

Supporting Information

for the manuscript

Sulfated zirconia: an efficient solid acid catalyst for esterification of myristic acid with short chain alcohols

K. Saravanan, Beena Tyagi* and H. C. Bajaj

Discipline of Inorganic Materials and Catalysis, Council of Scientific and Industrial Research (CSIR), Central Salt and Marine Chemicals Research Institute (CSMCRI), G. B. Marg, Bhavnagar, Gujarat 364 002, India.

Fax: +91-278-2566970

E-mail: btyagi@csmcri.org

Additional Figures and Figure Captions

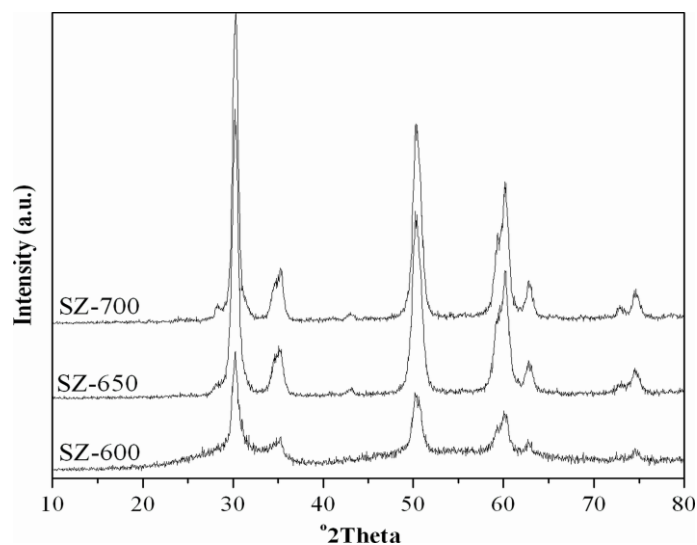


Figure S1: PXRD pattern of SZ catalysts calcined at different temperatures

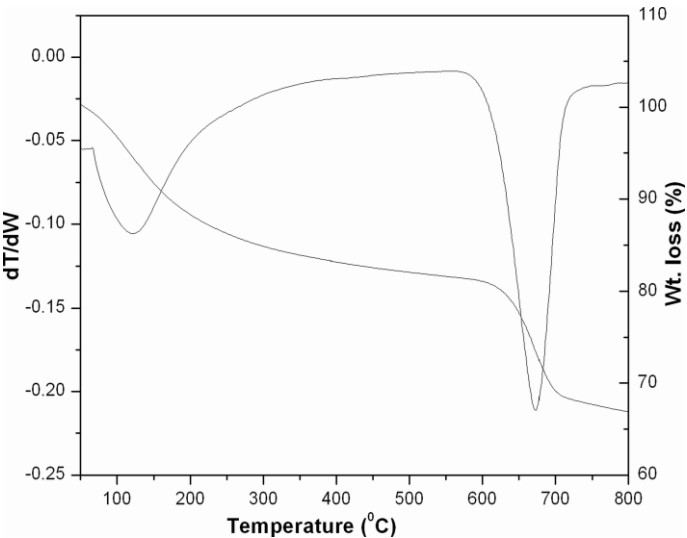


Figure S2: TGA and DTA profiles of sulfated zirconia sample

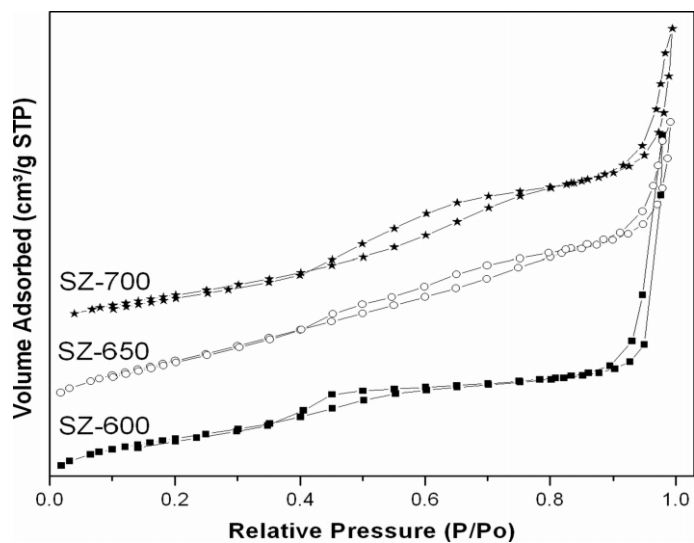


Figure S3: Nitrogen adsorption-desorption isotherms of SZ catalysts calcined at different temperatures

