

**Influence of structural and electronic properties of organomolybdenum(II)
compounds of the type [CpMo(CO)₃R] and [CpMo(O₂)(O)R]
(R = Cl, CH₃, CF₃) on the catalytic oxidation of olefins**

Simone A. Hauser,^{a,‡} Robert M. Reich,^{a,‡} János Mink,^{b,c} Alexander Pöthig,^d Mirza Cokoja^d and
Fritz E. Kühn^{a,e*}

a Molecular Catalysis, Department of Chemistry and Catalysis Research Center, Technische Universität München, Lichtenbergstraße 4, D-85747 Garching bei München, Germany. Tel: +49 89 289 13096; Fax: +49 89 289 13473. Corresponding author: E-mail: fritz.kuehn@ch.tum.de

b Hungarian Academy of Sciences, Chemical Research Center, Pusztaszeri u. 59-67, 1025 Budapest, Hungary

c Faculty of Information Technology, University of Pannonia, Egyetem u. 10 8200 Veszprém, Hungary

d Catalysis Research Center, Technische Universität München, Ernst-Otto-Fischer-Straße 1, D-85747 Garching bei München, Germany

e Chair of Inorganic Chemistry, Department of Chemistry and Catalysis Research Center, Technische Universität München, Lichtenbergstraße 4, D-85747 Garching bei München, Germany

‡ These authors contributed equally to this work.

Supporting Information

Table of Contents

S1. X-Ray single crystallography

Table S.1. Crystallographic data for compounds 3 and 6.

S2. Computational results - Visualization, coordinates and energies

Figure S2.1. LUMOs of the compounds [CpMo(CO)₃R]

Table S2.1. Summary of computational results

S2.2 Calculated coordinates for [CpMo(CO)₃Cl] (**1**)

S2.2 Calculated coordinates for [CpMo(CO)₃CH₃] (**2**)

S2.2 Calculated coordinates for [CpMo(CO)₃CF₃] (**3**)

S3. Experimental and calculated fundamental frequencies and force constants

Table S3.1. Experimental and calculated fundamental frequencies for [CpMo(CO)₃CH₃] and [CpMo(CO)₃CD₃].

Table S3.2. Calculated force constants for [CpMo(CO)₃CH₃] and [CpMo(CO)₃CD₃].

Table S3.3 IR and Raman frequencies assigned to the spectroscopic modes of vibration for the complexes **1 – 3** and **2a**.

S1. X-Ray single crystallography

Data were collected on an X-ray single crystal diffractometer equipped with a CCD detector (APEX II, κ -CCD), a fine-focused sealed tube with MoK α radiation ($\lambda = 0.71073 \text{ \AA}$) and a graphite monochromator, by using the APEX2 software package. [1] The measurements were performed on single crystals coated with perfluorinated ether. The crystals were fixed on the top of a glass fiber and transferred to the diffractometer. Crystals were frozen under a stream of cold nitrogen. A matrix scan was used to determine the initial lattice parameters. Reflections were merged and corrected for Lorenz and polarization effects, scan speed, and background using SAINT. [2] Absorption corrections, including odd and even ordered spherical harmonics were performed using SADABS. [2] Space group assignments were based upon systematic absences, E statistics, and successful refinement of the structures. Structures were solved by direct methods with the aid of successive difference Fourier maps [3], and were refined against all data using the APEX 2 software [1] in conjunction with SHELXL-97 [5] and SHELXLE [6]. Methyl hydrogen atoms were refined as part of rigid rotating groups, with a C–H distance of 0.98 \AA and $U_{\text{iso}(\text{H})} = 1.5 \cdot U_{\text{eq}(\text{C})}$. Other H atoms were placed in calculated positions and refined using a riding model, with methylene and aromatic C–H distances of 0.99 and 0.95 \AA , respectively, and $U_{\text{iso}(\text{H})} = 1.2 \cdot U_{\text{eq}(\text{C})}$. If not mentioned otherwise, non-hydrogen atoms were refined with anisotropic displacement parameters. Full-matrix least-squares refinements were carried out by minimizing $\sum w(F_o^2 - F_c^2)^2$ with SHELXL-97 [5] weighting scheme. Neutral atom scattering factors for all atoms and anomalous dispersion corrections for the non-hydrogen atoms were taken from *International Tables for Crystallography*. [4] Images of the crystal structures were generated by PLATON. [7]

