

## Supporting Information

### Acceptorless dehydrogenative coupling of primary alcohols to esters by heterogeneous Pt catalysts

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## Experimental

**General.** Commercially available organic compounds (from Tokyo Chemical Industry or Kanto Chemical) were used without further purification. The GC (Shimadzu GC-14B) and GCMS (Shimadzu GCMS-QP2010) analyses were carried out with Ultra ALLOY capillary column UA<sup>+</sup>-1 (Frontier Laboratories Ltd.) using nitrogen and He as the carrier gas.

**Catalyst preparation.** SnO<sub>2</sub> was prepared by calcination of H<sub>2</sub>SnO<sub>3</sub> (Kojundo Chemical Laboratory Co., Ltd.) at 500 °C for 3 h. Nb<sub>2</sub>O<sub>5</sub> was prepared by calcination of niobic acid (CBMMI) at 500 °C for 3 h. SiO<sub>2</sub> (Q-10, 300 m<sup>2</sup> g<sup>-1</sup>) was supplied from Fuji Silysia Chemical Ltd. HBEA zeolite (JRC-Z-HB25, SiO<sub>2</sub>/Al<sub>2</sub>O<sub>3</sub> = 25±5), MgO (JRC-MGO-3), TiO<sub>2</sub> (JRC-TIO-4), CeO<sub>2</sub> (JRC-CEO-3) was supplied from Catalysis Society of Japan. γ-Al<sub>2</sub>O<sub>3</sub> was prepared by calcination of γ-AlOOH (Catapal B Alumina purchased from Sasol) for 3 h at 900 °C. ZrO<sub>2</sub> was prepared by calcination (500 °C for 3 h) of ZrO<sub>2</sub>·nH<sub>2</sub>O prepared by hydrolysis of zirconium oxynitrate 2-hydrate in water by aqueous NH<sub>4</sub>OH solution, followed by filtration of precipitate, washing with water three times, and drying at 100 °C for 12 h.

Precursor of Pt/SnO<sub>2</sub> was prepared by an impregnation method; a mixture of SnO<sub>2</sub> and an aqueous HNO<sub>3</sub> solution of Pt(NH<sub>3</sub>)<sub>2</sub>(NO<sub>3</sub>)<sub>2</sub> was evaporated at 50 °C, followed by drying at 90 °C for 12 h. Before each catalytic experiment, the Pt/SnO<sub>2</sub> catalyst (with Pt loading of 5 wt%) was prepared by in situ pre-reduction of the precursor in a pyrex tube under a flow of H<sub>2</sub> (20 cm<sup>3</sup> min<sup>-1</sup>) at 150 °C for 0.5 h. Other supported Pt catalysts (Pt = 5 wt%) were prepared by the same method. SnO<sub>2</sub>-supported metal catalysts, M/SnO<sub>2</sub> (M = Ni, Cu, Co, Ag, Pd, Ru, Rh, Re, Ir) with metal loading of 5 wt% were prepared by the impregnation method in the similar manner as Pt/SnO<sub>2</sub> using aqueous solution of metal nitrates (for Ni, Cu, Co, Ag), RuCl<sub>3</sub>, IrCl<sub>3</sub>·nH<sub>2</sub>O or NH<sub>4</sub>ReO<sub>4</sub> or aqueous HNO<sub>3</sub> solution of Pd(NO<sub>3</sub>)<sub>2</sub>. A commercial Pt-loaded carbon catalyst (Pt/C, Pt = 5 wt%) was purchased from N.E. Chemcat, Corporation.

**Catalytic reactions.** Pt/SnO<sub>2</sub> was used as a standard catalyst. After the pre-reduction at 150 °C, we carried out catalytic tests using a batch-type reactor without exposing the catalyst to air as follows. The mixture of alcohol (1 mmol) and n-dodecane (0.2 mmol) was injected to the

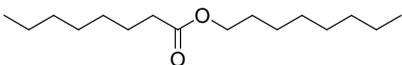
pre-reduced catalyst inside the reactor (cylindrical glass tube, 18 cm<sup>3</sup>) through a septum inlet, followed by filling 1 atm N<sub>2</sub>. Then, the resulting mixture was stirred and heated at 180 °C. Conversion and yields of products were determined by GC using n-dodecane as an internal standard adopting the GC-sensitivity estimated using the isolated product. The products were identified by GC-MS equipped with the same column as GC. The analysis of the gas phase product (H<sub>2</sub>) was carried out by the mass spectrometer (BELMASS). Another set of catalytic experiments was carried out to determine isolated yields of ester as follows. After the reaction, the catalyst was removed by filtration and the reaction mixture was concentrated under vacuum evaporator to remove the volatile compounds. Then, the esters in Table 3 were isolated by column chromatography using silica gel 60 (spherical, 63-210 μm, Kanto Chemical Co. Ltd.) with hexane/ethylacetate (2/98 or 5/95) as the eluting solvent, followed by analyses by <sup>1</sup>H NMR, <sup>13</sup>C NMR and GCMS.

**In situ IR.** In situ IR spectra were recorded using a JASCO FT/IR-4200 equipped with an MCT detector. The closed IR cell surrounded by the Dewar vessel was connected to an evacuation system. During the IR measurement, the IR cell was cooled by freezing mixture of dry ice/ethanol in the Dewar vessel, and the thermocouple near the sample showed -50 ± 3 °C. The sample was pressed into a 30 mg of self-supporting wafer (ϕ = 2 cm) and mounted into the IR cell with CaF<sub>2</sub> windows. Spectra were measured accumulating 15 scans at a resolution of 4 cm<sup>-1</sup>. After in situ pre-evacuation of the sample at 500 °C for 0.5 h, a reference spectrum of the sample disc (SnO<sub>2</sub>, γ-Al<sub>2</sub>O<sub>3</sub> or SiO<sub>2</sub>) was measured at -50 °C. Then, the sample was exposed to 2 Pa of benzaldehyde at -50 °C for 120 s, followed by evacuation for 1000 s. Then a differential IR spectrum, with respect to the reference spectrum, was recorded at -50 °C.

#### NMR and GCMS analysis

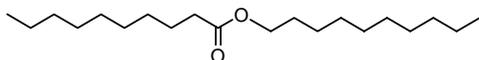
H and <sup>13</sup>C NMR spectra were recorded using at ambient temperature on JEOL-ECX 600 operating at 600.17 and 150.91 MHz, respectively with tetramethylsilane as an internal standard. All chemical shifts (δ) are reported in ppm and coupling constants (*J*) in Hz. All chemical shifts are reported relative to tetramethylsilane and *d*-solvent peaks (77.00 ppm, chloroform), respectively. Abbreviations used in the NMR experiments: s, singlet; d, doublet; t, triplet; m, multiplet. GC-MS spectra were recorded by SHIMADZU QP2010.

#### Octanoic acid octyl ester (C<sub>16</sub>H<sub>32</sub>O<sub>2</sub>):<sup>1</sup>



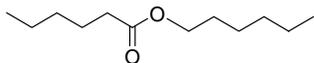
<sup>1</sup>H NMR (600.17 MHz, CDCl<sub>3</sub>), TMS: δ 4.04 (t, *J* = 6.84 Hz, 2H), 2.28 (t, *J* = 7.56 Hz, 2H), 1.62-1.58 (m, 4H), 1.33-1.24 (m, 18H), 0.86 (t, *J* = 6.90 Hz, 6H); <sup>13</sup>C NMR (150.91 MHz, CDCl<sub>3</sub>) δ 174.37, 64.53, 34.41, 31.90, 31.76, 31.64, 29.19, 29.16, 29.08, 28.90, 28.60, 25.91, 25.00, 22.61, 22.57, 14.01; GC-MS *m/e* 256.24.

#### Decanoic acid decyl ester (C<sub>20</sub>H<sub>40</sub>O<sub>2</sub>):<sup>2</sup>



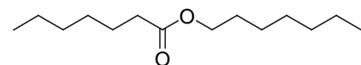
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**Heptanoic acid heptyl ester ( $\text{C}_{14}\text{H}_{28}\text{O}_2$ ) :**<sup>3</sup>



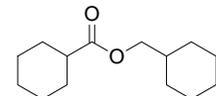
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**Hexanoic acid hexyl ester ( $\text{C}_{12}\text{H}_{24}\text{O}_2$ ) :**<sup>4</sup>



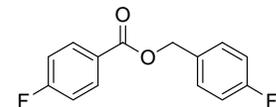
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**Cyclohexane methanoic acid cyclohexane methyl ester ( $\text{C}_{13}\text{H}_{22}\text{O}_2$ ) :**<sup>4</sup>



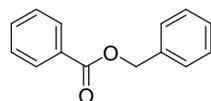
$^1\text{H}$  NMR (600.17 MHz,  $\text{CDCl}_3$ ); TMS:  $\delta$  3.86-3.84(m, 2H), 2.30-2.26 (m, 1H), 1.90-1.88 (m, 2H), 1.72-1.62 (m, 8H), 1.45-1.39 (m, 2H), 1.29-1.11 (m, 6H), 0.98-0.92 (m, 1H);  $^{13}\text{C}$  NMR (150.91 MHz,  $\text{CDCl}_3$ )  $\delta$  176.18, 69.23, 43.31, 37.15, 29.65 (C $\times$ 2), 29.06 (C $\times$ 2), 26.36, 25.76, 25.68 (C $\times$ 2), 25.46; GC-MS m/e 224.20.

