

SUPPORTING INFORMATION

Specification of Nitrogen Functional Group in Hydrotreated Petroleum Molecule by Hydrogen/Deuterium Exchange Electrospray Ionization High-Resolution Mass Spectrometry

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Table S1. Model compounds information.

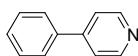
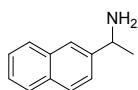
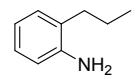
No.	Name	Structure	Formula	Exact mass	Purchased From
1	Quinoline		C ₉ H ₇ N ₁	129.05785	Sigma-Aldrich
2	5,6,7,8-tetrahydroquinoline		C ₉ H ₁₁ N ₁	133.08915	Aladdin
3	4-phenylpyridine		C ₁₁ H ₉ N ₁	155.07350	Aladdin
4	Decahydroquinoline		C ₉ H ₁₇ N ₁	139.13610	Aladdin
5	1,2,3,4-tetrahydroquinoline		C ₉ H ₁₁ N ₁	133.08915	Aladdin
6	Indoline		C ₈ H ₉ N ₁	119.07350	Aladdin
7	1-(naphthalen-2-yl)ethan-1-amine		C ₁₂ H ₁₃ N ₁	171.10480	Aladdin
8	2-propylaniline		C ₉ H ₁₃ N ₁	135.10480	TCI

Table S2. Elemental analysis of CGO and VR

Elemental Composition	CGO	Heavy Oil						
		VR I			VR II			
	HT Product	Feedstock	HT Product	Feedstock	VR II -R1	VR II -R2	VR II -R3	VR II -R4
C, wt%	85.37	86.40	85.57	87.03	86.78	87.03	86.91	86.81
H, wt%	13.15	12.76	13.22	11.36	11.49	11.57	11.87	12.34
O, wt%	1.41	1.03	0.86	1.07	1.14	0.84	0.80	0.58
N, ppm	709	3200	2700	5272	5813	5468	4159	2693
S, ppm	17	5100	1000	19262	14030	6064	2278	1396

Table S3. Relative abundance of model compounds dissolved in DT-DM, DT-DM+D₂O, DT-DM+DCOOD, DT-DM+DCOOD+D₂O.

No.	Class	Compounds	DM-DT				DM-DT+D ₂ O				DM-DT+DCOOD				DM-DT+D ₂ O+DCOOD			
			d ₀ H ⁺	d ₀ D ⁺	d ₁ D ⁺	d ₂ D ⁺	d ₀ H ⁺	d ₀ D ⁺	d ₁ D ⁺	d ₂ D ⁺	d ₀ H ⁺	d ₀ D ⁺	d ₁ D ⁺	d ₂ D ⁺	d ₀ H ⁺	d ₀ D ⁺	d ₁ D ⁺	d ₂ D ⁺
1		Quinoline	30.00	100			23.14	100			24.15	100			8.26	100		
2	Pyridine	5,6,7,8-tetrahydroquinoline	30.14	100			22.09	100			23.84	100			9.07	100		
3		4-phenylpyridine	33.35	100			23.79	100			26.77	100			10.48	100		
4	Non-aromatic cyclic amine	Decahydroquinoline	1.03	14.68	100		0.66	11.89	100		3.00	30.35	100		0.59	14.57	100	
5		1,2,3,4-tetrahydroquinoline	23.01	99.61	100		5.95	60.00	100		6.61	55.33	100		0.32	10.88	100	
6		Indoline	9.48	55.78	100		2.68	24.80	100		1.43	19.66	100		0.51	9.10	100	
7	non-aniline	1-(naphthalen-2-yl)ethan-1-amine	0.37	17.37	71.36	100	0.28	11.24	59.61	100	0.11	7.76	49.82	100	0.00	0.78	19.84	100
8	aniline	2-propylaniline	2.60	44.08	100	94.38	1.31	11.79	51.18	100	0.67	8.95	44.57	100	0.00	0.81	11.90	100