References:

- [1] APEX suite of crystallographic software. APEX 2 Version 2008.4. Bruker AXS Inc., Madison, Wisconsin, USA (2008).
- [2] SAINT, Version 7.56a and SADABS Version 2008/1. Bruker AXS Inc., Madison, Wisconsin, USA (2008).
- [3] Sheldrick, G. M. "**SHELXS-97**", Program for Crystal Structure Solution, Göttingen, (1997).
- [4] International Tables for Crystallography, Vol. C, Tables 6.1.1.4 (pp. 500-502), 4.2.6.8 (pp. 219-222), and 4.2.4.2 (pp. 193-199), Wilson, A. J. C., Ed., Kluwer Academic Publishers, Dordrecht, The Netherlands, 1992.
- [5] Sheldrick, G. M. "**SHELXL-97**", University of Göttingen, Göttingen, Germany, (1997).
- [6] Huebschle, C. B., Sheldrick, G. M. & Dittrich, B. "**SHELXLE**", *J. Appl. Cryst.* **2011**, *44*, 1281- 1284.
- [7] Spek, A. L. "**PLATON**", A Multipurpose Crystallographic Tool, Utrecht University, Utrecht, The Netherlands, (2010).

Table S.1. Crystallographic data for compounds **3** and **6**.

	3	6
CCDC	1030682	1030683
formula	C ₉ H ₅ F ₃ MoO ₃	C ₆ H ₅ F ₃ MoO ₃
fw	314.07	278.04
colour/habit	clearlight yellow needle	Intense yellow fragment
Cryst dimensions (mm ³)	0.07 × 0.11 × 0.55	0.08× 0.09× 0.34
crystsyst	triclinic	monoclinic
spacegroup	<i>P</i> 1	<i>C</i> 2/c
<i>a</i> , Å	6.4530(2)	20.5911(3)
<i>b</i> , Å	6.5440(2)	6.9325(1)
<i>c</i> , Å	6.7851(2)	12.3919(2)
α , deg	107.752(1)	90
β , deg	97.132(1)	117.372(1)
γ , deg	107.192(1)	90
<i>V</i> , Å ³	253.356(14)	1570.87(4)
<i>Z</i>	1	8
<i>T</i> , K	123(2)	123(2)
<i>D</i> _{calcd} , g cm ⁻³	2.059	2.351
μ , mm ⁻¹	1.326	1.693
<i>F</i> (000)	152	1072
θ range, deg	3.24 – 25.37	2.2 – 25.4
Index ranges (<i>h</i> , <i>k</i> , <i>l</i>)	±7, ±7, 0–8	±24, ±8,±14
no. of rflns collected	910	8754
no. of indep rflns/ <i>R</i> _{int}	910/0.0224	1447/0.0318
no. of obsd rflns (<i>I</i> >2σ(<i>I</i>))	905	1266
no. of data/restraints/params	910/3/145	1447/0/118
R1/wR2 (<i>I</i> >2σ(<i>I</i>)) ^a	0.0147/0.0381	0.0194/0.0431
R1/wR2 (all data) ^a	0.0147/0.0381	0.0247/0.0448
GOF (on <i>F</i> ²) ^a	1.111	1.043
Largest diff peak and hole (e Å ⁻³)	+0.468/-0.289	+0.595/-0.301

^a R1 = $\sum(|F_{\text{o}}| - |F_{\text{c}}|) / \sum |F_{\text{o}}|$; wR2 = $\{\sum [w(F_{\text{o}}^2 - F_{\text{c}}^2)^2] / \sum [w(F_{\text{o}}^2)^2]\}^{1/2}$; GOF = $\{\sum [w(F_{\text{o}}^2 - F_{\text{c}}^2)^2] / (n-p)\}^{1/2}$