**4-Fluoro benzoic acid 4-fluoro benzyl ester ( $\text{C}_{14}\text{H}_{10}\text{F}_2\text{O}_2$ ):**<sup>5</sup>



$^1\text{H}$  NMR (600.17 MHz,  $\text{CDCl}_3$ ); TMS:  $\delta$  8.11-8.06 (m, 2H), 7.42 (d,  $J = 5.52$  Hz, 2H), 7.12-7.05 (m, 4H), 5.31 (s, 2H);  $^{13}\text{C}$  NMR (150.91 MHz,  $\text{CDCl}_3$ )  $\delta$  165.84 (d,  $J = 254.50$  Hz), 165.39, 162.70 (d,  $J = 247.58$  Hz), 161.88, 132.23 (d,  $J = 9.20$  Hz), 131.70 (d,  $J = 3.1$  Hz), 130.27 (d,  $J = 8.07$ ), 126.24 (d,  $J = 3$  Hz), 115.56 (d,  $J = 21.88$  Hz), 66.12; GC-MS m/e 248.19.

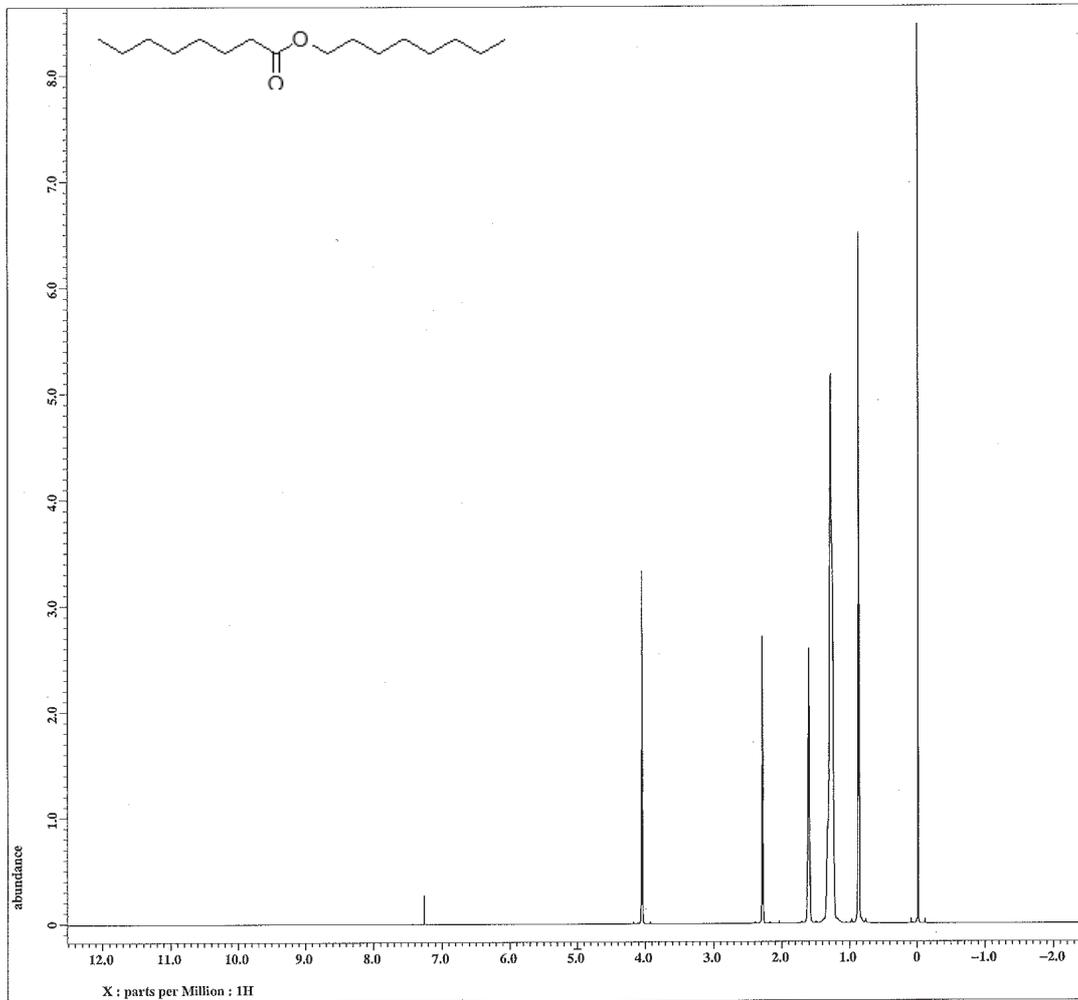
**Benzoic acid benzyl ester ( $\text{C}_{14}\text{H}_{12}\text{O}_2$ ) :**<sup>6</sup>



<sup>1</sup>H NMR (600.17 MHz, CDCl<sub>3</sub>); TMS: δ 8.10-8.09 (m, 2H), 7.56 (t, *J* = 6.90 Hz, 1H), 7.47-7.44 (m, 4H), 7.40 (t, *J* = 7.50 Hz, 2H), 7.37-7.34 (m, 1H), 5.36 (s, 2H); <sup>13</sup>C NMR (150.91 MHz, CDCl<sub>3</sub>) δ 166.38, 136.01, 132.98, 130.09, 129.66 (C×2), 128.55(C×2), 128.33(C×2), 128.20, 128.12(C×2), 66.64; GC-MS m/e 212.16.

#### References

- 1 M. Minakawa, H. Baek, Y. M. A Yamada, J. W Han and Y. Uozumi, *Org. Lett.*, 2013, **15**, 5798.
- 2 J. Malineni, C. Merkens, H. Keul, M. Möller, *Catal. Commun.*, 2013, **40**, 80.
- 3 A. Amati, G. Dosualdo, L. Zhao, A. Bravo, F. Fontana, F. Minisci and H. Bjørsvik, *Org. Proc. Res. Dev.*, 1998, **2**, 261.
- 4 T. Ooi, T. Miura, K. Takaya, K. Maruoka, *Tetrahedron Lett.* 1999, **40**, 7695.
- 5 C. Liu, S. Tang and A. Lei, *Chem. Commun.*, 2013, **49**, 1324.
- 6 K. Ishihara, M. Niwa and Y. Kosugi, *Org. Lett.*, 2008, **10**, 2187.



