

Supplemental Information

Selective oxidation of benzyl alcohol using *in situ* generated H₂O₂ over hierarchical Au-Pd titanium silicalite catalysts.

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Table S1 Low temperature benzyl alcohol oxidation in autoclave reactor over various catalysts and the product distribution.

Catalyst	Conversion (%)	Selectivity (%)					Benzaldehyde production (mol/h/Kg _{cat}) ¹
		benzene	toluene	benzaldehyde	benzoic acid	benzyl benzoate	
TS-1 (XG-REF)	1.0	0	0	100	0	0	5
TS-1 (LG-12%)	1.0	0	0	100	0	0	5
AX	0.1	0	0	100	0	0	1
A12	0	-	-	-	-	-	0
APX	0.2	0	0	100	0	0	1
AP12	0.2	0	0	100	0	0	1
PX	0.1	0	0	100	0	0	1
P12	0.1	0	0	100	0	0	1

Conditions; 0-2°C, methanol (5.6 g), water (2.9 g), 2.3 mmol benzyl alcohol (0.25 g), 30 minutes, catalyst (10 mg), 25% O₂/CO₂ (10.4 bar, 150 psi). ¹ ±0.5 mol/h/Kg_{cat}.

Table S2 H₂O₂ productivity at 2°C of various catalysts.

Catalyst	Productivity (mol _{H₂O₂} /h/Kg _{cat}) ¹	H ₂ O ₂ (wt.%)
TS-1 (XG-REF)	0	0
TS-1 (LG-12%)	0	0
AX	0	0
A12	0	0
APX	34	0.07
AP12	25	0.05
PX	6	0.01
P12	15	0.03

Conditions; 0-2°C, methanol (5.6 g), water (2.9 g), 30 minutes, catalyst (10 mg), 5 % H₂/CO₂ (28.96 bar, 420 psi), 25%O₂/CO₂ (10.4 bar, 150 psi). ¹ ±0.5 mol/h/Kg_{cat}.

Table S3 H₂O₂ productivity at 30°C of various catalysts.

Catalyst	Productivity (mol _{H₂O₂} /h/Kg _{cat}) ¹	H ₂ O ₂ (wt.%)
TS-1 (XG-REF)	0	0
TS-1 (LG-12%)	0	0
AX	0	0
A12	0	0
APX	124	0.25
AP5	96	0.19
AP8	84	0.17
AP12	95	0.19
DAPX	12	0.02
DAP12	40	0.08
PX	16	0.03
P12	77	0.15

Conditions; 30°C, methanol (5.6 g), water (2.9 g), 30 minutes, catalyst (10 mg), 5 % H₂/CO₂ (28.96 bar, 420 psi), 25%O₂/CO₂ (10.4 bar, 150 psi). ¹±0.5 mol/h/Kg_{cat}.

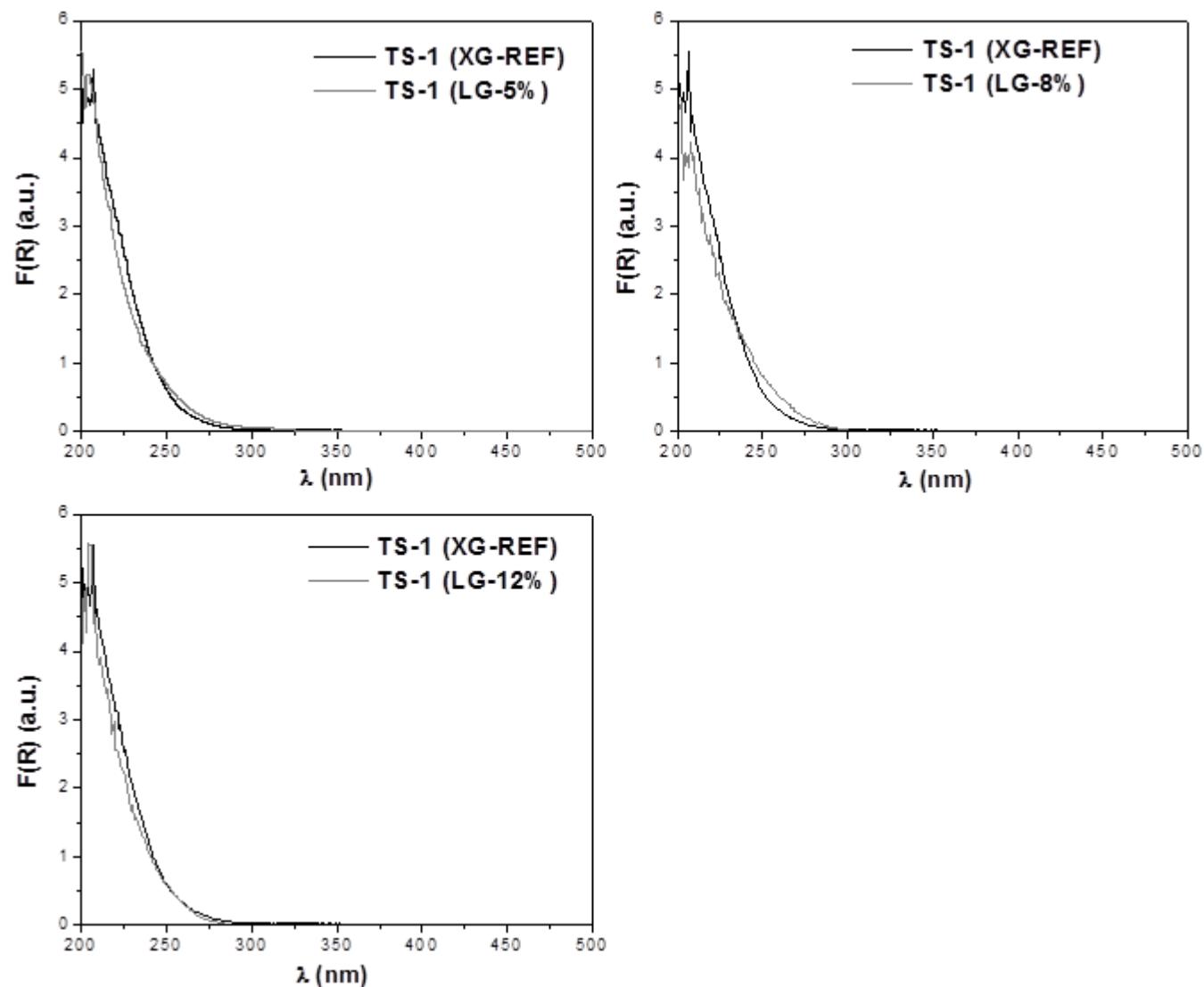


Fig. S1 Comparison DR UV/Vis spectroscopy of hierarchical TS-1 samples and TS-1 reference zeolite (calcined).