1	Н	-6.1167	-1.5005	-1.7171	0.229	0	0
50	1	Н	-6.1434	-3.1341	-1.0450	0.230	0	0
51	6	С	-8.7721	-2.1782	-1.1815	-0.454	0	0
52	1	Н	-7.7662	-2.3837	0.7252	0.227	0	0
53	1	Н	-7.7335	-0.7452	0.0654	0.227	0	0
54	6	С	-10.1699	-1.9222	-0.6070	-0.676	0	0
55	1	Н	-8.6473	-1.6036	-2.1156	0.229	0	0
56	1	Н	-8.6778	-3.2419	-1.4602	0.229	0	0
57	1	Н	-10.9580	-2.1928	-1.3256	0.240	0	0
58	1	Н	-10.3363	-2.5108	0.3090	0.227	0	0
59	1	Н	-10 3044	-0.8600	-0 3480	0 227	0	0
60	6	C	-0.0013	2 3611	0.9506	-0.451	Ő	0
61	1	н	0 8841	0.8786	-0 3538	0.242	Ő	0
62	1	Н	-0.8875	0.8788	-0 3539	0.242	0	0
63	6	C	0.0010	3 5025	-0.0740	-0.451	0	0
64	1	н	-0 8858	2.3025 2.4566	1 6037	0.736	0	0
65	1	и Ц	0.8872	2.7500	1.60.52	0.230	0	0
66	6	C II	0.0023	2.7550 1 8010	0 5725	-0 152	0	0
67	1	с ц	0.0042	7.024U 3 /01Q	-0 7292	0.432	0	0
60	1	11 U	0.0040	2 1056	-0.7203	0.231	0	0
00	1	п	-0.0029	5.4030	-0.1203	0.231	U	U

69	6	С	0.0082	6.0422	-0.4446	-0.454	0	0
70	1	Н	-0.8794	4.9923	1.2279	0.229	0	0
71	1	Н	0.8872	4.9873	1.2294	0.229	0	0
72	6	С	0.0142	7.4346	0.2006	-0.454	0	0
73	1	Н	0.8908	5.9434	-1.1017	0.227	0	0
74	1	Н	-0.8755	5.9505	-1.1012	0.227	0	0
75	6	С	0.0190	8.5761	-0.8224	-0.676	0	0
76	1	Н	-0.8679	7.5325	0.8566	0.229	0	0
77	1	Н	0.8970	7.5249	0.8569	0.229	0	0
78	1	Н	0.0246	9.5601	-0.3301	0.240	0	0
79	1	Н	0.9067	8.5225	-1.4722	0.227	0	0
80	1	Н	-0.8705	8.5317	-1.4705	0.227	0	0
Total						1.000	0.0000	0.0638

**Table S10.** Atom coordinates, charges from Natural Population Analysis (NPA), and  $\sigma$ -moment<br/>calculated at the BP-TZVP level of theory for  $[P_{1888}]^+$  cation.



Centre	Atomic	Atom	Coordinates/Å			σ-moment		
number	Number	Atom	X	У	Z	NPA -	Hb_acc3	Hb_don3
1	15	Р	0.0318	-1.5349	0.7073	1.689	0	0
2	6	С	0.0113	0.1720	1.3813	-0.805	0	0
3	6	С	0.0459	-2.7043	2.1028	-1.044	0	0
4	6	С	-1.4466	-1.8377	-0.3372	-0.800	0	0
5	6	С	1.5168	-1.8023	-0.3376	-0.800	0	0
6	6	С	-2.7969	-1.6521	0.3729	-0.461	0	0
7	1	Н	0.0623	-3.7305	1.7145	0.287	0	0.0127
8	6	С	2.8631	-1.6016	0.3760	-0.461	0	0
9	6	С	-0.0143	1.2852	0.3221	-0.461	0	0
10	6	С	4.0443	-1.8689	-0.5688	-0.451	0	0
11	6	С	-0.0304	2.6783	0.9693	-0.451	0	0
12	6	С	-3.9728	-1.9567	-0.5672	-0.451	0	0
13	6	С	-5.3384	-1.8031	0.1145	-0.452	0	0
14	6	С	-6.5188	-2.1125	-0.8151	-0.452	0	0
15	6	С	-7.8866	-1.9844	-0.1320	-0.454	0	0
16	6	С	5.4072	-1.7143	0.1181	-0.452	0	0
17	6	С	6.5918	-1.9799	-0.8198	-0.452	0	0
18	6	С	7.9577	-1.8641	-0.1307	-0.454	0	0
19	6	С	-0.0679	3.8156	-0.0594	-0.452	0	0
20	6	С	-0.0838	5.2106	0.5793	-0.452	0	0
21	6	С	-0.1318	6.3525	-0.4442	-0.454	0	0
22	1	Н	-1.3371	-2.8669	-0.7146	0.280	0	0.0120

