

Supporting Information

Crystal Phase Effects on the Structure and Performance of Ruthenium Nanoparticles for the CO₂ Hydrogenation

Qingquan Lin,^{a,b,c} Xiao Yan Liu,^a Ying Jiang,^d Yong Wang,^d Yanqiang Huang*^a and Tao Zhang^a

^a State Key Laboratory of Catalysis, Dalian Institute of Chemical Physics,
Chinese Academy of Sciences, Dalian 116023, China.

^b State Key Laboratory of Fine Chemicals, School of Chemical engineering,
Dalian University of Technology, Dalian 116024, China.

^c Graduate University of Chinese Academy of Sciences, Beijing 10049, China.

^d Center of Electron Microscopy and State Key Laboratory of Silicon Materials,
Department of Materials Science and Engineering,
Zhejiang University, Hangzhou 310027, China.

† *Electronic Supplementary Information (ESI) available: the characterization data of XRD, Raman, HAADF-STEM, HRTEM and H₂-TPR results. See DOI: 10.1039/b000000x/*

* *Corresponding author: (+) 86 411 84379416, E-mail address: yqhuang@dicp.ac.cn*

Supporting figures

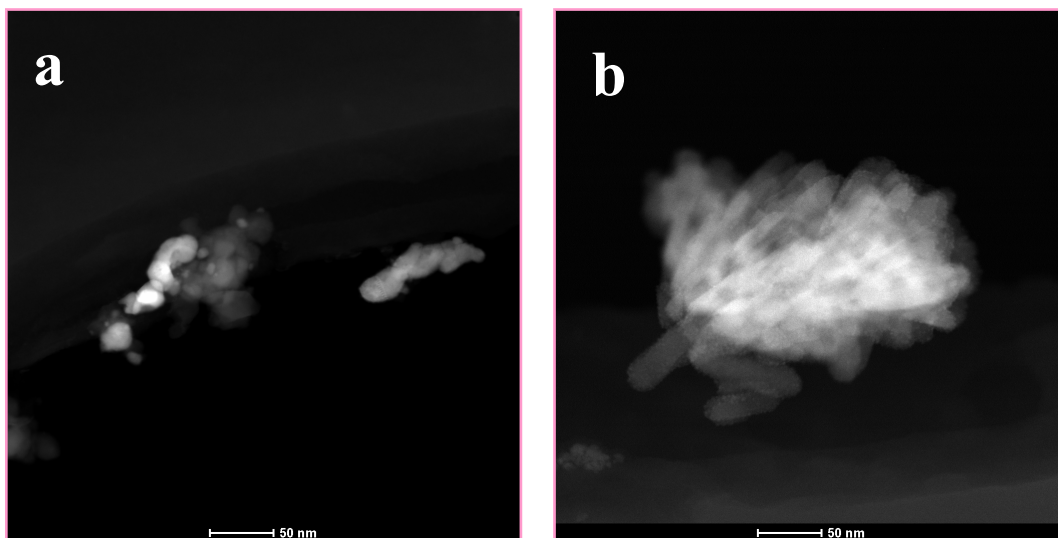


Fig. S1 HAADF-STEM images of Ru catalysts with relatively low magnification

(a), Ru/a-TiO₂; (b), Ru/r-TiO₂.

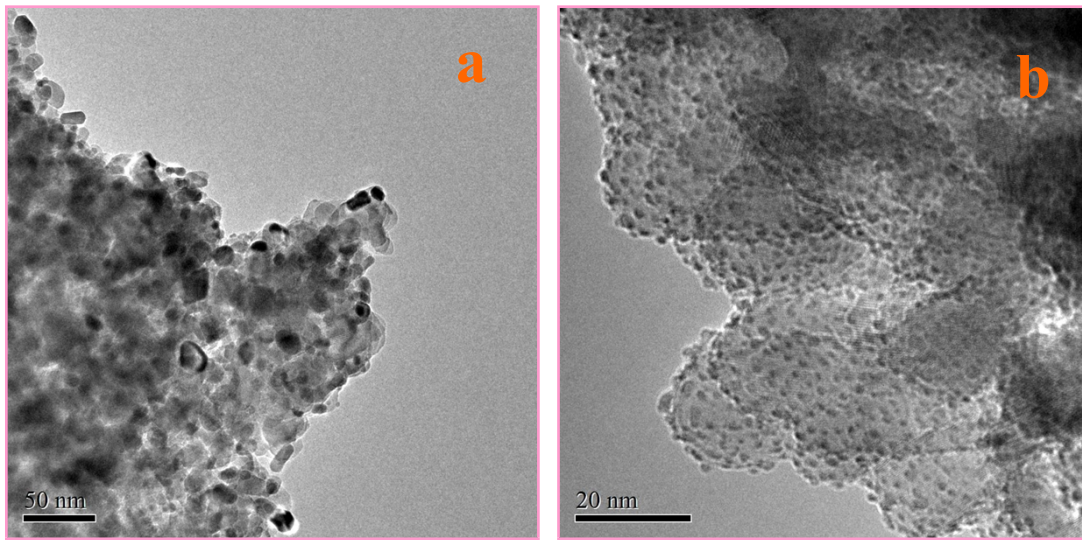


Fig. S2 HRTEM images of (a) Ru/a-TiO₂ and (b) Ru/r-TiO₂ catalysts

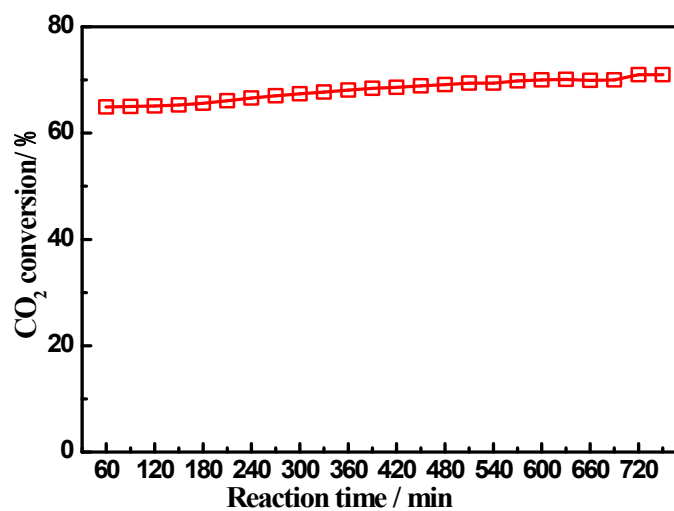


Fig. S3 CO₂ conversion as a function of the reaction time at 300 °C for Ru/r-TiO₂ catalysts

Reaction conditions: 18 vol% CO₂ + 72% H₂ + 10% N₂, catalyst: Each 0.040g diluted with 0.400g SiO₂, total space velocity: 75,000 mL g_{cat}⁻¹ h⁻¹.

Table S1 The calculated results of H₂-TPR results

	RuO _x /r-TiO ₂	RuO _x /a-TiO ₂
H ₂ consumption (mL g _{cat} ⁻¹)	15.4	16.8