

High CO₂ and CO conversion to hydrocarbons using bridged Fe nanoparticles on carbon nanotubes

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Supporting Information

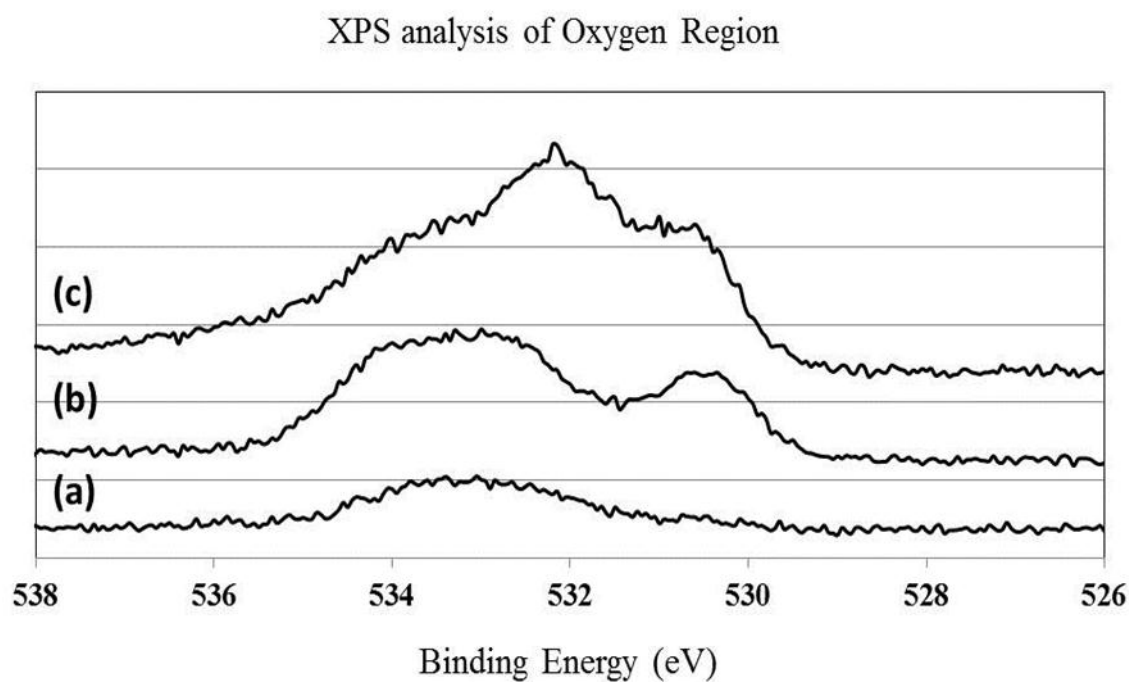


Figure S1 X-ray photoelectron spectroscopy of oxygen region of (a) as produced Fe@CNT, (b) oxidised (570 °C in air) CNTs and (c) reduced Fe@CNTs (50 sccm H₂ at 400 °C)