23	1	Н	-1.3575	-1.1677	-1.2064	0.282	0	0.0004
24	1	Н	0.9352	-2.5287	2.7206	0.289	0	0.0021
25	1	Н	-0.8518	-2.5558	2.7156	0.289	0	0.0019
26	1	Н	1.4253	-2.8292	-0.7256	0.280	0	0.0137
27	1	Н	1.4176	-1.1253	-1.2003	0.282	0	0.0008
28	1	Н	-0.8668	0.2313	2.0430	0.281	0	0.0015
29	1	Н	0.8974	0.2581	2.0293	0.281	0	0.0009
30	1	Н	2.9365	-0.5739	0.7666	0.239	0	0
31	1	Н	2 9335	-2 2784	1 2421	0.238	0	0
32	1	Н	0.8654	1 2018	-0 3357	0.239	0	0
33	1	Н	-0.9019	1 1744	-0 3207	0.239	Ő	ů 0
34	1	Н	-2 8549	-2 3160	1 2499	0.238	Ő	Ő
35	1	Н	-2 8898	-0.6199	0 7470	0.230	0	0
36	1	Н	-3 9203	-1 2876	-1 4433	0.235	0	0
37	1	Н	-3 8689	-2 9858	-0.9524	0.235	0	0
38	1	Н	-5 3823	-2.9050	0.9924	0.230	0	0
30	1	н Н	-5 / 396	-2.4707	0.5022	0.230	0	0
<i>4</i> 0	1	н Н	-6 4822	-0.7757	-1 6876	0.230	0	0
40 //1	1	н Н	-6 4070	-3.1366	-1.0070	0.229	0	0
41 1	1	и П	7 0231	-5.1500	0 7388	0.227	0	0
42	1	и П	7 0002	-2.0034	0.7500	0.227 0.227	0	0
45	1	и П	-7.9992	-0.901/ 2 8808	0.2709	0.227 0.225	0	0
44 15	1	и П	3.9341	-2.0090	-0.9780	0.235	0	0
45	1 1	п u	5.9643	-1.1001	-1.4290	0.230	0	0
40	1	и П	5.4933	-0.0955	0.3332	0.230	0	0
4/ 10	1 1	п u	5.4000	-2.4004	0.9709	0.230	0	0
40	1	11 U	6 5 5 0 3	-2.9900	-1.2555	0.229	0	0
49 50	1	11 U	0.5505	-1.2/31	-1.0070	0.229	0	0
51	1	11 U	8.0392	-0.0373	0.3133	0.227	0	0
52	1 1	п u	0.0003	-2.3772	0.7122	0.227 0.225	0	0
52 52	1	П	-0.9001	2.7309	1.0302	0.235	0	0
55 54	1	П	0.8012	2.7902	1.0090	0.230	0	0
54	1	П	0.8066	3./310 2.(070	-0.7280	0.230	0	0
55 57	1 1	H	-0.9599	5.09/9	-0.6994	0.230	0	0
50 57	1	H	-0.9542	5.2907	1.2544	0.229	0	0
57 50	1	H	0.8113	5.3292	1.2155	0.229	0	0
58 50	1 1	H	0.7372	6.2720	-1.1214	0.227	0	0
59 (0	l	H	-1.0280	6.2345	-1.0/94	0.227	0	0
60	6	C	-10.4307	-2.1/44	-0.3645	-0.6/6	0	0
61	6	C	-9.0692	-2.2918	-1.0596	-0.454	0	0
62	l	H	-10.5054	-2.8/59	0.4814	0.227	0	0
63	l	H	-10.5883	-1.1585	0.0306	0.227	0	0
64	l	H	-11.2575	-2.3955	-1.0561	0.239	0	0
65	l	H	-8.9516	-3.3104	-1.4687	0.229	0	0
66	1	H	-9.0371	-1.6079	-1.9253	0.229	0	0
67	6	C	-0.1997	8.8827	-0.8392	-0.676	0	0
68	6	C	-0.1463	7.7490	0.1914	-0.454	0	0
69	1	H	0.6715	8.8484	-1.5119	0.227	0	0
70	1	Н	-1.1044	8.8094	-1.4626	0.227	0	0

