

*Supplementary materials*

**Fischer-Tropsch Synthesis on cobalt/Al<sub>2</sub>O<sub>3</sub>-modified SiC catalysts: effect of cobalt-alumina interactions**

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Supplementary **Table S1** shows the concentration of SiC support itself with the main elements of Si, C, and O, and it found to be formed with SiC species coated on SiOxCy layers with the surface area of ~ 25.2 m<sup>2</sup>/g, which are well matched with the results of SICAT reports in their homepage (<http://www.sicatcatalyst.com/>). The Fourier-transformed infrared-spectroscopy (FT-IR) in supplementary **Figure S1** showed the characteristic vibration peaks of Si-O, Si-C and Si-O-Si species are assigned to the wavenumbers of 697, 845 and 1100 cm<sup>-1</sup>, respectively,<sup>14</sup> and it supports that the structures of beta-SiC can be in the phases of SiC species coated on SiOxCy layers as well.

Supplementary **Figure S2** shows the profiles of NH<sub>3</sub>-TPD patterns of Al<sub>2</sub>O<sub>3</sub>-modified SiC supports after 500°C calcination without cobalt component, which are assigned to AS (in main manuscript) and Al<sub>2</sub>O<sub>3</sub>(x)/SiC (in supplementary data) with the digit of x for Al<sub>2</sub>O<sub>3</sub>wt% on beta-SiC support. The number of acid sites is only assigned to the integrated area (corresponding to the desorbed mmolNH<sub>3</sub>/g) of the peak below 300°C, which can be assigned to the weak acid sites, and the higher desorption temperature above 300°C seems to be the possible desorption of surface hydroxyl groups or strongly adsorbed H<sub>2</sub>O molecules. The number of acid sites is also summarized in supplementary **Table S2** with the amounts of

weak acid sites (mmolNH<sub>3</sub>/g) below the desorption temperature of 300°C. The strong acid sites assigned to the desorption peak above 300°C can be attributed to the desorption of water molecules, which is strongly adsorbed on the surfaces or it is desorbed from Al<sub>2</sub>O<sub>3</sub> structures at high temperature.

Supplementary **Table S1**. Results of XRF analysis of beta-SiC support itself

Formula	Concentration (wt%)
Si	24.79
C	32.99
O	41.95
Fe	0.17
Al	0.07
Ca	0.03

Supplementary **Table S2**. Summarized results of NH<sub>3</sub>-TPD of Al<sub>2</sub>O<sub>3</sub>-modified SiC support

Support	Amount of acidic sites (adsorbed NH <sub>3</sub> ; cm <sup>3</sup> /g)		
	Weak (<300°C)	Strong (>300°C)	Total
Al <sub>2</sub> O <sub>3</sub> (2.5)/SiC	0.325	2.006	2.331
Al <sub>2</sub> O <sub>3</sub> (5)/SiC	0.787	1.360	2.147
Al <sub>2</sub> O <sub>3</sub> (10)/SiC	0.813	2.245	3.058
Al <sub>2</sub> O <sub>3</sub> (15)/SiC	0.621	0.725	1.346

