

Supporting Information

Aqueous dependent sensing of hydrazine and phosphate anions using bis-heteroleptic Ru(II) Complex with phthalimide anchored pyridine-triazole ligand

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Instruments

The NMR spectra were carried out in Bruker Avance 500 spectrometers using CDCl_3 , CD_3CN , and DMSO-d₆ solvent (TMS as internal standard) in ppm. The FT-IR spectra were recorded on a Thermo Scientific Nicolet iS5 FT IR Spectrometer. The CHN elemental analysis was carried out from the Vario EL III microanalyzer. UV-Vis and Fluorescence spectra were carried out on Specord S 600 diode-array UV-Vis spectrophotometer and Shimadzu RF-5301 PC spectrofluorophotometer, respectively in RT. The luminescence lifetime was studied with the Time-Correlated Single Photon Counting method (TC-SPC) in a Horiba FluoroMax-4 instrument. The CV and DPV experiment were recorded on an Autolab three-electrode system, where Ag/AgCl as a reference electrode, glassy carbon as a working electrode, and platinum wire as a counter electrode in an acetonitrile solvent containing 0.1 M tetrabutylammonium hexafluorophosphate ($^4\text{Bu}_4\text{N}^+\text{PF}_6^-$) as the supporting electrolyte with 100 mV scan rate and the system was calibrated using ferrocene/ferrocenium ion (Fc/Fc^+) standard.

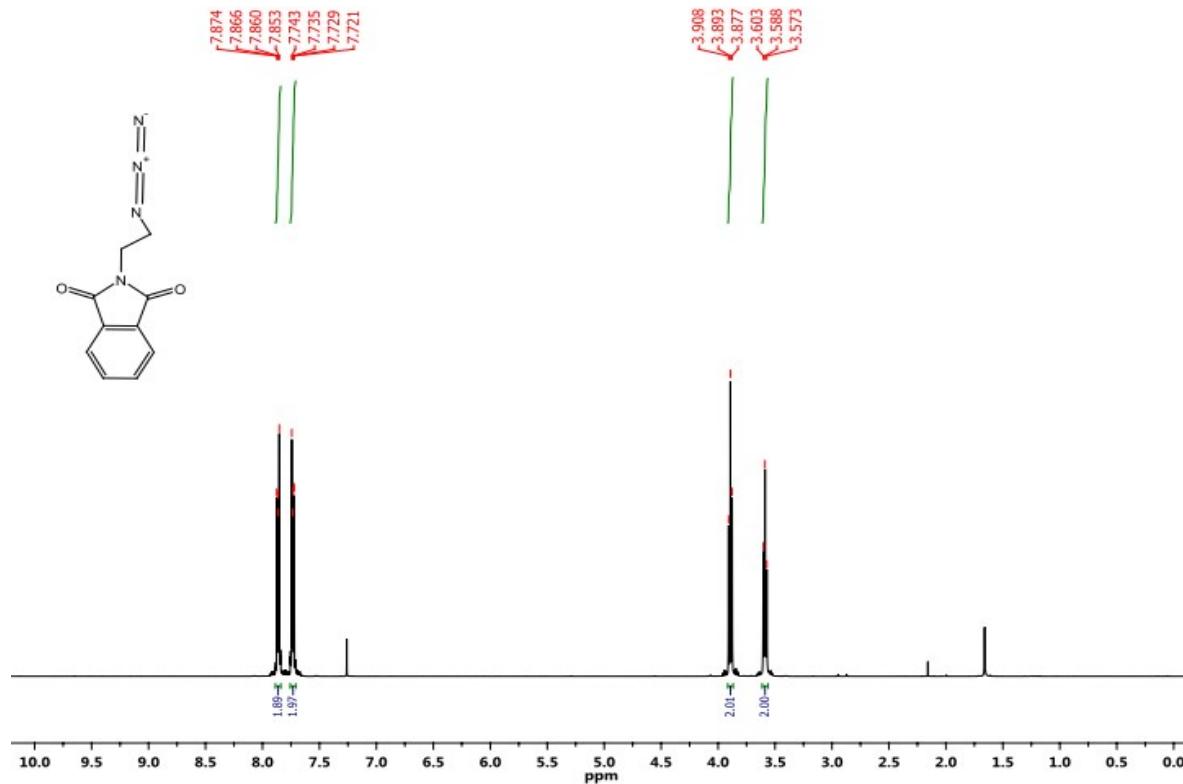


Fig.S1 ^1H NMR spectrum of TpN_3 in CDCl_3

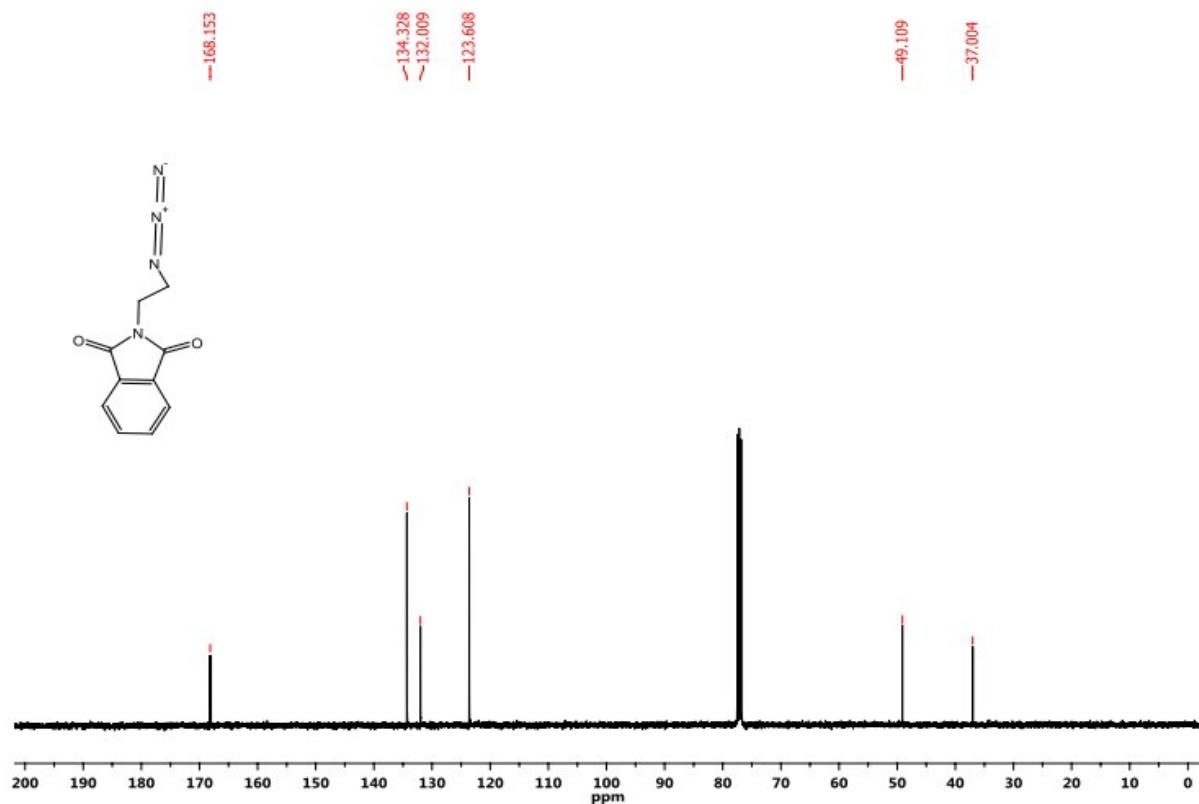


Fig.S2 ^{13}C NMR spectrum of TpN_3 in CDCl_3

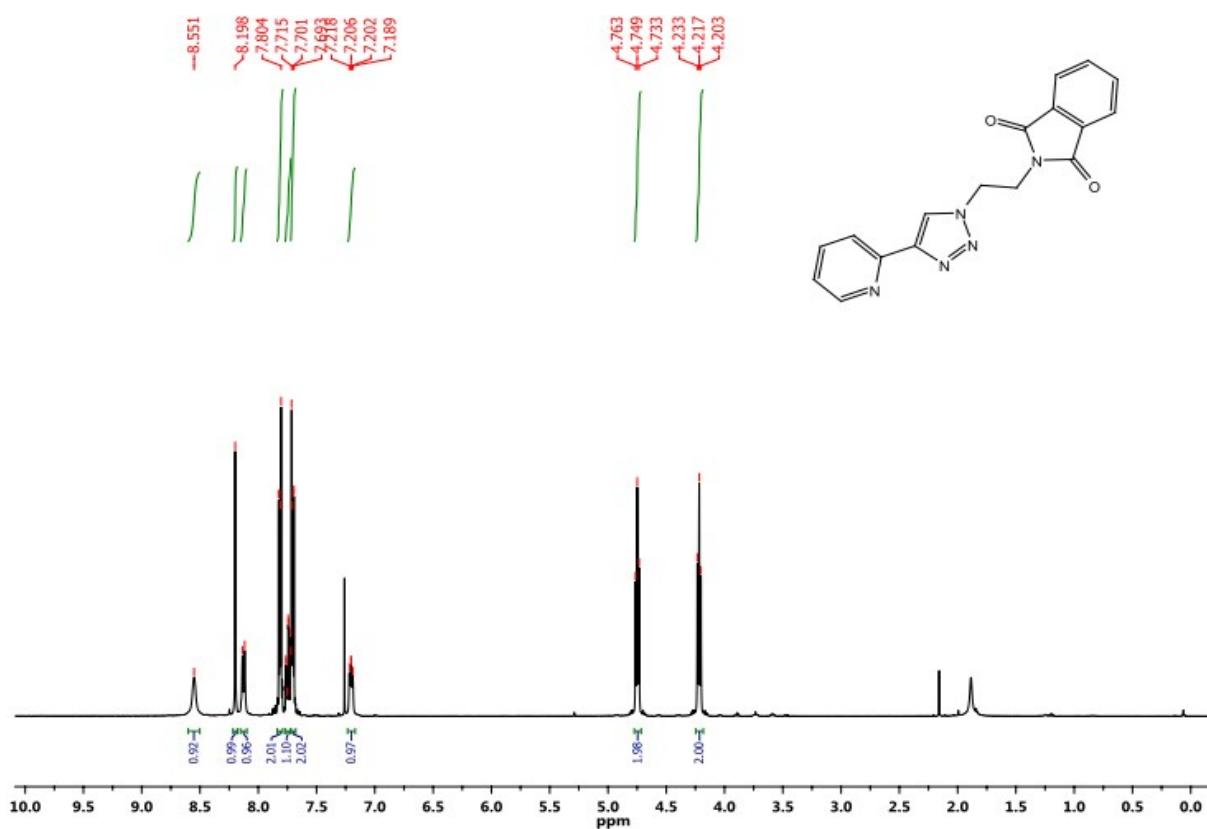
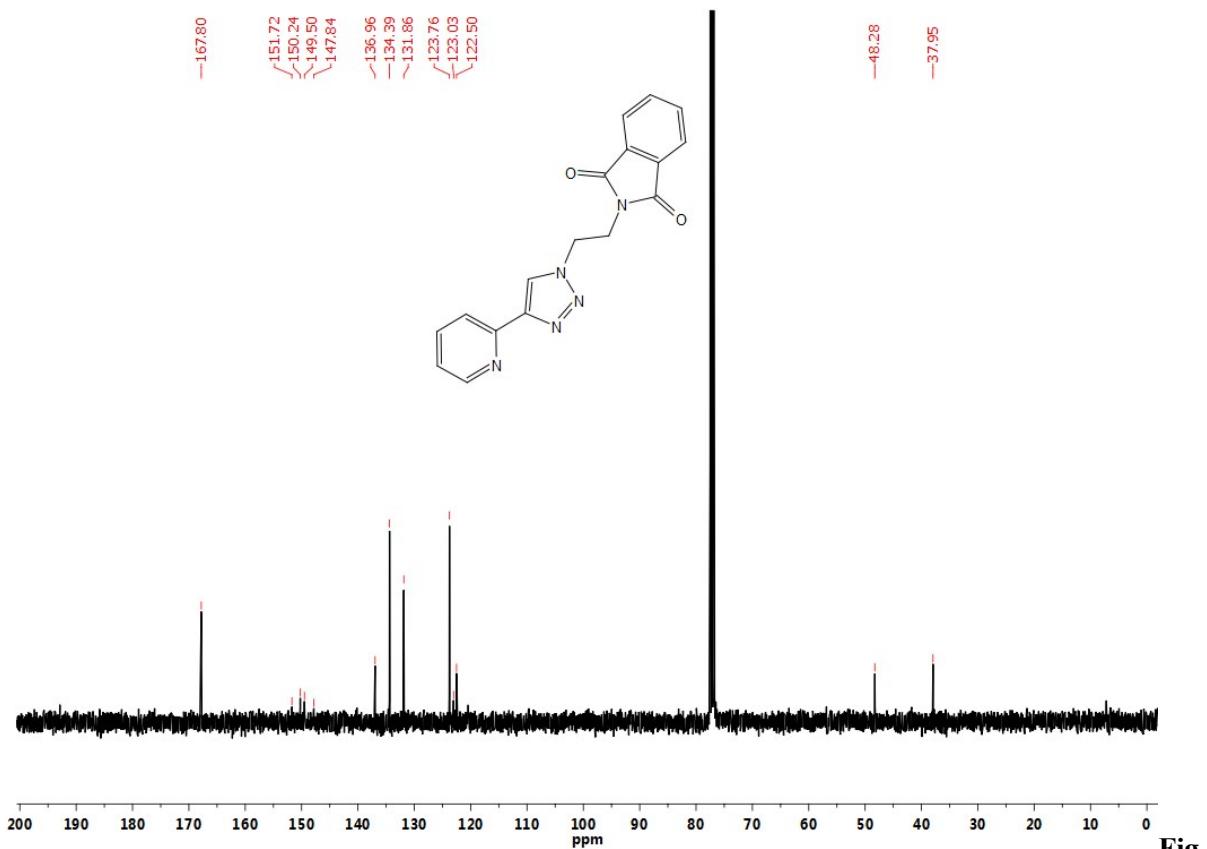