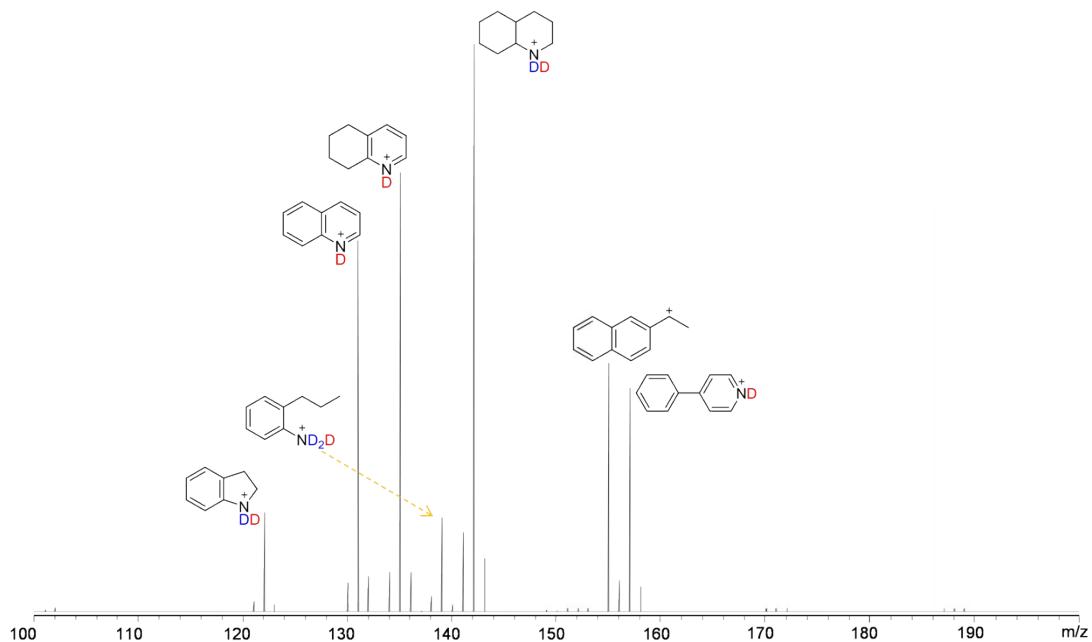


Figure S1. The (+) ESI Orbitrap mass spectra of mixed model compounds with the same molar mass. The aliphatic cyclic amine has higher relative abundance than others, but anilines have lower ionization efficiencies. Due to the easier fragmentation of the 1-(naphthalen-2-yl)ethan-1-amine's C-N bond, the relative abundance of N1H3 is extremely low, and the mainly expressed of the amine as $[\text{M}-\text{NH}_2]^+$.

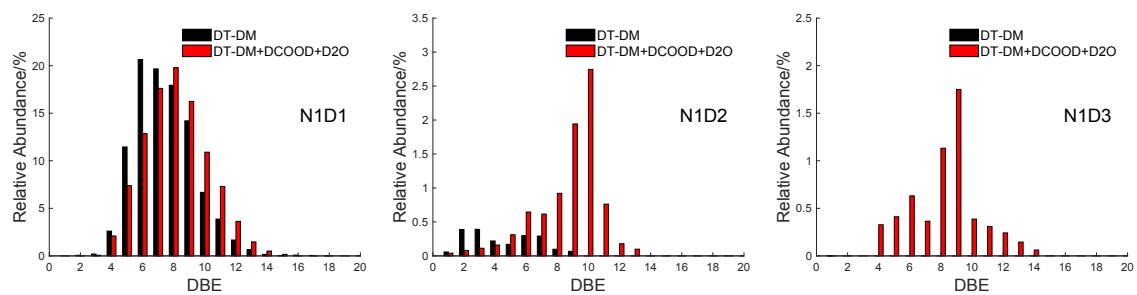


Figure S2. DBE distributions of (A) N1D1, (B) N1D2 and (C) N1D3 of the CGO dissolved in DT-DM or DT-DM+DCOOD+D₂O detected by (+) ESI FT-ICR MS.

