## **Supporting Information**

# Sulphated Yttria-Zirconia as a regioselective catalyst system for the alcoholysis of epoxides

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#### **Preparation of Catalysts:**

The catalyst was prepared by mixing aqueous solutions of yttrium nitrate and zirconyl nitrate in the appropriate mole ratio to which aqueous ammonia (28%) was added under vigorous stirring until a pH of the solution reached to 8.5 and precipitate was formed. Washing with deionized water, drying at 110° C for 24 h. further treating with sulfuric acid (4 M), drying at 120° C and subsequent calcinations of 500° C for 3 h. resulted in a highly acidic material.

#### Thermogravimetric analysis:

The thermal stability of the prepared sample was investigated by a TGA method. The thermogram obtained for the sulphated yttria-zirconia is presented in Fig. 5. Thermogravimetric analysis (TGA) was carried out at  $10^{\circ}$  C min<sup>-1</sup>. The TGA profile was characterized by weight loss till 223° C. The weight loss occurs at 67° C to 223° C (9.15 % weight loss). Further upto 600° C catalyst show good thermal stability and no remarkable weight loss were observed.



Figure 1. Thermogravimetric analysis curve for  $SO_4^{2-}/Y_{0.16}Zr_{0.84}O_2$  catalyst.

#### Potentiometric titrations for acidity measurement:

Potentiometric titrations were carried out by using Equip-tronic model (EQ-614A) instruments. Catalysts were dried at  $120^{\circ}$  C for 2 h., prior to use for titrations. The 0.1 g of catalysts amount dispersed in acetonitrile as non-aqueous solvent and stirred for 1 h. Further the catalyst titrated with 0.05 M solution of n-butylamine in acetonitrile. The initial electrode potential ( $E_i$ ) indicates the maximum acid strength of the sites, and the end point value in mmol/g of n-butylamine, indicates the total number of acid sites.



**Figure 2.** Potentiometric titration curves for all the mixed oxides and surface modified mixed oxides used in the study of epoxide ring opening reaction (**Please See table no. 1**).



## X-Ray diffraction analysis:

Figure 3 X-ray diffraction analysis of different compositions of sulphated yttria-zirconia catalysts.

## **FT-IR** analysis



Figure 4 FT-IR analysis of different compositions of sulphated yttria-zirconia catalysts.