

## Supporting Information

### **Aqueous dependent sensing of hydrazine and phosphate anions using bis-heteroleptic Ru(II) Complex with phthalimide anchored pyridine-triazole ligand**

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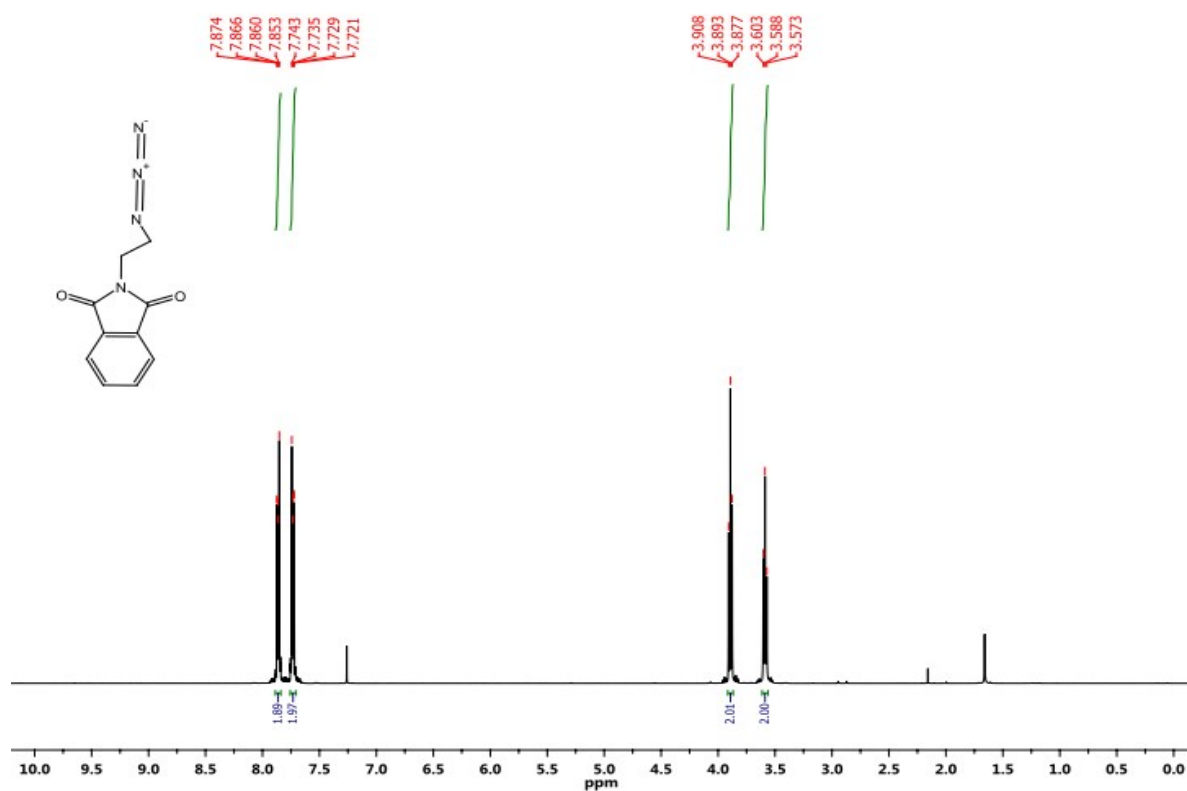
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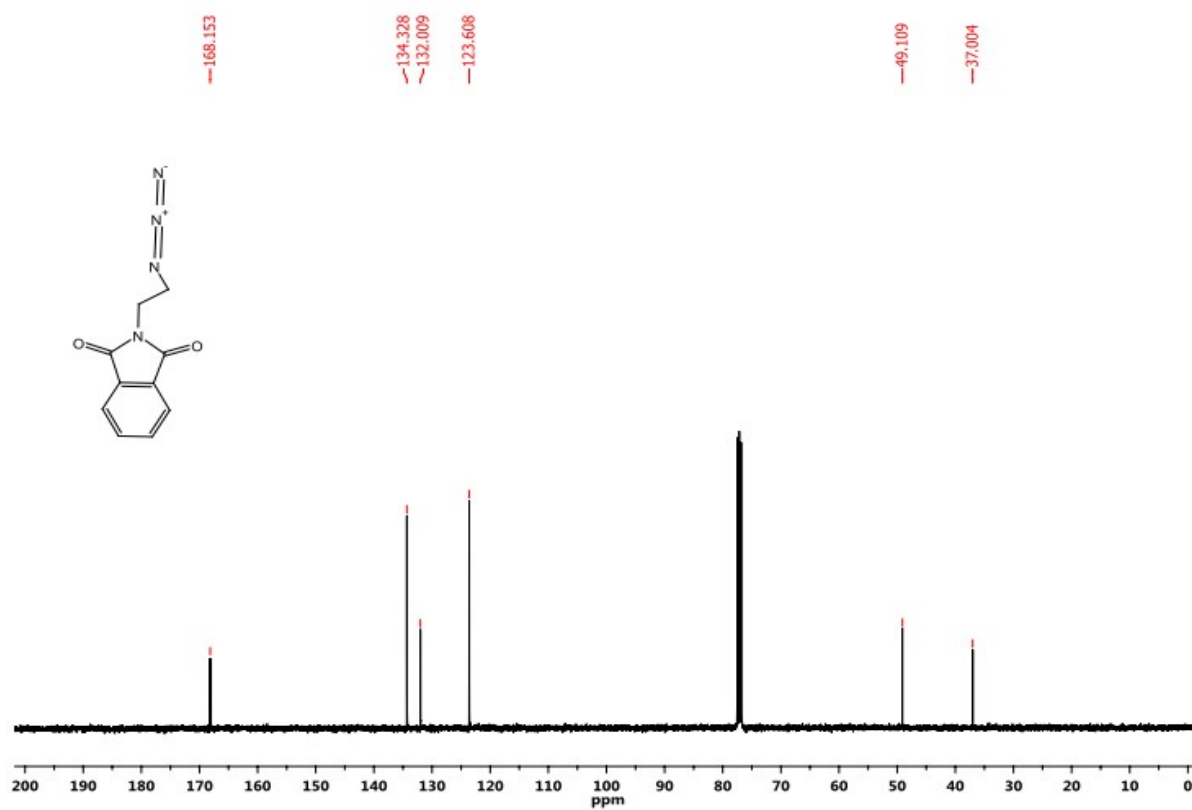
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## Instruments