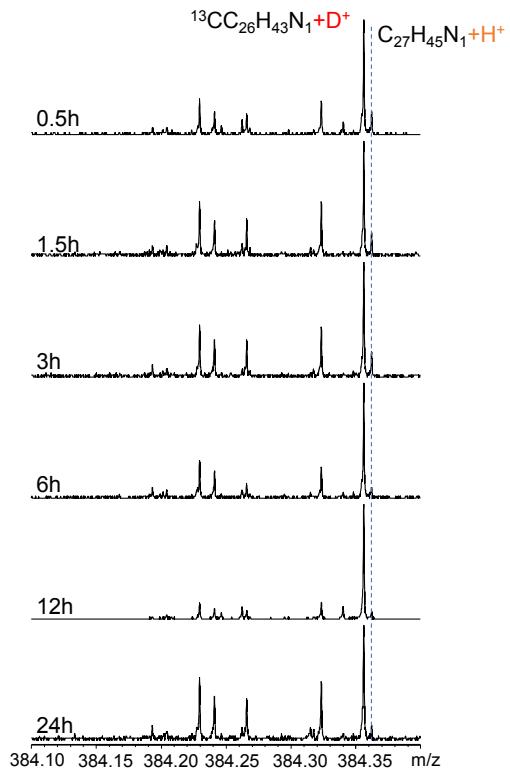


Figure S3. Expanded (+) ESI FT-ICR MS spectra for VGO dissolved in DT-DM with different HDX reaction time. Reaction time is 0.5 h, 1.5 h, 3 h, 6 h, 12 h or 24 h. Prolonging the reaction time can slightly reduce the relative abundance of N1H1.

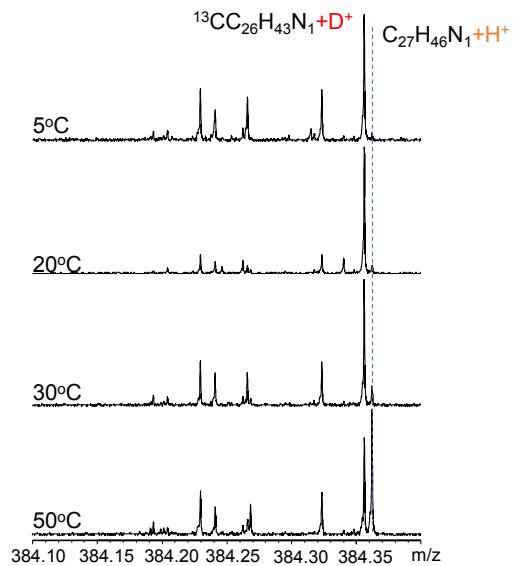


Figure S4. Expanded (+) ESI FT-ICR MS spectra and the relative abundance of N1H1 for crude oil's distillation cut dissolved in DT-DM with different reaction temperature. 5 °C means reacted in refrigerator at 5 °C for 30 minutes. 20 °C means reacted at room temperature for 30 minutes. 30 °C and 50 °C means reacted on hot plates of 30 °C and 50 °C for 30 minutes, respectively. The increase of the reaction temperature makes relative abundance of N1H1 increase significantly, but the relative abundance change of the N1H1 is negligible at room temperature and refrigeration conditions.