Fig.S3 ^1H NMR spectrum of TpH in CDCl_3



.S4 ^{13}C NMR spectrum of TpH in CDCl_3

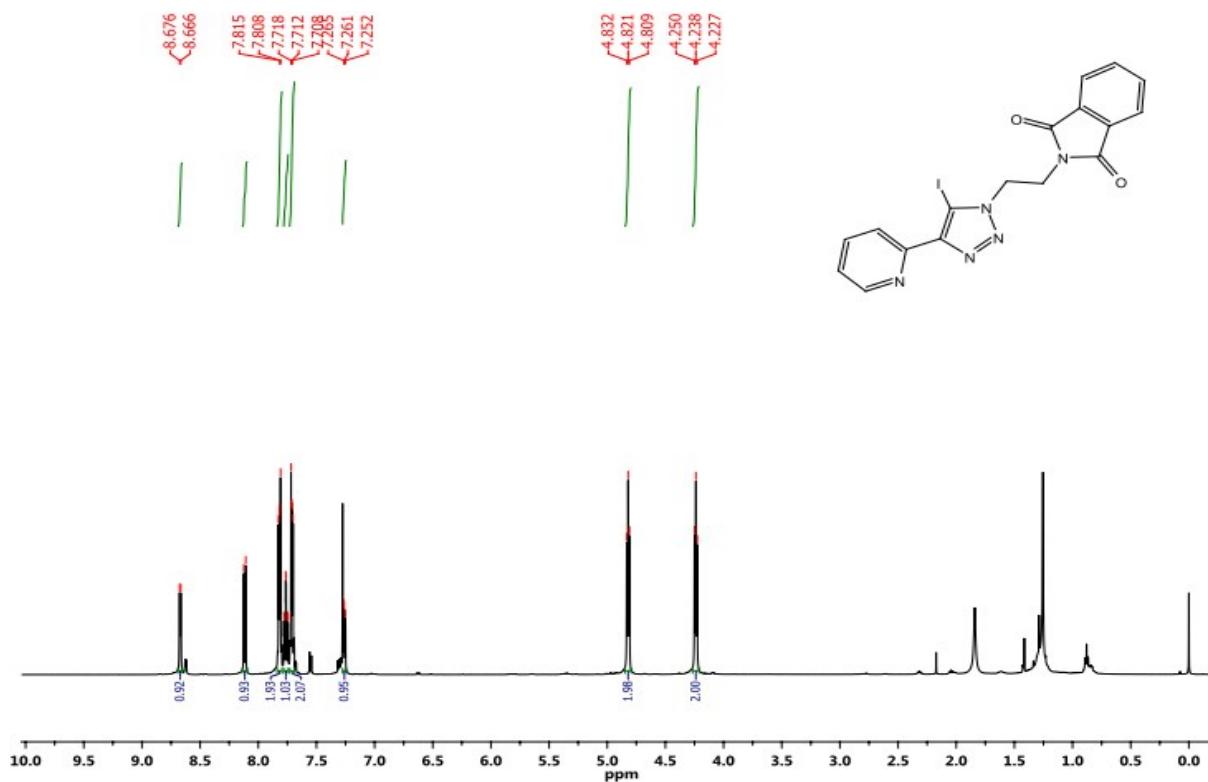


Fig.S5 ^1H NMR spectrum of TpI in CDCl_3

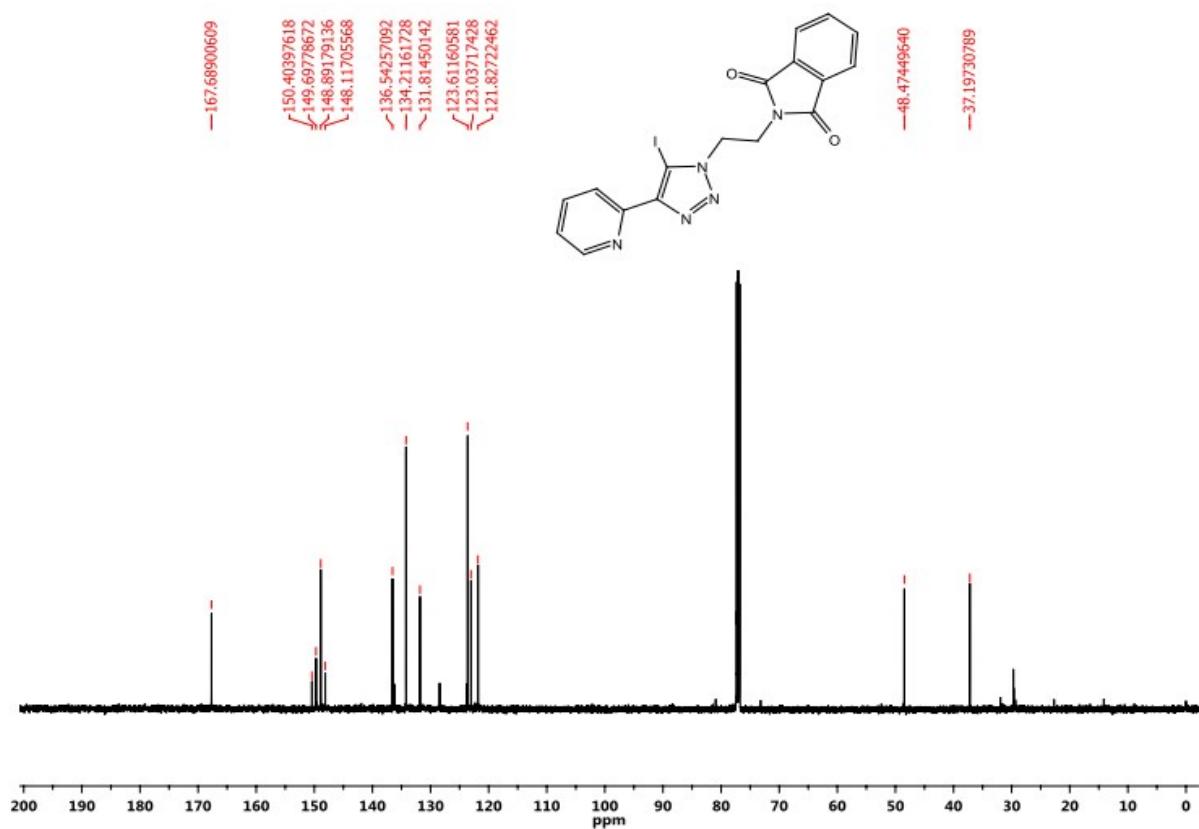


Fig.S6 ^{13}C NMR spectrum of TpI in CDCl_3

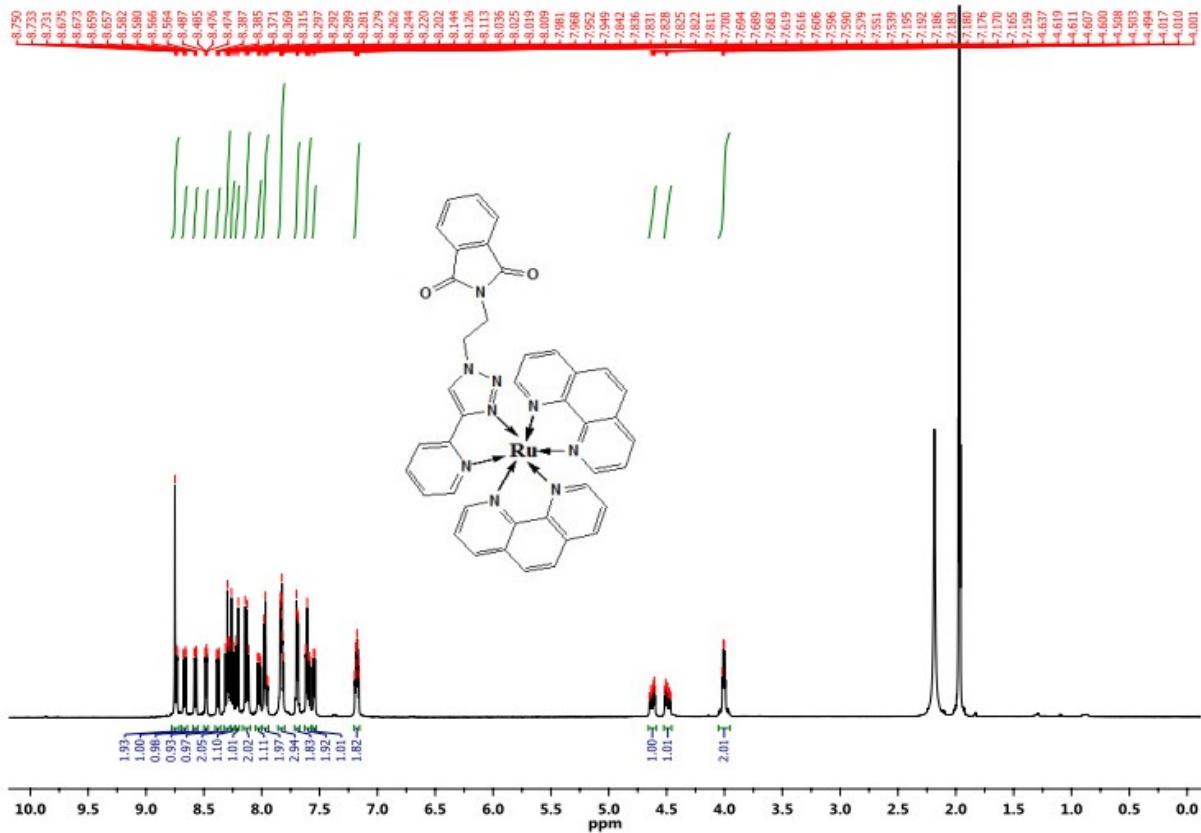


Fig.S7 ^1H NMR spectrum of RtpH in CD_3CN

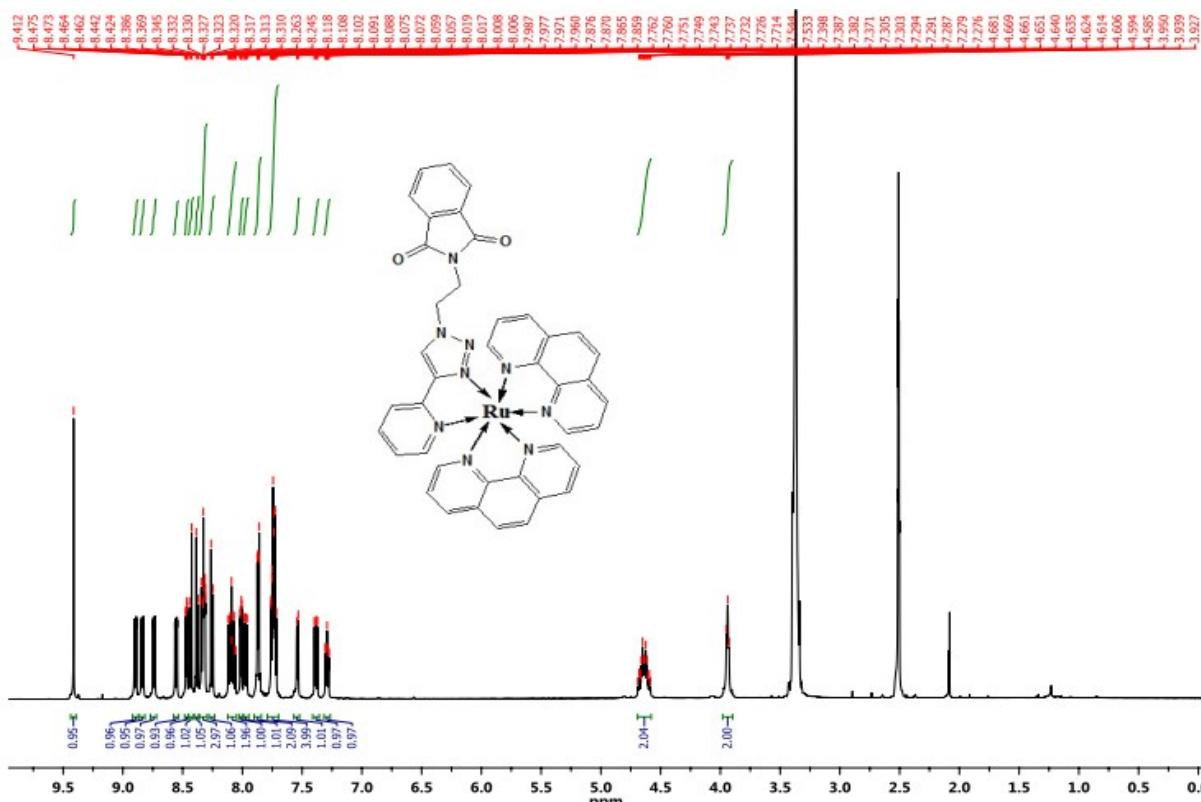


Fig.S8 ^1H NMR spectrum of RtpH in DMSO-d_6

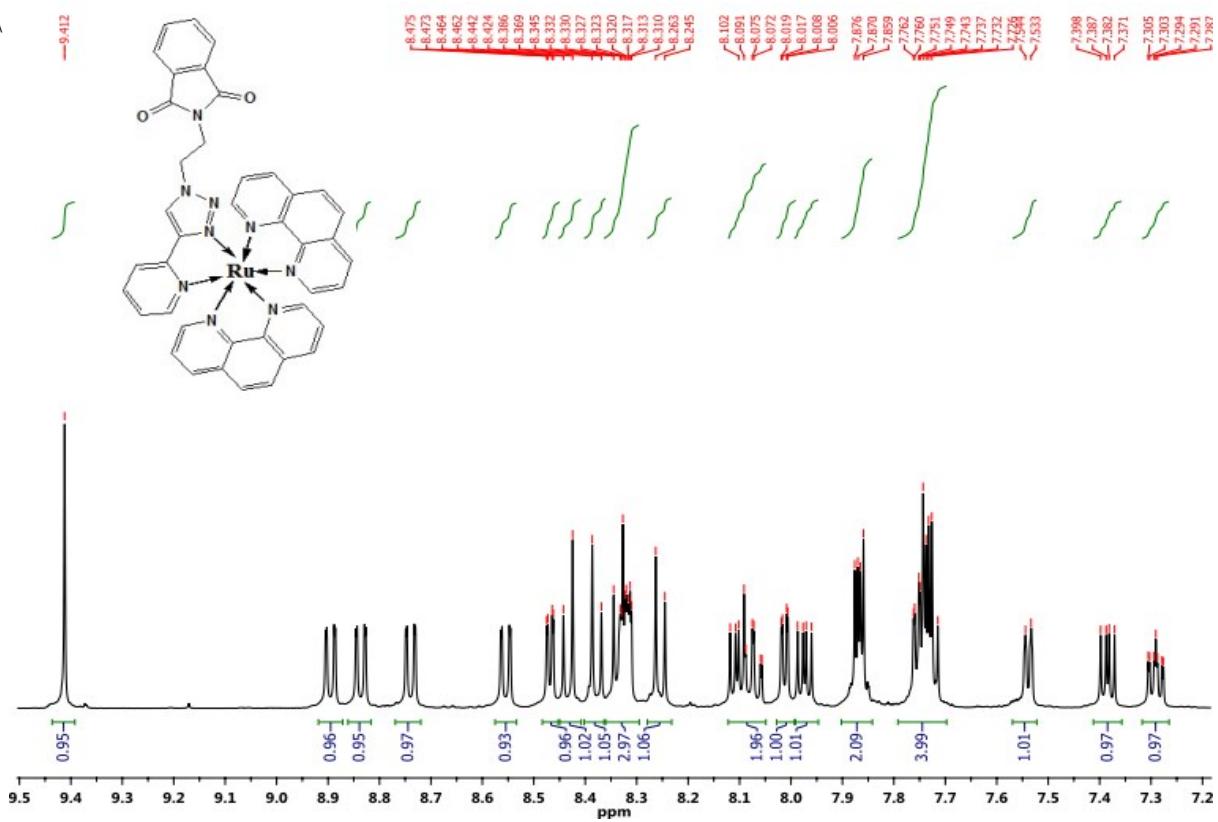


Fig.S9 Expanded ^1H NMR spectrum of RtpH in DMSO-d_6

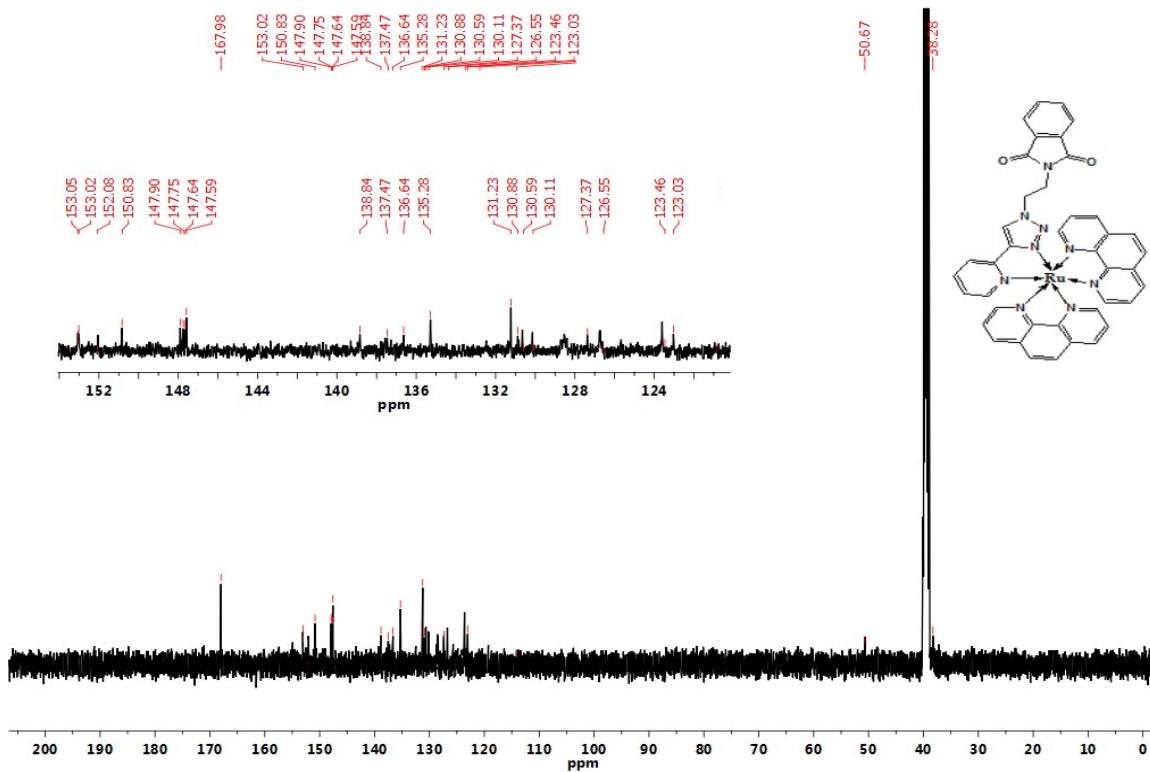


Fig.S10 ^{13}C NMR spectrum of RtpH in DMSO-d₆

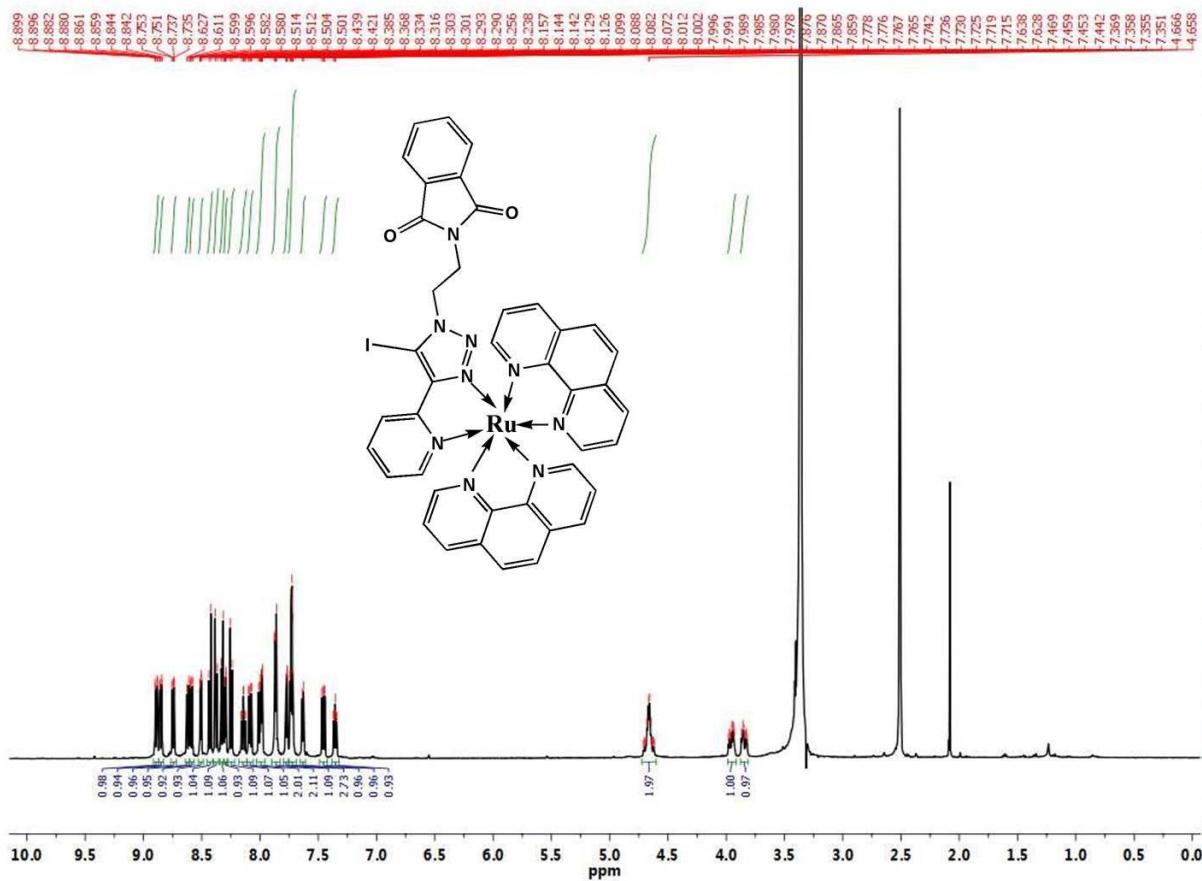


Fig.S11 ^1H NMR spectrum of RtpI in DMSO-d_6