**JEOL**

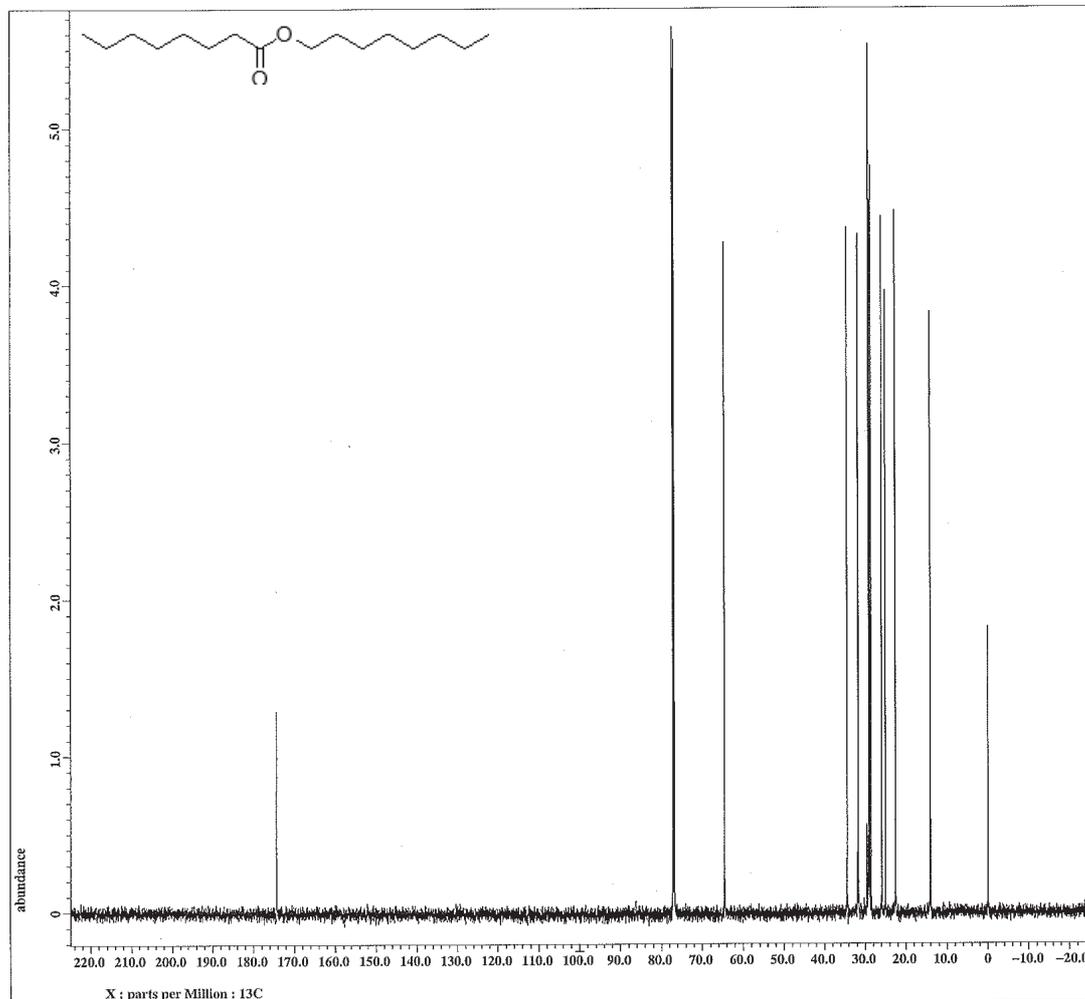
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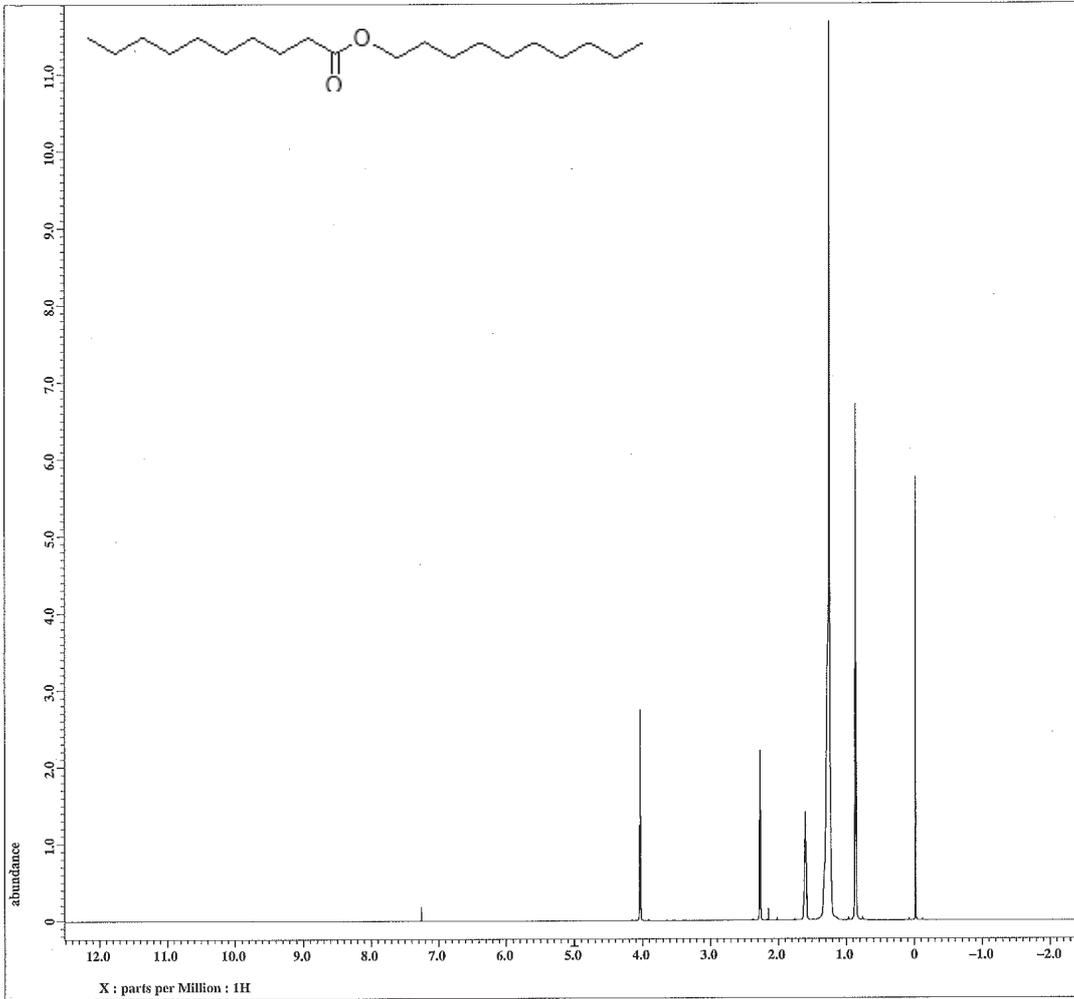
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**JEOL**

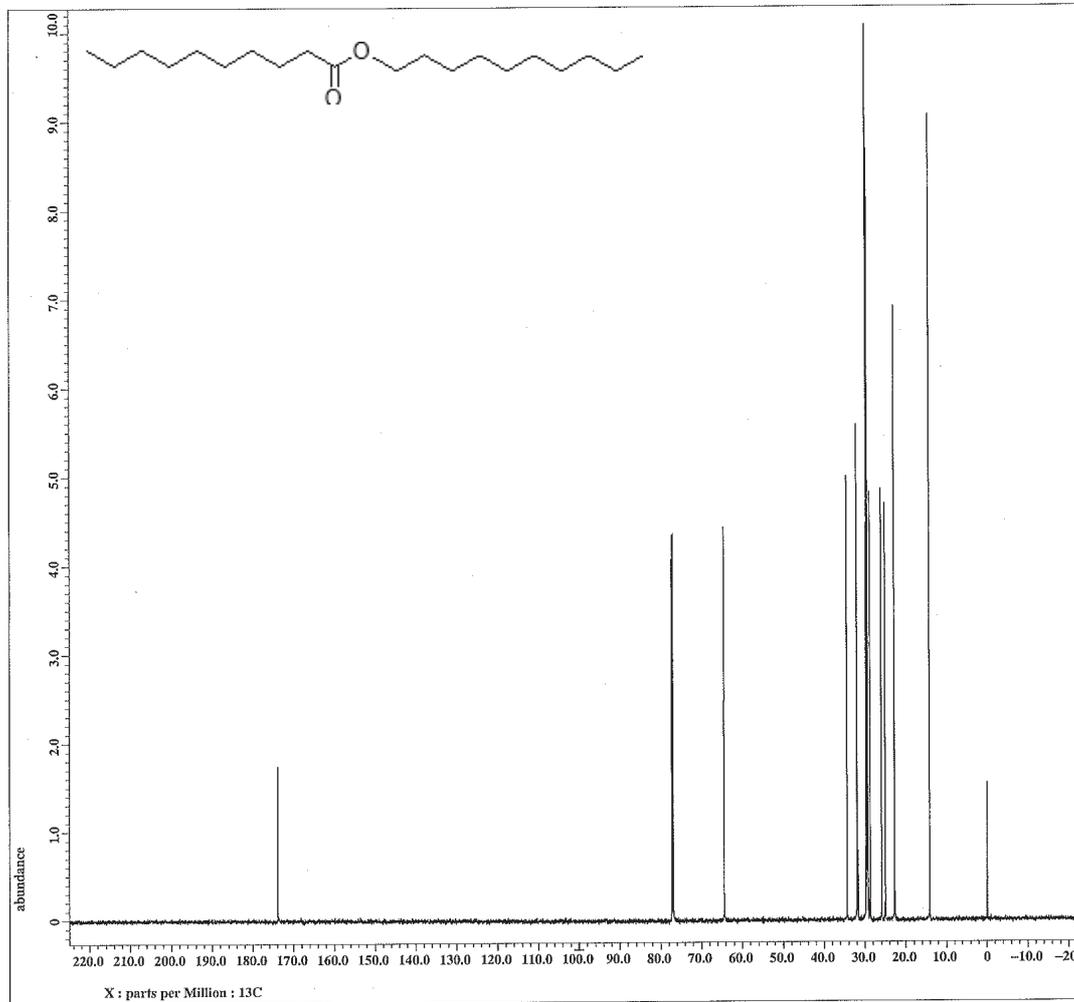
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**JEOL**