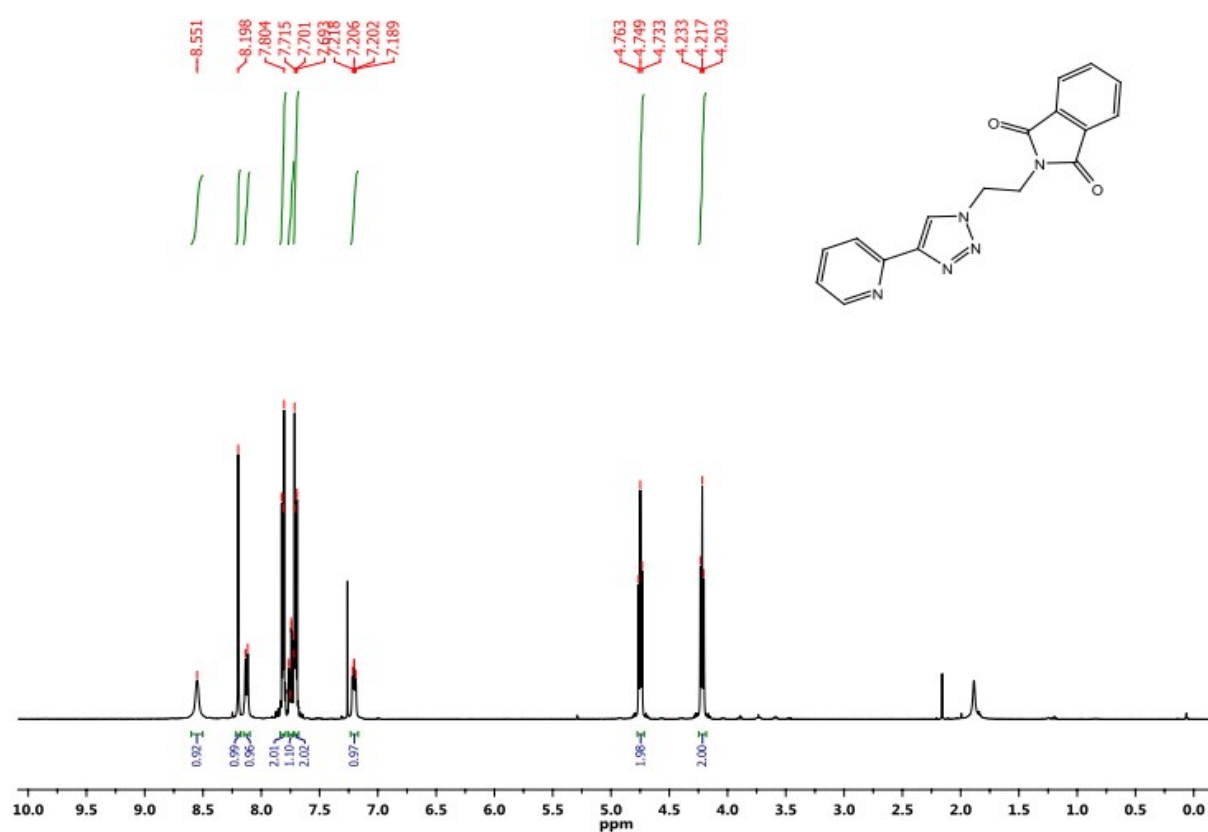
The NMR spectra were carried out in Bruker Avance 500 spectrometers using CDCl<sub>3</sub>, CD<sub>3</sub>CN, and DMSO-d<sub>6</sub> solvent (TMS as internal standard) in ppm. The FT-IR spectra were recorded on a Thermo Scientific Nicolet iS5 FT IR Spectrometer. The CHN elemental analysis was carried out from the Vario EL III microanalyzer. UV-Vis and Fluorescence spectra were carried out on Specord S 600 diode-array UV-Vis spectrophotometer and Shimadzu RF-5301 PC spectrofluorophotometer, respectively in RT. The luminescence lifetime was studied with the Time-Correlated Single Photon Counting method (TC-SPC) in a Horiba FluoroMax-4 instrument. The CV and DPV experiment were recorded on an Autolab three-electrode system, where Ag/AgCl as a reference electrode, glassy carbon as a working electrode, and platinum wire as a counter electrode in an acetonitrile solvent containing 0.1 M tetrabutylammonium hexafluorophosphate (tBu<sub>4</sub>N<sup>+</sup>PF<sub>6</sub><sup>-</sup>) as the supporting electrolyte with 100 mV scan rate and the system was calibrated using ferrocene/ferrocenium ion (Fc/Fc<sup>+</sup>) standard.