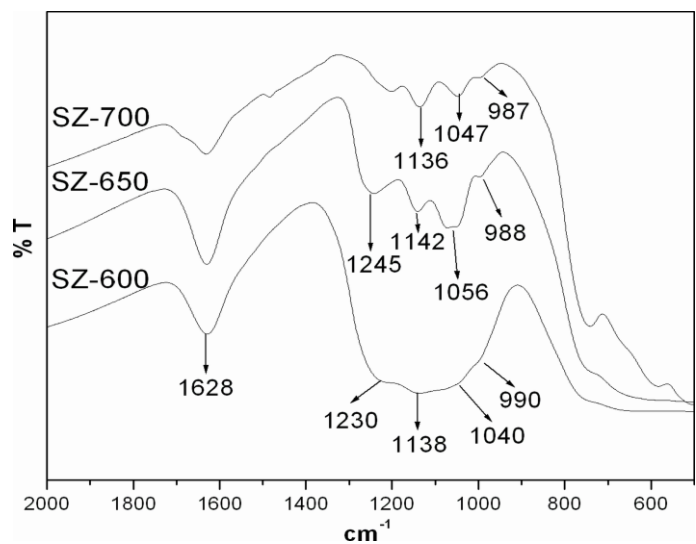


Figure S4: IR spectra of SZ catalysts calcined at different temperatures

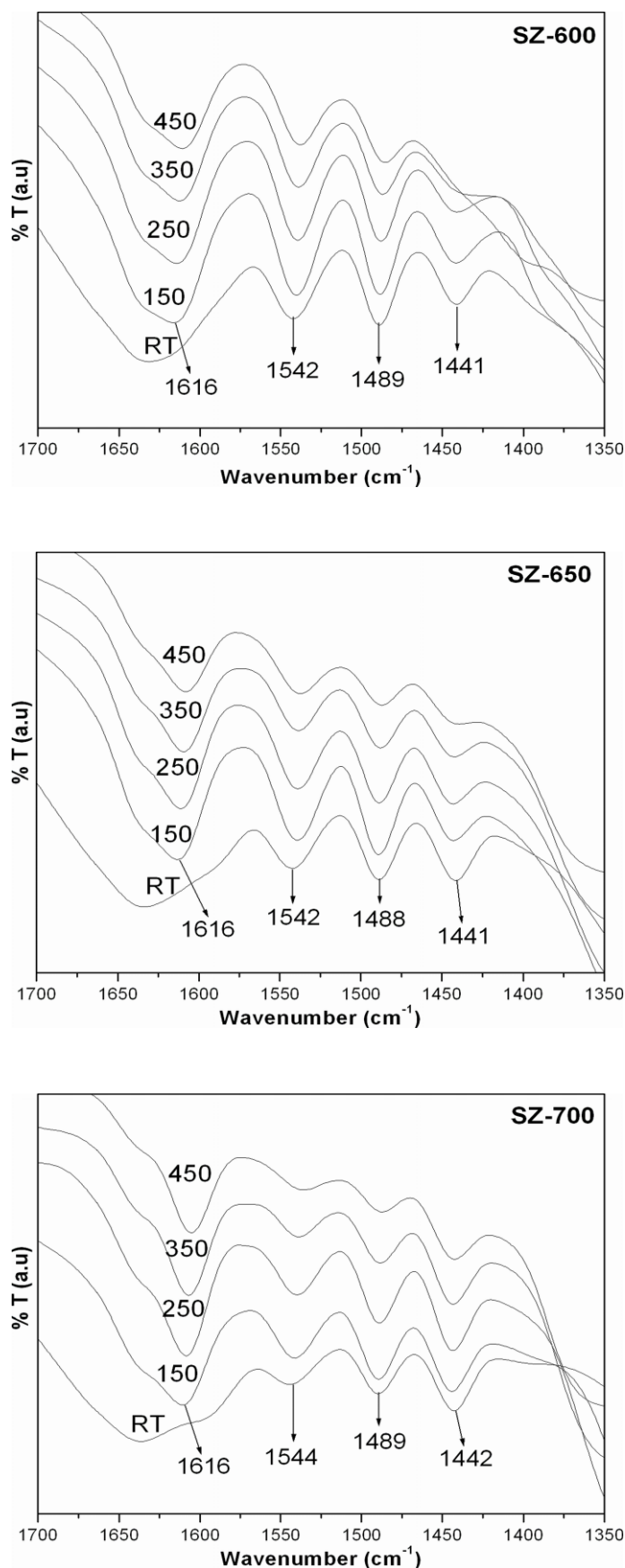


Figure S5: DRIFT spectra of SZ catalysts after pyridine desorption at different temperatures (from RT to 450°C)

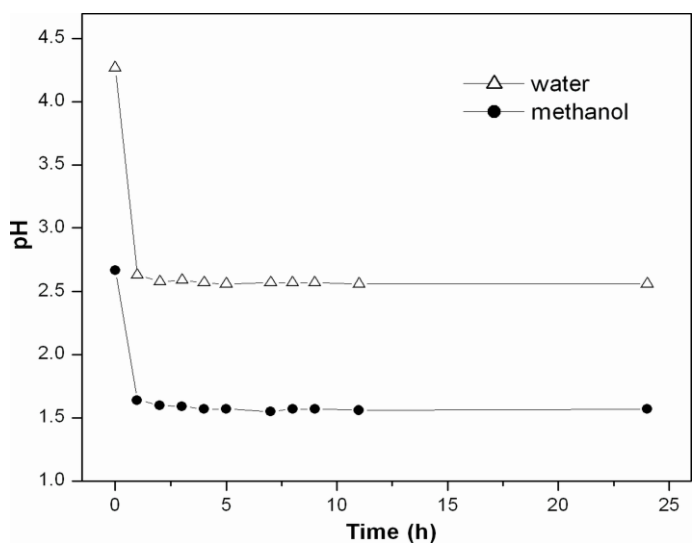