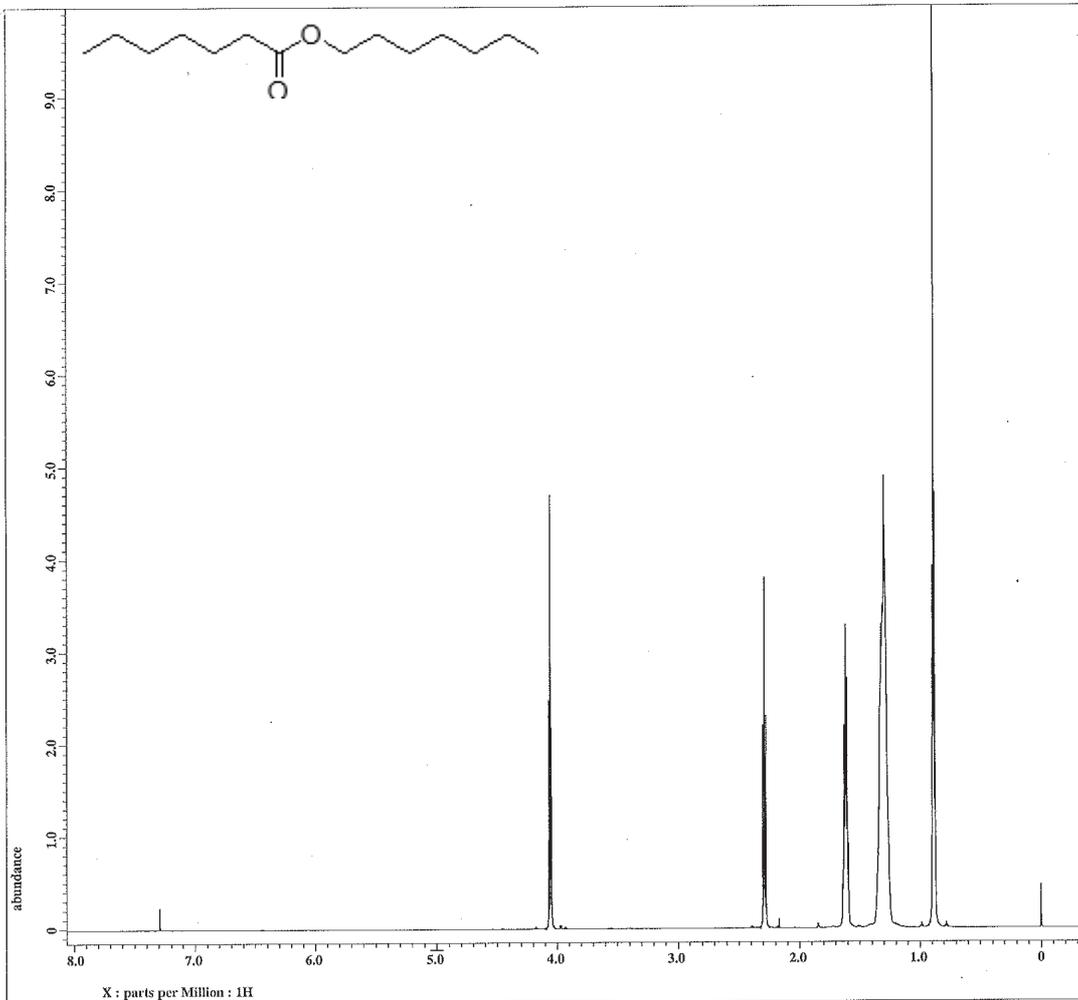
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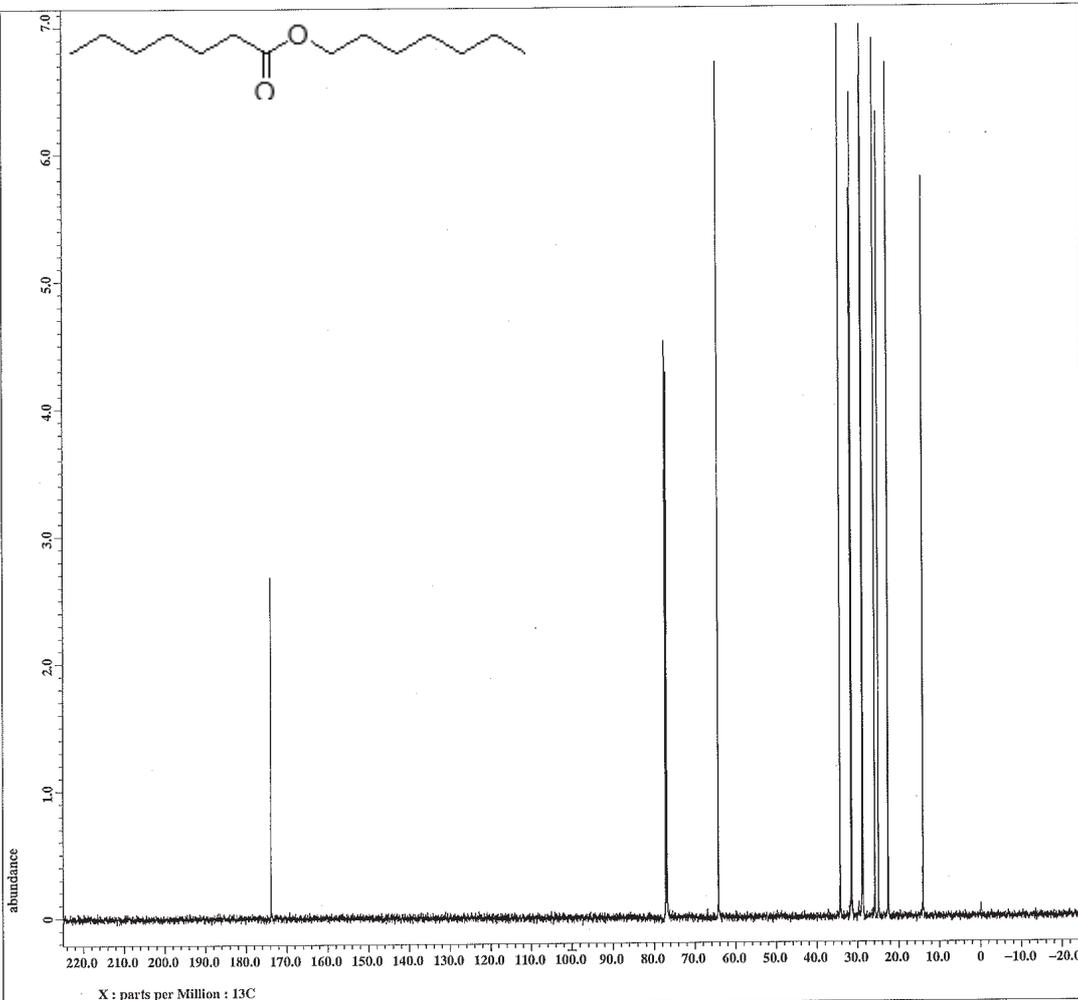


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 Repetition\_time = 6.4548992[s]  
 Temp\_get = 21.5[dc]



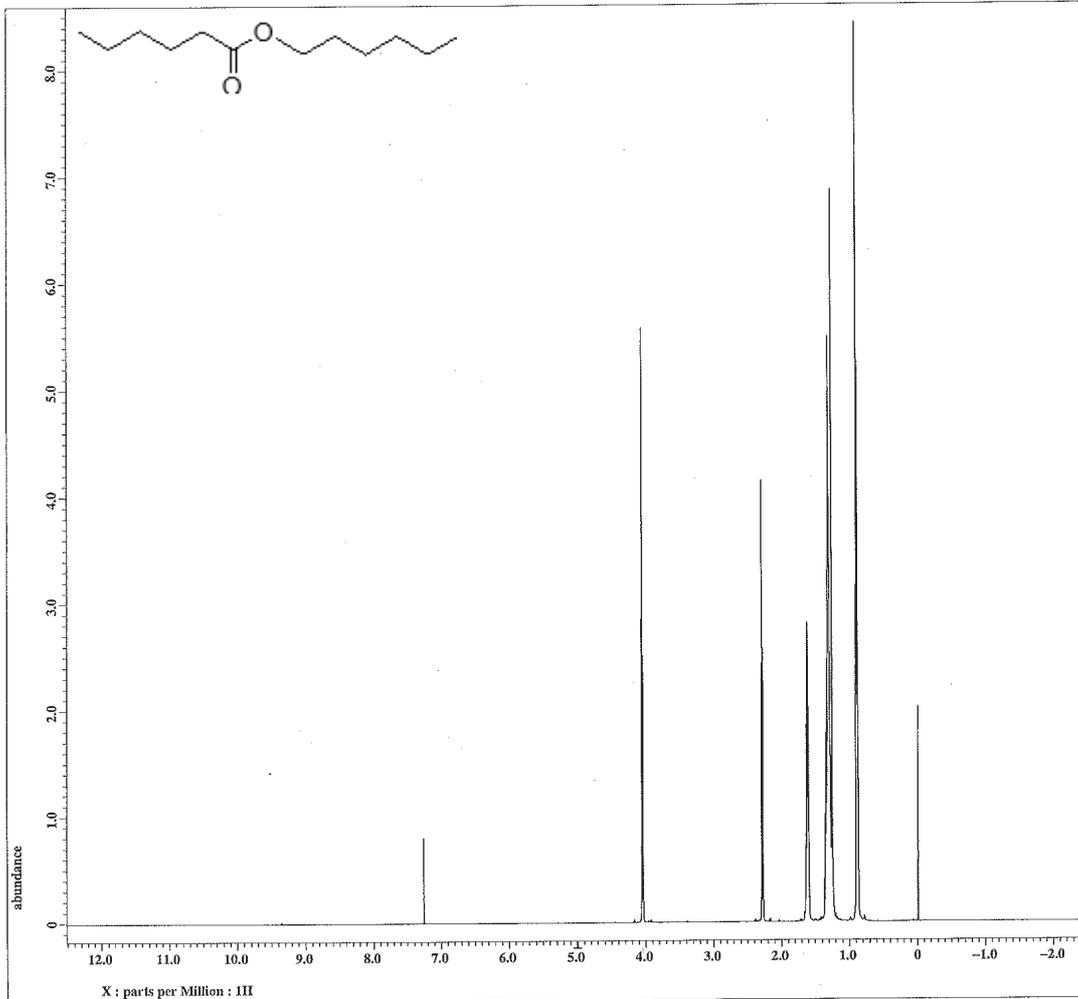
Filename = Exp-Konika-48-4-carbo  
 Author = delta  
 Experiment = single\_pulse\_dec  
 Sample\_id = Exp-Konika-48-4-carbo  
 Solvent = CHLOROFORM-D  
 Creation\_time = 6-FEB-2014 17:34:28  
 Revision\_time = 6-FEB-2014 17:47:44  
 Current\_time = 6-FEB-2014 17:47:53

Content = Exp-Konika-48-4-carbo  
 Data\_format = 1D COMPLEX  
 Dim\_size = 32768  
 Dim\_title = 13C  
 Dim\_units = [ppm]  
 Dimensions = X  
 Site = ECA 600  
 Spectrometer = DELTA2\_NMR