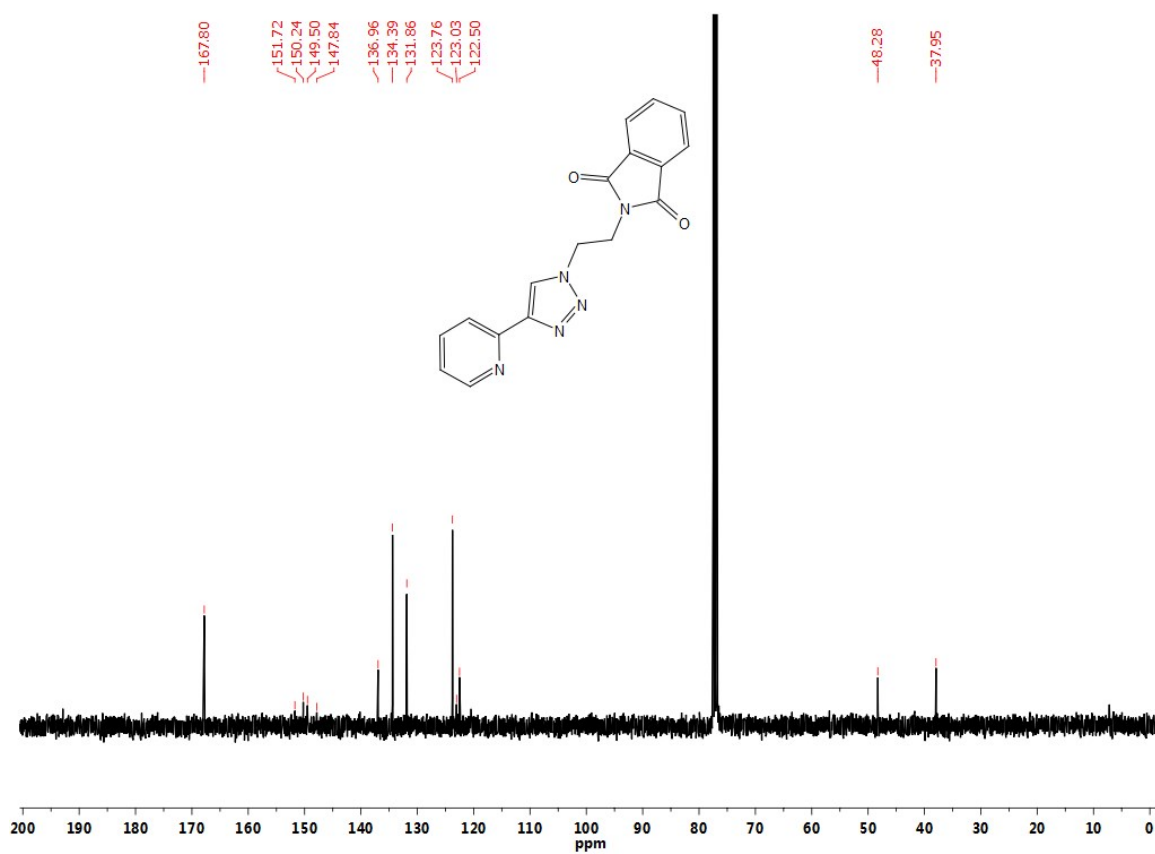
**Fig.S1** <sup>1</sup>H NMR spectrum of TpN<sub>3</sub> in CDCl<sub>3</sub>