Figure S6: pH Vs. time after stirring SZ-600 catalyst in water and methanol

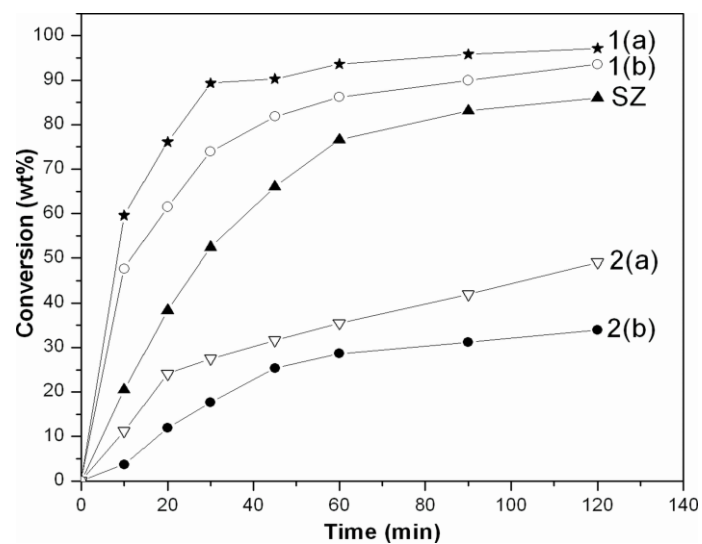


Figure S7: Esterification of myristic acid with liquid and solid acid catalysts.
Reaction conditions: acid: methanol=1:10; SZ = 0.5 wt%; 1 = H₂SO₄: (a) 0.5 wt% and (b) 0.25 wt%; 2 = Amberlyst-15: (a) 1 wt% and (b) 0.5 wt%