S2. Computational results - Visualization, coordinates and energies

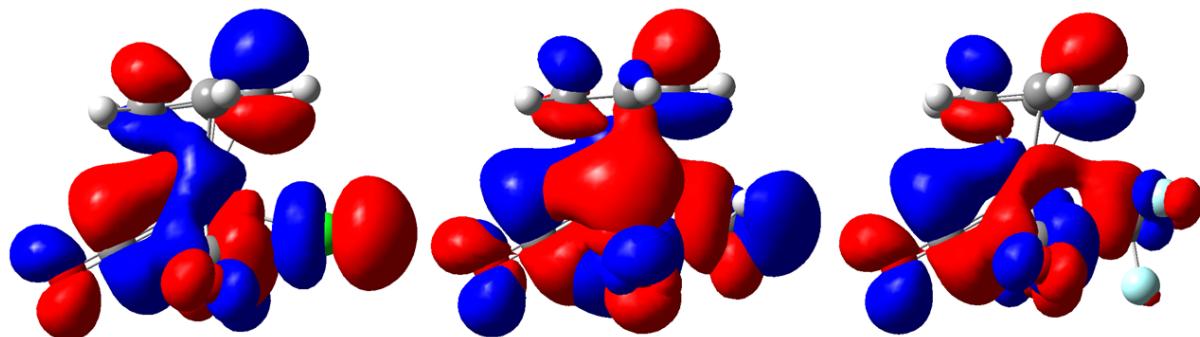


Figure S2.1.LUMOs of the compounds $[\text{CpMo}(\text{CO})_3\text{R}]$ with $\text{R} = \text{Cl}$ (**1**), CH_3 (**2**) and CF_3 (**3**) (from left to right) in gas phase (B3LYP/6-31+G** (d,p) level of theory.

Table S2.1.Summary of computational results

	compounds		
	1	2	3
<i>Mulliken charge on Mo</i>	1.025	1.702	1.891
<i>HOMO-LUMO-gap [eV]</i>	3.6635	4.4782	4.3740
<i>Sum of electronic and thermal Enthalpies [eV]</i>	-28897.4	-17458.7	-25561.2
<i>Sum of electronic and thermal Free Energies [eV]</i>	-28898.9	-17460.2	-25562.9
<i>Mo-Ligand-Polarity</i>	0.789	1.616	1.771

S2.2 Calculated coordinates for [CpMo(CO)₃Cl] (1)

C	-1.98799900	1.05031000	0.71919000
H	-1.98508600	1.92502200	1.35501300
C	-1.98771000	1.05159400	-0.71694400
C	-2.06754900	-0.30715800	1.15686900
H	-1.98474000	1.92741900	-1.35122900
C	-2.06703400	-0.30513400	-1.15698800
H	-2.10110700	-0.65004000	2.18174600
C	-2.12405500	-1.13024100	-0.00077400
H	-2.09986800	-0.64629600	-2.18246100
H	-2.13743400	-2.21087100	-0.00172400
O	1.77326100	-0.05259700	-2.63421800
O	1.25133800	2.97414800	0.00328700
O	1.77463300	-0.06037000	2.63328300
C	1.16564500	-0.01750400	1.65715000
C	0.79357700	1.91196600	0.00223500
C	1.16480900	-0.01283900	-1.65759400
Mo	0.00132100	0.06871700	0.00014700
Cl	0.85416300	-2.32745600	-0.00266000

S2.3 Calculated coordinates for [CpMo(CO)₃CH₃] (2)

C	-1.82019783	-0.71713523	1.25051445
H	-1.68775007	-1.35336194	2.11480344
C	-1.82030749	0.71630747	1.25077626
C	-2.07475634	-1.15596256	-0.08195241
H	-1.68796713	1.35224152	2.11529403
C	-2.07494141	1.15560462	-0.08152904
H	-2.15552611	-2.18289472	-0.41142726
C	-2.23331364	-0.00003801	-0.89774864
H	-2.15579313	2.18265678	-0.41061061
H	-2.43308789	0.00014462	-1.96024613
O	1.76516347	2.55224243	-0.68997555
O	1.89313294	0.00001597	2.52688384
O	1.76620962	-2.55153250	-0.69039205
C	1.16093503	-1.59677189	-0.43760916
C	1.21573630	-0.00000203	1.58706085
C	1.16031965	1.59716295	-0.43734229
Mo	0.03279605	-0.00003560	-0.03279794
C	0.45995179	0.00019976	-2.33987667
H	-0.00689862	-0.89233949	-2.76219103
H	-0.00705368	0.89273007	-2.76204857
H	1.52003777	0.00032076	-2.59195059