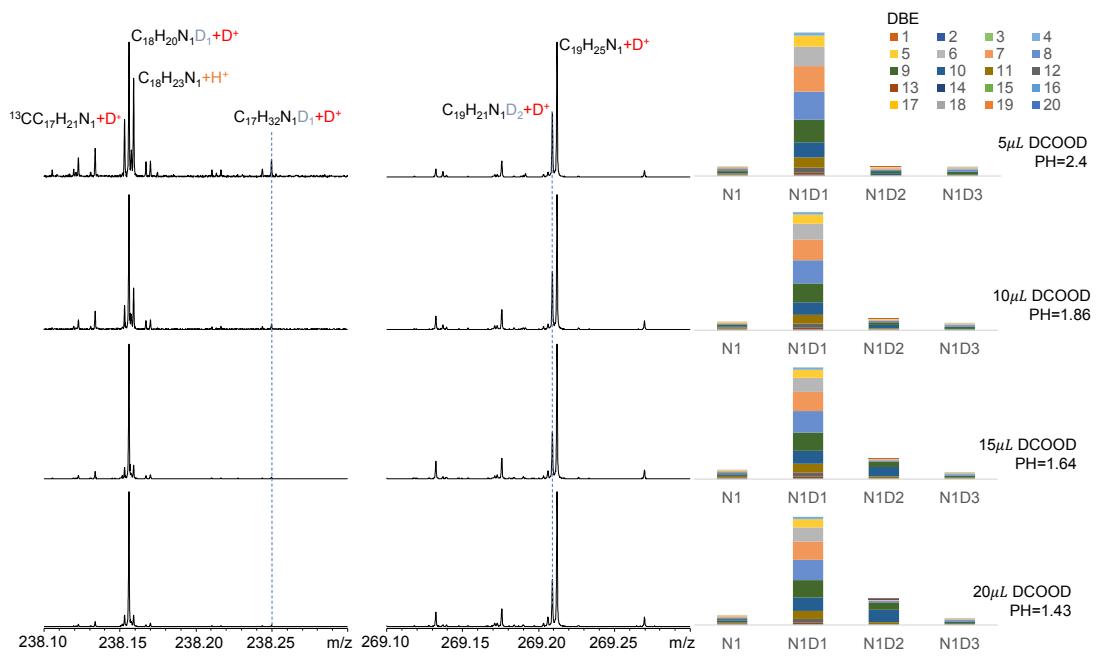


Figure S5. Expended mass spectra and the relative abundance distribution of N1 class species of the CGO dissolved in DT-DM+DCOOD+D₂O with different DCOOD addition detected by (+) ESI FT-ICR MS. The increase of the DCOOD addition significantly increase the relative abundance of N1D2 class speices.