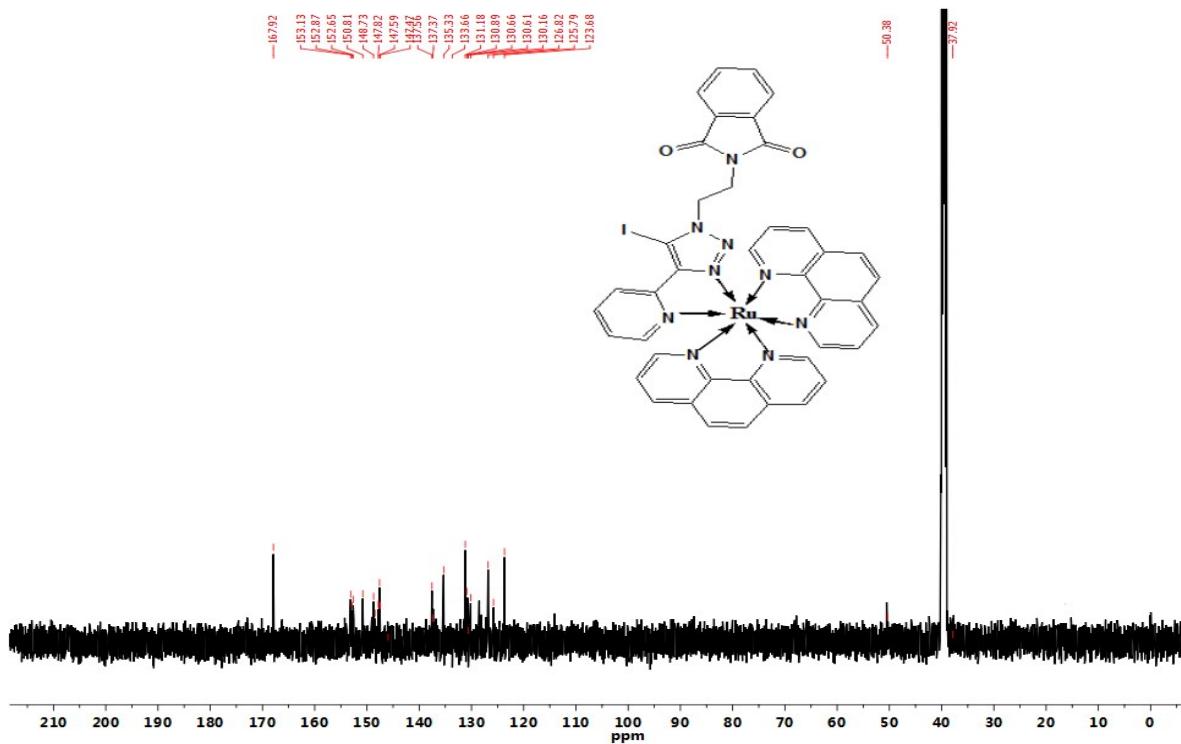
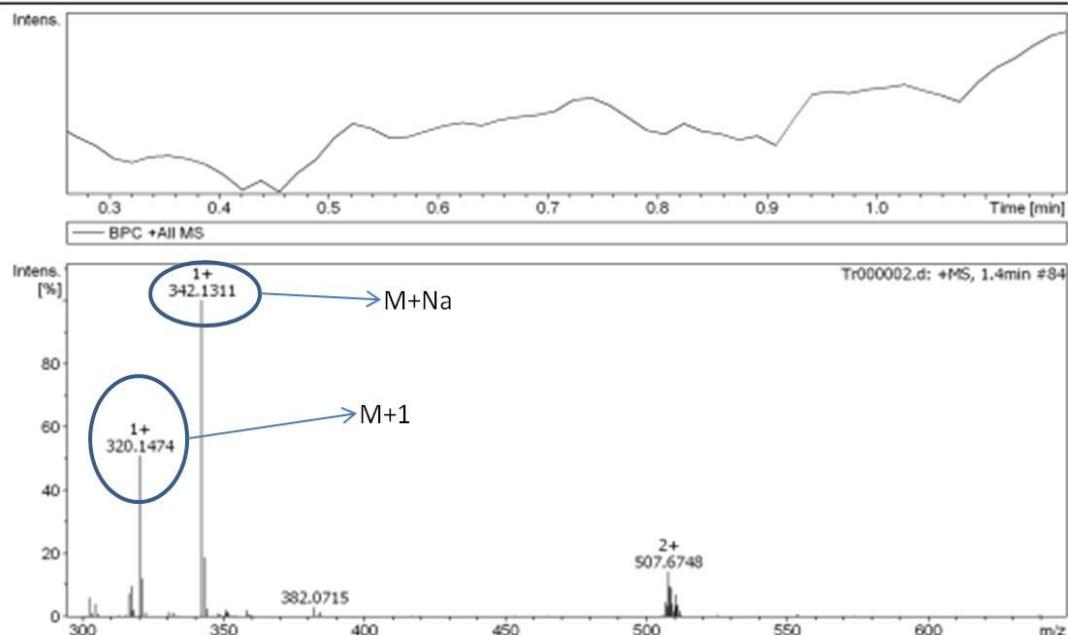


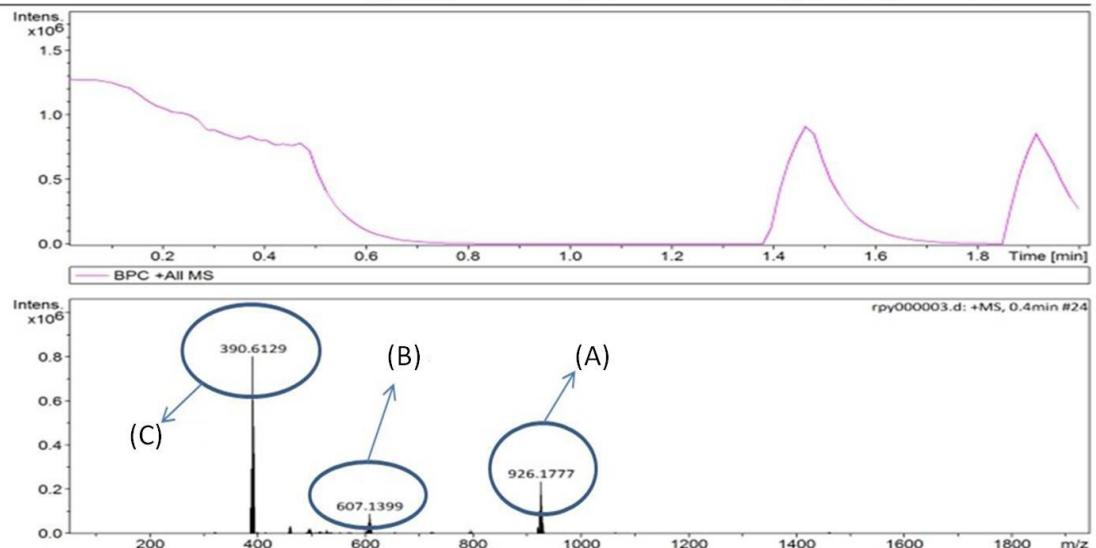
Fig.S12 ^{13}C NMR spectrum of RtpI in DMSO-d_6

Acquisition Parameter

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	1800 m/z	Set Charging Voltage	0 V	Set Divert Valve	Source
		Set Corona	0 nA	Set APCI Heater	0 °C

**Fig.S13** HR ESI-Mass spectrum of TpH**Acquisition Parameter**

Source Type	ESI	Ion Polarity	Positive	Set Nebulizer	0.4 Bar
Focus	Not active	Set Capillary	4500 V	Set Dry Heater	180 °C
Scan Begin	50 m/z	Set End Plate Offset	-500 V	Set Dry Gas	4.0 l/min
Scan End	3000 m/z	Set Charging Voltage	0 V	Set Divert Valve	Source
		Set Corona	0 nA	Set APCI Heater	0 °C