S2.4 Calculated coordinates for [CpMo(CO)₃CF₃] (3)

C	2.10216800	-1.37084900	-0.71527100
H	2.90086600	-1.01419600	-1.35121600
C	2.10117500	-1.36991600	0.72018900
C	0.88852200	-1.97754700	-1.15390000
H	2.89906300	-1.01259500	1.35677400
C	0.88686900	-1.97602300	1.15786000
H	0.58835200	-2.14202300	-2.17967100
C	0.14523900	-2.35034700	0.00168700
H	0.58508000	-2.13921900	2.18336400
H	-0.82615200	-2.82405600	0.00131900
O	-0.47600200	1.58099100	2.64711600
O	2.36994600	2.52566200	-0.00217300
O	-0.47411800	1.57210200	-2.65203800
C	-0.19743600	1.04759300	-1.66325000
C	1.64175200	1.62863600	-0.00125700
C	-0.19889300	1.05292500	1.66036400
Mo	0.37604900	0.05731300	0.00023700
C	-1.89849100	-0.13691100	-0.00011400
F	-2.36409900	-0.84309200	1.08911800
F	-2.36384100	-0.84526800	-1.08793700
F	-2.61931600	1.02275400	-0.00136000

S3. Experimental and calculated fundamental frequencies and force constants

Table S3.1. Experimental and calculated fundamental frequencies for [CpMo(CO)₃CH₃] and [CpMo(CO)₃CD₃].

	[CpMo(CO) ₃ CH ₃]		[CpMo(CO) ₃ CD ₃]		Potential Energy Distribution ^d (%)	Description of mode
	Exper.	Calcd.	Exper.	Calcd.		
A'₁	2012 ^a	2010	2010	2004	83 v _s CO + 10 v _s MoC	CO sym. stretching
	1903	1890	1900	1890	81v _a CO + 11 v _a MoC	CO asym. stretching
	598	592	602	592	55 β _s MoCO + 25 β _a MoCO ^c + 8 δ _s CMoC	MoCO bending
	562	562	562	562	61 β _a MoCO + 21 β _s MoCO + v _s MoC	MoCO bending
	502	503	498	503	47 v _s MoC + 14 γ _a MoCO + 13 v _a MoC	MoC stretching
	485	484	489	484	48 γ _a MoCO + 15 v _s MoC + 10 vMoCp	MoCO bending
	452	450	450	447	53 v _a MoC + 22 γ _a MoCO + 12 vMoCp	MoC stretching
	406	408	392	388	70 vMo-Me + 17 v _a MoC + 9 v _s MoC	Mo-Me stretching
	335	335	333	333	83 vMoCp + 10 v _s MoC + 5 vMo-Me	Mo-Cp stretching
	131 ^b	145	130	139	61 δ _s CpMoMe + 13 δCMoCp + 10 βMoCO	CpMoMe deformation
	107	110	104	108	43 δ _s MoC ₃ + 32 δCMoCp	MoC ₃ sym. deformation
	90	99	90	99	41 δ _a CMoC + 32 δ _a CMoCp + 16 δCpMoMe	CMoC deformation
	~60	66	~60	66	57 δCMoCp + 26 δCpMoMe	CMoCp deformation
A''₁₄	1920	1911	1907	1900	92 v _a CO + 8 v _a MoC	CO asym. stretching
	562	560	562	560	87 β _a MoCO + 6 δ _a CMoCp	MoCO deformation
	502	508	498	508	72 v _a MoC + 11 γ _a MoCO + 6 v _a CO	MoC stretching
	465	463	470	463	61 γ _a MoCO + 19 δ _a CMoC + 9 δ _s MoCO	MoCO deformation
	438	432	429	432	85 γ _a MoCO + 5 v _a CO + 4 δ _a CMoC	MoCO deformation
	130	144	130	136	δ _a CMoMe + 43 δ _a CMoC	CMoMe deformation
	-	120	~120	120	91 δCMoCp + 5 β _a MoCO	CMoCp deformation
	107	104	90	104	76 δ _a CMoC + 21 γ _a MoCO	CMoC deformation

^aAveraged IR and Raman frequencies were used as fundamentals when the wavenumbers were slightly different or split due to the solid-state effects.

