Supporting Information for

## Ionophore-based pH Independent Detection of Ions Utilizing Aggregation-Induced Effects

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TPE<sup>+</sup>

TMRE<sup>+</sup>

 $F_3C$   $F_3C$   $F_3C$   $F_3C$   $F_3C$   $F_3C$   $F_3$   $F_3C$   $F_3$   $F_3C$   $F_3$   $F_3C$   $F_3$   $F_3C$   $F_3$   $F_3$   $F_3C$   $F_3$   $F_3$ 

TFPB







K<sup>+</sup> ionophoreNa<sup>+</sup> ionophorePb<sup>2+</sup> ionophoreCa<sup>2+</sup> ionophoreScheme S1. Chemical structures of the sensing components incorporated into the ion-selective<br/>nanospheres.Na<sup>+</sup> ionophoreNa<sup>+</sup> ionophore

## **Table S1.** Composition of the cocktail solutions used to prepare the eight ion-selective nanosensors.(a) AIE-based nanosensors

	Ionophore		TFPB		TPE		F127	PVC	DOS
	mg /mL	mmol /mL	mg /mL	mmol /mL	mg /mL	mmol /mL	mg /mL	mg /mL	mg /mL
K+	0.90	0.81	0.5	0.56	0.25	0.53	2	2	2
Na+	1.00	1.00	0.5	0.56	0.25	0.53	2	2	2
Ca <sup>2+</sup>	0.50	1.08	0.5	0.56	0.25	0.53	2	2	2
Pb <sup>2+</sup>	1.00	0.95	0.5	0.56	0.25	0.53	2	2	2

## (b) ACQ-based nanosensors

	lonophore		TFPB		TM	IRE	F127	PVC	DOS
	mg /mL	mmol /mL	mg /mL	mmol /mL	mg /mL	mmol /mL	mg /mL	mg /mL	mg /mL
K⁺	0.90	0.81	0.5	0.56	0.25	0.49	2	2	2
Na⁺	1.00	1.00	0.5	0.56	0.25	0.49	2	2	2
Ca <sup>2+</sup>	0.50	1.08	0.5	0.56	0.25	0.49	2	2	2
Pb <sup>2+</sup>	1.00	0.95	0.5	0.56	0.25	0.49	2	2	2



**Figure S1.** Fluorescence spectra of suspensions containing 1  $\mu$ M of TPE<sup>+</sup> (a) or TMRE<sup>+</sup> (b) upon gradual increase of TFPB in the nanospheres. (c) Normalized emission ( $I/I_0$ ) as a function of the ratio between the number of moles of added TFPB and the number of moles of TPE<sup>+</sup> (or TMRE<sup>+</sup>). Arrows indicate the increase of TFPB.



**Figure S2.** Fluorescence  $K^+$  responses in de-ionized water of the nanospheres containing the  $K^+$  ionophore valinomycin, TFPB, TPE<sup>+</sup>.  $I_0$  is the maximum emission intensity in blank solutions.



**Figure S3.** Responses of the ion-selective nanospheres containing different ionophores including sodium ionophore X, calcium ionophore II, and lead ionophore IV. Nanospheres containing  $TMRE^+$  are shown in a), c), and e), while nanospheres containing  $TPE^+$  are shown in b), d), and f).  $I_0$  represents the maximum emission intensity in blank solutions without target ions. Error bars represent standard deviations from three separate measurements.



**Figure S4.** Calibrations for the measurements of sodium and potassium ion concentrations in urine and blood serum samples using the Na<sup>+</sup> and K<sup>+</sup> nanosensors in the AIE and ACQ modes, respectively. The orange dots represent signals from the samples. Error bars represent standard deviations from three separate measurements.