## **Electronic Supplementary Information**

## Donor-Acceptor Porphyrin Functionalized Pt Nano-Assemblies for Artificial Photosynthesis: A Simple and Efficient Homogeneous Photocatalytic Hydrogen Production System

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Fig. S1 FT-IR spectra of TPPAN and Pt-TPPAN nanocomposite.

## **FT-IR discussion:**

Fourier transform infrared (FT–IR) research on Pt–TPPAN reveals the interaction between the Pt nanoparticle and TPPAN in the nanohybrid. As shown in Fig. S1, the C=N stretching vibration of free TPPAN occurs at 1468 cm<sup>-1</sup>, while there is *ca*. 11 cm<sup>-1</sup> shift to low wavenumbers for the peak of C=N vibration of porphyrin in Pt–TPPAN (*ca*.1457 cm<sup>-1</sup>) and the intensity of this peak decreases conspicuously. The fact of the peak shift and intensity decrease for Pt–TPPAN can be ascribed to the interaction between Pt nanoparticle and the N atom of the porphyrin ring.<sup>S1-S3</sup> On the other hand, the peak of N–H at 966 cm<sup>-1</sup> disappear since the hydrogen atom in the N–H bonding is replaced by metallic Pt species.<sup>S4</sup> The results give a evidence that there is coordination bond interaction between the Pt nanocomposite and the nitrogen atoms of the porphyrin ring.

## **References:**

- S1 M. Zhu, M. Han, Y. Du, P. Yang and X. Wang, *Dyes Pigment.*, 2010, 86, 81–86.
- S2 M. Zhu, Y. Dong, Y. Du, Z. Mou, J. Liu, P. Yang and X. Wang, *Chem. Eur. J.*, 2012, 18, 4367–4374.
- S3 P. Yang, W. Zhang, Y. Du and X. Wang, J. Mol. Catal. A: Chem., 2006, 260, 4–10.
- S4 E. Sun, Y. Shi, P. Zhang, M. Zhou, Y. Zhang, X. Tang and T. Shi, J. Mol. Struct., 2008, 889, 28–34.



**Fig. S2** B3LYP/6–31G(\*) optimized structures of TPPAN with different model. a):

ball & stick model; b): space filling model.



Fig. S3 B3LYP/6-31G (\*) optimized geometry parameters. The distances are given in

Å.



Fig. S4 Spectra overlap of the fluorescence spectrum of 9-ClMA (black line) and the absorption spectrum of TPPAN (red line).



Fig. S5 Fluorescence spectrum of 9-ClMA excited at 420 nm (0.05 mM) at room temperature.



Fig. S6 Absorption spectrum (—) and corrected fluorescence excitation spectrum ( $\lambda_{em}$ = 656 nm) (- - -) of TPPAN in CH<sub>2</sub>Cl<sub>2</sub> solution.