^bExperimental frequencies below 150 cm⁻¹ were assigned on the basis of normal coordinate calculations.

^c The β(MoCO) are the linear bendings of Mo-C-O groups in the plane of CMoCp, the notation of γ(MoCO) requests the linear bendings perpendicular to the plane of CMoCp.

^dPotential Energy Distributions (PED) were taken from the result of [CpMo(CO)₃CH₃] calculations.

Table S3.2. Calculated force constants for $[\text{CpMo}(\text{CO})_3\text{CH}_3]$ and $[\text{CpMo}(\text{CO})_3\text{CD}_3]$.

Force constants	Groups involved and descriptions	Numerical values of force constants	Units
$K_1(\text{C}'\text{O})$	$\text{C}'\text{O}$ opposite to CH_3/CD_3	13.969	a
$K_2(\text{CO})$	CO close to CH_3/CD_3	14.396	a
$F_s(\text{CO},\text{CO})$	Interaction between CO, CO groups	0.555	a
$F_l(\text{C}'\text{O},\text{CO})$	Interaction between $\text{C}'\text{O},\text{CO}$ groups	0.278	a
$K(\text{MoC})$	Mo-CO	3.108	a
$F(\text{MoC},\text{MoC})$	Mo-CO	0.011	a
$K(\text{Mo-Me})$	Mo- CH_3 , Mo- CD_3	1.527	a
$K(\text{Mo-Cp})$	Mo-Cp	3.110	a
$H(\text{MoCO})$	MoCO linear bending (in plane)	0.865	b
$h(\text{MoCO},\text{MoCO})$	MoCO , MoCO interaction	0.008	b
$H'(\text{MoCO})$	MoCO linear bending (out of plane)	0.534	b
$h'(\text{MoCO},\text{MoCO})$	MoCO, MoCO interaction	0	b
$H(\text{CMoMe})$	skeletal bending	(0.55) ^c	b
$H(\text{CMoC})$	skeletal bending	(0.62)	b
$H(\text{CMoCp})$	skeletal bending	(0.48)	b
$H(\text{CpMoMe})$	skeletal bending	(0.50)	b
$F(\text{MoMe},\text{MoC}')$	stretch-stretch interaction	(0.10)	a
$F(\text{MoMe},\text{MoC})$	stretch-stretch interaction	(0.05)	a

Units of force constants are:

a, 10^2 N m^{-1} ;b, $10^{-18} \text{ N m rad}^{-2}$;

c, The constrained values are listed in brackets.

Table S 3.3 IR and Raman frequencies assigned to the spectroscopic modes of vibration for the complexes **1 – 3** and **2a** [$\text{CpMo}(\text{CO})_3\text{CD}_3$].

