

Electronic Supplementary Information

Bi-functional NH₂-MIL-101(Fe) for One-Pot Tandem Photo-oxidation/Knoevenagel Condensation between Aromatic Alcohols and Active Methylene Compounds

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Fig. S1 XRD patterns of (a) calculated MIL-101(Fe); (b) as-synthesized NH₂-MIL-101(Fe).

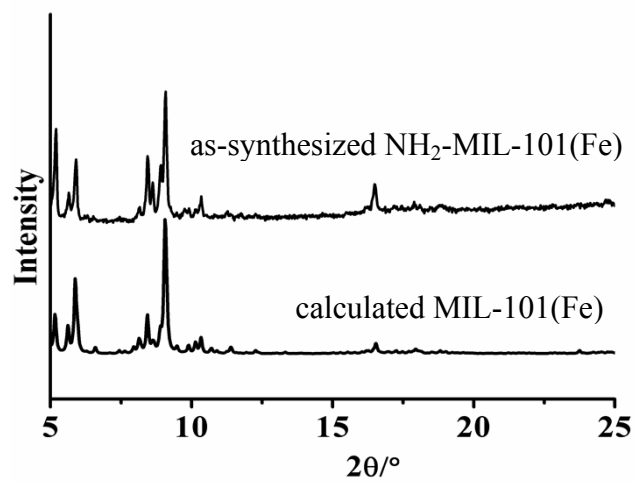


Fig. S2 IR spectrum of fresh $\text{NH}_2\text{-MIL-101(Fe)}$.

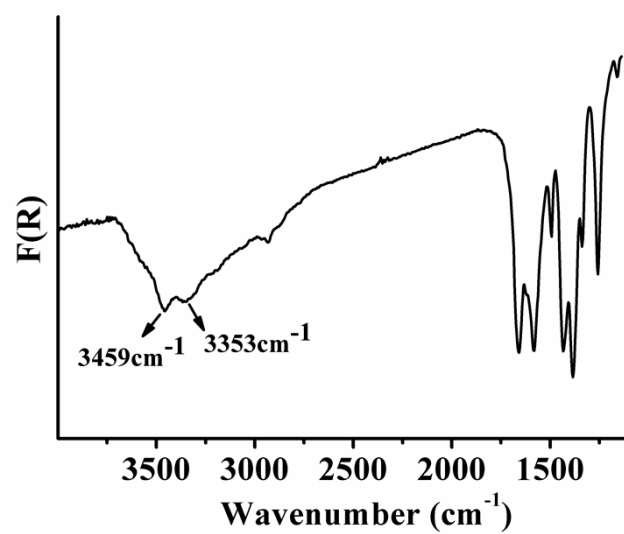


Fig. S3 N₂ adsorption/desorption isotherms of fresh NH₂-MIL-101(Fe).

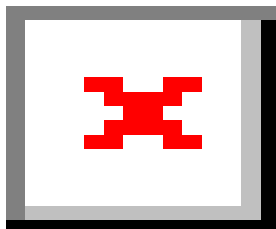


Fig. S4 UV-vis spectrum of $\text{NH}_2\text{-MIL-101(Fe)}$.

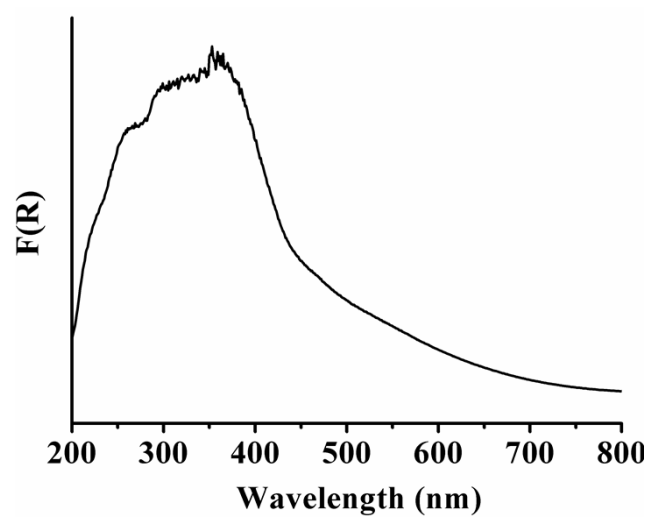


Fig. S5 The XRD patterns of fresh and used $\text{NH}_2\text{-MIL-101(Fe)}$.

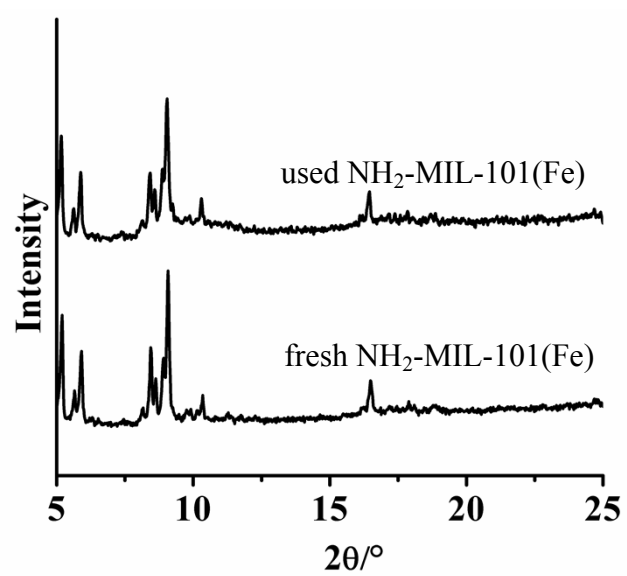


Fig. S6 The IR spectra of fresh and used $\text{NH}_2\text{-MIL-101(Fe)}$.