**Fig.S2** <sup>13</sup>C NMR spectrum of TpN<sub>3</sub> in CDCl<sub>3</sub>

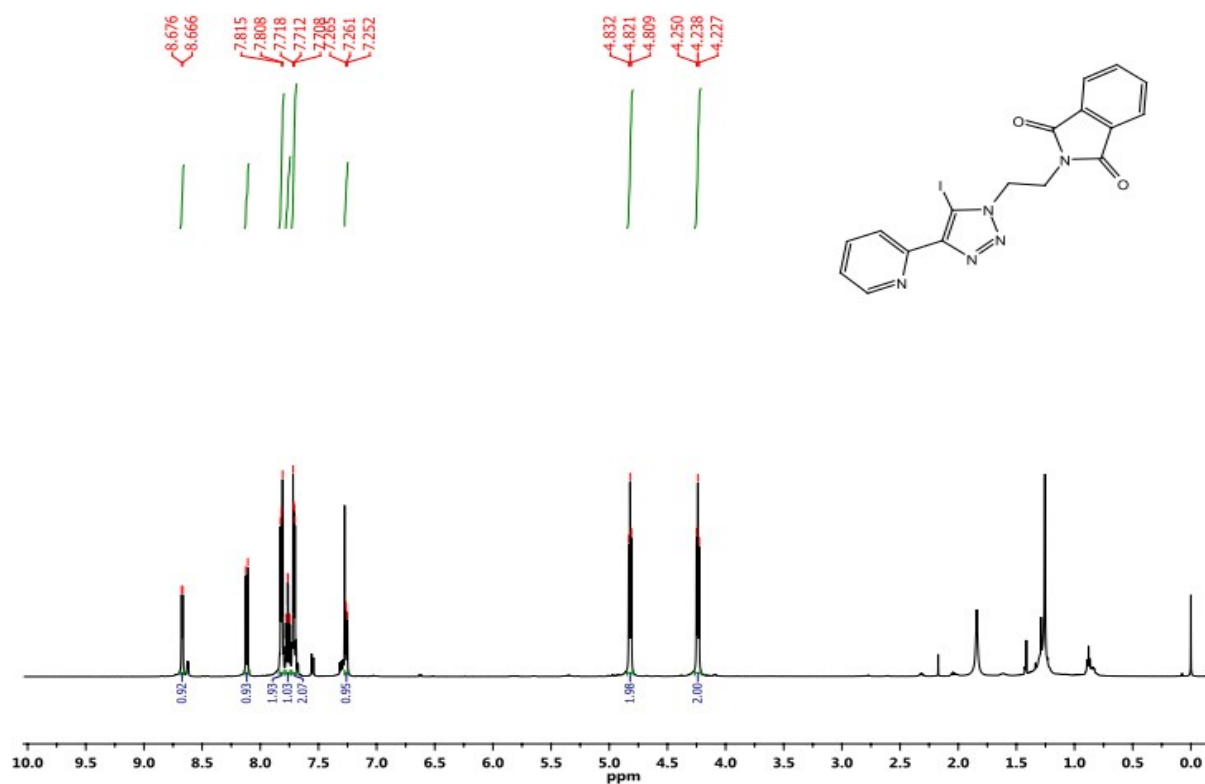


**Fig.S3** <sup>1</sup>H NMR spectrum of TpH in CDCl<sub>3</sub>

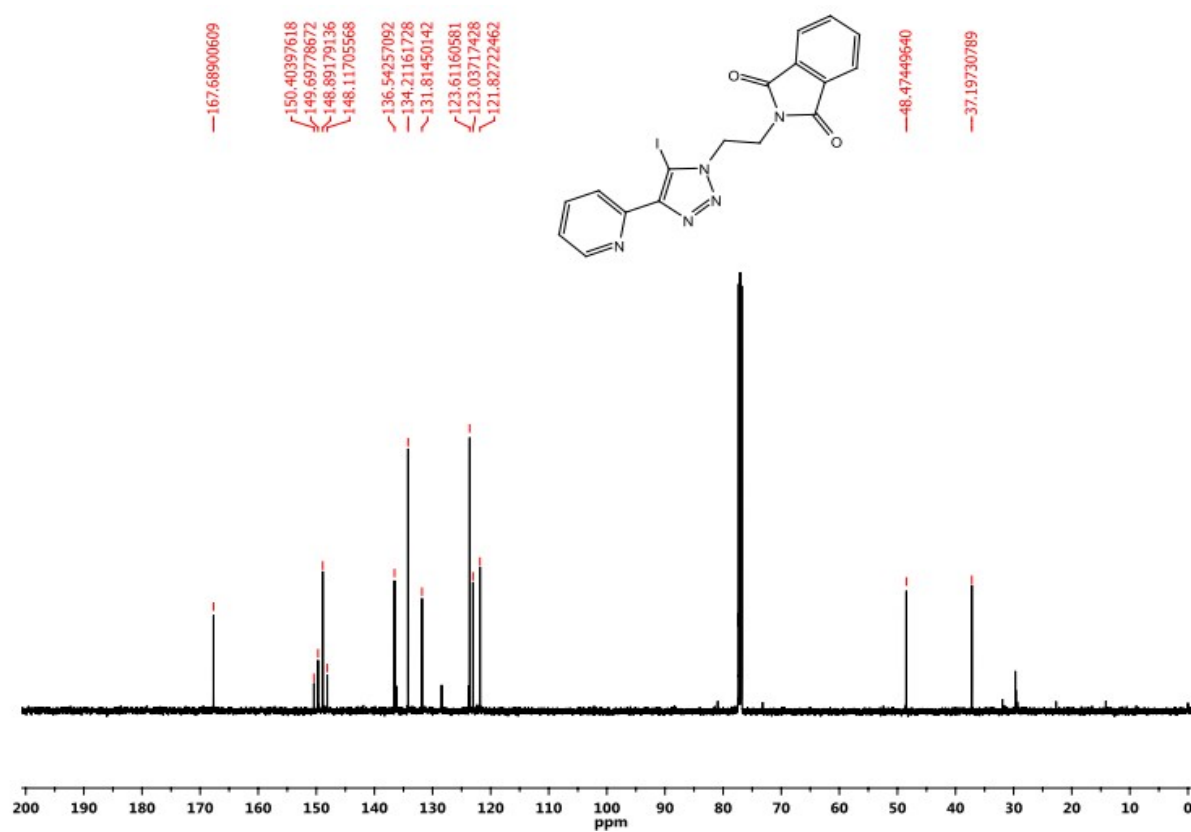


**.S4** <sup>13</sup>C NMR spectrum of TpH in CDCl<sub>3</sub>

**Fig**