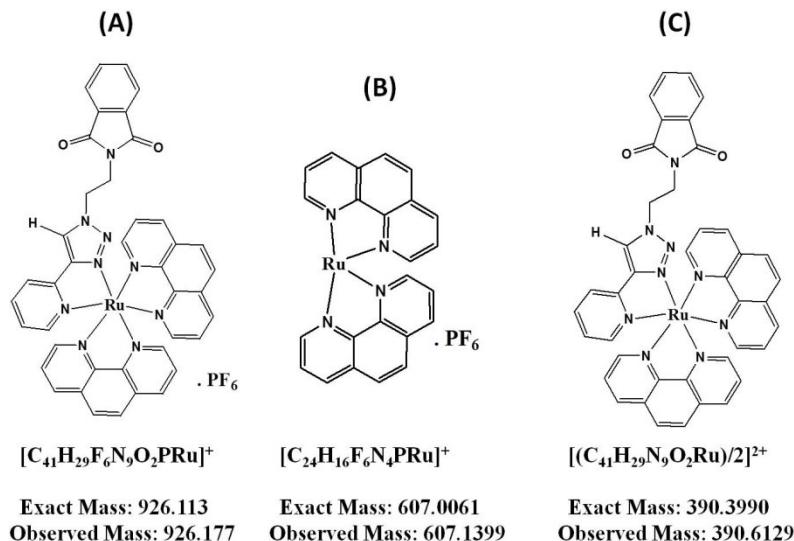


Fig.S14 HR ESI-Mass spectrum of RtpH

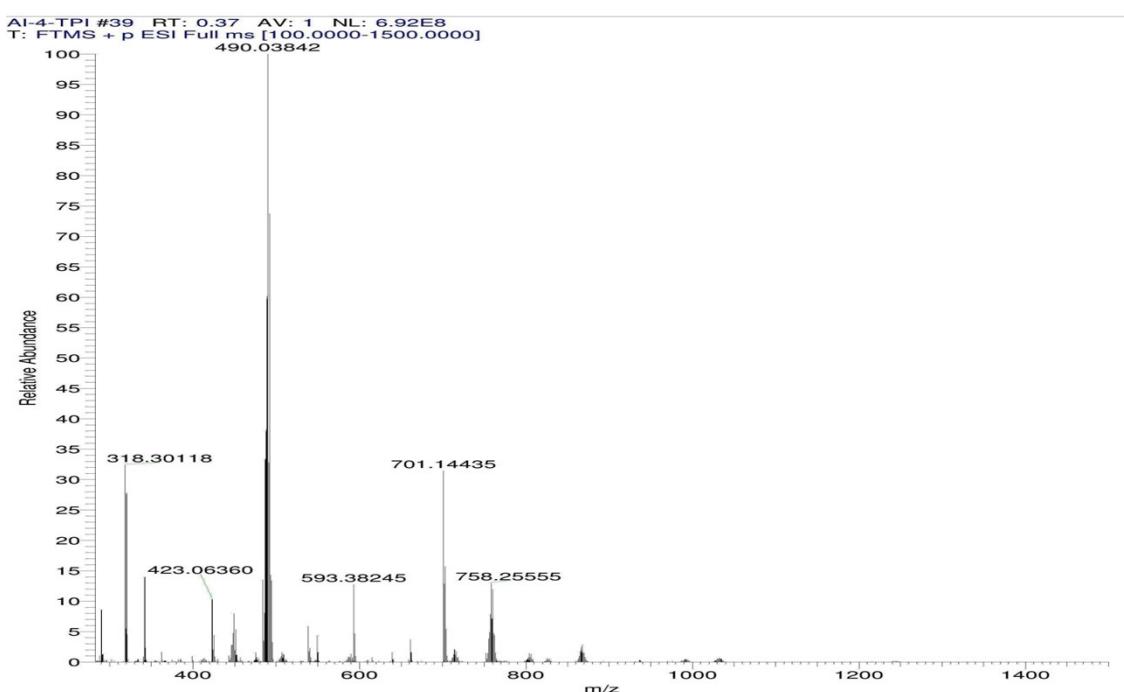


Fig. S15 HR-ESI Mass spectrum of TpI (Calculated value for $[\text{C}_{17}\text{H}_{12}\text{IN}_5\text{O}_2\text{Ru}]^+ + 2\text{Na} - \text{H} = 489.9747$, observed = 490.0384)

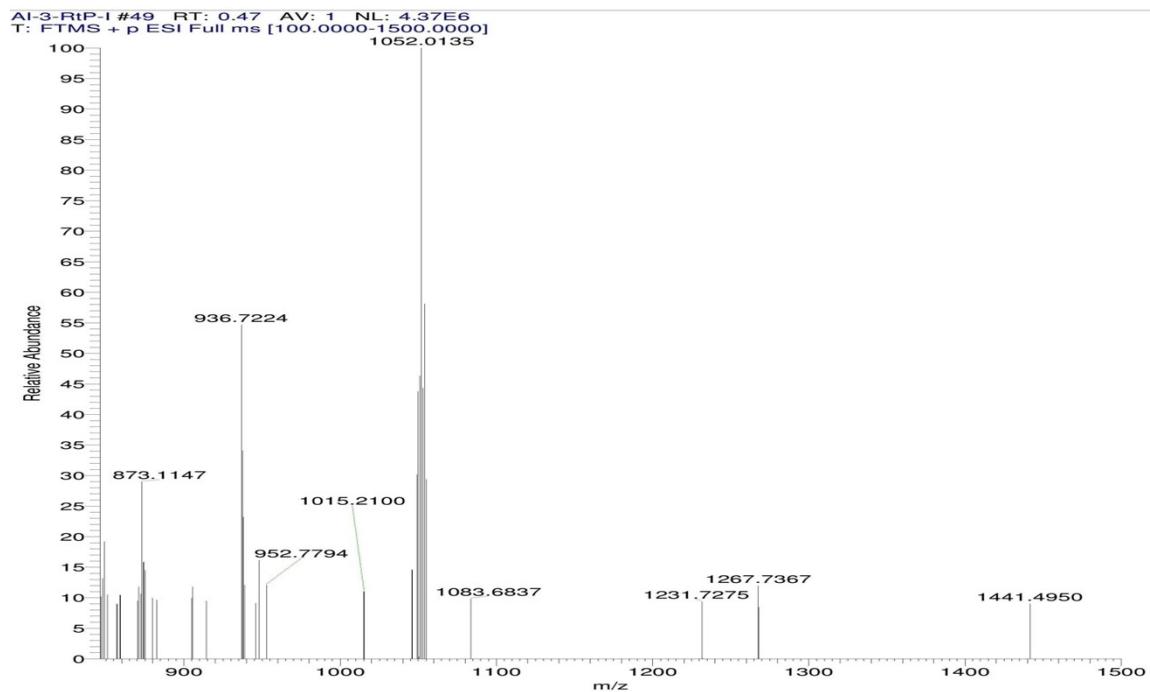


Fig. S16 HR-ESI Mass spectrum of RtpI (Calculated value for $[C_{41}H_{28}F_6IN_9O_2PRu]^+ = 1052.0101$, observed = 1052.0135)

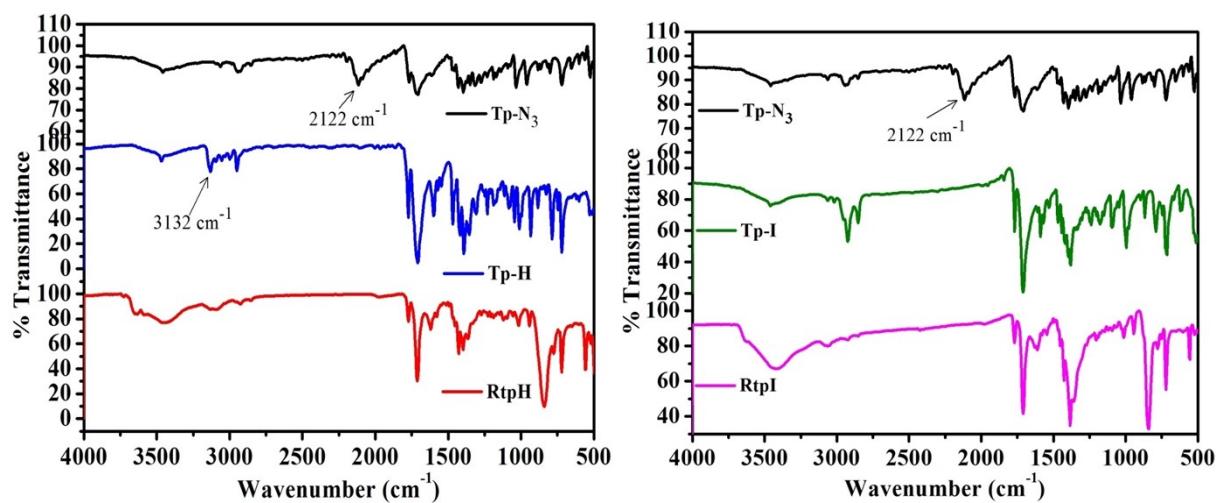


Fig. S17 FT-IR spectra of TpN₃, TpH, TpI, RtpH and RtpI.

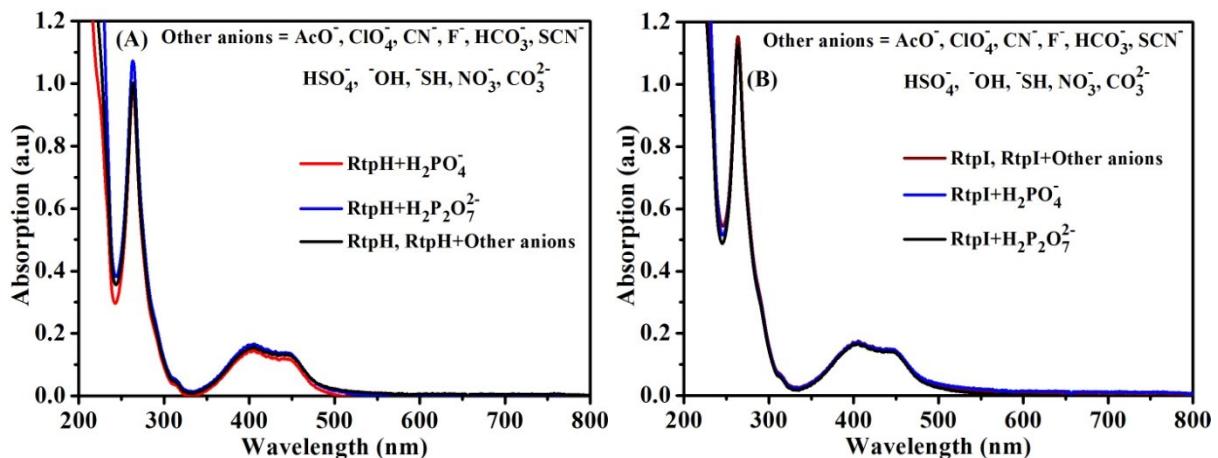


Fig.S18 (A) UV-Vis spectra of RtpH (10 μ M) with various anions (5 equiv.) and phosphate anions (H_2PO_4^- and $\text{H}_2\text{P}_2\text{O}_7^{2-}$) (1 equiv.), (B) Uv-Vis spectra of RtpI (10 μ M) with various anions (5 equiv.) and phosphate anions (H_2PO_4^- and $\text{H}_2\text{P}_2\text{O}_7^{2-}$) (1 equiv.) in CH_3CN .

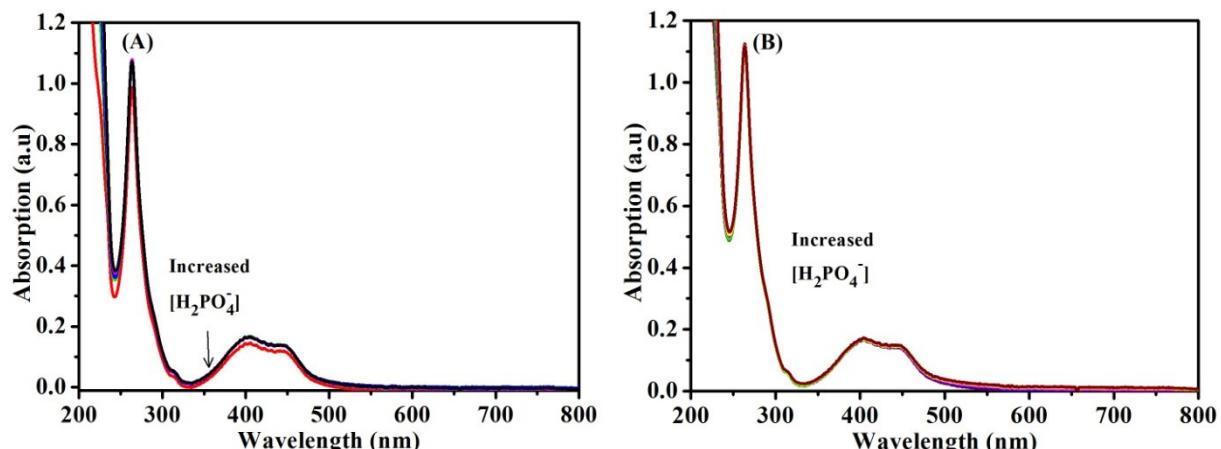


Fig.S19 (A) UV-Vis spectra of RtpH (10 μ M) with the addition of H_2PO_4^- anions (0-1 equiv.), (B) Uv-Vis spectra of RtpI (10 μ M) with the addition of H_2PO_4^- anions (0-1 equiv.) in CH_3CN .

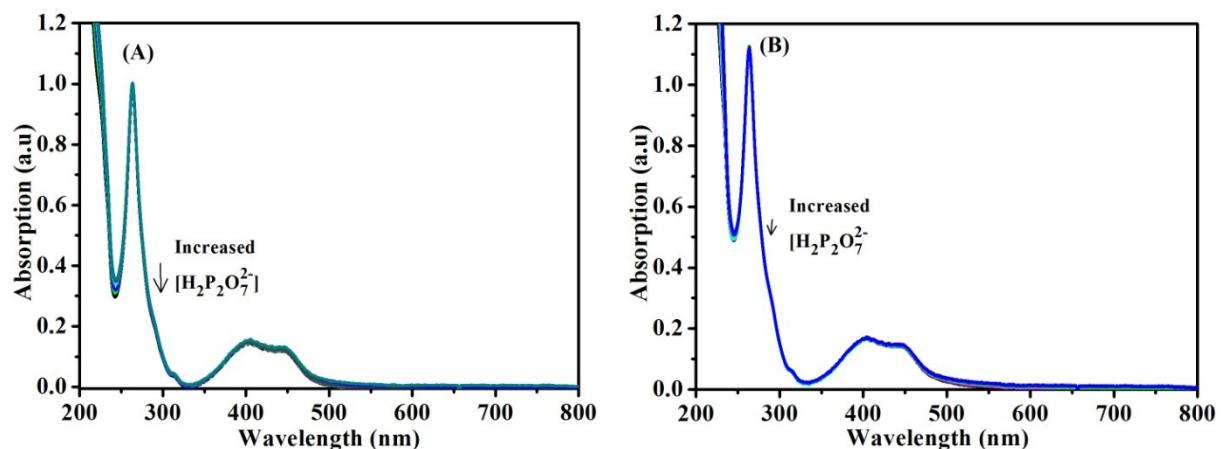


Fig.S20 (A) UV-Vis spectra of RtpI (10 μ M) with the addition of $\text{H}_2\text{P}_2\text{O}_7^{2-}$ -anions (0-1 equiv.), (B) Uv-Vis spectra of RtpI (10 μ M) with the addition of $\text{H}_2\text{P}_2\text{O}_7^{2-}$ -anions (0-1 equiv.) in CH_3CN .