Field\_strength = 14.09636928[T] (600[M]  
 X\_acq\_duration = 0.868352[s]  
 X\_domain = 13C  
 X\_freq = 150.91343039[MHz]  
 X\_offset = 100[ppm]  
 X\_points = 32768  
 X\_prescans = 4  
 X\_resolution = 1.15160672[Hz]  
 X\_sweep = 37.73584906[KHz]  
 Irr\_domain = 1H  
 Irr\_freq = 600.1723046[MHz]  
 Irr\_offset = 5[ppm]  
 Clipped = FALSE  
 Mod\_return = 1  
 Scans = 228.0  
 Total\_scans = 228.0

X\_90\_width = 11.4[us]  
 X\_acq\_time = 0.868352[s]  
 X\_angle = 30[deg]  
 X\_atn = 7.5[db]  
 X\_pulse = 3.8[us]  
 Irr\_atn\_dec = 19.391[db]  
 Irr\_atn\_noc = 19.391[db]  
 Irr\_noise = WALFZ  
 Decoupling = TRUE  
 Initial\_wait = 1[s]  
 Noc = TRUE  
 Noc\_time = 2.5[s]  
 Recvr\_gain = 76  
 Relaxation\_delay = 2.5[s]  
 Repetition\_time = 3.368352[s]  
 Temp\_get = 22.4[dc]

(A)



**JEOL**

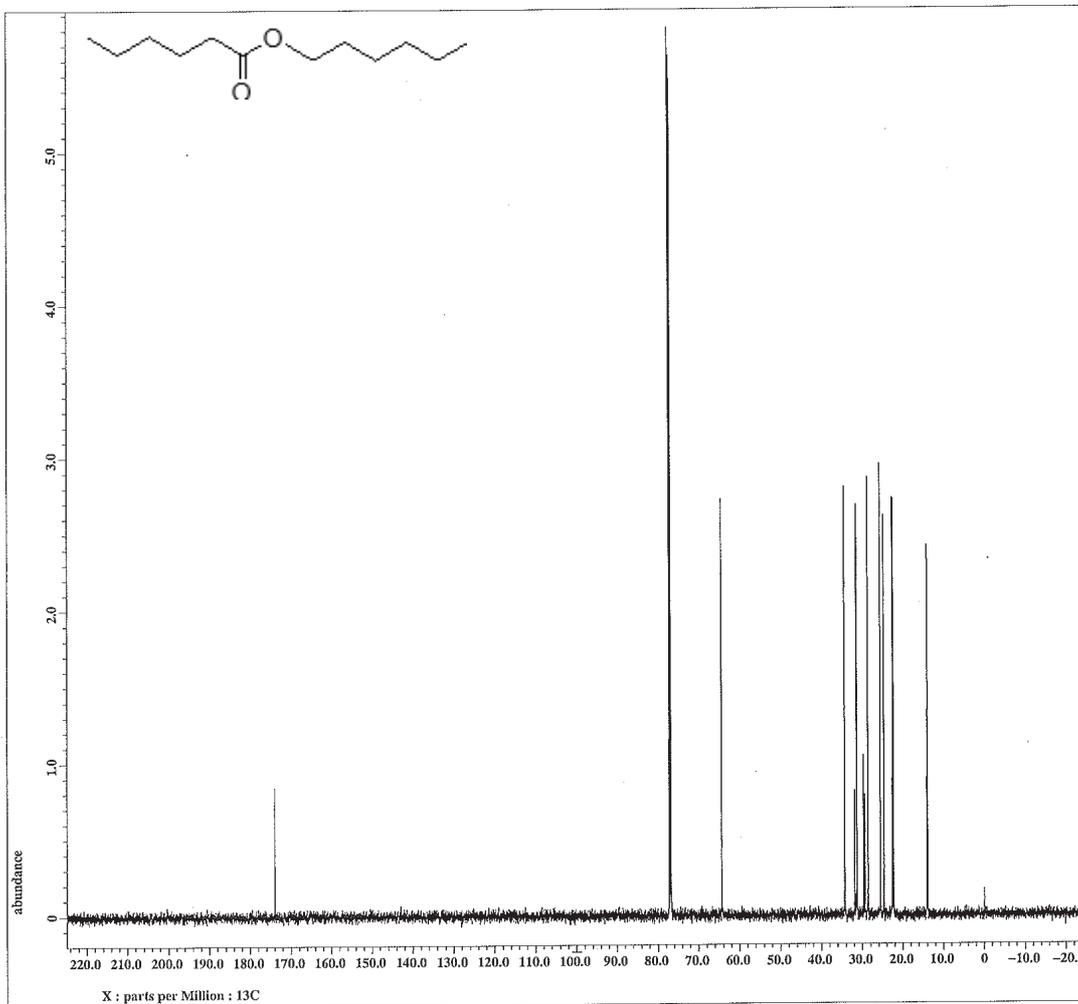
```

Filename      = Exp-Konika-48-3-proto
Author       = delta
Experiment   = single_pulse.ex2
Sample_id    = Exp-Konika-48-3-proto
Solvent      = CHLOROFORM-D
Creation_time = 6-FEB-2014 16:56:08
Revision_time = 6-FEB-2014 17:29:07
Current_time = 6-FEB-2014 17:29:23

Content      = Exp-Konika-48-3-proto
Data_format  = 1D COMPLEX
Dim_size     = 13107
Dim_title    = 1H
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 1.4548992[s]
X_domain       = 1H
X_freq         = 600.1723046[MHz]
X_offset       = 5[ppm]
X_points       = 16384
X_prescans     = 1
X_resolution   = 0.68733284[Hz]
X_sweep        = 11.26126126[kHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Tri_domain     = 1H
Tri_freq       = 600.1723046[MHz]
Tri_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 8
Total_scans    = 8

X_90_width    = 13[us]
X_acq_time     = 1.4548992[s]
X_angle        = 45[deg]
X_atn          = 3.6[db]
X_pulse        = 6.5[us]
Irr_mode       = Off
Tri_mode       = Off
Dante_presat  = FALSE
Initial_wait   = 1[s]
Recvr_gain     = 36
Relaxation_delay = 5[s]
Repetition_time = 6.4548992[s]
Temp_get       = 21.7[dc]
  
```



**JEOL**

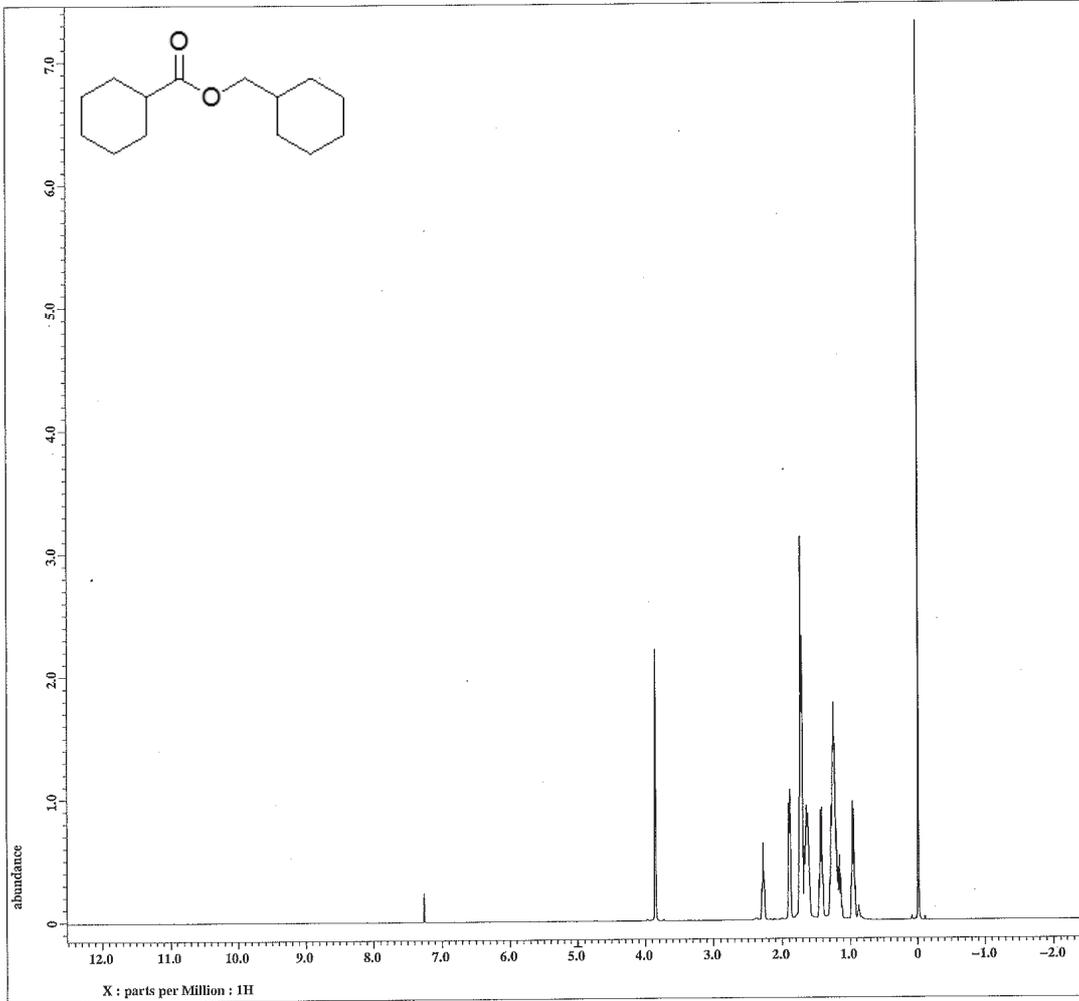
```

Filename      = Exp-Konika-48-3-carbo
Author       = delta
Experiment   = single_pulse_dec
Sample_id    = Exp-Konika-48-3-carbo
Solvent      = CHLOROFORM-D
Creation_time = 6-FEB-2014 17:11:21
Revision_time = 6-FEB-2014 17:38:21
Current_time = 6-FEB-2014 17:38:32

Content      = Exp-Konika-48-3-carbo
Data_format  = 1D COMPLEX
Dim_size     = 32768
Dim_title    = 13C
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 0.868352[s]
X_domain       = 13C
X_freq         = 150.91343039[MHz]
X_offset       = 100[ppm]
X_points       = 32768
X_prescans     = 4
X_resolution   = 1.15160672[Hz]
X_sweep        = 37.73584306[kHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 252
Total_scans    = 252

X_90_width    = 11.4[us]
X_acq_time     = 0.868352[s]
X_angle        = 30[deg]
X_atn          = 7.5[db]
X_pulse        = 3.8[us]
Irr_atn_dec   = 19.391[db]
Irr_atn_noe   = 19.391[db]
Irr_noise     = WALTZ
Decoupling     = TRUE
Initial_wait   = 1[s]
Noe            = TRUE
Noe_time       = 2.5[s]
Recvr_gain     = 78
Relaxation_delay = 2.5[s]
Repetition_time = 3.368352[s]
Temp_get       = 22[dc]
  