**Fig.S5** <sup>1</sup>H NMR spectrum of TpI in CDCl<sub>3</sub>



**Fig.S6** <sup>13</sup>C NMR spectrum of TpI in CDCl<sub>3</sub>

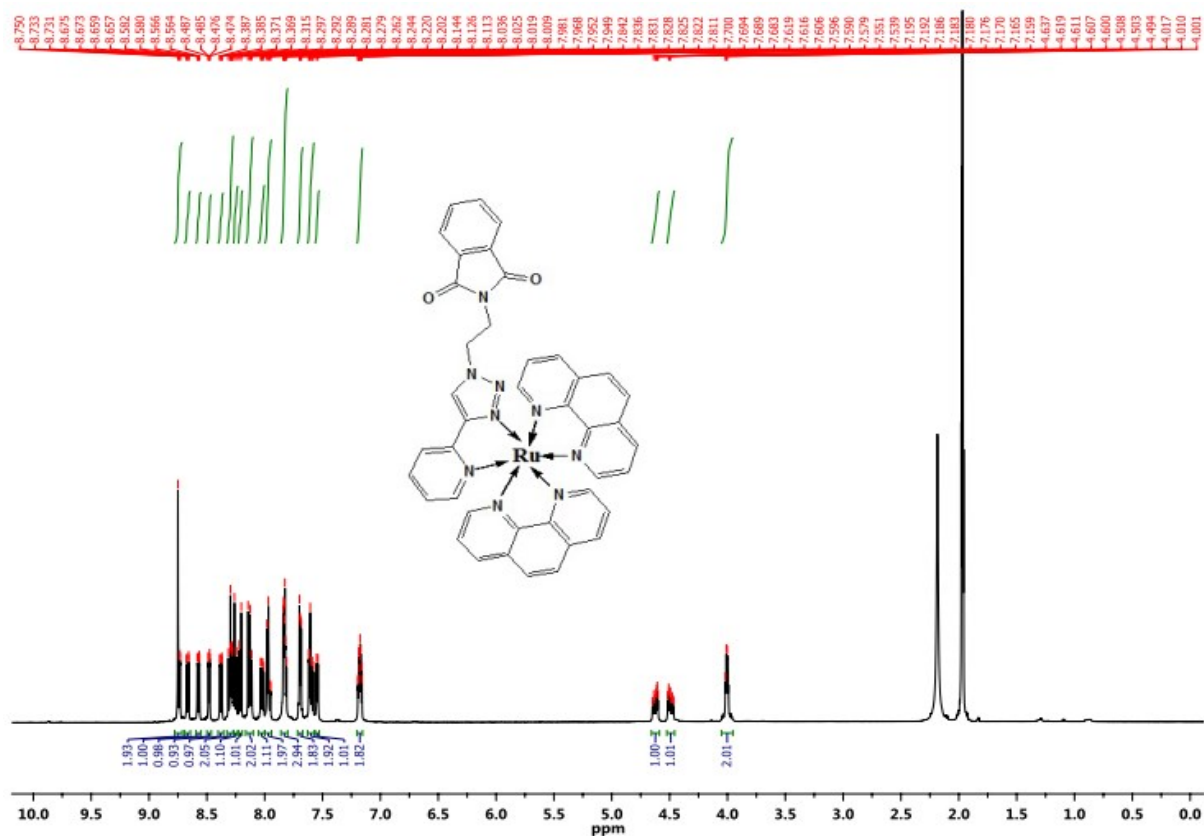


Fig.S7  $^1\text{H}$  NMR spectrum of RtpH in  $\text{CD}_3\text{CN}$

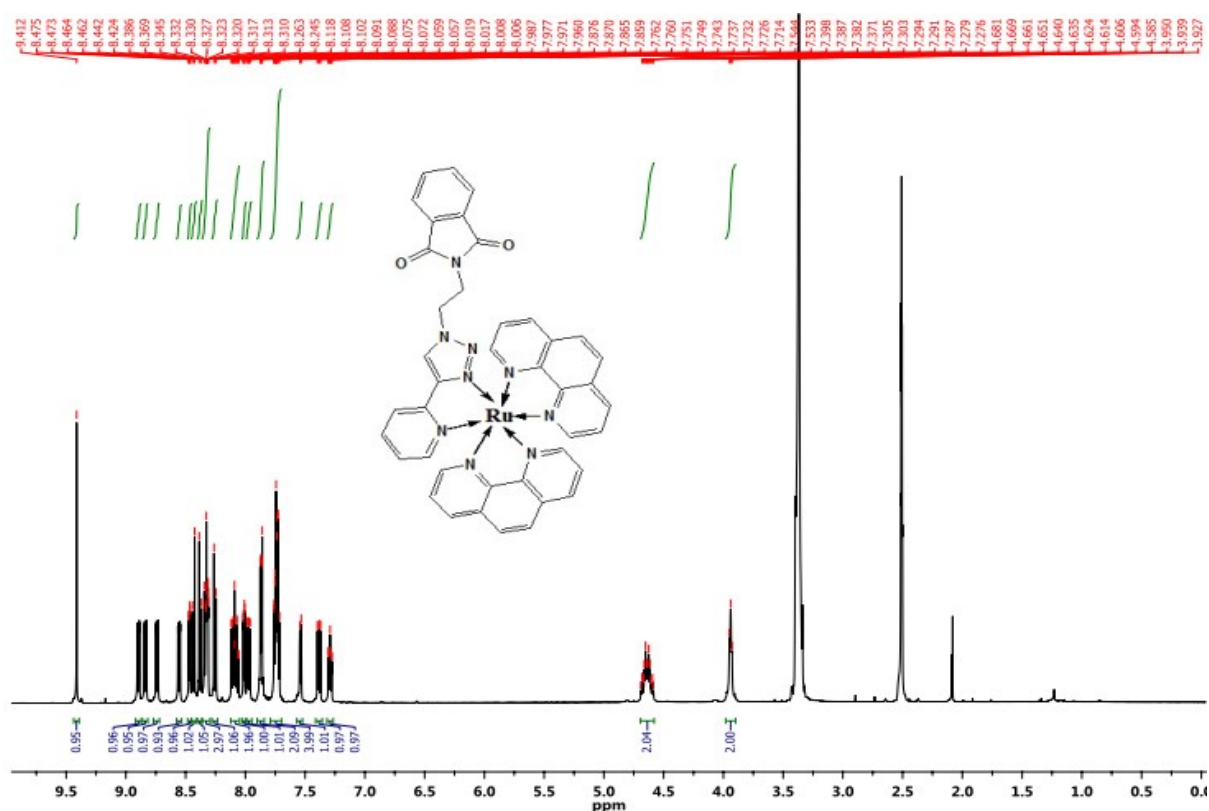