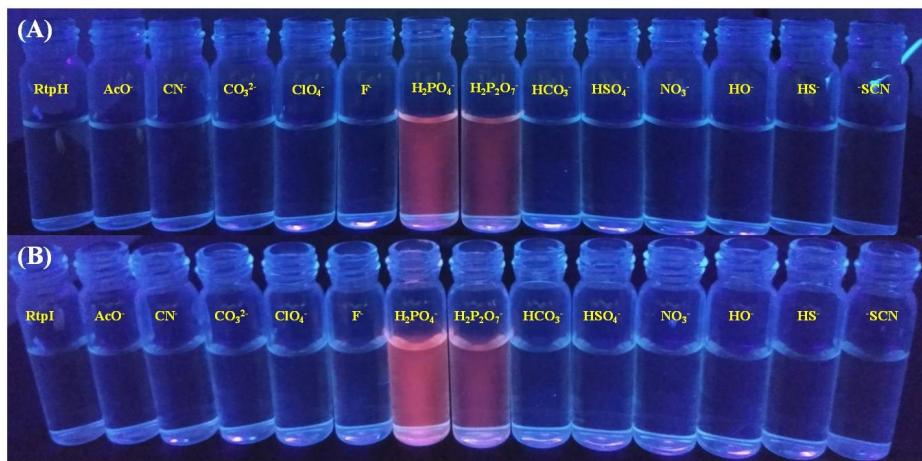


Fig.S21 Under UV light illumination, (A) RtpH and (B) RtpI with various types of anions in CH₃CN

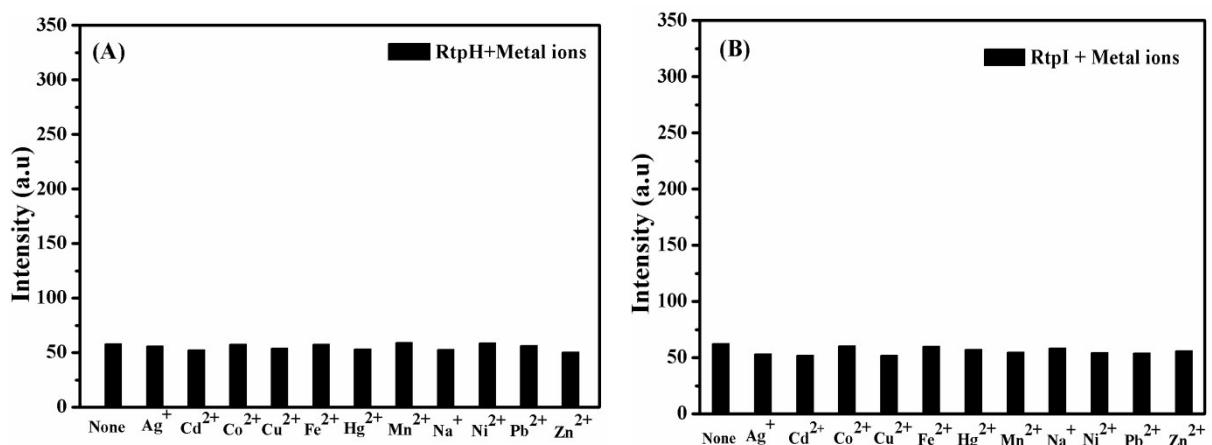


Fig.S22 (A) Emission intensity of RtpH (10 μ M) with different metal ions (10 equiv.), (B) Emission intensity of RtpI (10 μ M) with different metal ions (10 equiv.) in CH₃CN.

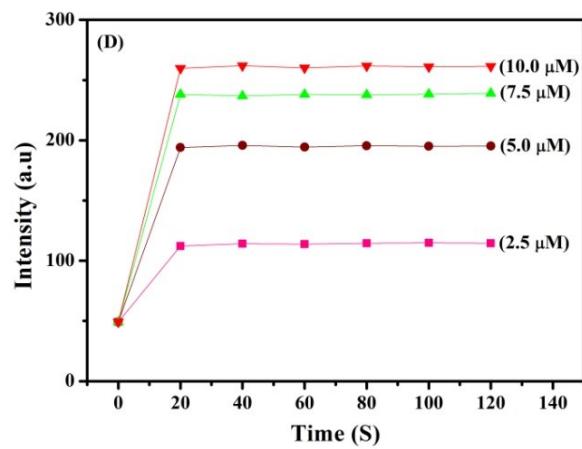
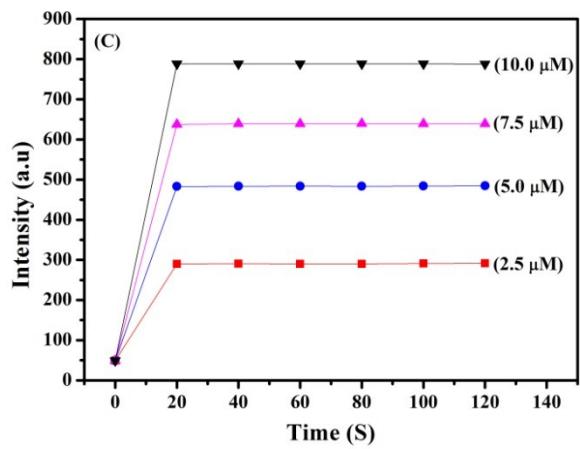
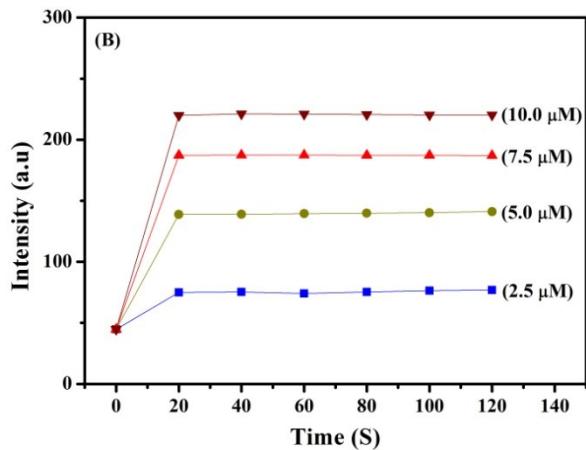
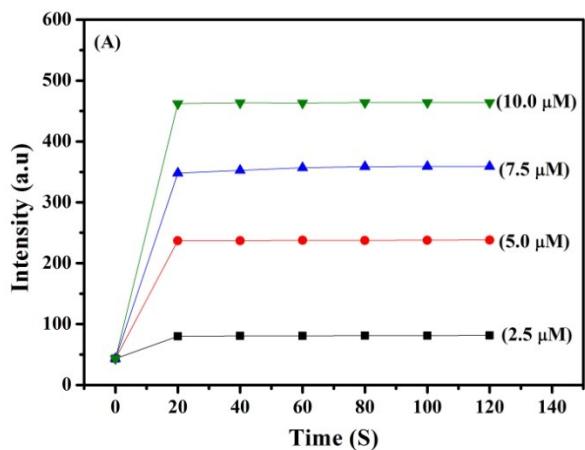


Fig.S23 (A) Time course study of RtpH (10 μ M) upon addition different equivalent of H_2PO_4^- (B) $\text{H}_2\text{P}_2\text{O}_7^{2-}$ in CH_3CN and time course study of RtpI (10 μ M) upon addition different equivalent of H_2PO_4^- (C) $\text{H}_2\text{P}_2\text{O}_7^{2-}$ (D) in CH_3CN .

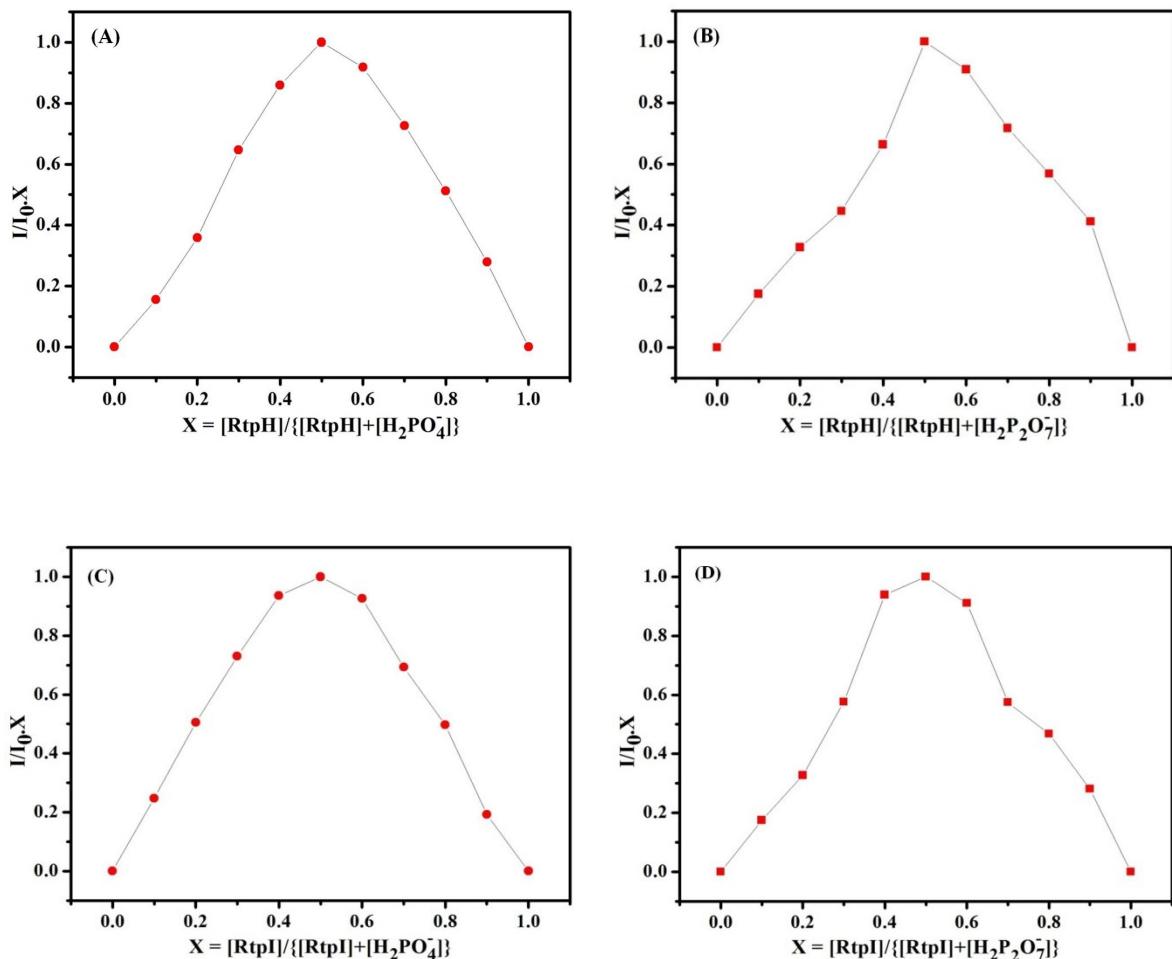


Fig.S24 Job's plot of RtpH with (A) $H_2PO_4^-$ (B) $H_2P_2O_7^{2-}$, Job's plot of RtpI with (A) $H_2PO_4^-$ (B) $H_2P_2O_7^{2-}$

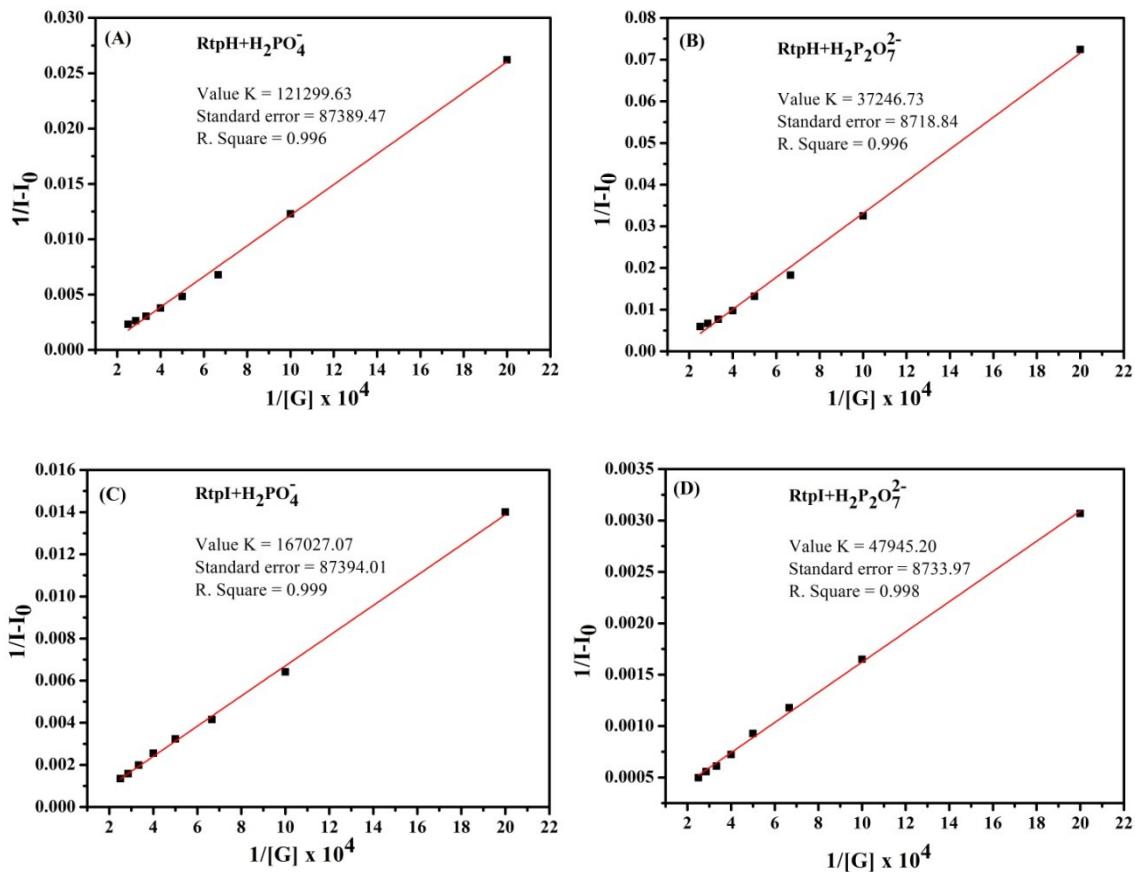


Fig. S25. Benesi-Hildebrand plot from emission titration data of RtpH (10 μM) with H_2PO_4^- (A) and $\text{H}_2\text{P}_2\text{O}_7^{2-}$ (B) and of RtpI with H_2PO_4^- (C) and $\text{H}_2\text{P}_2\text{O}_7^{2-}$ (D).

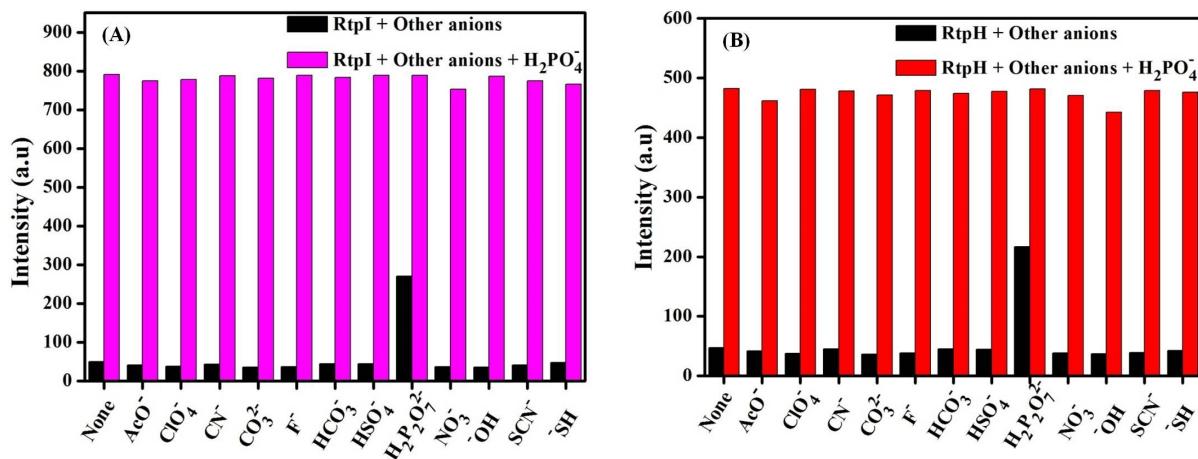


Fig. S26 (A) Interference study of RtpI (10 μM) with and without H_2PO_4^- anions in presence of different anions (10 equiv.), (B) Interference study of RtpH (10 μM) with and without H_2PO_4^- anions in presence of different anions (10 equiv.).