1		2		2a		3		Assignment
IR	Raman	IR	Raman	IR	Raman	IR	Raman	
3123m	3127m	3113m	3123m	3113m	3123m	3126m	1303m	$\nu_s(\text{CH})$, Cp
3118m								$\nu_a(\text{CH})$, Cp
3103m	3111w		3100w		3102w		3108w	$\nu_a(\text{CH})$, Cp
		2981m	2982m	2233m	2236m			$\nu_a(\text{CH}_3/\text{CD}_3)$
		2902m	2905m	2116m	2114m			$\nu_s(\text{CH}_3/\text{CD}_3)$
		2020w,sh	2019m	2018w,sh	2018m			$\nu_s(\text{C}\equiv\text{O})$
						2060w,sh	2056m	$\nu_s(\text{C}\equiv\text{O})$
2040vs	2041vs	2006s	2003s	2004s	2002s	2048s	2042s	$\nu_s(\text{C}\equiv\text{O})$
	1974w			1948w,sh	1945w,sh	1980w,sh	1977s	$\nu_a(\text{C}\equiv\text{O})$
1969vs	1960s	1920m,sh	1921s,sh	1920m,sh	1920s,sh	1963s	1962vs	$\nu_a(\text{C}\equiv\text{O})$
	1949vs							$\nu_a(\text{C}\equiv\text{O})$
1932vs	1930m	1903vs	1900vs	1901vs	1889vs	1934vs	1932s	$\nu_a(\text{C}\equiv\text{O})$
1924s,sh						1916m,sh		$\nu_a(\text{C}\equiv\text{O})$
1421m	1422vw	1423m	1425m	1423m	1421m,b	1430m	1427m	$\nu_a(\text{CC})$, Cp
		1423m	1420m	1062m	1061m			$\delta_a(\text{CH}_3/\text{CD}_3)$
1354w	1355vw	1354w	1352w	1353w	1352w		1356w	$\nu_a(\text{CC})$, Cp
		1161m	883m	881s				$\delta_a(\text{CH}_3/\text{CD}_3)$
1110vw	1109s	1109vw	1107s	1109vw	1107s	1113vw	1109s	$\nu_s(\text{CC})$, Cp
1064m	1061m	1060m	1061m	1062m	1061m		1069m	$\beta(\text{CH})$, Cp
						1050s	1057m	$\nu_a(\text{CF}_3)$
						1043s		$\nu_a(\text{CF}_3)$
1014m		1011m	1011vw	1012m	1009vw	1012s		$\beta(\text{CH})$, Cp
1005m	1007vw				1003vw	1004s	1005vw	$\beta(\text{CH})$, Cp
						982s	978vw	$\nu_s(\text{CF}_3)$
822s	816vw	825s	822vw	822s	821vw	831m	827m	$\gamma(\text{CH})$, Cp
		613w	608vw	488vs		695m	692m	$\rho_a(\text{CF}_3)$
				475s	475s			$\rho(\text{CH}_3/\text{CD}_3)$
601w	~600vw	613w	608vw	614w,sh		613w		$\delta_a(\text{MoCO})$
561s	560vw	587s	584vw	597s	596w	573s	571w	$\delta_a(\text{MoCO})$
524s	~520vw	562vs	562m	561vs	563vw	546vs	544w	$\beta_a(\text{MoCO})$
		755w	758vw	548s,sh	549vw			$\rho(\text{CH}_3/\text{CD}_3)$
						525w	521w	$\delta_a(\text{CF}_3)$
						512w		$\delta_a(\text{CF}_3)$
471s	464w	502w,sh		498m,sh	498w,sh	479s	477w	$\nu_s(\text{MoC})$
471s	464w	489vs	482m	488vs	490w,sh	479s	477w	$\gamma_a(\text{MoCO})$
		465w,sh		463w,sh	476ms	455ms	458ms	$\beta_a(\text{MoCO})$
432w	428vw							$\beta_a(\text{MoCO})$
432w	428vw	451s	453m	448s	451s	430vs	432m	$\nu_a(\text{MoC})$
415m	420vw	437m	438vs	429ms	429s	404m	402m	$\gamma(\text{MoCO})$
		405w	408ms	392m	391m			$v(\text{MoC})$, CH_3/CD_3
	383w,sh							$\nu_a(\text{MoCp})$, tilt
	371m	355w	355s	353m	352vs	366w	361w,sh	$\nu_a(\text{MoCp})$, tilt
361s ^a	350w,sh	335w	334vs	334w	333vs			$\nu_s(\text{MoCp})$
337vw ^a	338ms					352vw	350vs	$\nu_s(\text{MoCp})$
281vs ^a	278w							$v(\text{MoCl})$
						251s	249s	$v(\text{MoC})$, CF_3
						234m	233m	$\rho(\text{CF}_3)$
		173s	125vw,sh	160s				$\tau(\text{CH}_3/\text{CD}_3)$
		160m		148m				$\tau(\text{CH}_3/\text{CD}_3)$
	130w,sh	133w		130w,sh		137w	137w,m	$\delta_a(\text{CpMoR})$
	107w		104w			119w	119w	$\delta_a(\text{CMoC})$
		90w,sh		90w,sh		106vw		$\delta_a(\text{CMoC})$
	60vw					60vw		$\delta_a(\text{CMoCp})$

a Far infrared data taken from ref. 25

b Notation of fundamental modes: v – stretching; δ – bending or deformation; ρ – rocking; β – in plane, γ – out of plane deformation; τ – torsion; subscript 's' symmetric, subscript 'a' anti-symmetric.