Fig.S8  $^1\text{H}$  NMR spectrum of RtpH in  $\text{DMSO-d}_6$



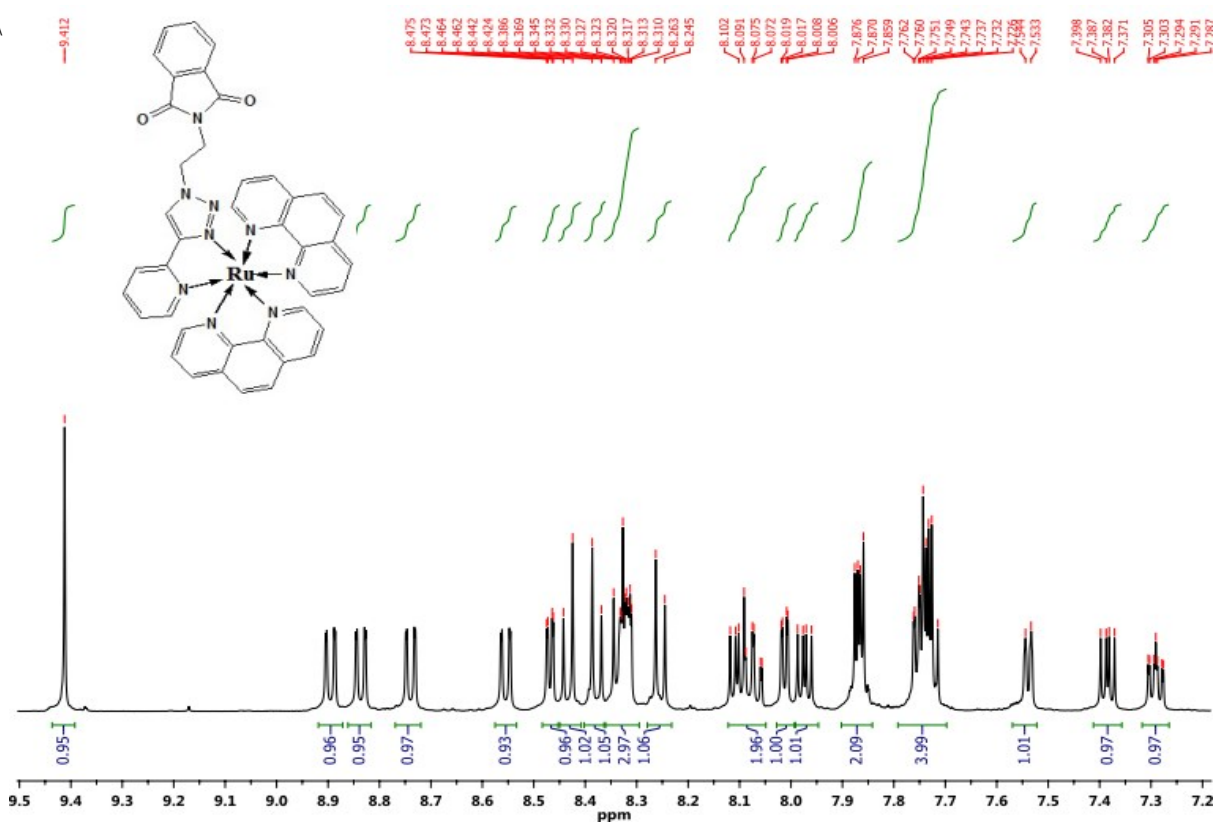


Fig.S9 Expanded  $^1\text{H}$  NMR spectrum of RtpH in  $\text{DMSO-d}_6$

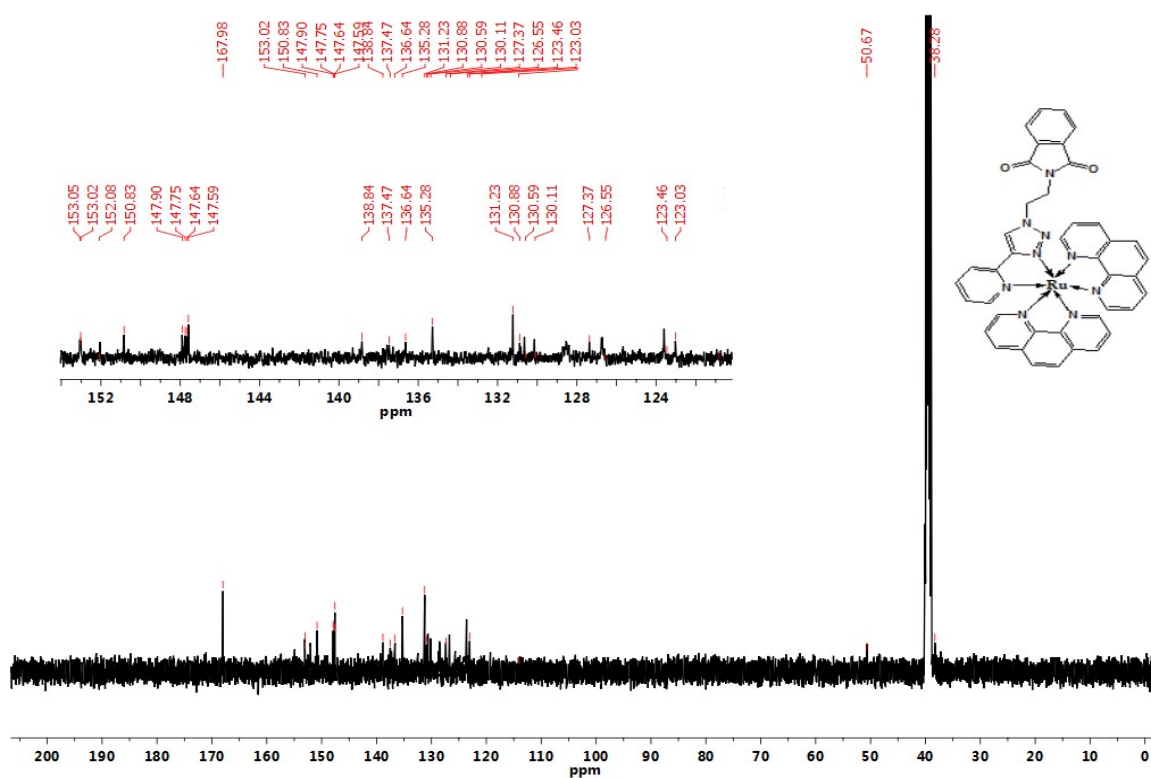
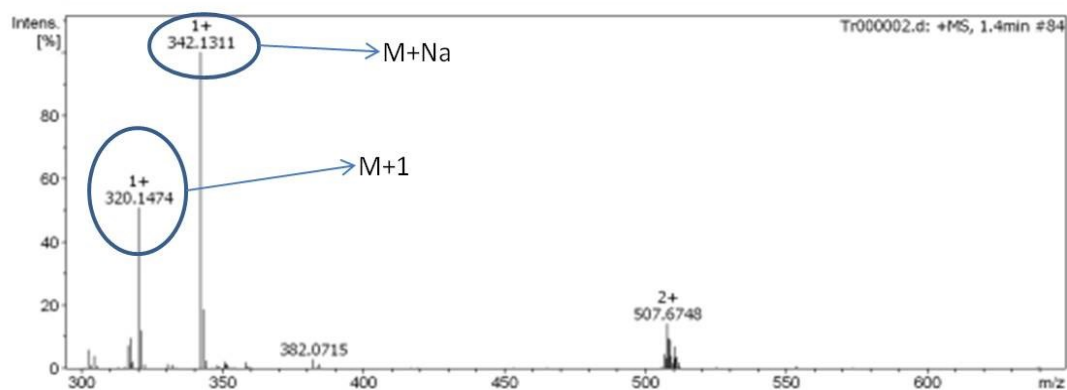
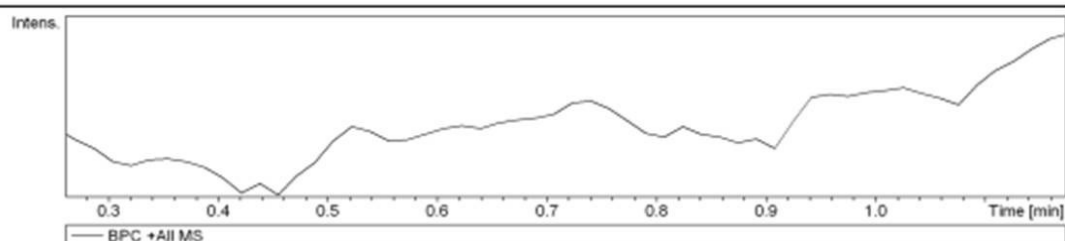