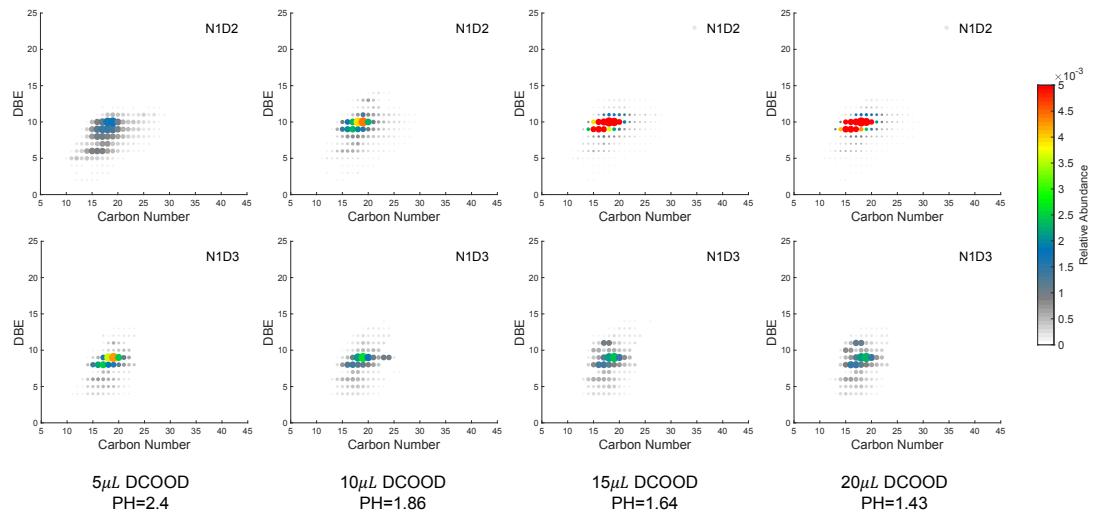


Figure S6. Plots of DBE versus carbon number for N1D2 and N1D3 classes species of the CGO dissolved in DT-DM+DCOOD+D₂O with different DCOOD addition detected by (+) ESI FT-ICR MS.

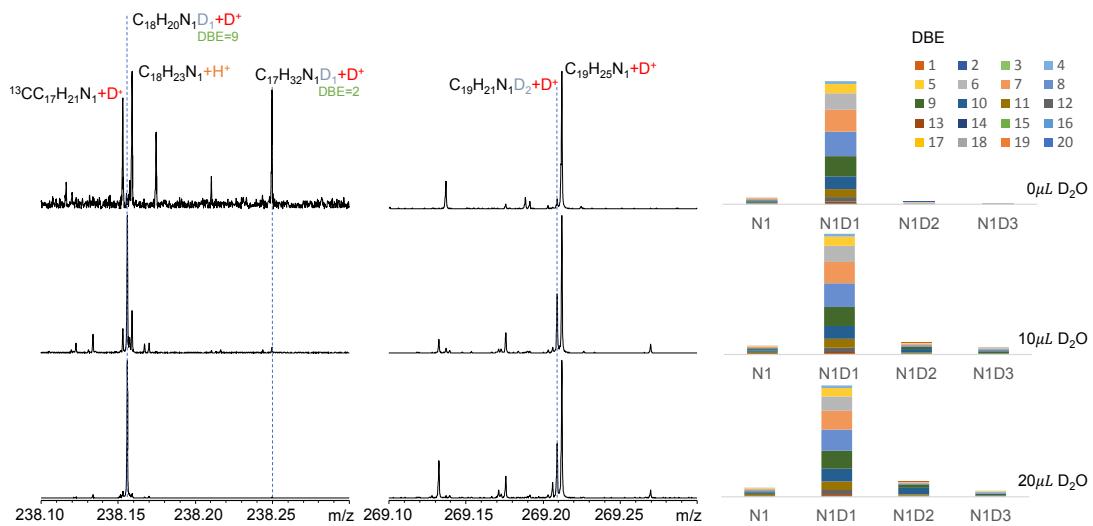


Figure S7. Expanded mass spectra and the relative abundance distribution of N1 class species of the CGO dissolved in DT-DM+DCOOD+ D_2O with different D_2O addition detected by (+) ESI FT-ICR MS. The increase of the D_2O addition makes relative abundance of N1D2 and N1D3 increase significantly, but when the addition is 20 μL , the heavy oil solution will be opacified.