```



**JEOL**

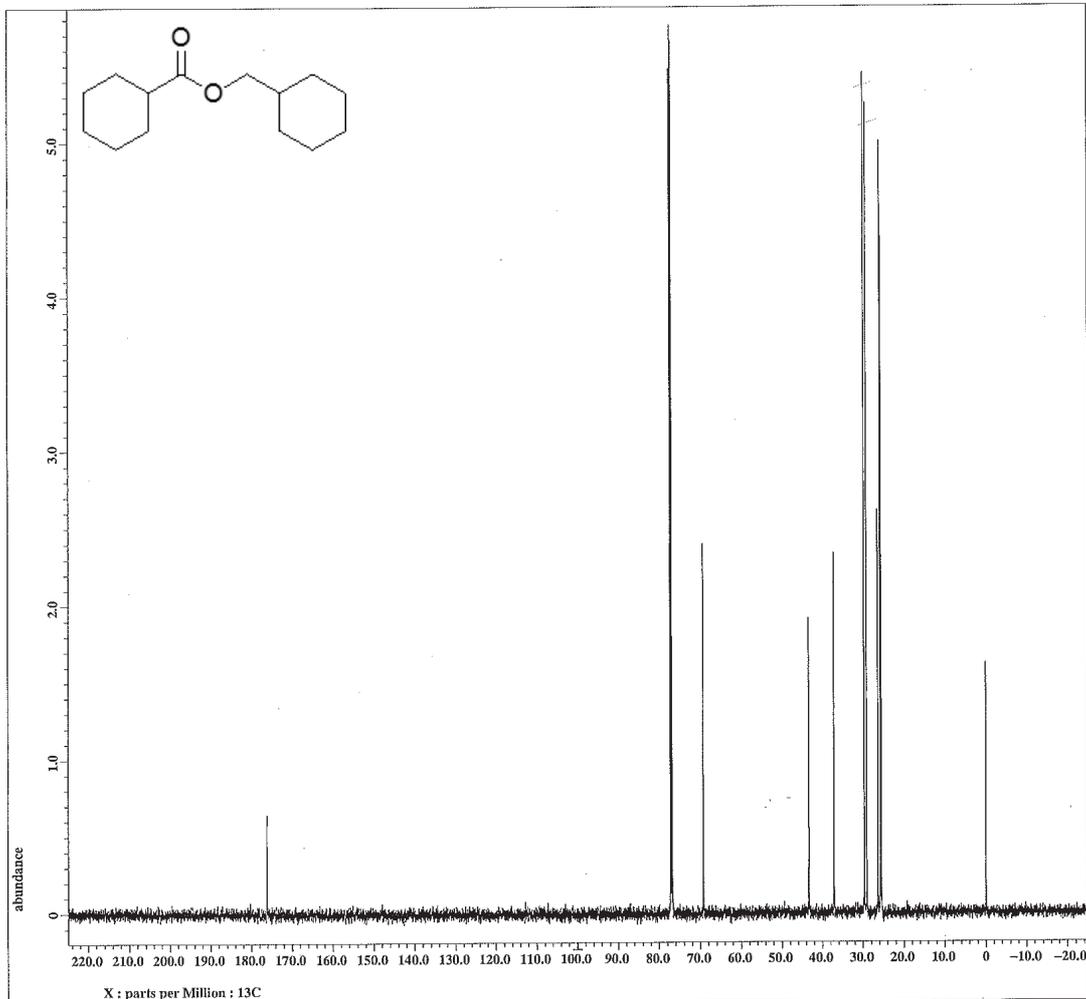
```

Filename      = Exp-Konika-20-4-proto
Author       = delta
Experiment   = single_pulse.ex2
Sample_id    = Exp-Konika-20-4-proto
Solvent      = CHLOROFORM-D
Creation_time = 14-FEB-2014 16:02:53
Revision_time = 14-FEB-2014 16:17:03
Current_time  = 14-FEB-2014 16:17:27

Content      = Exp-Konika-20-4-proto
Data_format  = 1D COMPLEX
Dim_size     = 13107
Dim_title    = 1H
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 1.4548992[s]
X_domain       = 1H
X_freq         = 600.1723046[MHz]
X_offset       = 5[ppm]
X_points       = 16384
X_prescans     = 1
X_resolution   = 0.68733284[Hz]
X_sweep        = 11.26126126[MHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Tri_domain     = 1H
Tri_freq       = 600.1723046[MHz]
Tri_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 8
Total_scans    = 8

X_90_width    = 13[us]
X_acq_time     = 1.4548992[s]
X_angle        = 45[deg]
X_atn          = 3.6[db]
X_pulse        = 6.5[us]
Irr_mode       = OFF
Tri_mode       = OFF
Dante_presat   = FALSE
Initial_wait   = 1[s]
Recvr_gain     = 36
Relaxation_delay = 5[s]
Repetition_time = 6.4548992[s]
Temp_get       = 20.4[dc]
  
```



**JEOL**

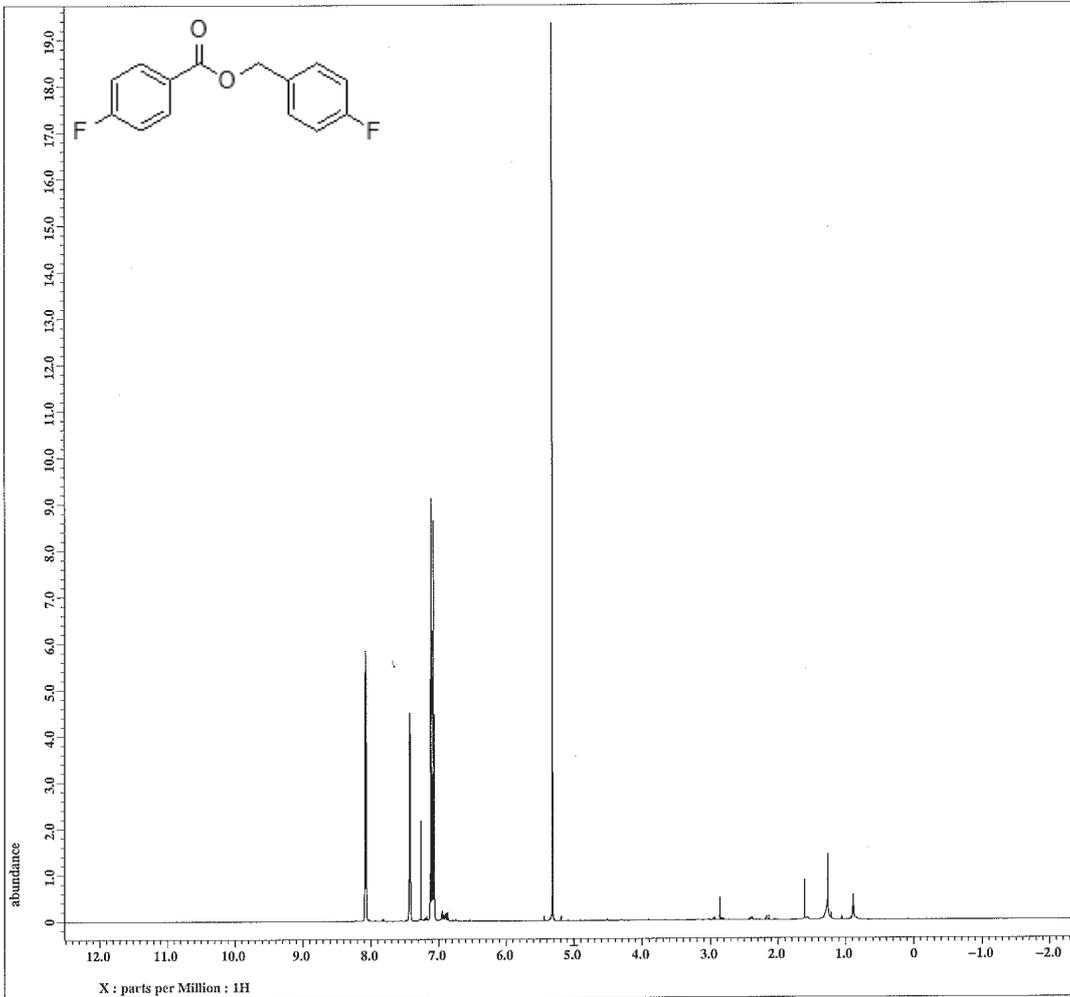
```

Filename      = Exp-Konika-20-4-carbo
Author       = delta
Experiment   = single_pulse_dec
Sample_id    = Exp-Konika-20-4-carbo
Solvent      = CHLOROFORM-D
Creation_time = 14-FEB-2014 16:15:40
Revision_time = 14-FEB-2014 16:43:01
Current_time  = 14-FEB-2014 16:43:07

Content      = Exp-Konika-20-4-carbo
Data_format  = 1D COMPLEX
Dim_size     = 32768
Dim_title    = 13C
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 0.868352[s]
X_domain       = 13C
X_freq         = 150.91343039[MHz]
X_offset       = 100[ppm]
X_points       = 32768
X_prescans     = 4
X_resolution   = 1.15160672[Hz]
X_sweep        = 37.73584906[kHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 209
Total_scans    = 209

X_90_width    = 11.4[us]
X_acq_time     = 0.868352[s]
X_angle        = 30[deg]
X_atn          = 7.5[db]
X_pulse        = 3.8[us]
Irr_atn_dec    = 19.391[db]
Irr_atn_noe    = 19.391[db]
Irr_noise      = WALTZ
Decoupling     = TRUE
Initial_wait   = 1[s]
Noe            = TRUE
Noe_time       = 2.5[s]
Recvr_gain     = 78
Relaxation_delay = 2.5[s]
Repetition_time = 3.368352[s]
Temp_get       = 21.7[dc]
  
