

Supplementary Information for

**Highly efficient and robust Cu/SiO₂ catalyst prepared by ammonia
evaporation hydrothermal method for glycerol hydrogenolysis to 1,2-
propanediol**

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Table S1 Comparison for Cu/SiO₂ catalysts prepared by different preparation methods in glycerol hydrogenolysis to 1,2-PDO.

Preparation methods	Operation	T (°C)	P (MPa)	Conversion (%)	Selectivity (%)	Stability	Reference
Ion exchange	continuous	245	1.5	92	86	-	1
Impregnation	batch	240	8	10	83.6	deactivation	2
Homogeneous deposition-precipitation	batch	180	9	22.1	98.0	deactivation	3
Heterogeneous deposition-precipitation	batch	180	9	35.0	93.7	deactivation	3
Precipitation-gel	batch	180	9	73.4	94.3	200 h ^a	4
Precipitation-gel ^b	batch	200	8	73.3	92.2	deactivation ^a	5
Precipitation-gel	continuous	200	5	62.1	89.5	deactivation ^c	6
Ammonia evaporation hydrothermal method	continuous	200	5	100	98.3	300 h	this work

^a Performed in fixed-bed reactor; solvent was 10 wt% water and 50 wt% methanol.

^b Cu/SiO₂ calcined at 450 °C

^c 90 wt% water as solvent.

The catalytic performances of Cu/SiO₂ catalysts with different preparation methods are summarized in [Table S1](#). Compared to previous catalysts, the AEH method presented higher glycerol conversion and 1,2-PDO selectivity. Notably, most of the previous Cu/SiO₂ catalysts (except precipitation-gel method) deactivated rapidly due to the serious agglomeration of copper nanoparticles. The Cu/SiO₂ prepared by precipitation-gel method stabilized 200 h, while our AEH presented 300 h long-term

performance.⁴ The above results indicated that AEH method was highly efficient for fabricating active, selective and robust Cu-based catalyst for glycerol hydrogenolysis to 1,2-PDO.

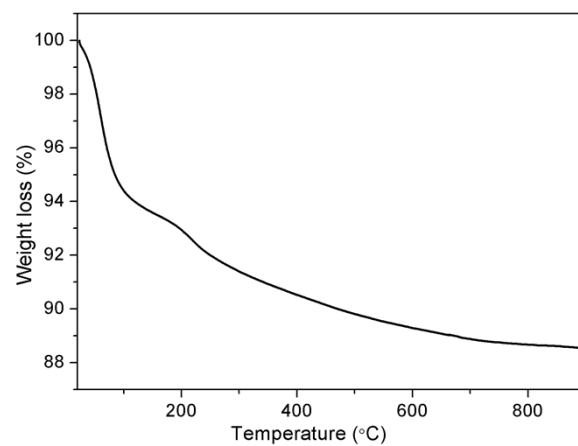


Fig. S1 Thermogravimetric analysis in oxidative atmosphere of the spent 25Cu/SiO₂-IM catalyst during long-term test.

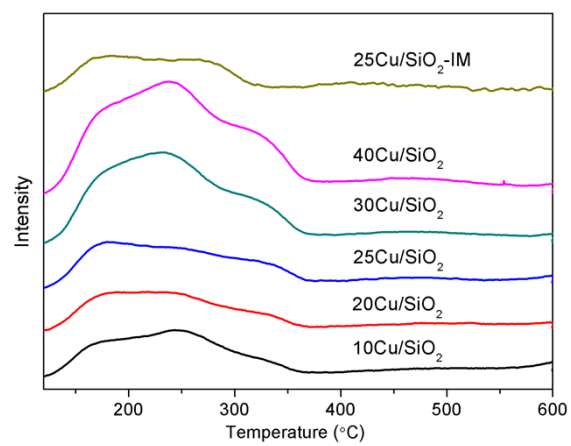


Fig. S2 NH₃-TPD profiles of various catalysts.

Reference

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