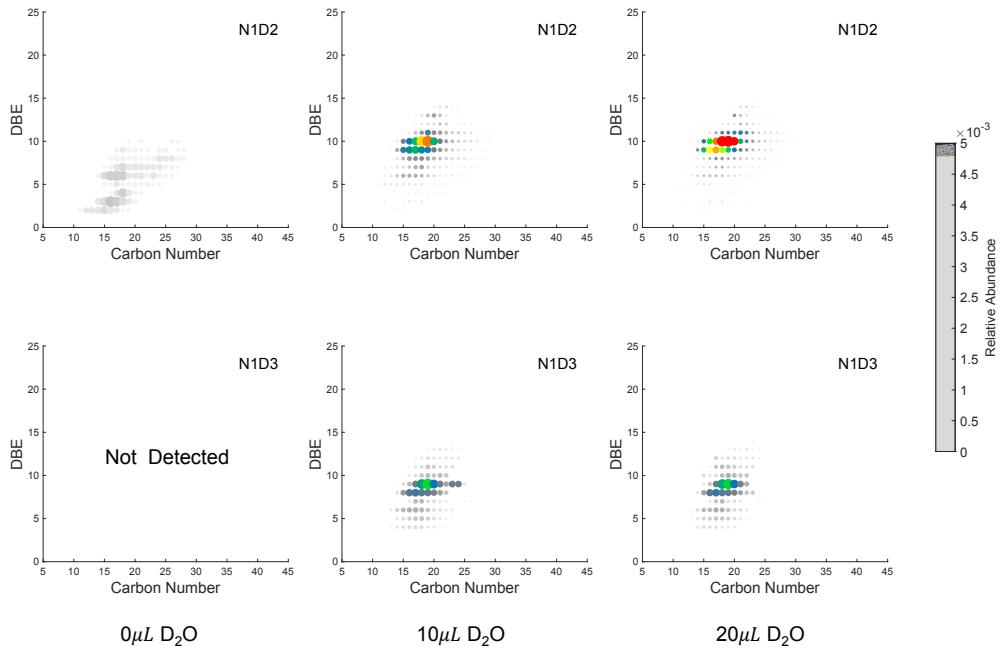


Figure S8. Plots of DBE versus carbon number for N1D2 and N1D3 classes species of the CGO dissolved in DT-DM+DCOOD+D₂O with different D₂O addition detected by (+) ESI FT-ICR MS. In order to detected more N1D2 and N1D3 peaks, the 10 μL DCOOD and 10 μL D₂O were chosen to be added.

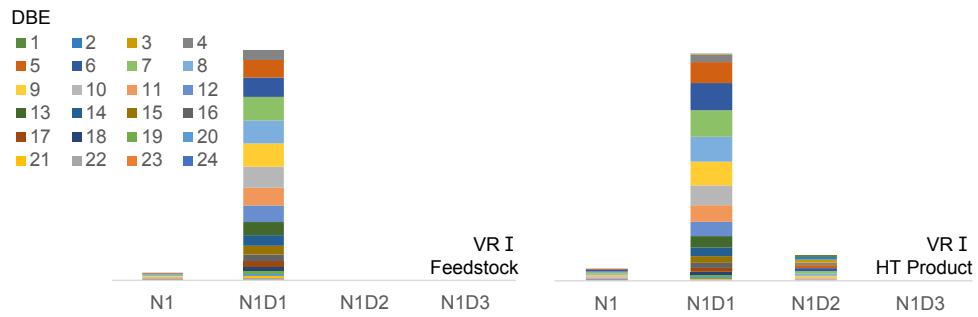


Figure S9. The relative abundance distribution of N1 class species of the VR I before and after hydrotreating detected by HDX (+) ESI FT-ICR MS.