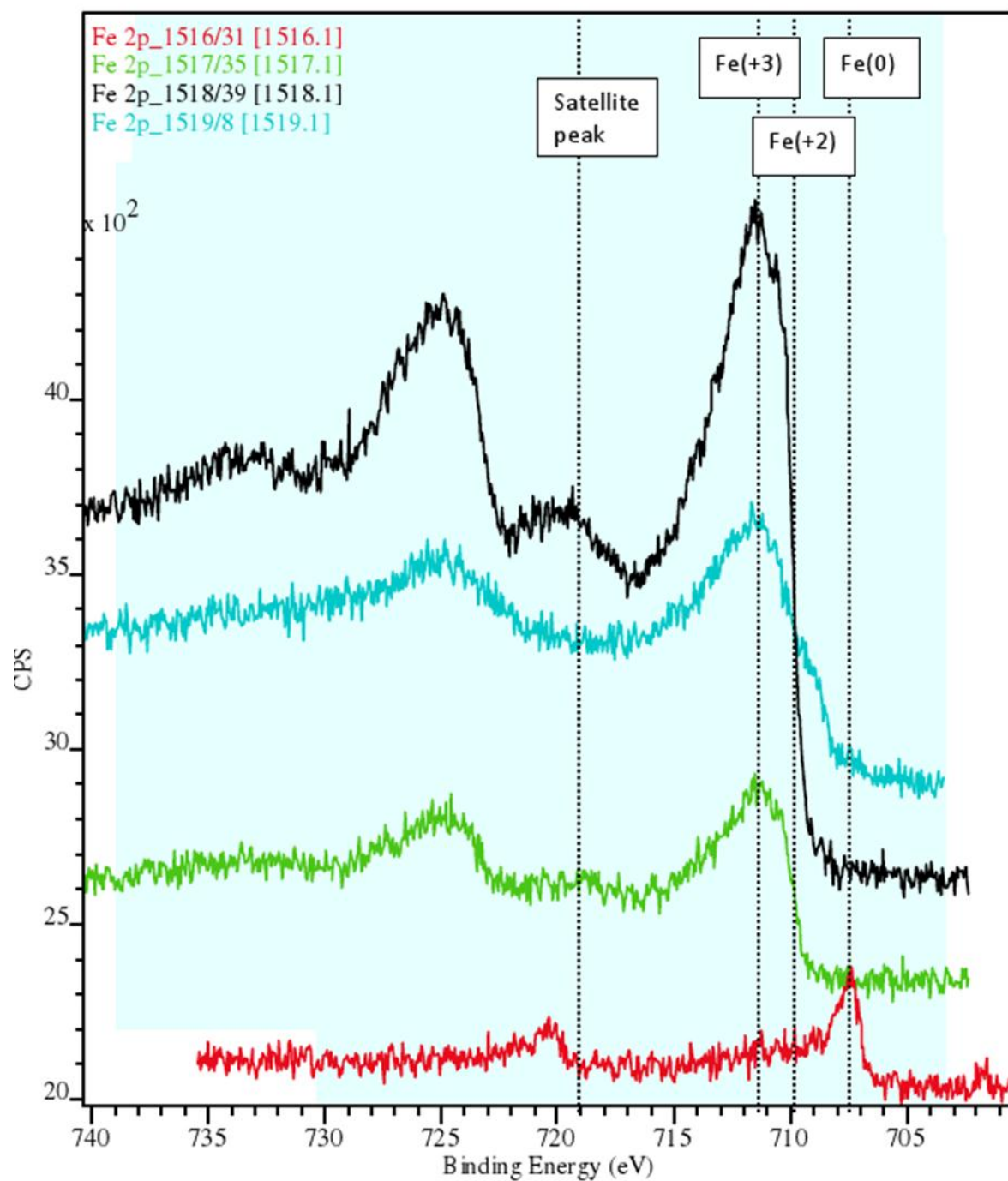


Figure S2 X-ray photoelectron spectroscopy of Fe regions

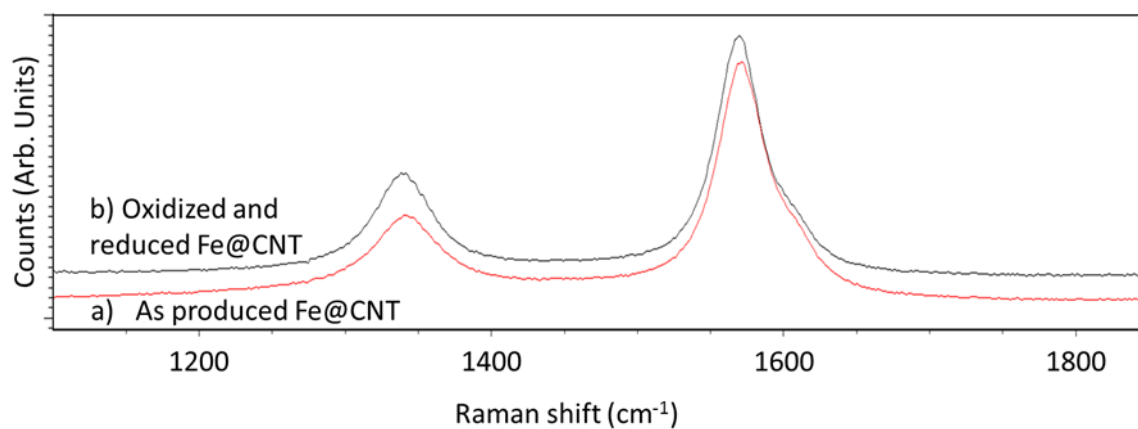


Figure S3 Raman spectra showing (a) as produced Fe@CNT catalyst and (b) Oxidized and reduced Fe@CNT