71	1	Н	-0.2079	9.8699	-0.3529	0.240	0	0
72	1	Н	0.7506	7.8676	0.8235	0.229	0	0
73	1	Н	-1.0127	7.8275	0.8704	0.229	0	0
74	6	С	10.5032	-2.0276	-0.3662	-0.676	0	0
75	6	С	9.1440	-2.1222	-1.0688	-0.454	0	0
76	1	Н	10.6525	-1.0304	0.0769	0.227	0	0
77	1	Н	10.5815	-2.7677	0.4456	0.227	0	0
78	1	Н	11.3328	-2.2094	-1.0659	0.239	0	0
79	1	Н	9.1092	-1.3998	-1.9027	0.229	0	0
80	1	Н	9.0336	-3.1217	-1.5241	0.229	0	0
Total						1.000	0.0000	0.0460

T/K	$x_{\mathrm{IL}}$	x <sub>H2O</sub>	$H_{\mathrm{MF}}/\mathrm{kJ}\cdot\mathrm{mol}^{-1b}$	$H_{\rm HB}/{\rm kJ}\cdot{\rm mol}^{-1c}$	$H_{\rm VdW}/{\rm kJ}\cdot{\rm mol}^{-1d}$	$H_{\rm INT}/{\rm kJ}{\cdot}{ m mol}^{-1e}$
[S <sub>221</sub> ][NTf <sub>2</sub> ]						
288.15	0.2284	0.7716	0.0577	1.6020	-0.0989	1.5607
293.15	0.2420	0.7580	0.0688	1.7253	-0.1056	1.6885
298.15	0.2561	0.7439	0.0809	1.8486	-0.1121	1.8175
303.15	0.2705	0.7295	0.0942	1.9712	-0.1184	1.9470
308.15	0.2854	0.7146	0.1086	2.0925	-0.1247	2.0764
313.15	0.3007	0.6993	0.1246	2.2315	-0.1331	2.2230
318.15	0.3163	0.6837	0.1406	2.3278	-0.1367	2.3318
$[N_{4111}][NTf_2]$						
288.15	0.2104	0.7896	0.0735	1.5724	-0.1038	1.5421
293.15	0.2243	0.7757	0.0865	1.6997	-0.1115	1.6747
298.15	0.2396	0.7604	0.1007	1.8273	-0.1192	1.8089
303.15	0.2513	0.7487	0.1154	1.9317	-0.1240	1.9232
308.15	0.2723	0.7277	0.1339	2.1024	-0.1368	2.0994
313.15	0.2857	0.7143	0.1520	2.2261	-0.1441	2.2340
318.15	0.2995	0.7005	0.1702	2.3258	-0.1486	2.3475
[S <sub>222</sub> ][NTf <sub>2</sub> ]						
288.15	0.1841	0.8159	0.0809	1.4201	-0.0784	1.4226
293.15	0.1936	0.8064	0.0935	1.5220	-0.0829	1.5326
298.15	0.2065	0.7935	0.1070	1.6243	-0.0873	1.6440
303.15	0.2177	0.7823	0.1229	1.7514	-0.0940	1.7802
308.15	0.2324	0.7676	0.1387	1.8529	-0.0982	1.8934
313.15	0.2455	0.7545	0.1569	1.9782	-0.1047	2.0303
318.15	0.2559	0.7441	0.1765	2.1013	-0.1112	2.1666
$[N_{4441}][NTf_2]$						
288.15	0.0979	0.9021	0.0927	0.9069	-0.0484	0.9511
293.15	0.1023	0.8977	0.1093	1.0100	-0.0542	1.0652
298.15	0.1089	0.8911	0.1236	1.0776	-0.0571	1.1440
303.15	0.1179	0.8821	0.1388	1.1452	-0.0601	1.2238
308.15	0.1252	0.8748	0.1597	1.2510	-0.0658	1.3448
313.15	0.1329	0.8671	0.1822	1.3569	-0.0715	1.4675
318.15	0.1399	0.8601	0.2009	1.4236	-0.0744	1.5501
$[N_{1888}][NTf_2]$						
288.15	0.0179	0.9821	0.0308	0.1959	-0.0079	0.2188
293.15	0.0215	0.9785	0.0506	0.3182	-0.0136	0.3552
298.15	0.0272	0.9728	0.0640	0.3831	-0.0166	0.4305
303.15	0.0354	0.9646	0.0787	0.4487	-0.0197	0.5077
308.15	0.0426	0.9574	0.1054	0.5723	-0.0257	0.6520
313.15	0.0508	0.9492	0.1231	0.6388	-0.0289	0.7330
318.15	0.0597	0.9403	0.1537	0.7621	-0.0352	0.8806
$[P_{1888}][NTf_2]$						
288.15	0.0112	0.9888	0.0293	0.1942	-0.0079	0.2155

