Supporting Information

Heterolytic cleavage of Si–H bonds: Reduction of imines using silane/high-valent oxo-molybdenum MoO₂Cl₂ as a catalyst

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- 1. Complete reference of 9.
- 2. Table S1 The relative free energies of the key transition states obtained at M06(6-311++G(d,p)+ LANL2DZ), M06(6-311++G(d,p)+ SDD) and M06(6-311++G(d,p)+ QVZP) and MP2(6-311G(d,p)+ LANL2DZ) calculation level for MoO_2Cl_2 mediated the hydrosilylation reaction are shown.
- 3. Table S2 The relative free energies of the key transition states obtained at B3LYP-D(6-311++G(2d,p)+ QVZP), M06(6-311++G(2d,p)+ QVZP) calculation level for MoO₂Cl₂ mediated the hydrosilylation reaction are shown.
- 4. Table S3 The relative free energies with different solvation, $(CH_2Cl_2, THF and Toluene)$ of the key transition states and the intermediates obtained at M06(6-311++G(2d,p)+ QVZP) calculation level for MoO₂Cl₂ mediated the hydrosilylation reaction are shown.
- 5. Scheme S1 The energy profile for MoO₂Cl₂ catalyzing the hydrosilylation of imines process through the [2+2] addition pathway. The solvation energies are included in parentheses, and the gas-phase energies are the values without parentheses.
- 6. Figure S1. The calculated geometric structures of the transition states along the [2+2] addition pathway.
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- 9. Figure S4 The detailed scan plot for the ionic hydrosilylation pathway from the adduct **3+im**.
- 10. Figure S5 The IRC plot for the ionic hydrosilylation pathway from the transition state **TS4anti**.
- 11. Table S4 Cartesian coordinates for all optimized structures in XYZ format.

Complete reference of 9

Frisch, M. J. et al. *Gaussian 09*; Gaussian, Inc. Frisch, M. J.; Trucks, G. W.; Schlegel, H. B.; Scuseria, G. E.; Robb, M. A.; Cheeseman, J. R.; Montgomery, Jr., J. A.; Vreven, T.; Kudin, K. N.; Burant, J. C.; Millam, J. M.; Iyengar, S. S.; Tomasi, J.; Barone, V.; Mennucci, B.; Cossi, M.; Scalmani, G.; Rega, N.; Petersson, G. A.; Nakatsuji, H.; Hada, M.; Ehara, M.; Toyota, K.; Fukuda, R.; Hasegawa, J.; Ishida, M.; Nakajima, T.; Honda, Y.; Kitao, O.; Nakai, H.; Klene, M.; Li, X.; Knox, J. E.; Hratchian, H. P.; Cross, J. B.; Bakken, V.; Adamo, C.; Jaramillo, J.; Gomperts, R.; Stratmann, R. E.; Yazyev, O.; Austin, A. J.; Cammi, R.; Pomelli, C.; Ochterski, J. W.; Ayala, P. Y.; Morokuma, K.; Voth, G. A.; Salvador, P.; Dannenberg, J. J.; Zakrzewski, V. G.; Dapprich, S.; Daniels, A. D.; Strain, M. C.; Farkas, O.; Malick, D. K.; Rabuck, A. D.; Raghavachari, K.; Foresman, J. B.; Ortiz, J. V.; Cui, Q.; Baboul, A. G.; Clifford, S.; Cioslowski, J.; Stefanov, B. B.; Liu, G.; Liashenko, A.; Piskorz, P.; Komaromi, I.; Martin, R. L.; Fox, D. J.; Keith, T.; Al-Laham, M. A.; Peng, C. Y.; Nanayakkara, A.; Challacombe, M.; Gill, P. M. W.; Johnson, B.; Chen, W.; Wong, M. W.; Gonzalez, C.; and Pople, J. A.; Gaussian, Inc., Pittsburgh PA,, 2009. **Table S1** The relative free energies of the intermediates and transition states obtained at M06(6-311++G(d,p)+ LANL2DZ), M06(6-311++G(d,p)+ SDD) and M06(6-311++G(d,p)+ QVZP) calculation level for MoO_2Cl_2 mediated the hydrosilylation reaction are shown. The solvation energies are included in parentheses, and the gas-phase energies are the values without parentheses.

	M06 6-311+++G(d,p) + LANL2DZ)	M06 (6-311+++G(d,p)+ SDD	M06 (6-311++G(2d,p) + QVZP)	MP2 (6-311G(d,p)+ LANL2DZ
TS[2+2]	28.7(26.4)	29.1	30.9(29.0)	38.8 (35.9)
TS4anti	18.1(14.4)	20.4	25.4(21.4)	20.0(19.7)

These results have indicated that the ionic hydrosilylation pathway passing through **TS4anti** is the most favorable pathway at all level of calculation.

