

Supporting Information

**Shape-Controlled Synthesis of Pt nanostructures and
Evaluation of Catalytic and Electrocatalytic Performance**

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Figure S1

TEM image of spherical Pt nanoparticles (A) and the corresponding histogram (B).

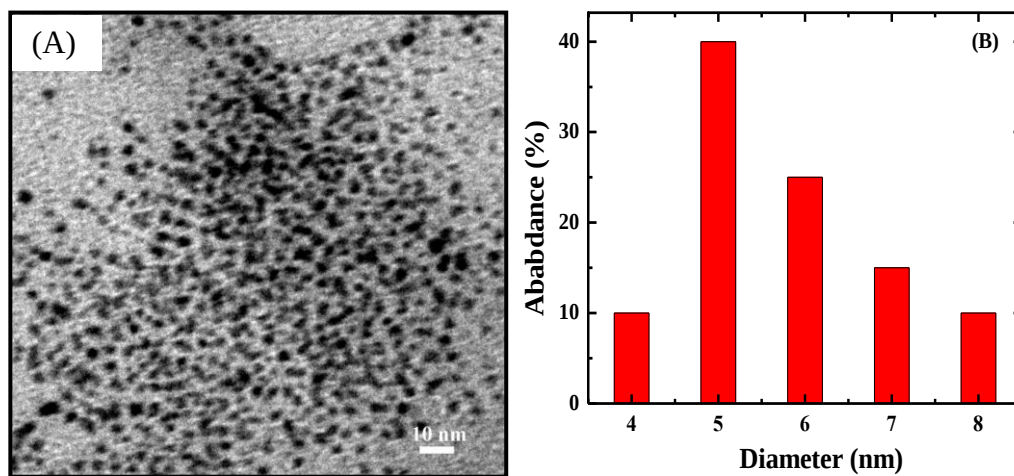


Figure S2

Magnified TEM images of peanut-like Pt nanostructures

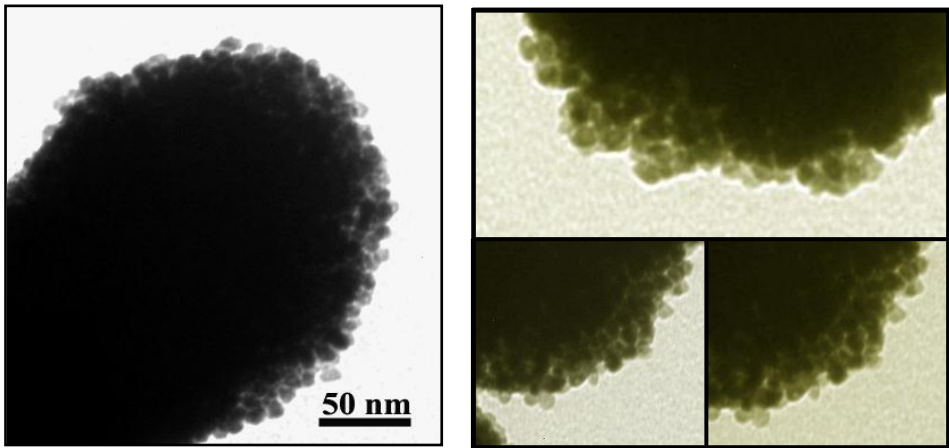


Figure S3

TEM image for the nanoparticles obtained at an intermediate stage (30 min) during the synthesis of dendritic Pt nanostructures.

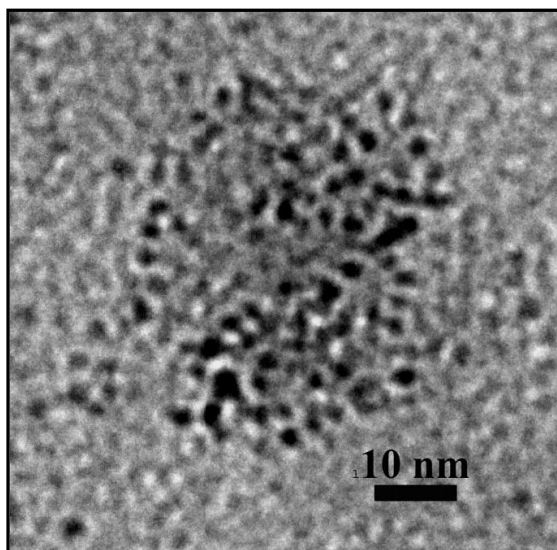


Figure S4

TEM images of MCNT-supported (A) dendritic and (B, C) peanut-like nanoparticles.