 Table S11. Partial molar excess enthalpy of binary mixtures (water + IL) at equilibrium concentration predicted using COSMO-RS.<sup>a</sup>

293.15	0.0153	0.9847	0.0506	0.3182	-0.0136	0.3552
298.15	0.0195	0.9805	0.0640	0.3831	-0.0166	0.4305
303.15	0.0240	0.9760	0.0787	0.4487	-0.0197	0.5077
308.15	0.0316	0.9684	0.1054	0.5723	-0.0257	0.6520
313.15	0.0416	0.9584	0.1231	0.6388	-0.0289	0.7330
318.15	0.0474	0.9526	0.1537	0.7621	-0.0352	0.8806

<sup>a</sup>The partial excess molar enthalpy was estimated using the method describe in reference 1. <sup>b</sup>Partial excess molar enthalpy as contribution of electrostatic-misfit interaction,  $H_{\rm MF}$ .

<sup>c</sup>Partial excess molar enthalpy as contribution of viroletostate inisite interaction,  $H_{\text{HB}}$ . <sup>d</sup>Partial excess molar enthalpy as contribution of Van der Waals forces,  $H_{\text{VdW}}$ . <sup>e</sup>Total Partial excess molar enthalpy as contribution of electrostatic-misfit interaction,  $H_{\text{INT}} = H_{\text{MF}} +$  $H_{\rm HB} + H_{\rm VdW}$ 

IL	$A \pm \sigma$	$B \pm \sigma$	$C \pm \sigma$	$D \pm \sigma$	$E \pm \sigma$
[S <sub>221</sub> ][NTf <sub>2</sub> ]	$1.98\pm0.01$	-995 ± 3	-87 ± 1	$3081 \pm 44$	$12 \pm 1$
$[S_{222}][NTf_2]$	$1.90\pm0.07$	$-1036 \pm 20$	$-92 \pm 1$	$3331\pm32$	$13 \pm 1$
$[N_{4111}][NTf_2]$	$2.24\pm0.09$	$-1096 \pm 26$	$-103 \pm 1$	$4016\pm36$	$14 \pm 1$
$[N_{4441}][NTf_2]$	$1.60\pm0.11$	$-1135 \pm 34$	$-135 \pm 2$	$4361\pm90$	$20 \pm 1$
$[N_{1888}][NTf_2]$	$9.51\pm0.35$	$-3911 \pm 107$			
$[P_{1888}][NTf_2]$	$9.67\pm0.54$	$-4044 \pm 162$			

**Table S12.** Parameters for the mole fraction solubility correlations as a function of temperature along with the respective standard deviation,  $\sigma$ .



Figure S1. Sigma profile (a) and sigma-potential (b) of H<sub>2</sub>O computed by COSMO-RS.

Reference

1 Kurnia, K. A.; Coutinho, J. A. P. Ind. Eng. Chem. Res. 2013, 52, 13862.