## $H_2$ dissociation over Ag/Al<sub>2</sub>O<sub>3</sub>: The first step in hydrogen assisted selective catalytic reduction of NO<sub>X</sub>

Simon Klacar<sup>1\*</sup> and Henrik Grönbeck<sup>1</sup>

<sup>1</sup>Department of Applied Physics and Competence Centre for Catalysis, Chalmers University of Technology, SE-412 96 Göteborg, Sweden.

\*<u>klacar@chalmers.se</u>

## **Electronic Supplementary Information**

Herein we present the calculated low energy structures in the case of molecular and dissociated  $H_2$  adsorption. These are referred to as the initial (I) and final (F) states, respectively. Moreover, the calculated transition state (TS) structures are reported. For the Ag@Al<sub>2</sub>O<sub>3</sub>(110)-b structure, only the final state is reported as no barrier for dissociation is found in this case. Finally, for clarity, we present the bare Ag@Al<sub>2</sub>O<sub>3</sub>(110)-a structures. All structures are presented in a top view, besides the Ag@Al<sub>2</sub>O<sub>3</sub>(110)-a structure which also is given in a side view. Color codes: blue, purple, red, white balls correspond to Ag, Al, O, and H respectively.



**Figure S1.**  $H_2$  dissociation over Ag(111). The H-H distance is reported in Å.



1.42



**Figure S2.**  $H_2$  dissociation over Ag(100). The H-H distance is reported in Å.

0.75

1.49



**Figure S3.**  $H_2$  dissociation over Ag(211). The H-H distance is reported in Å.



**Figure S4.**  $H_2$  dissociation over p(4x4). The H-H distance is reported in Å. Note that the barrier for  $H_2$  dissociation is related to H diffusion over the surface, hence the H-H distance is long in the TS. The H-H distance is reported in Å.



**Figure S5.**  $H_2$  dissociation over  $Ag_2O(111)$ . In the final state (F) formation of two OH groups is observed. The H-H distance is reported in Å.

## 0.76 1.05

**Figure S6.**  $H_2$  dissociation over  $Al_2O_3(110)$ . In the initial state (I), the  $H_2$  molecule is adsorbed over the  $Al_{III}$  site. Upon dissociation (F) electron pairing leads to formation of an Ag-H and O-H pair. The H-H distance is reported in Å.



**Figure S7.** The final state for dissociated  $H_2$  over  $Ag@Al_2O_3(110)$ -b.



0.76

**Figure S8.** Top view of molecular and dissociated  $H_2$  adsorption over  $Ag_{NPM}/Al_2O_3(110)$ . Molecular adsorption occurs over the  $Al_{III}$  site and upon dissociation an Al-H and O-H pair is formed owing to electron pairing.



**Figure S9.** Top and side view of  $Ag@Al_2O_3(110)$ -a.