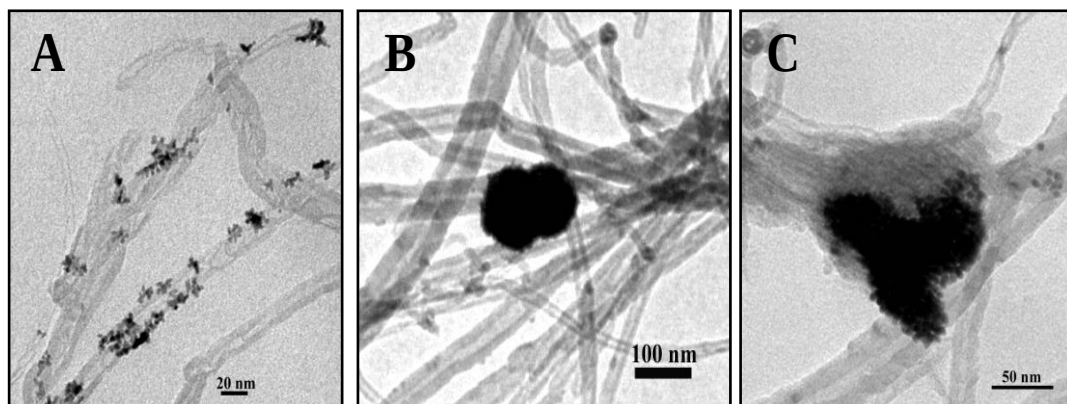


Figure S5

Voltammetric profile obtained for (a) dendritic and (b) peanut-like Pt nanoparticle modified electrodes in 0.5 M H₂SO₄. Scan rate: 100 mV/s.

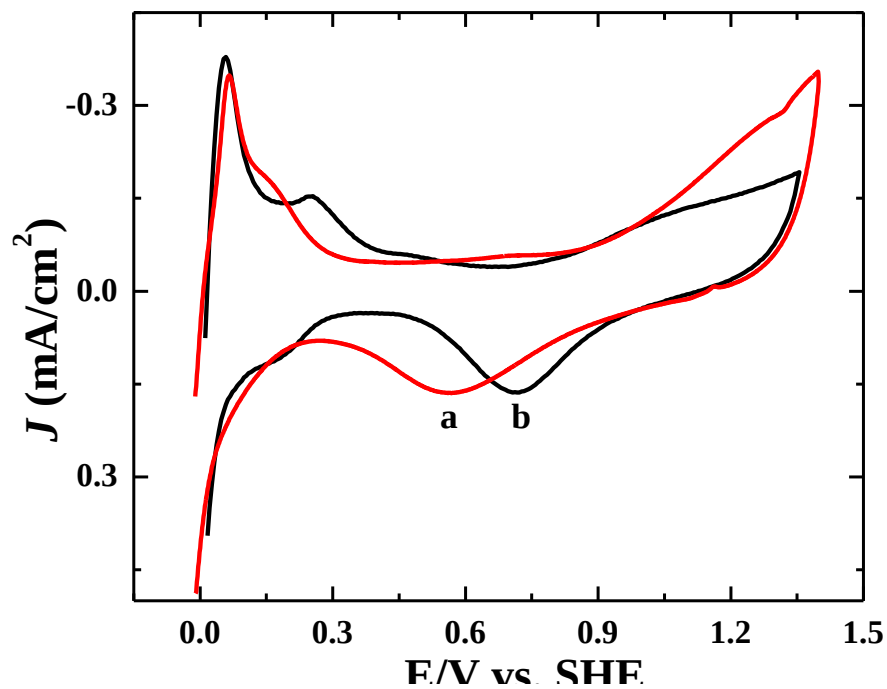


Figure S6

Plot illustrating the catalytic activity of (a) peanut-like, (b) dendritic and (c) spherical nanoparticles towards hydrogenation of unsaturated alcohols

