

SBA-15-Pr-SO₃H as nanoreactor catalyzed oxidation of sulfides into sulfoxides

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Experimental section

Materials and Methods: All starting materials were available and used without any purification. Melting points were determined in a capillary tube and are not corrected. ¹H NMR spectra were recorded on a 200-MHz spectrometer using TMS as internal standard.

General procedure for the oxidation of sulfides: In a sealed tube equipped with a stir bar, a solution of sulfide (1 mmol) in acetonitrile (10 mL) was prepared. Aqueous 30% H₂O₂ (2 mmol, 0.2 mL) and SBA-15-Pr-SO₃H (0.6 mmol, 0.6 g, –SO₃H groups) were added and the mixture was stirred at 40 °C for the time period specified in Table 1. After the completion of the reaction, the reaction mixture was diluted with ethanol and filtered to remove the catalyst. Evaporation gave the corresponding sulfoxide as the only product. All of the products are known compounds and were easily characterized by comparison with authentic samples (¹H NMR, mp).

^1H NMR spectra of sulfoxides (S2-S7)