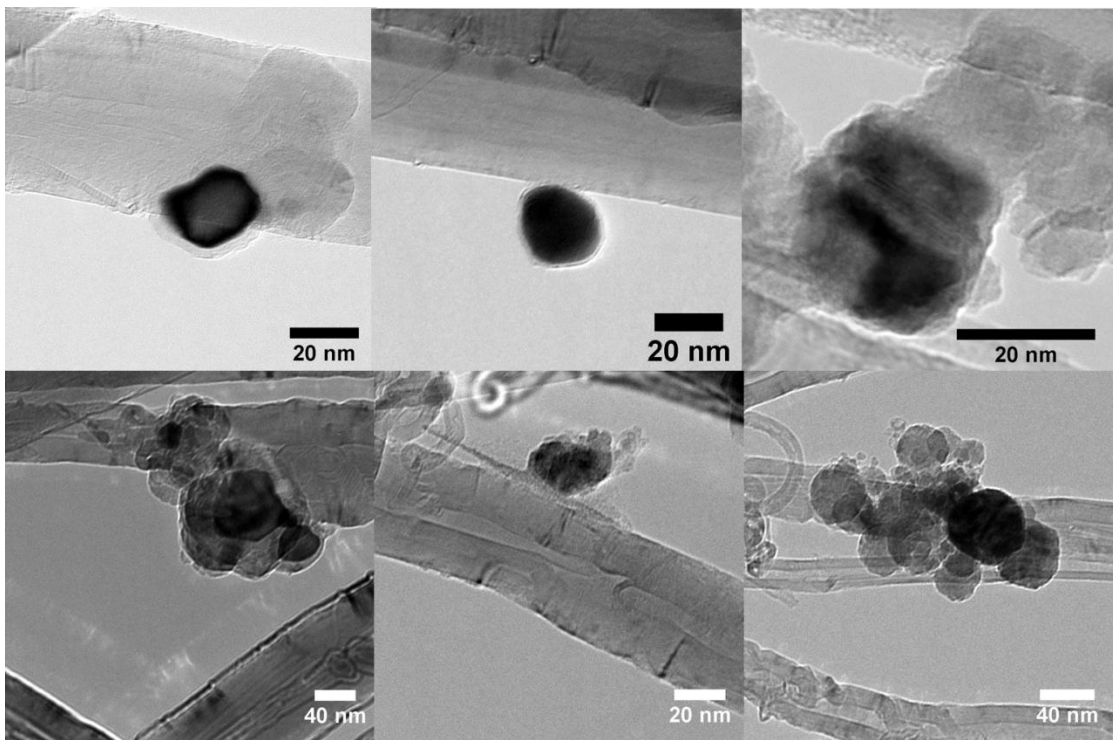


Figure S4 TEM micrographs of Fe decorated CNTs

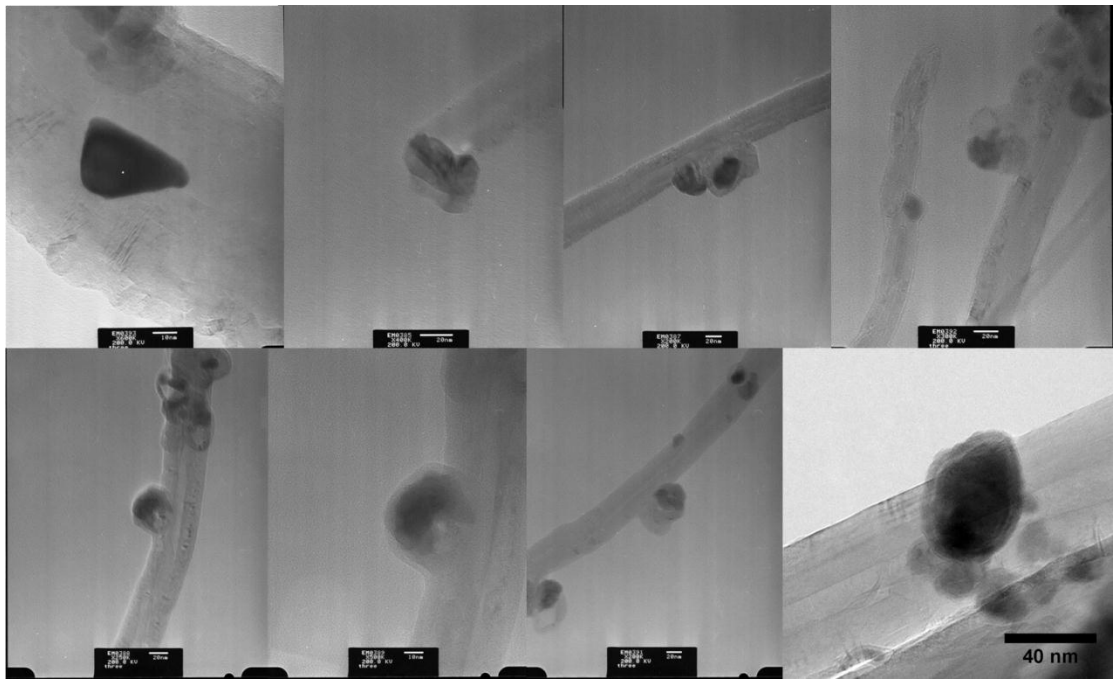


Figure S5 TEM micrographs of Fe@CNTs

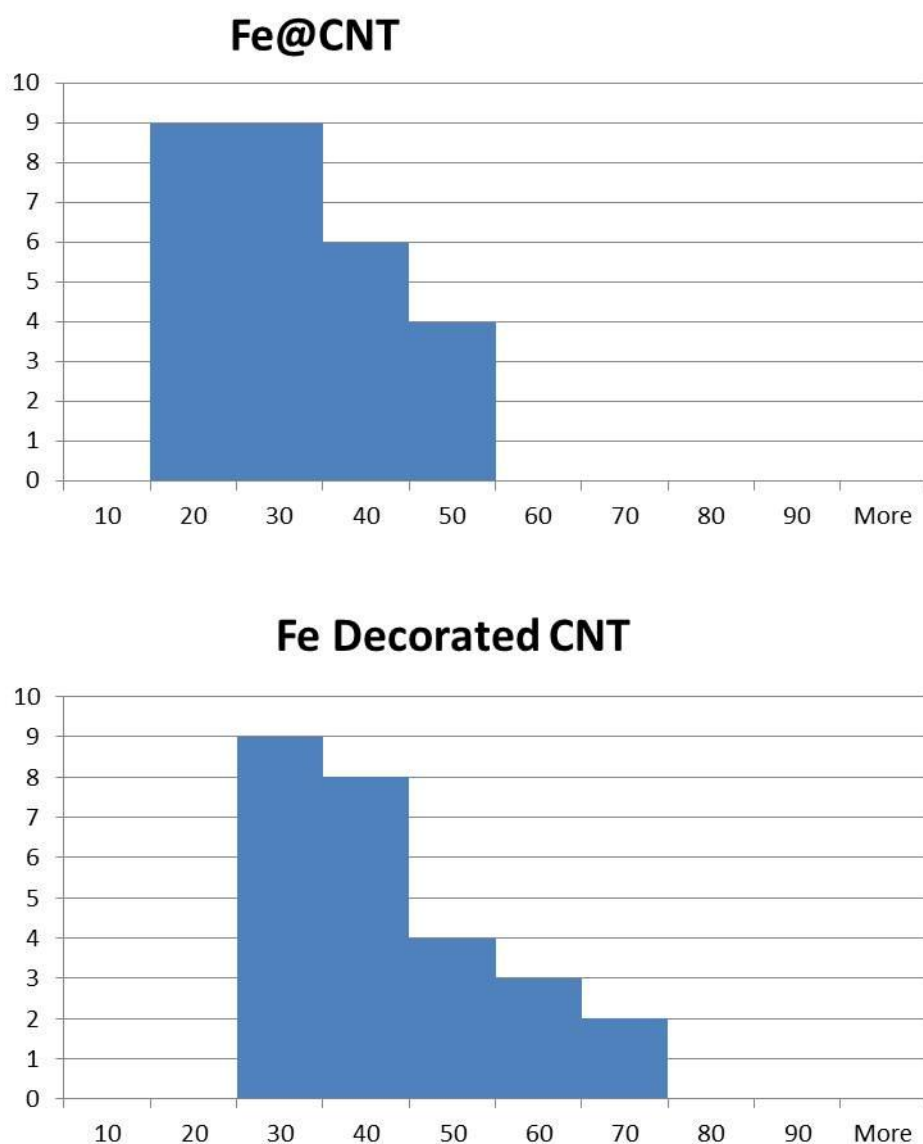


Figure S6 Histograms showing size distribution of iron nanoparticles of each catalyst

Equation 1 $FTY_{CO} = \text{mol CO converted to hydrocarbon} / \text{grams iron second}$

Equation 2 $FTY_{CO_2} = \text{mol CO}_2 \text{ converted to product} / \text{grams iron second}$