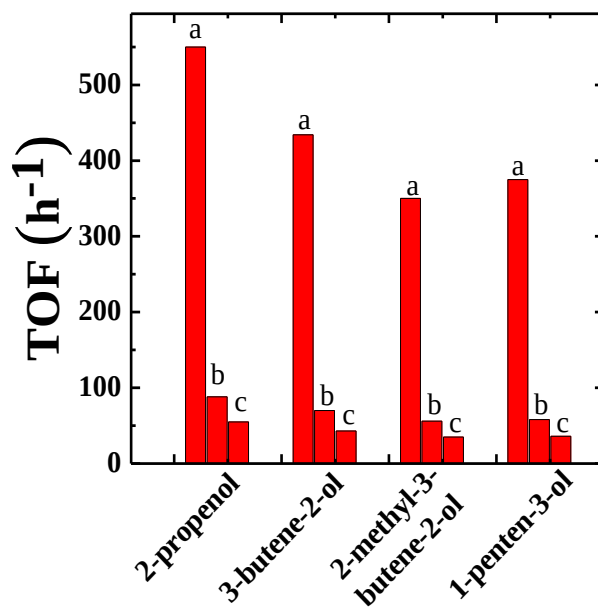
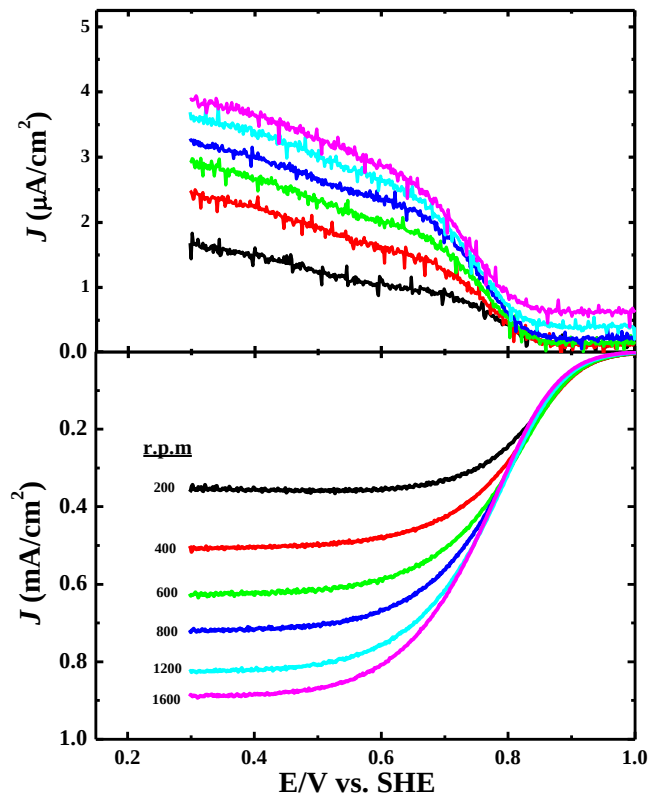


Figure S7

The figure represents the rotation-dependent polarization curves registered for ORR with dendritic Pt nanoelectrocatalyst-modified electrode in 0.5 M H₂SO₄. Scan rate: 5 mV/s. Ring potential: 0.85 V.



$$\frac{1}{J} = \frac{1}{J_d} + \frac{1}{J_k} \dots\dots\dots(1)$$

$$\frac{1}{J} = \frac{1}{0.62nFC_0D^{2/3}\nu^{-1/6}\omega^{1/2}} + \frac{1}{nFkC_0} \dots\dots\dots(2)$$

Polarization curves were analyzed by Koutecky-Levich equations (1) and (2), where J_d is the diffusion limited current density and J_k is the kinetic current density, respectively; ' ω ' is the angular velocity, ' n ' is the number of electrons transferred, ' F ' is Faraday's constant, ' C ' is the concentration of dissolved oxygen, ' D ' is the diffusion coefficient of

oxygen, ' ν ' is the kinematic viscosity of solution and ' k ' is the apparent electron transfer rate constant. The flow of current at the ring is negligible with respect to the current observed at the disk. For instance, the flow of ring-current at potential > 0.75 V was observed ≤ 0.5 μA and the ring current at < 0.7 V was $\sim 1\text{--}4$ μA . This is negligible with respect to the disk current (~ 1 mA) obtained. Thus it is considered that the 4 electron pathway is the main reaction on the nanoparticle-based electrodes.

Figure S8

Plot comparing the kinetic current density of the three catalysts at 0.9 V

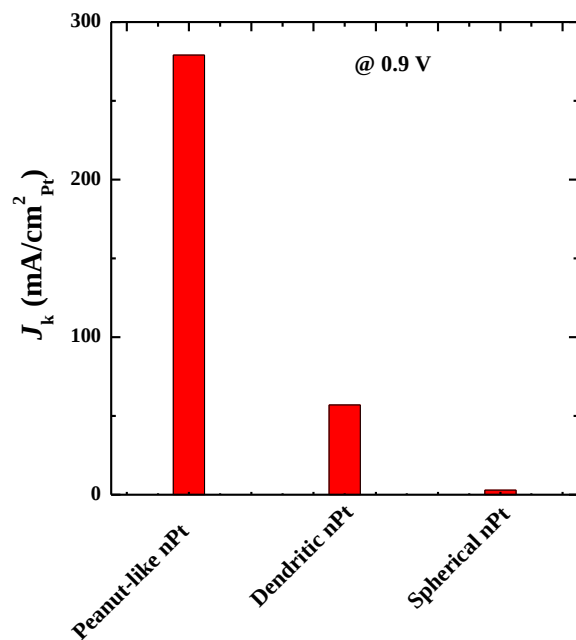


Figure S9

Polarization curve illustrating the effect of catalyst support in the ORR. The polarization curve for ORR with MCNT-supported (a) and unsupported (b) dendritic Pt nanostructure. The mass loading was kept constant in both the cases.