```



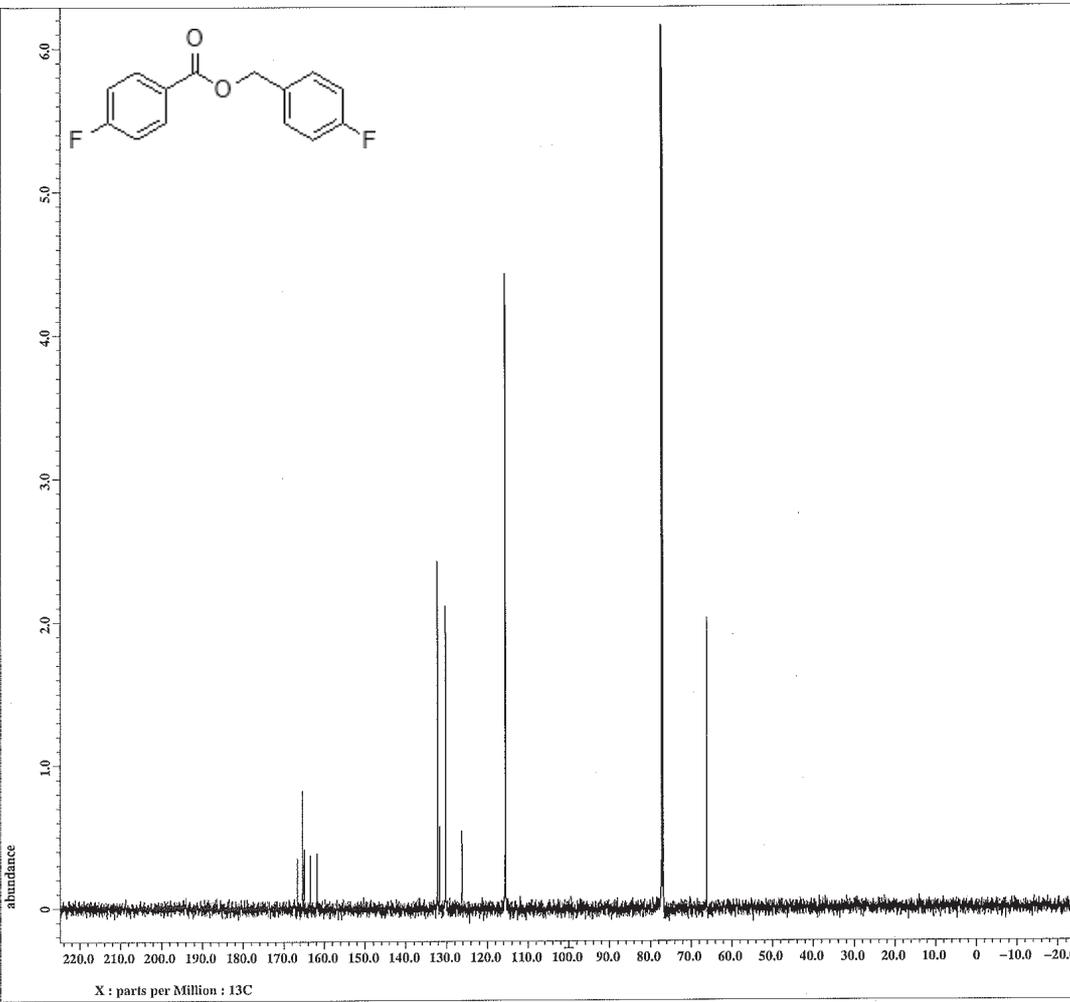
```

Filename      = Exp-konika-24-1-proto
Author       = delta
Experiment   = single_pulse.ex2
Sample_id    = Exp-konika-24-1-proto
Solvent      = CHLOROFORM-D
Creation_time = 28-JAN-2014 15:53:30
Revision_time = 28-JAN-2014 16:05:34
Current_time  = 28-JAN-2014 16:05:37

Content      = Exp-konika-24-1-proto
Data_format  = 1D COMPLEX
Dim_size     = 13107
Dim_title    = 1H
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 1.4548992[s]
X_domain       = 1H
X_freq         = 600.1723046[MHz]
X_offset       = 5[ppm]
X_points       = 16384
X_prescans     = 1
X_resolution   = 0.68733284[Hz]
X_sweep        = 11.26126126[kHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Tri_domain     = 1H
Tri_freq       = 600.1723046[MHz]
Tri_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 8
Total_scans    = 8

X_90_width    = 13[us]
X_acq_time     = 1.4548992[s]
X_angle        = 45[deg]
X_atn          = 3.6[db]
X_pulse        = 6.5[us]
Irr_mode       = Off
Tri_mode       = Off
Dante_presat  = FALSE
Initial_wait   = 1[s]
Recvr_gain     = 44
Relaxation_delay = 5[s]
Repetition_time = 6.4548992[s]
Temp_get       = 20.7[dc]
  
```



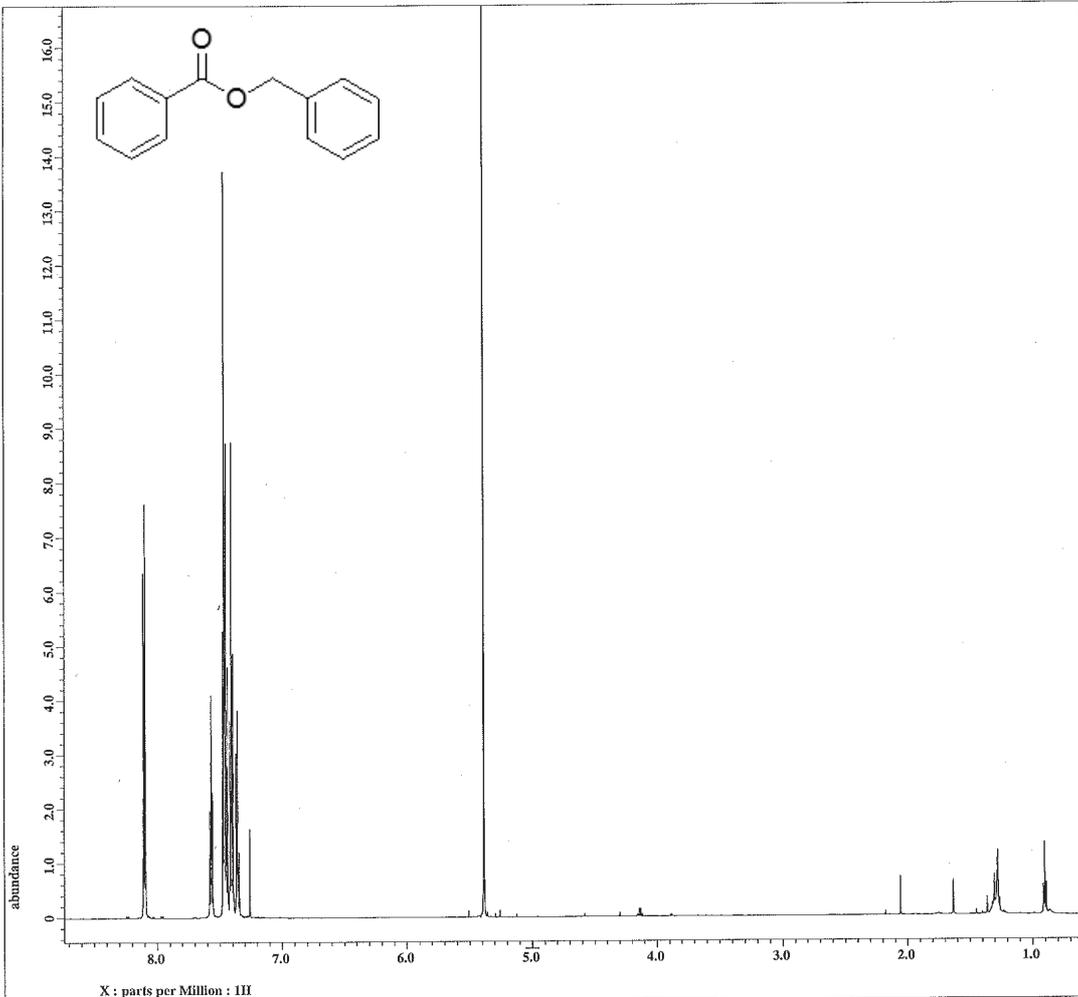
```

Filename      = Exp-konika-24-1-carbo
Author       = delta
Experiment   = single_pulse_dec
Sample_id    = Exp-konika-24-1-carbo
Solvent      = CHLOROFORM-D
Creation_time = 28-JAN-2014 16:01:51
Revision_time = 28-JAN-2014 16:14:55
Current_time  = 28-JAN-2014 16:14:59

Content      = Exp-konika-24-1-carbo
Data_format  = 1D COMPLEX
Dim_size     = 32768
Dim_title    = 13C
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 0.868352[s]
X_domain       = 13C
X_freq         = 150.91343039[MHz]
X_offset       = 100[ppm]
X_points       = 32768
X_prescans     = 4
X_resolution   = 1.15160672[Hz]
X_sweep        = 37.73584906[kHz]
Irr_domain     = 1H
Irr_freq       = 600.1723046[MHz]
Irr_offset     = 5[ppm]
Clipped        = FALSE
Mod_return     = 1
Scans          = 130
Total_scans    = 130

X_90_width    = 11.4[us]
X_acq_time     = 0.868352[s]
X_angle        = 30[deg]
X_atn          = 7.5[db]
X_pulse        = 3.8[us]
Irr_atn_dec   = 19.391[db]
Irr_atn_noe   = 19.391[db]
Irr_noise     = WALZ2
Decoupling     = TRUE
Initial_wait   = 1[s]
Noe           = TRUE
Noe_time       = 2.5[s]
Recvr_gain     = 78
Relaxation_delay = 2.5[s]
Repetition_time = 3.368352[s]
Temp_get       = 20.9[dc]
  
