Supporting Information

Dependence of cracking activity upon the Brønsted acidity of Y zeolite: DFT study and experimental confirmation

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Num.	System ^(a)	Cation Site	Chemical composition of unit cell
1	H⁺(1Al)		$H_1Al_1Si_{47}O_{96}$
2	3H*(3Al)		$H_3Al_3Si_{45}O_{96}$
3	H ⁺ Ba ²⁺ (3Al)	Ι	$Ba_1H_1Al_3Si_{45}O_{96}$
4		II	
5	H ⁺ Ca ²⁺ (3Al)	Ι	Ca1H1Al3Si45O96
6		II	
7	H ⁺ La(OH) ²⁺ (3Al)	Ι'	$La_1H_2Al_3Si_{45}O_{97}$
8	H ⁺ La ³⁺ (4Al)	Ι	La1H1Al4Si44O96
9		Ι'	
10	H ⁺ Al(OH) ²⁺ (3Al)	Ι'	H2Al4Si45O97
11		II'	

Table S1. Description of the eleven faujasite systems employed in this study. See also the corresponding figures below which include the protonated octane.

(a) This notation indicates the protons, the extraframework cations, and the number of framework Al.

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Figure S1. Deprotonated faujasite $H^+(1A1)$ (num. 1 in Table 3) with the octane carbocation.



Figure S2. Deprotonated faujasite $3H^+(3A1)$ (num. 2 in Table 3) with the octane carbocation.



Figure S3. Deprotonated faujasite $H^+Ba^{2+}(3Al)$ (num. 3 in Table 3) with the octane carbocation.

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Figure S4. Deprotonated faujasite $H^+Ba^{2+}(3A1)$ (num. 4 in Table 3) with the octane carbocation.



Figure S5. Deprotonated faujasite $H^+Ca^{2+}(3A1)$ (num. 5 in Table 3) with the octane carbocation.



Figure S6. Deprotonated faujasite $H^+Ca^{2+}(3A1)$ (num. 6 in Table 3) with the octane carbocation.



Figure S7. Deprotonated faujasite H⁺La(OH)²⁺(3Al) (num. 7 in Table 3) with octane carbocation.



Figure S8. Deprotonated faujasite $H^+La^{3+}(4A1)$ (num. 8 in Table 3) with the octane carbocation.

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Figure S9. Deprotonated faujasite $H^+La^{3+}(4Al)$ (num. 9 in Table 3) with the octane carbocation.



Figure S10. Deprotonated faujasite $H^+La^{3+}(4Al)$ (num. 10 in Table 3) with the octane carbocation.



Figure S11. Deprotonated faujasite $H^+Al(OH)^{2+}(3Al)$ (num. 11 in Table 3) with octane carbocation.