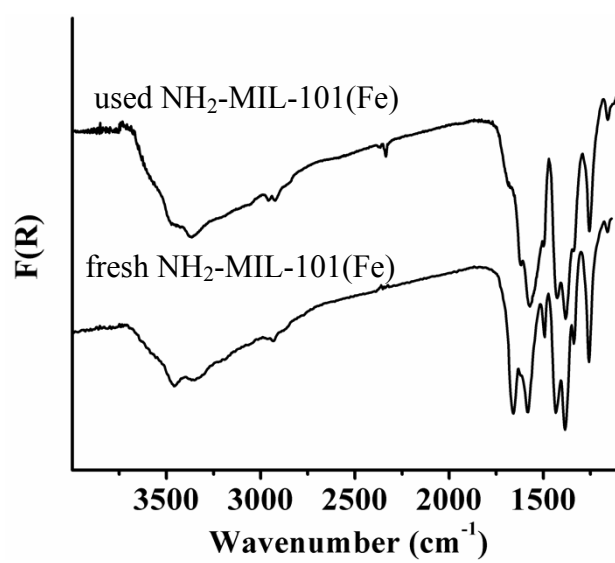


Fig. S7 The BET of fresh and used $\text{NH}_2\text{-MIL-101(Fe)}$.

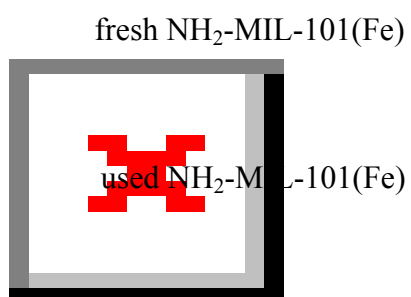


Fig. S8

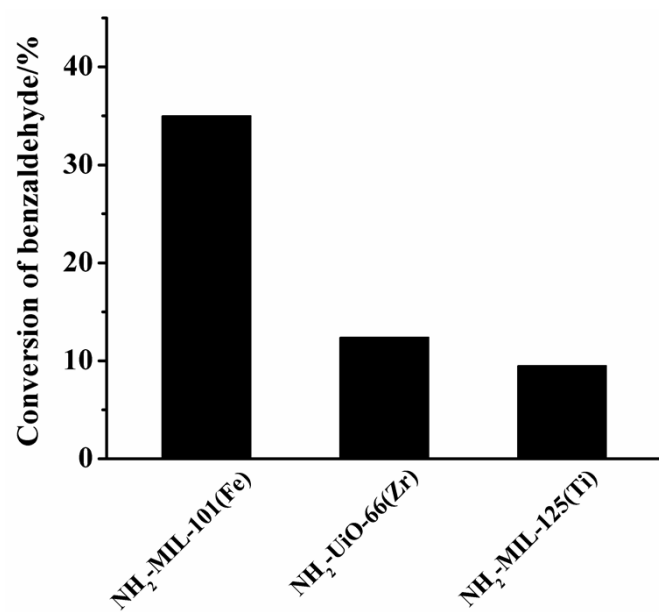


Table S1 Influence of the solvent on the photocatalytic oxidation of benzyl alcohol to form benzyl aldehyde.

entry	catalyst	solvent	conv. of alcohol (%)	selectivity (%)
1	NH ₂ -MIL-101(Fe)	benzotrifluoride	27	93
2	NH ₂ -MIL-101(Fe)	toluene	15	95
3	NH ₂ -MIL-101(Fe)	acetonitrile	21	92
4	NH ₂ -MIL-101(Fe)	B/A (10:1)	39	95
5 ^[a]	NH ₂ -MIL-101(Fe)	B/A (10:1)	0	0
6 ^[b]	No catalyst	B/A (10:1)	0	0

Reaction conditions: benzyl alcohol (0.1mmol), solvent (3mL), catalyst (20mg), O₂ (1atm), light irradiation ($\lambda \geq 420\text{nm}$), 40h. [a] Without visible light irradiation. [b] No catalyst.

Table S2 Photocatalytic oxidation of benzyl alcohol to form benzyl aldehyde over various catalysts.

entry	catalyst	substrate	conv. of alcohol (%)	selectivity (%)
1	NH ₂ -MIL-101(Fe)	benzyl alcohol	39	95
2	NH ₂ -UiO-66(Zr)	benzyl alcohol	28	91
3	NH ₂ -MIL-125(Ti)	benzyl alcohol	37	90

Reaction conditions: benzyl alcohol (0.1mmol), B/A (10:1, 3mL), catalyst (20mg), O₂ (1atm), light irradiation ($\lambda \geq 420\text{nm}$), 40h.