```



**JEOL**

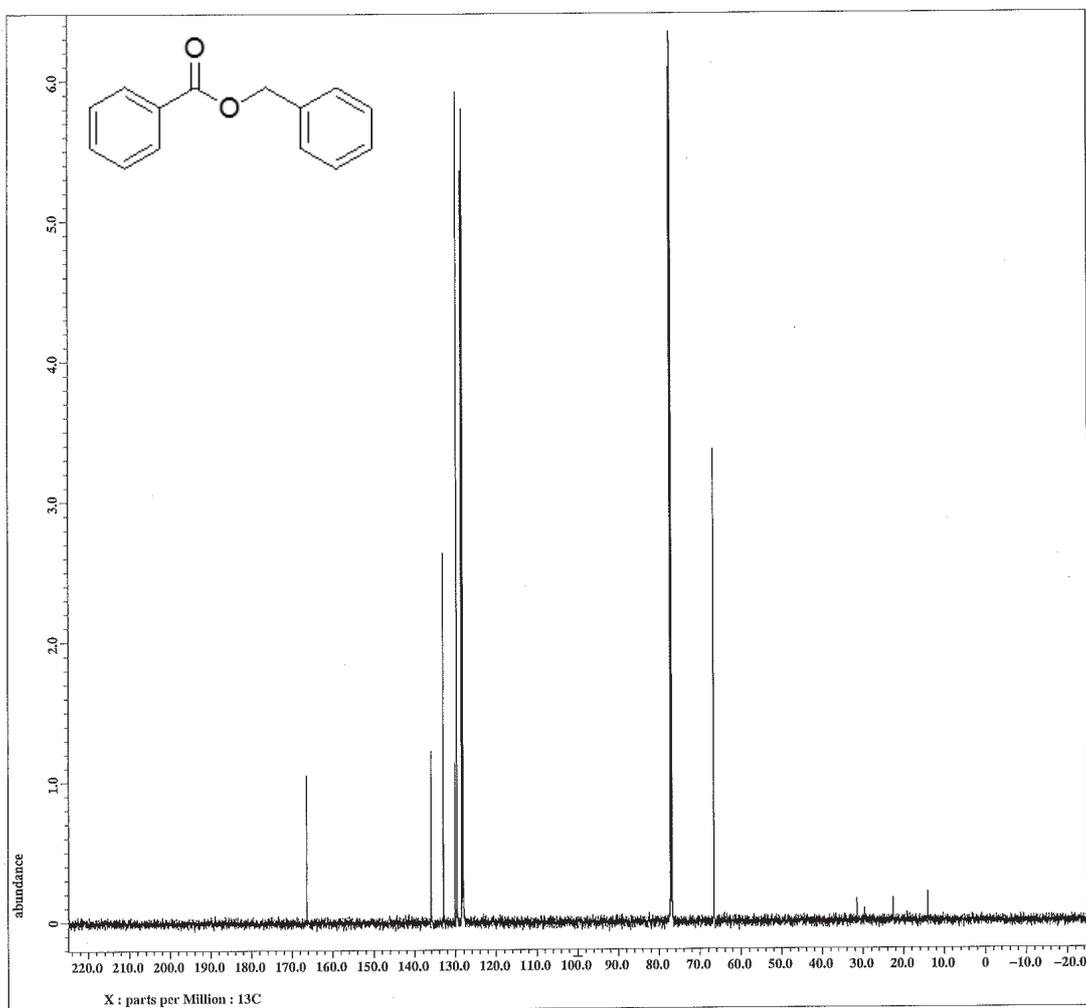
```

Filename      = Exp-Konika-20-5-proto
Author       = delta
Experiment   = single_pulse_ex2
Sample_id    = Exp-Konika-20-5-proto
Solvent      = CHLOROFORM-D
Creation_time = 21-JAN-2014 16:01:34
Revision_time = 21-JAN-2014 16:13:35
Current_time = 21-JAN-2014 16:14:29

Content      = Exp-Konika-20-5-proto
Data_format  = 1D COMPLEX
Dim_size     = 13107
Dim_title    =
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 1.4548992[s]
X_domain      = 1H
X_freq        = 600.1723046[MHz]
X_offset      = 5[ppm]
X_points      = 16384
X_prescans    = 1
X_resolution  = 0.68733284[Hz]
X_sweep       = 11.26126126[kHz]
Irr_domain    = 1H
Irr_freq      = 600.1723046[MHz]
Irr_offset    = 5[ppm]
Tri_domain    = 1H
Tri_freq      = 600.1723046[MHz]
Tri_offset    = 5[ppm]
Clipped       = FALSE
Mod_return    = 1
Scans         = 8
Total_scans   = 8

X_90_width    = 13[us]
X_acq_time     = 1.4548992[s]
X_angle        = 45[deg]
X_atn          = 3.6[db]
X_pulse        = 6.5[us]
Irr_mode       = Off
Tri_mode       = Off
Dante_presat   = FALSE
Initial_wait   = 1[s]
Recvr_gain     = 42
Relaxation_delay = 5[s]
Repetition_time = 6.4548992[s]
Temp_get       = 21.5[dc]
  
```



**JEOL**

```

Filename      = Exp-Konika-20-5-carbo
Author       = delta
Experiment   = single_pulse_dec
Sample_id    = Exp-Konika-20-5-carbo
Solvent      = CHLOROFORM-D
Creation_time = 21-JAN-2014 16:17:37
Revision_time = 21-JAN-2014 16:31:09
Current_time = 21-JAN-2014 16:31:28

Content      = Exp-Konika-20-5-carbo
Data_format  = 1D COMPLEX
Dim_size     = 32768
Dim_title    = 13C
Dim_units    = [ppm]
Dimensions   = X
Site         = ECA 600
Spectrometer = DELTA2_NMR

Field_strength = 14.09636928[T] (600[M]
X_acq_duration = 0.868352[s]
X_domain      = 13C
X_freq        = 150.91343039[MHz]
X_offset      = 100[ppm]
X_points      = 32768
X_prescans    = 4
X_resolution  = 1.15160672[Hz]
X_sweep       = 37.73584906[kHz]
Irr_domain    = 1H
Irr_freq      = 600.1723046[MHz]
Irr_offset    = 5[ppm]
Clipped       = TRUE
Mod_return    = 1
Scans         = 267
Total_scans   = 267

X_90_width    = 11.4[us]
X_acq_time     = 0.868352[s]
X_angle        = 30[deg]
X_atn          = 7.5[db]
X_pulse        = 3.8[us]
Irr_atn_dec   = 19.391[db]
Irr_atn_noe   = 19.391[db]
Irr_noise     = WALTZ
Decoupling    = TRUE
Initial_wait   = 1[s]
Noe           = TRUE
Noe_time       = 2.5[s]
Recvr_gain     = 78
Relaxation_delay = 2.5[s]
Repetition_time = 3.368352[s]
Temp_get       = 21.9[dc]
  
```