

Supporting Information

Tuning the selectivity of supported gold catalyst in solvent- and radical initiator-free aerobic oxidation of cyclohexene

Daniil S. Ovoshchnikov,^{†a} Baira G. Donoeva,^{†ab} Bryce E. Williamson^a and Vladimir B. Golovko^{*ab}

^a Department of Chemistry, University of Canterbury, Christchurch 8140, New Zealand. +64 3 364 2110; Tel: +64 3 364 2442; E-mail: vladimir.golovko@canterbury.ac.nz

^b The MacDiarmid Institute for Advanced Materials and Nanotechnology.

1. Characterization of [Au₉(PPh₃)₈](NO₃)₃ and Au₁₀₁(PPh₃)₂₁Cl₅.

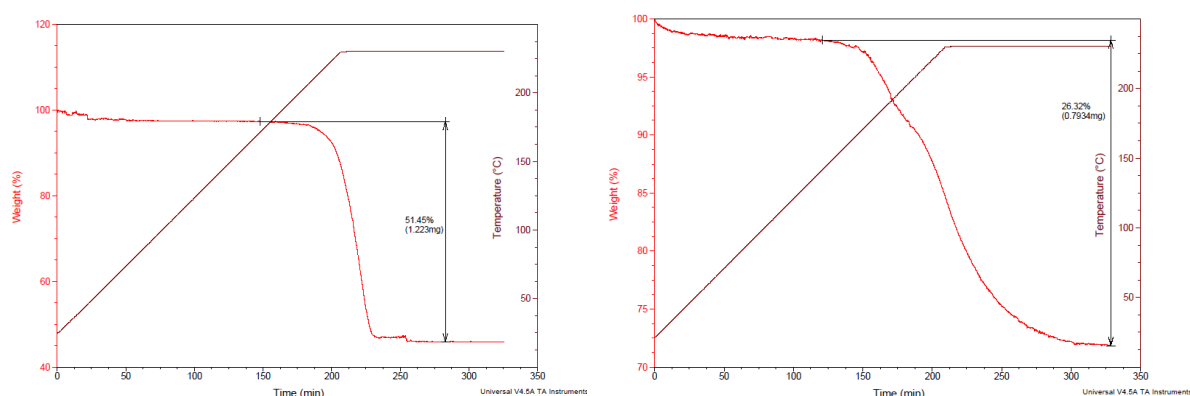


Fig. S1 TGA of [Au₉(PPh₃)₈](NO₃)₃ (left) and Au₁₀₁(PPh₃)₂₁Cl₅ (right).

2. Characterization of MIL-101.

MIL-101 was synthesized following the procedure described by Hatton et. al.^[1] X-ray powder diffraction was performed on Philips PW 1700 automated diffractometer with CoK α radiation and graphite monochromator. Surface area measurements were performed using Micromeritics ASAP 2010. BET surface area of the sample was found to be 3265 m²/g, Langmuir Surface area – 4050 m²/g.

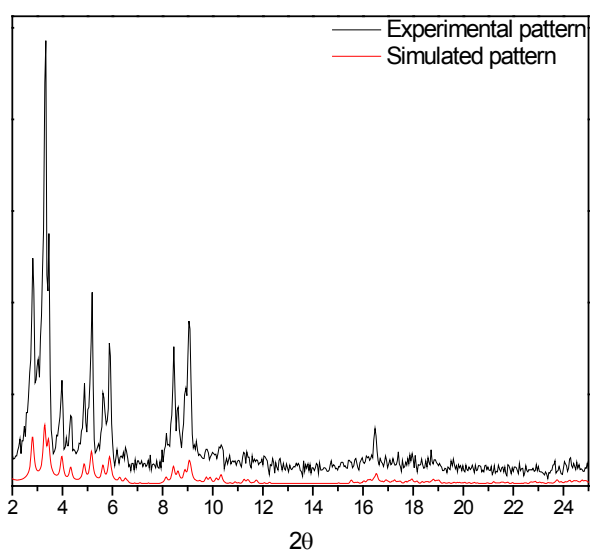


Fig. S2 Powder X-Ray diffraction pattern of MIL-101: experimental (black) and simulated (red). Patterns are re-plotted for CuK α radiation source.

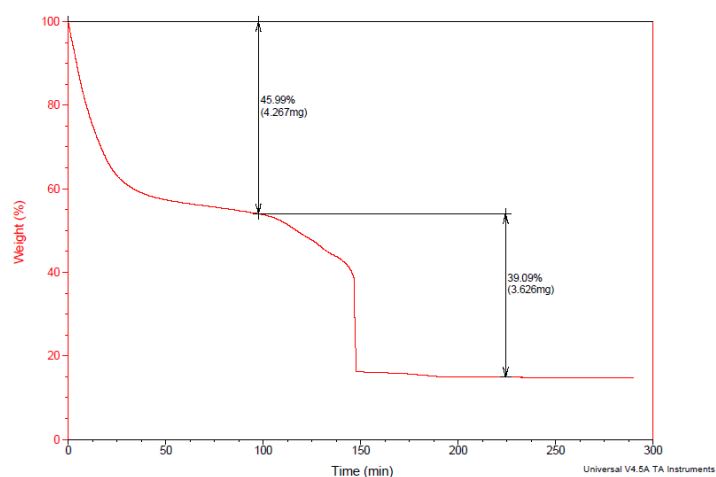


Fig. S3 TGA of MIL-101.

- [1] L. Bromberg, Y. Diao, H. Wu, S. A. Speakman, T. A. Hatton, *Chem. Mater.* **2012**, 24, 1664-1675.