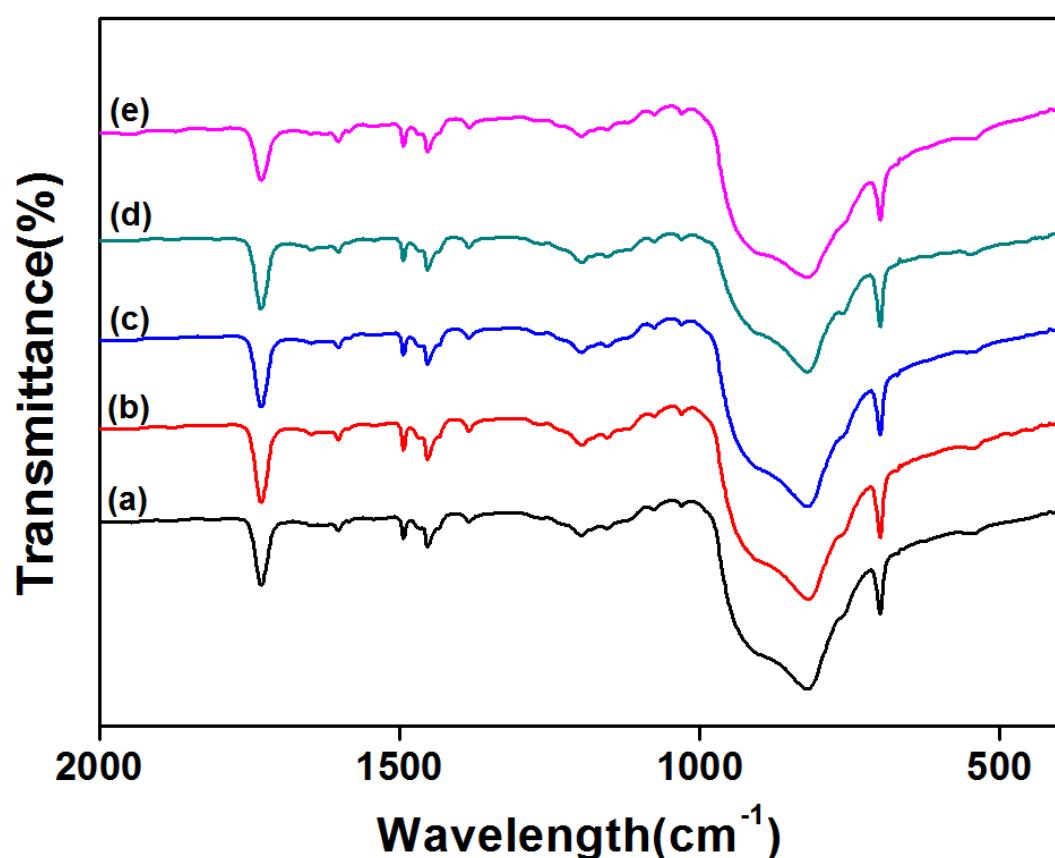


Figure S1. FT-IR spectra of Al<sub>2</sub>O<sub>3</sub>-modified SiC supports

(a) Beta-SiC, (b) Al<sub>2</sub>O<sub>3</sub>(2.5)/SiC, (c) Al<sub>2</sub>O<sub>3</sub>(5)/SiC, (d) Al<sub>2</sub>O<sub>3</sub>(10)/SiC, (e) Al<sub>2</sub>O<sub>3</sub>(15)/SiC

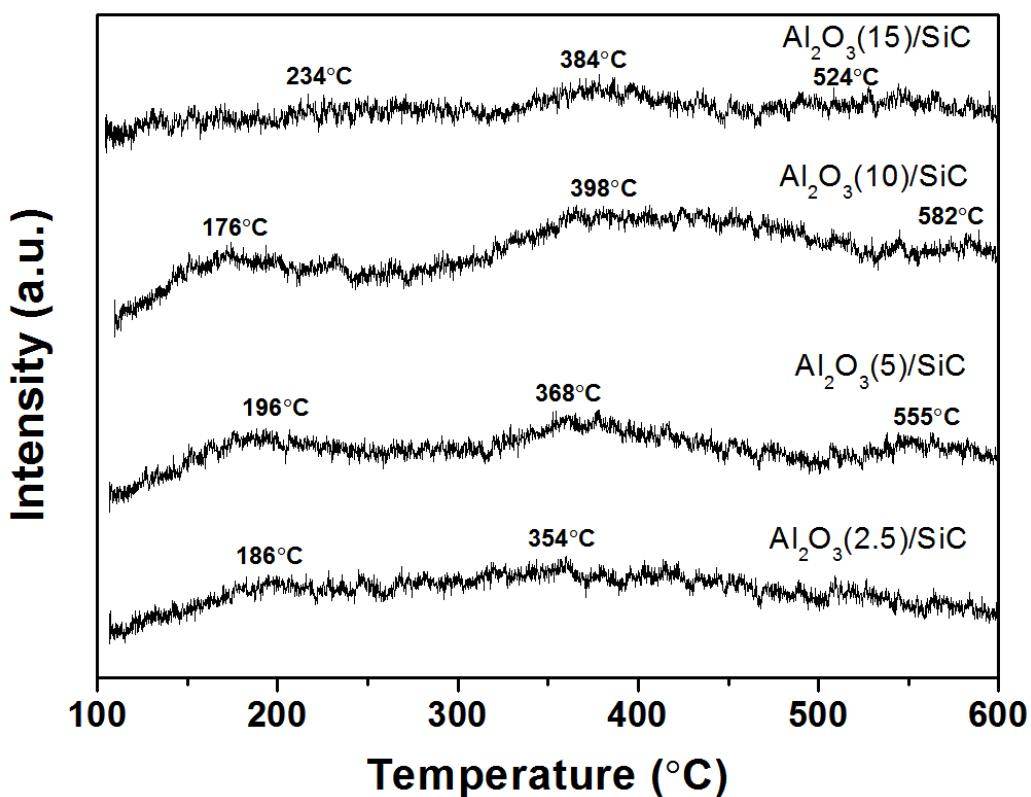


Figure S2. Profiles of  $\text{NH}_3$ -TPD on  $\text{Al}_2\text{O}_3$ -modified SiC supports