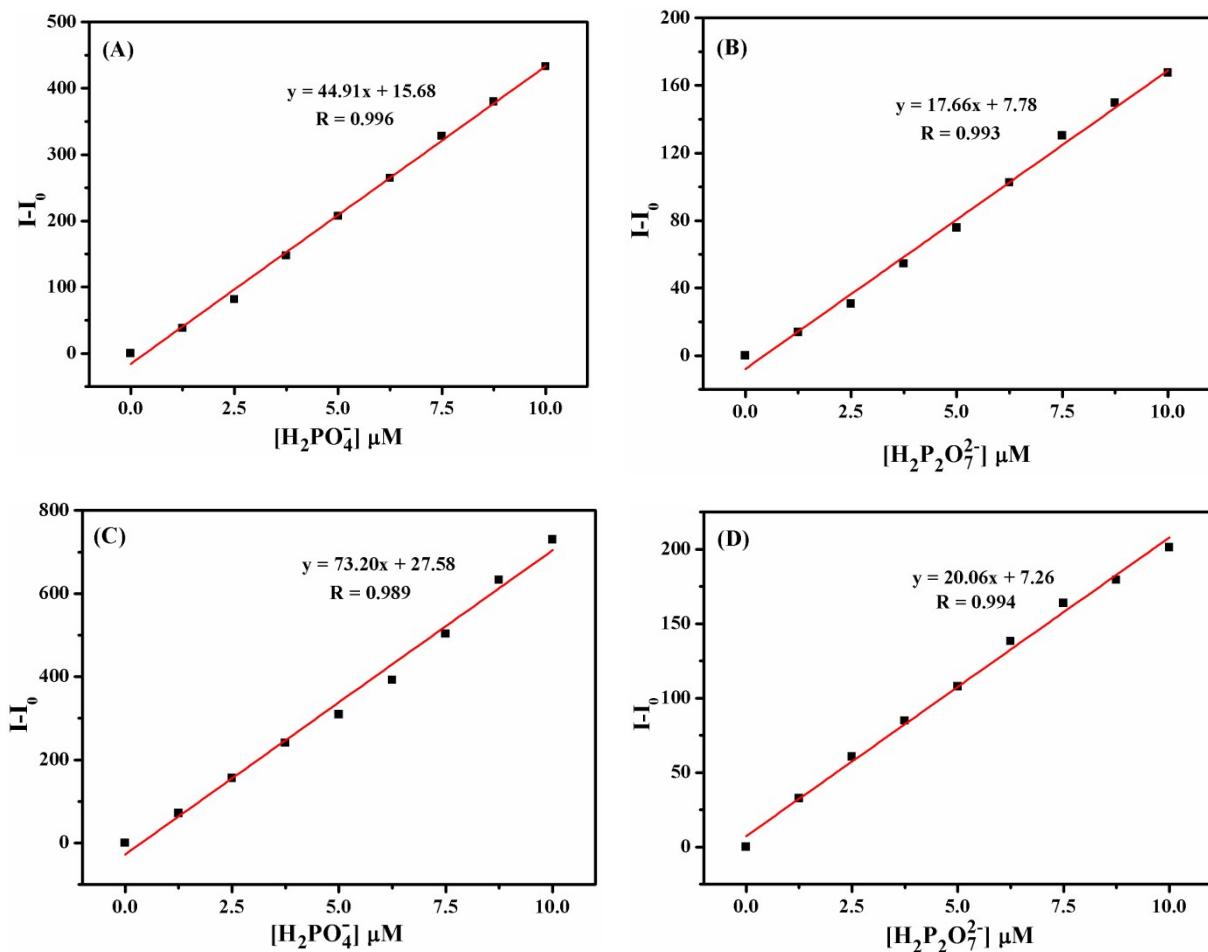


Fig.S27 A linear plot of RtpH (A, B) and RtPI (C, D) with $H_2PO_4^-$ and $H_2P_2O_7^{2-}$ in CH_3CN .

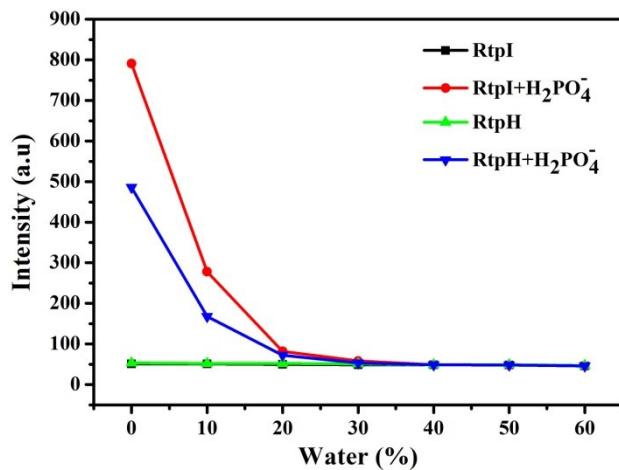


Fig. S28 The emission intensity of metalloreceptors (RtpH and RtpI, 10 μM) with phosphate anions ($H_2PO_4^-$ and $H_2P_2O_7^{2-}$, 10 μM) upon varying water percentage with CH_3CN .

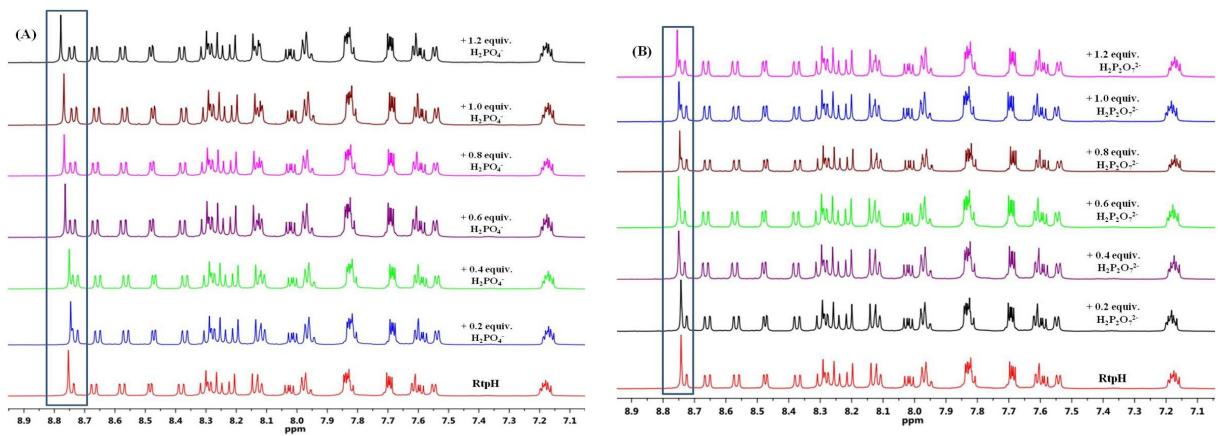


Fig.S29 (A) ¹H NMR titration of RtpH (7.5 mM) with H_2PO_4^- (0-1.2 equiv.), (B) ¹H NMR titration of RtpI (7.5 mM) with H_2PO_4^- (0-1.2 equiv.) in $\text{CD}_3\text{CN}/\text{D}_2\text{O}$ (9/1, v/v).

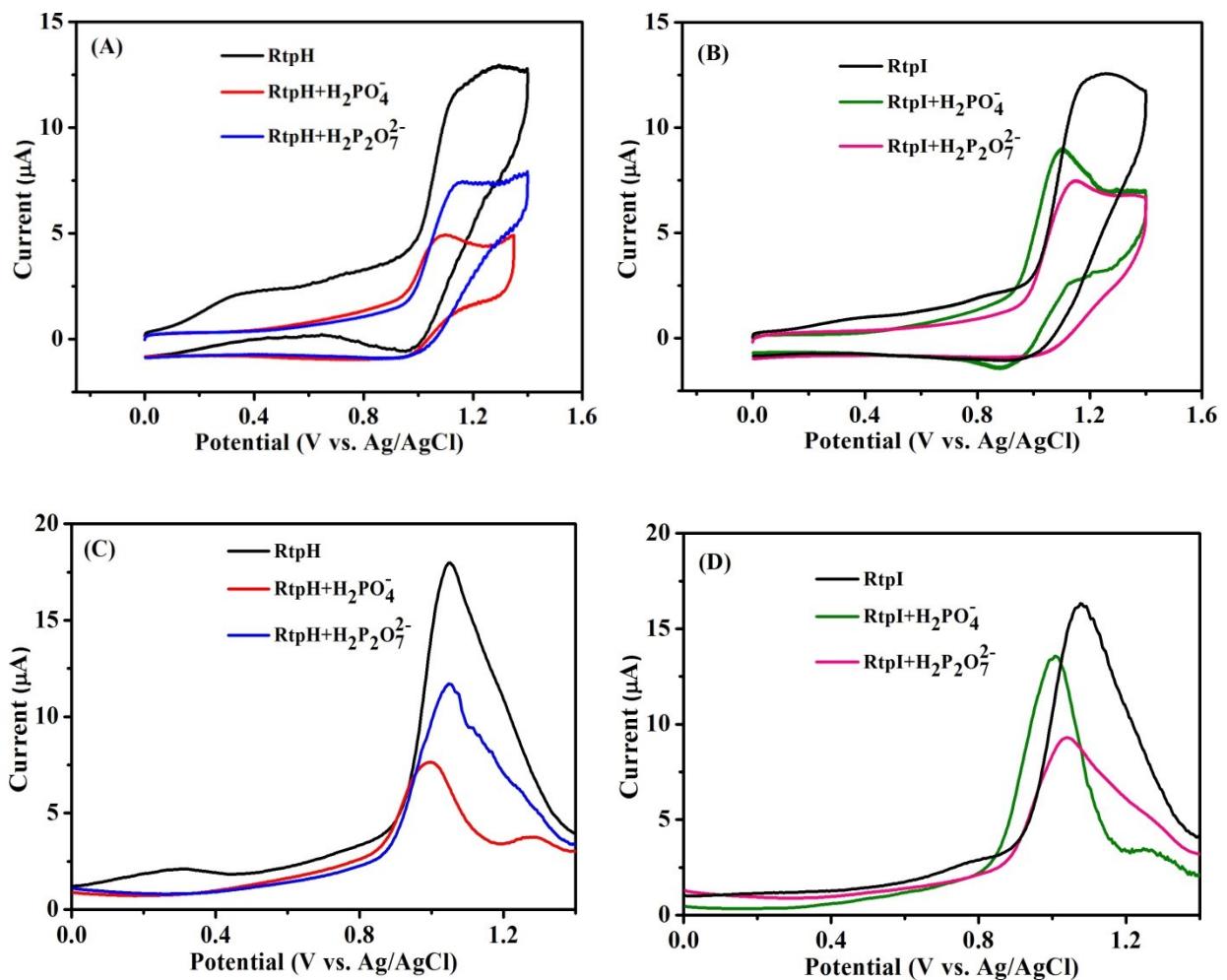


Fig.S30 (A) Cyclic Voltammetry (A and B) and Differential Pulse Voltammetry (C and D) of metalloreceptors (RtpH and RtpI) with and without phosphate anions.

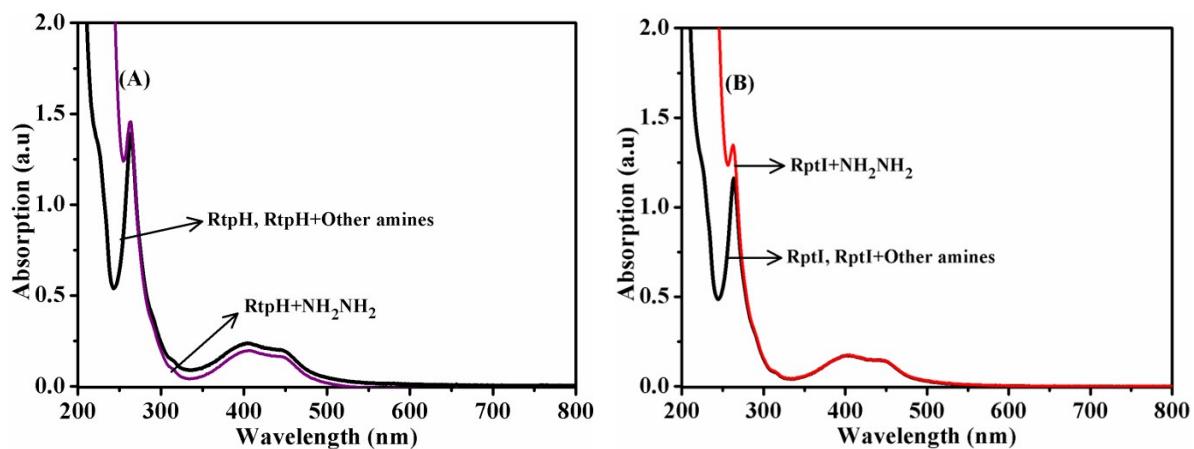


Fig.S31 (A) UV-Vis spectra of RtpH (25 μ M) with various anions (10 equiv.) and hydrazine (4 equiv.), (B) UV-Vis spectra of RtpI (25 μ M) with various anions (10 equiv.) and hydrazine (4 equiv.) in $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (6/4, v/v).

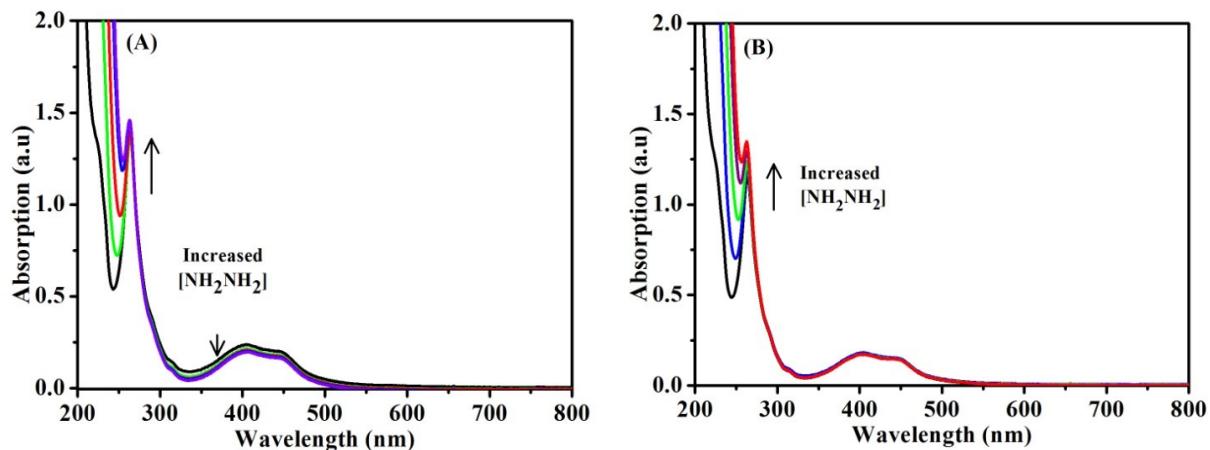


Fig.S32 (A) UV-Vis spectra of RtpH (25 μ M) with the gradual addition of hydrazine (4 equiv.), (B) UV-Vis spectra of RtpI (25 μ M) with the gradual addition of hydrazine (4 equiv.) in $\text{CH}_3\text{CN}/\text{H}_2\text{O}$ (6/4, v/v).