Fig.S10  $^{13}\text{C}$  NMR spectrum of RtpH in  $\text{DMSO-d}_6$



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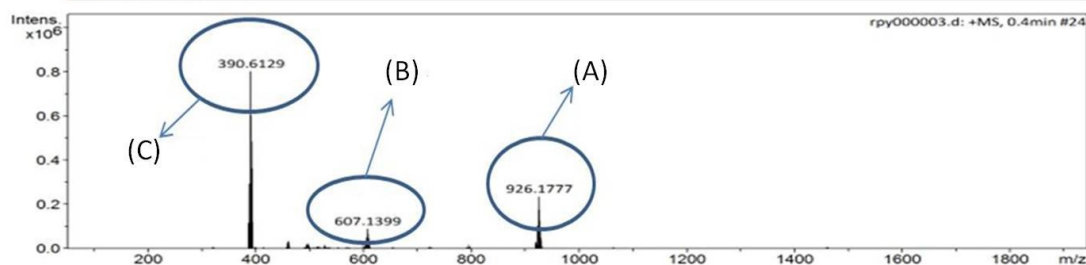
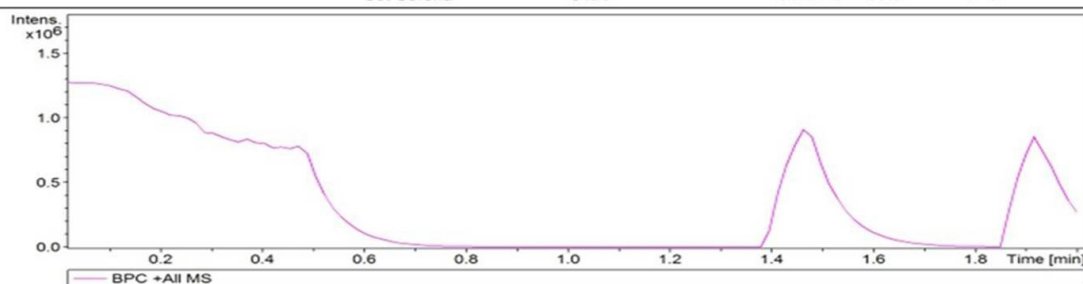
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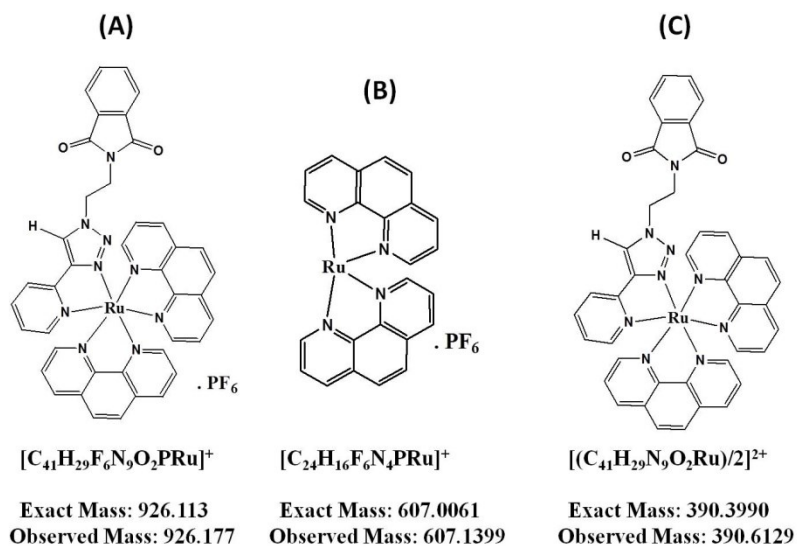


**Fig.S13** HR ESI-Mass spectrum of TpH

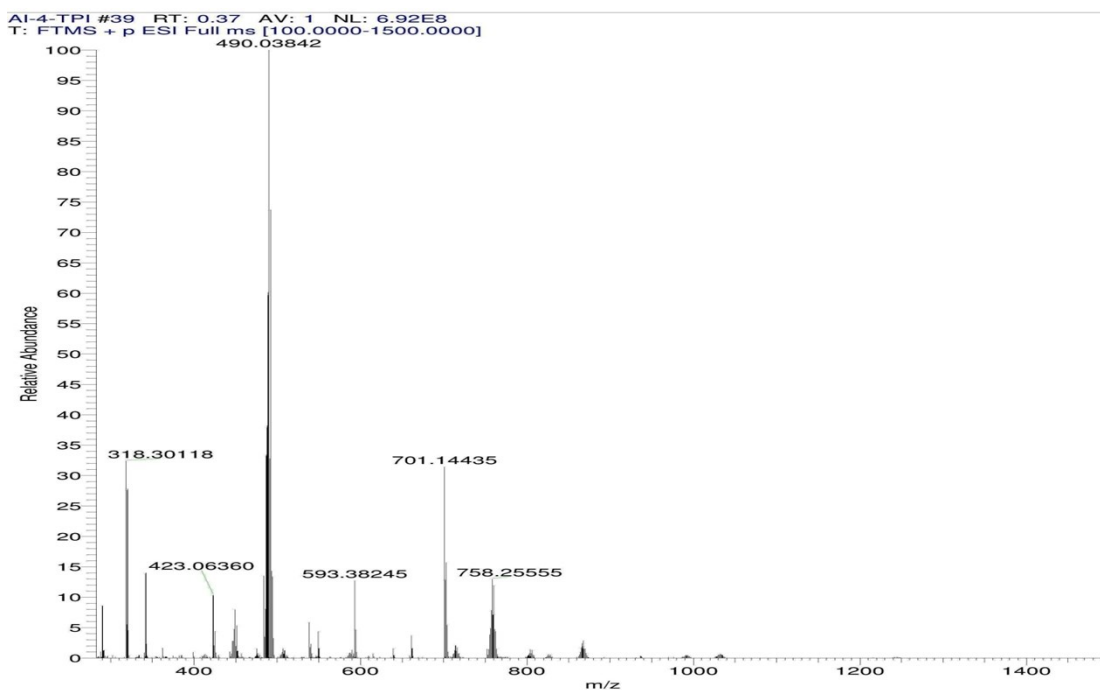
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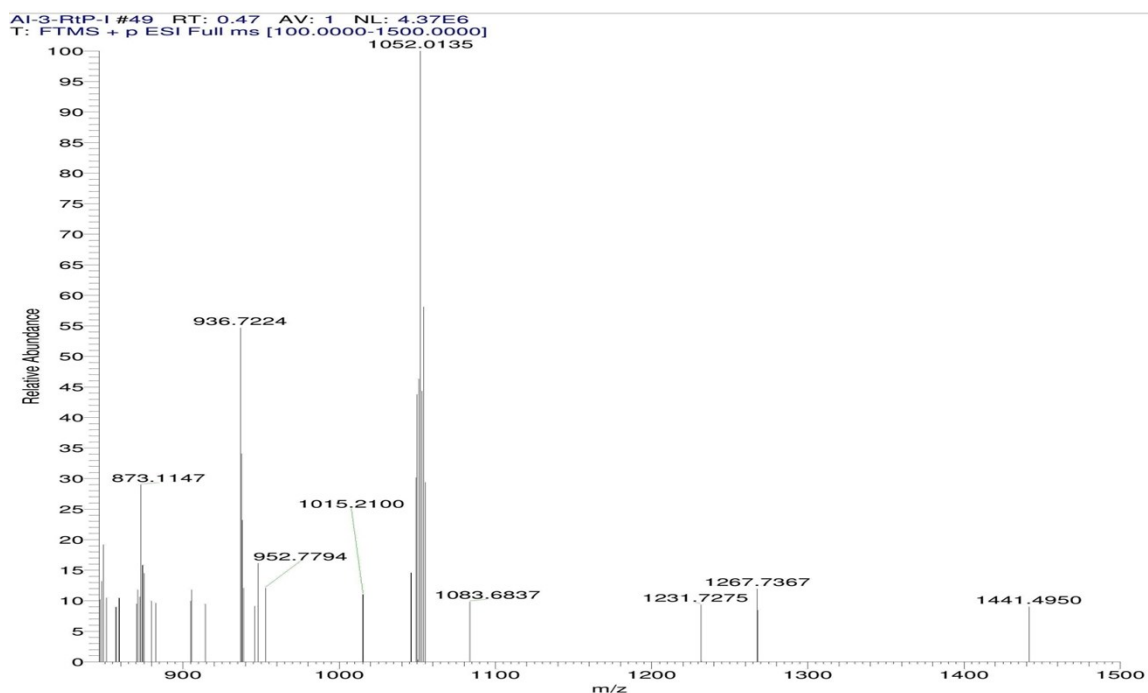