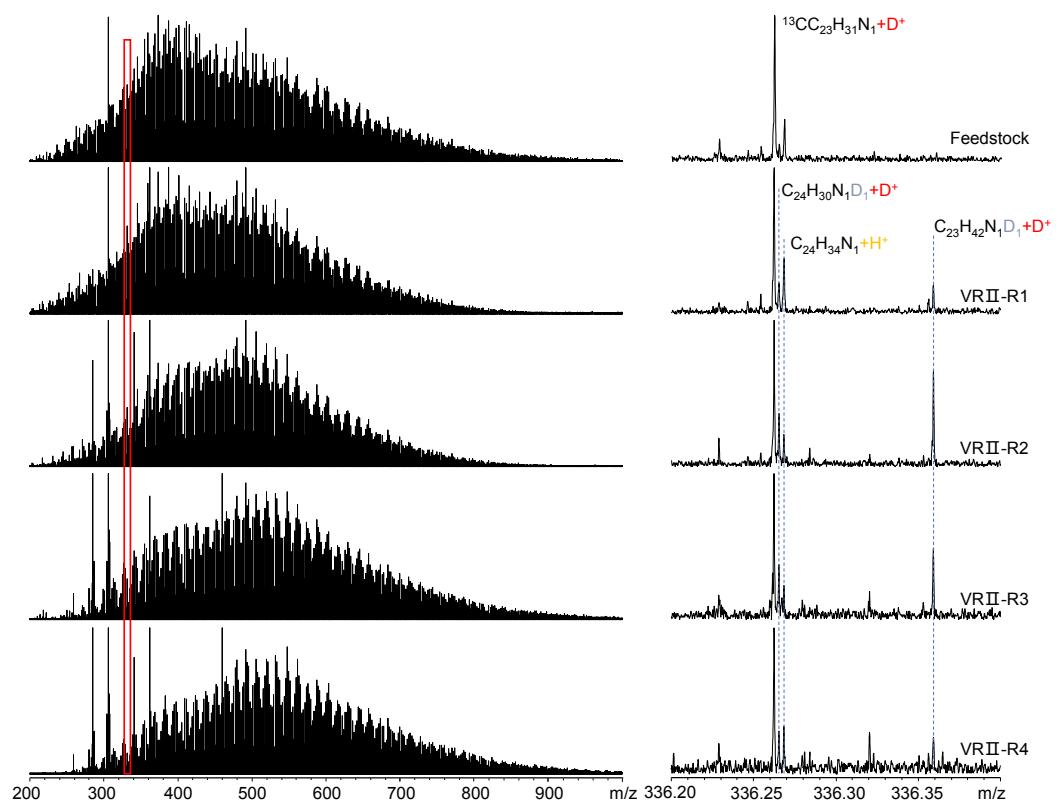


Figure S10. Broadband and expanded HDX (+) ESI FT-ICR MS spectra of the VR II before and after hydrotreating.

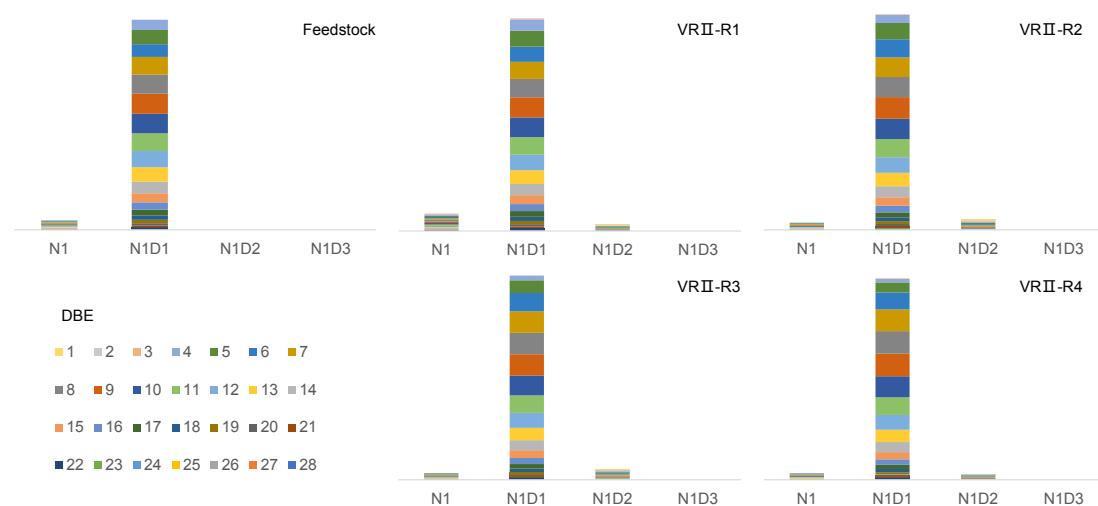


Figure S11. The relative abundance distribution of N1 class species of the VR II before and after hydrotreating detected by HDX (+) ESI FT-ICR MS.