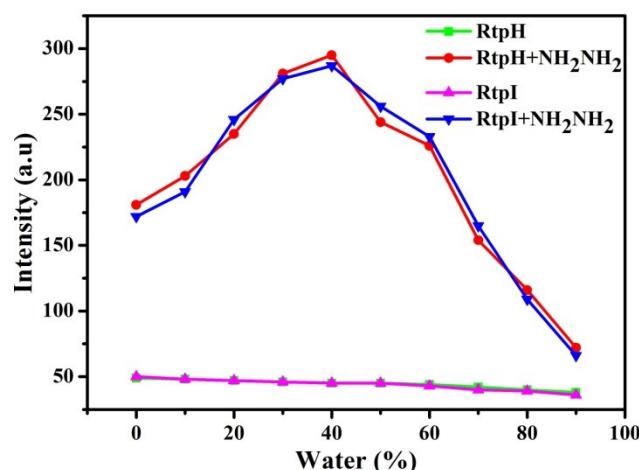


Fig. S33 The emission intensity of metalloreceptors (RtpH and RtpI, 25 μ M) with hydrazine (100 μ M) upon varying water percentage with CH_3CN

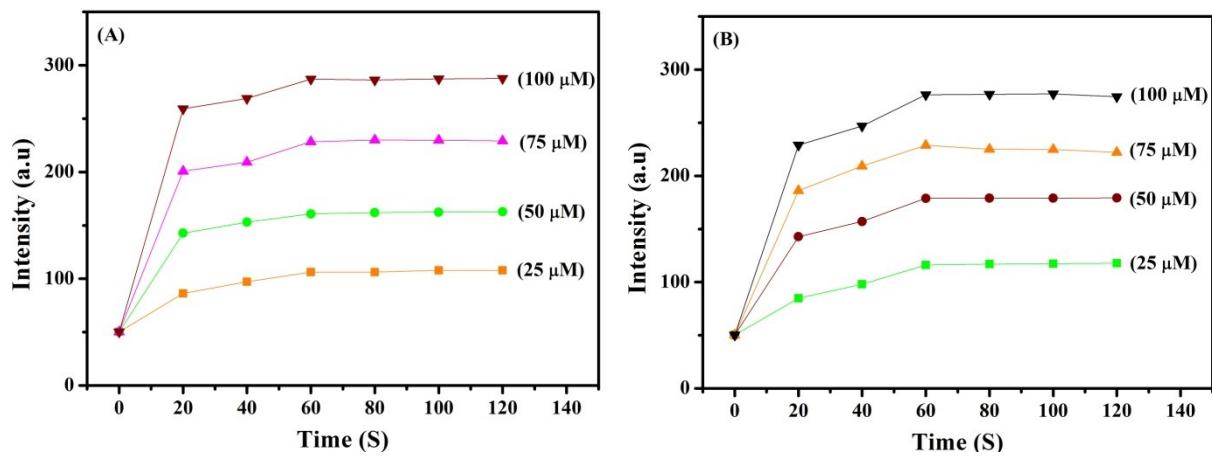


Fig. S34 (A) Time course study of RtpH (25 μ M) upon addition different equivalent of hydrazine (B) Time course study of RtpI (25 μ M) upon addition different equivalent of hydrazine

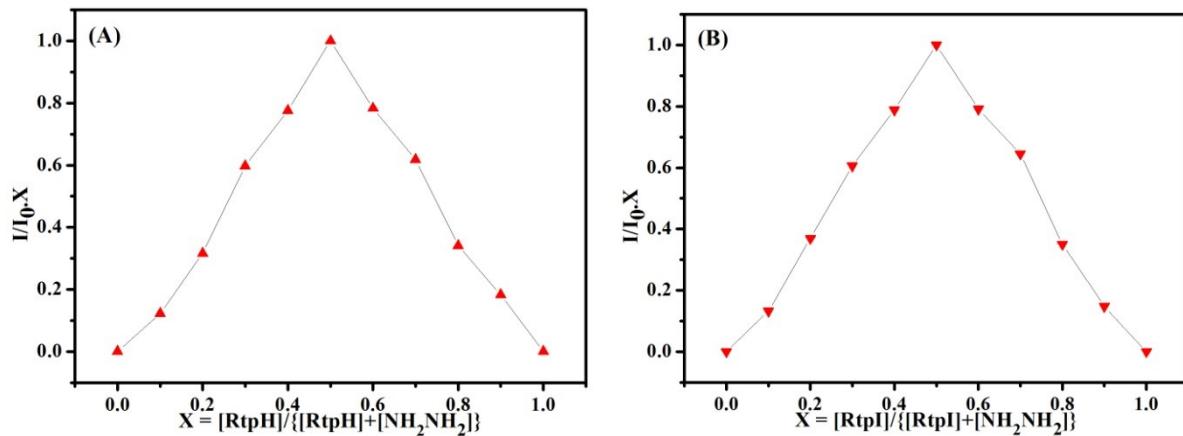


Fig.S35 (A) Job's plot of RtpH with hydrazine, (B) Job's plot of RtpI with hydrazine in CH_3CN/H_2O (6/4, v/v)

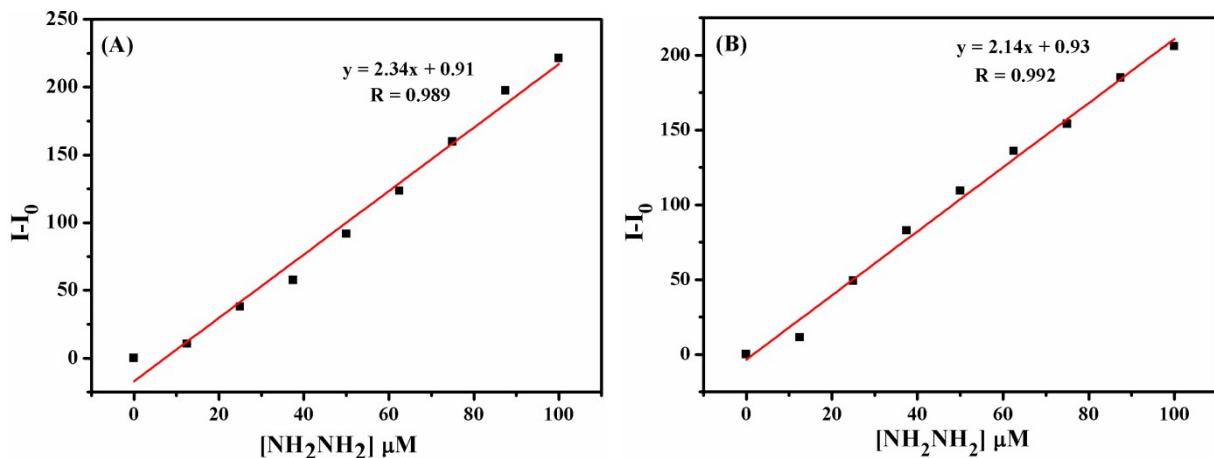


Fig.S36 (A) A linear plot of RtpH with gradual addition of hydrazine (B) A linear plot of RtpI with gradual addition of hydrazine in CH_3CN/H_2O (6/4, v/v).

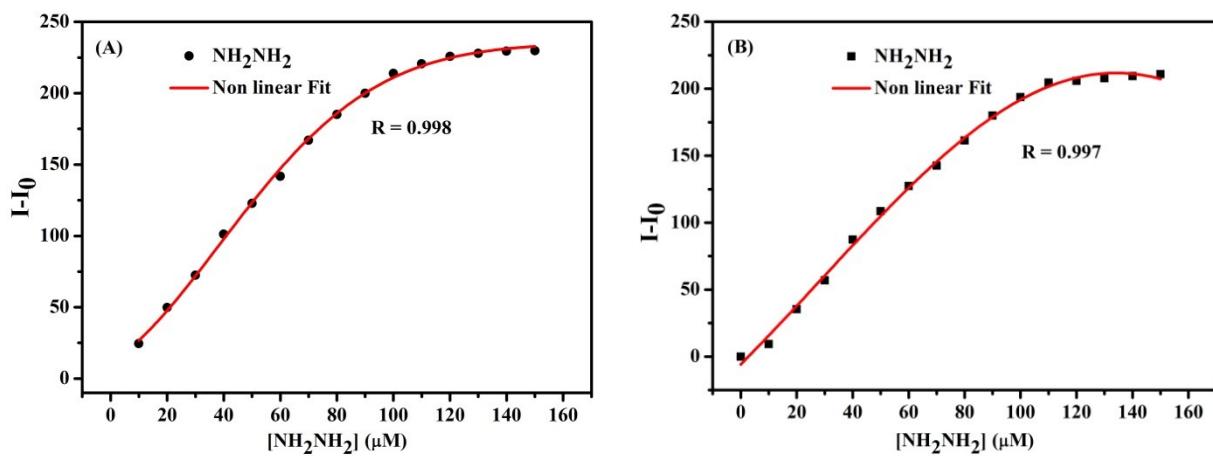


Fig. S37 (A) Non-linear fitting of RtpH (A)/ RtpI (B) (25 μM) upon addition different concentration of hydrazine

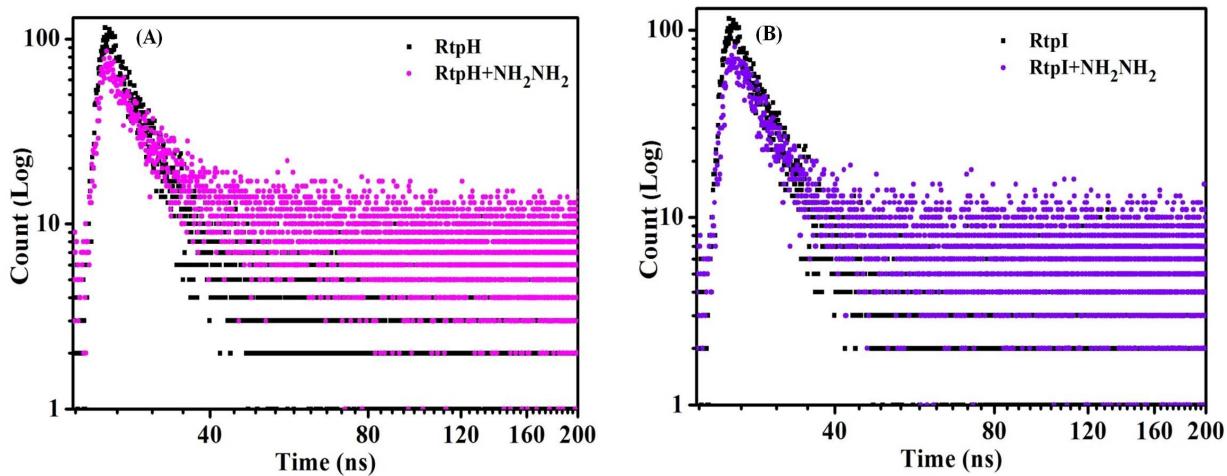


Fig.S38 (A) Lifetime decay of RtpH (10 μM) with and without hydrazine (4 equiv.), (B) Lifetime decay of RtpI (10 μM) with and without hydrazine (4 equiv.) in CH_3CN/H_2O (6/4, v/v).

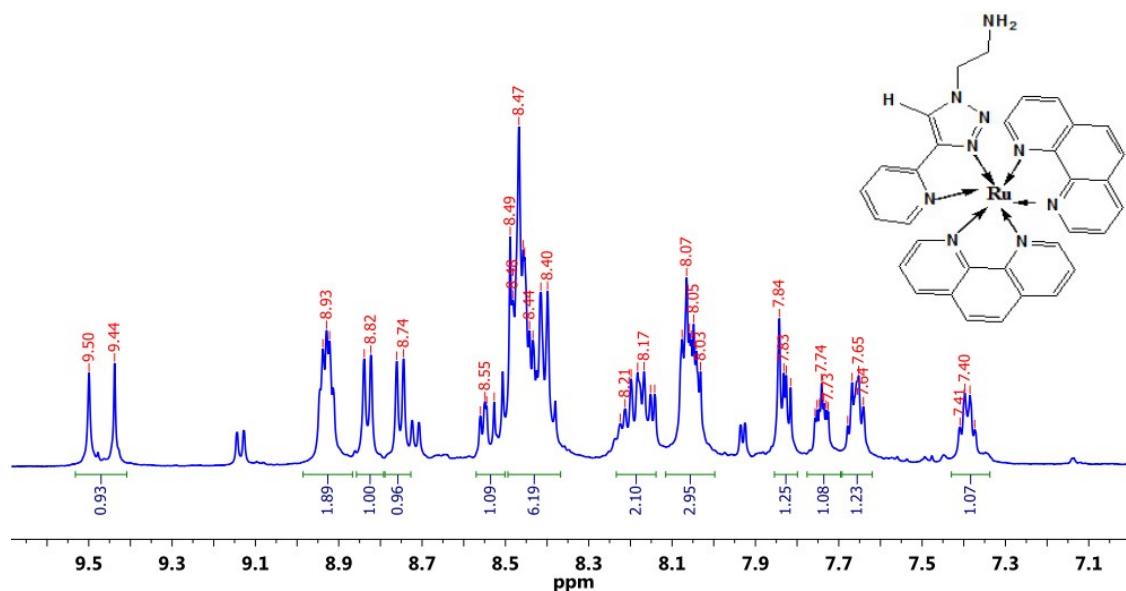


Fig. S39. Partial ¹H NMR spectrum of RtpH-NH₂ in DMSO-d₆

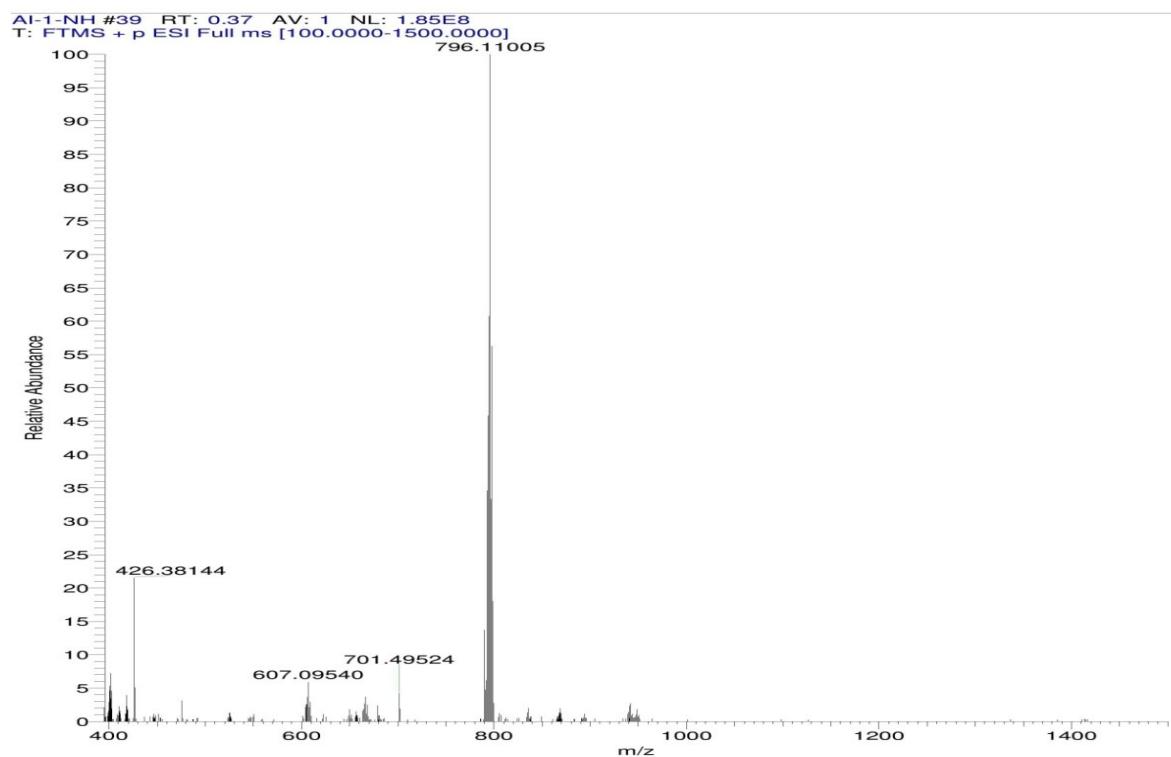


Fig. S40 HR-ESI Mass spectra of RtpH-NH₂ (Calculated value for [C₃₃H₂₆F₆IN₉PRu]⁺ = 796.1080, observed = 796.1100)