Table S2 The relative free energies of the key transition states obtained at B3LYP-D(6-311++G(2d,p)+ QVZP), M06(6-311++G(2d,p)+ QVZP) calculation level for MoO_2Cl_2 mediated the hydrosilylation reaction are shown.

	B3LYP	M06	B3LYP-sol	M06-sol
	(6-311++G(2d,p) + QVZP)	(6-311++G(2d,p) + QVZP)	(6-311++G(2d,p) + QVZP)	(6-311++G(2d,p) + QVZP)
3+im	11.1	14.3	13.0	17.2
TS4anti	25.7	25.4	17.0	21.4
4anti	25.4	25.3	13.4	16.5

Table S3 The relative free energies with different solvation, $(CH_2Cl_2, THF and Toluene)$ of the key transition states and the intermediates obtained at M06(6-311++G(2d,p)+ QVZP) calculation level for MoO₂Cl₂ mediated the hydrosilylation reaction are shown.

	M06-gas (6-311++G(2d,p) + QVZP)	M06- CH ₂ Cl ₂ (6-311++G(2d,p) + QVZP)	M06-THF (6-311++G(2d,p) + QVZP)	M06-Toluene (6-311++G(2d,p) + QVZP)
3+im	14.3	17.2	16.4	20.0
TS4anti	25.4	21.4	21.9	27.9
4anti	25.3	16.5	17.8	25.9
5anti	28.6	12.0	12.8	23.7
TS5syn	33.6	30.7	32.0	37.7
5syn	15.9	7.5	8.4	16.8
6syn	16.6	7.8	8.4	17.0
TS7	19.6	12.5	13.4	21.2



Scheme S1 The energy profile for MoO_2Cl_2 catalyzing the hydrosilylation of imines process through the [2+2] addition pathway. The solvation energies are included in parentheses, and the gas-phase energies are the values without parentheses.

Figure S1. The calculated geometric structures of the transition states along the [2+2] addition pathway.



Figure S2 The calculated geometric structures of the transition states for the transformation between the ion pair of **5anti**, **5syn** and **6syn**. The solvation energies are included in parentheses, and the gas-phase energies are the values without parentheses.

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Figure S3 The calculated geometric structures of other two isomer of **TS4syn**. The solvation energies are included in parentheses, and the gas-phase energies are the values without parentheses.

Figure S4 The detailed scan plot for the ionic hydrosilylation pathway from the adduct **3+im**. (The first maximum point corresponds to the transition state TS4anti, the first minimum point corresponds to the first intermediate 4anti. The maximum point corresponds to the second transition state 5anti, and then the laste point corresponds to the syn intermediate 5syn)



Figure S5 The IRC plot for the ionic hydrosilylation pathway from the transition state **TS4anti**. (From the last point of the IRC calculation (both forward and reverse calculations), we optimization the structures, which leads to the adduct of 3+im, and 4anti, respectively.



Table S4 Cartesian coordinates for each structure calculated. (XYZ format).

1				Н	-3.600965	-1.708118	-3.510147
Мо	0.109824	-0.384989	0.246467	Н	-2.569668	2.396042	-2.401461
0	-0.062374	0.719808	1.475190	Н	-3.335423	1.611702	-3.793056
Õ	1 662329	-0 219766	-0 321355	Н	-4.098122	1.525656	-2.191283
CĬ	-1 365225	0.078476	-1 412072	Н	-5.080090	0.330534	-4.832612
Cl	-0.190556	-2 482006	1.054927	Н	-4.414982	1.194350	-6.209382
CI	-0.170550	-2.402000	1.054727	Н	-5 450984	-0 227246	-6 484346
TC[2]	21			Н	-1 424796	-2 458649	-7 184201
15[2+	-2] 1 1 492 40	1 255720	0 255704	н	-1.424790	-2.456045	-5 442128
0	1.148240	-1.255/20	-0.355/94	11 U	2 2 1 0 9 0 9	2.030247	-5.442128
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и Ц	2.974203	1 420082	1 62/2134	Č	-6 646747	-0 550896	1 217805
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Ĉ	-9 550020	3 759319	-2 215815	н	-1 977309	0 713482	4 048260
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С	-3.511677	0.528540	0.216556	Н	-2.003347	5.944258	5.523083
č	-2 403825	1 415548	-2 802847	н	-2 462454	4 369054	3 711/1/
CI	-2.403023	2 2 (2 2 7 (-2.002047		-2.402434	2.075(00	1 2710(0
CI	-0.42/048	-2.2033/0	-0.43//30	C	-0.890248	5.8/5090	1.2/1809
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С	-6.076145	0.221118	-2.867818	Н	-1.851498	4.303127	0.967205
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Si	-3.373742	-0.227181	-3.130062	Н	-4.455626	-1.922003	-4.558032
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