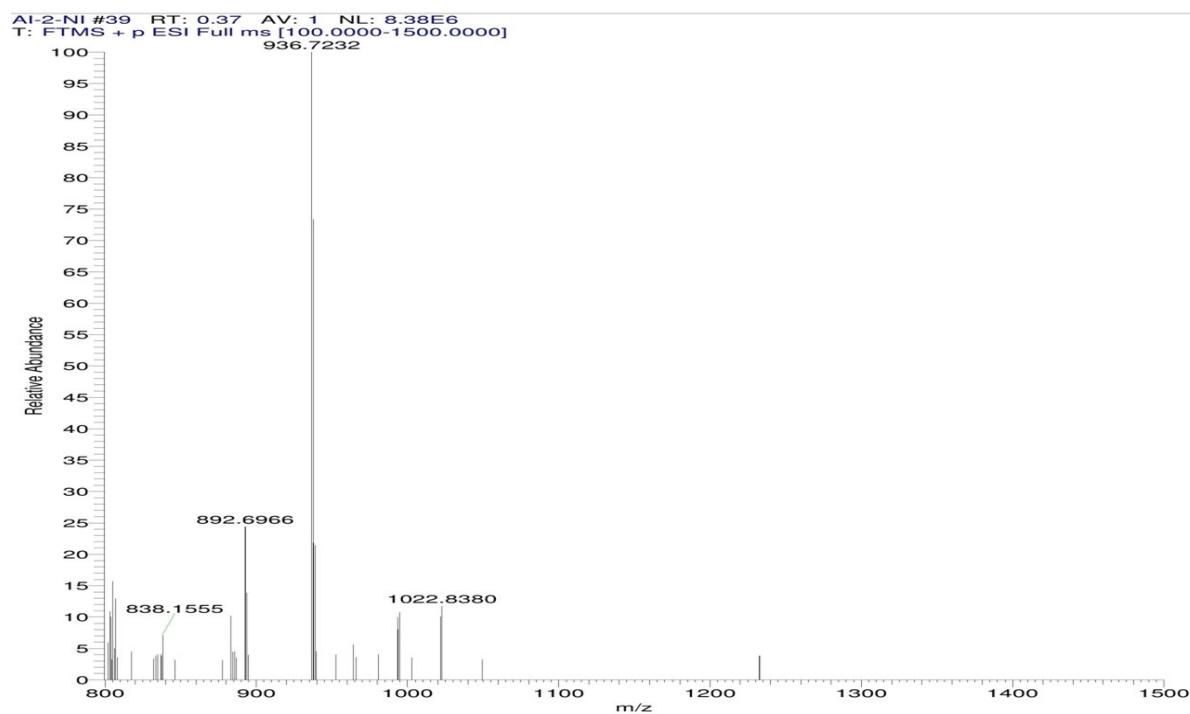


Fig. S41 HR-ESI Mass spectra of RtpI-NH₂ (Calculated value for [C₃₃H₂₇F₆IN₉PRu]⁺ + H + 2Na = 936.7646, observed = 936.7232)

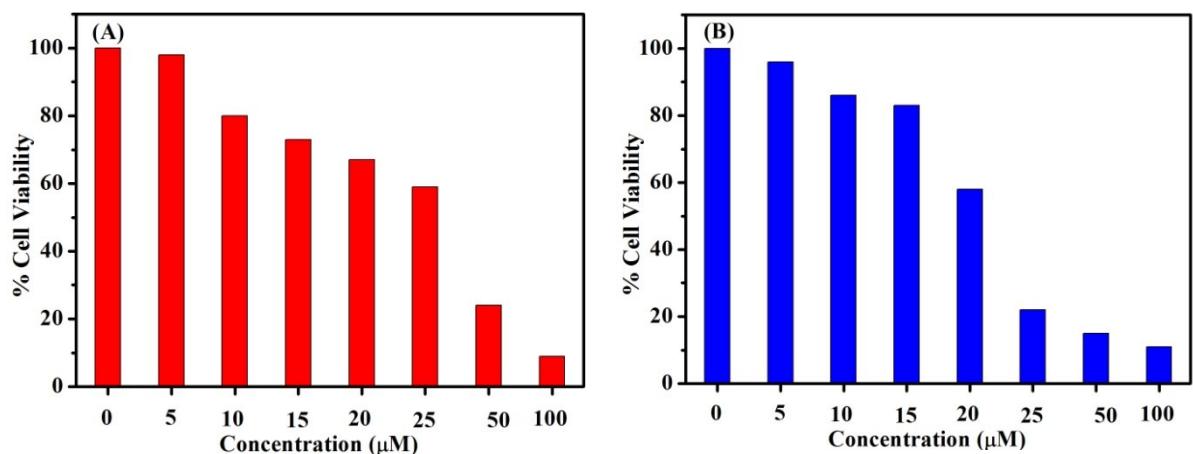


Fig.S42 Cell viability of HeLa cells in the presence of various concentrations of (A) RtpH and (B) RtpI

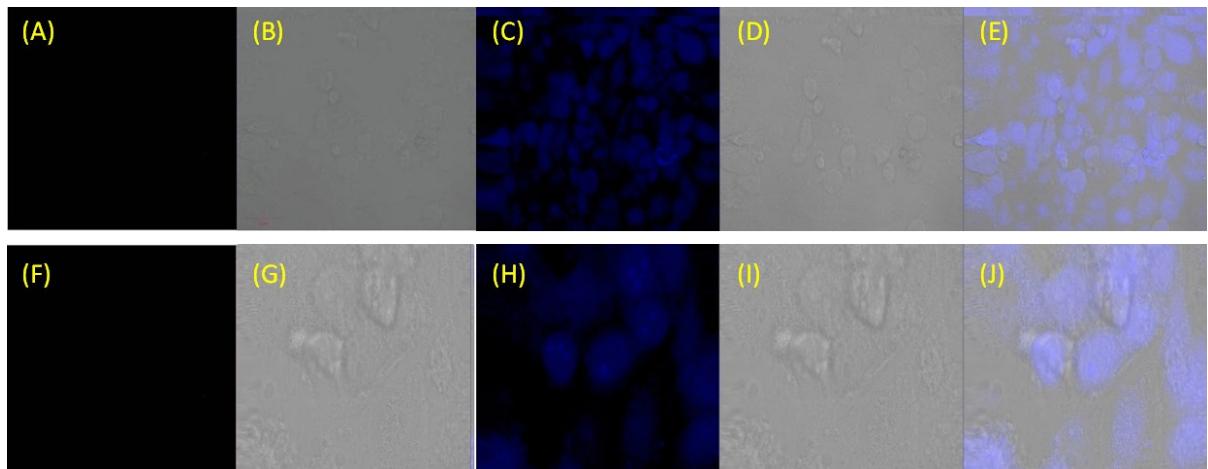


Fig. S43 Cellular uptake (HeLa) image of $25\mu\text{M}$ RtpH in dark (A), bright field (B), 4 equiv. of hydrazine treated RtpH (blue filter) (C), 4 equiv. of hydrazine treated RtpH (bright field) (D), 4 equiv. of hydrazine treated RtpH merged with RtpI (E), and $25\mu\text{M}$ RtpI in dark (F), bright field (G), 4 equiv. of hydrazine treated RtpI (blue filter) (H), 4 equiv. of hydrazine treated RtpI (bright field) (I), 4 equiv. of hydrazine treated RtpI merged with RtpH (J).

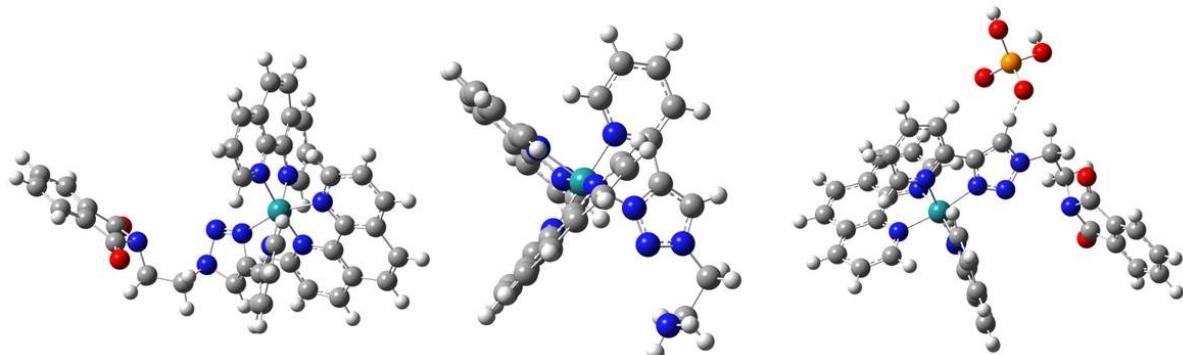


Fig.S44 Optimized structure of RtpH, RtpH- NH_2 , RtpH- H_2PO_4^- (from left to right)

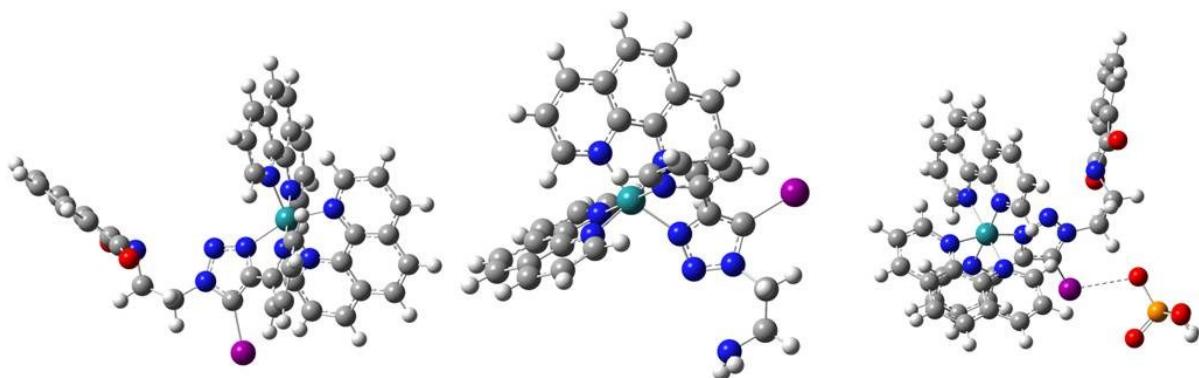


Fig.S45 Optimized structure of RtpI, RtpI-NH₂NH₂, RtpI-H₂PO₄⁻ (from left to right)

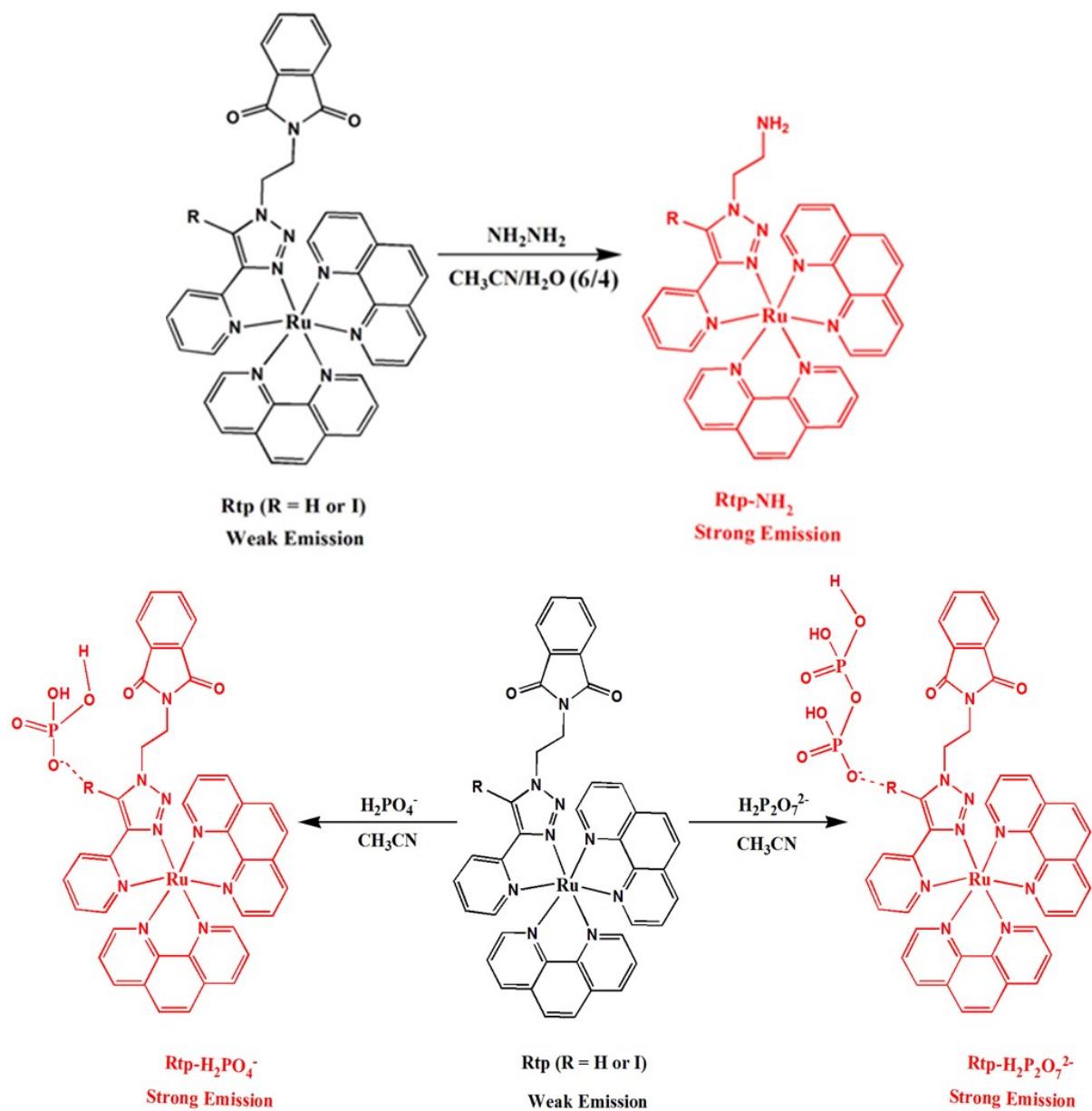


Fig.S46 Probable reaction mechanism of metallocceptor (RtpH/RtpI) with hydrazine (above) and phosphate anions (proposed anion binding mechanism, below)

<i>Complex</i>	<i>IC₅₀ Value (against HeLa cells)</i>
<i>Cis-Platin</i>	$7.2 \pm 1.2 \mu M$
<i>RtpH</i>	$28.1 \pm 3.0 \mu M$
<i>RtpI</i>	$22.5 \pm 1.7 \mu M$

Table S1. IC₅₀ Values for cis-Platin and metalloreceptors against HeLa cells