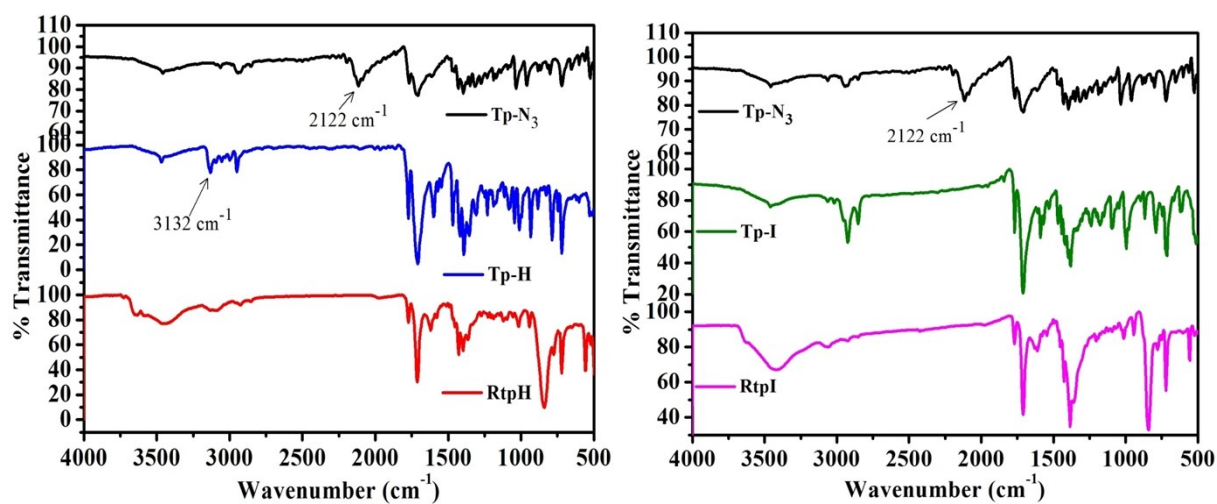
**Fig.S14** HR ESI-Mass spectrum of RtpH



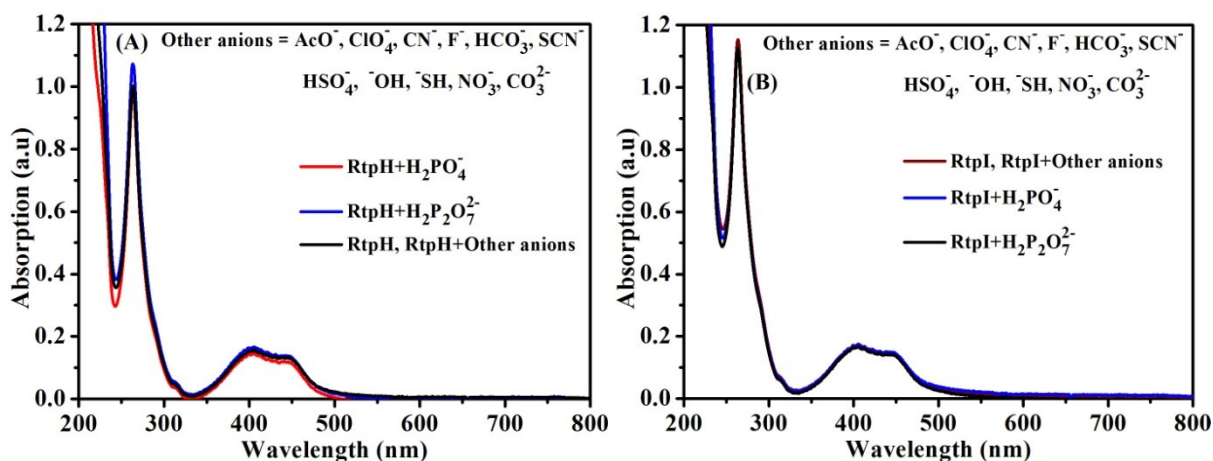
**Fig. S15** HR-ESI Mass spectrum of TplI (Calculated value for  $[C_{17}H_{12}IN_5O_2Ru]^+ + 2Na - H = 489.9747$ , observed = 490.0384)



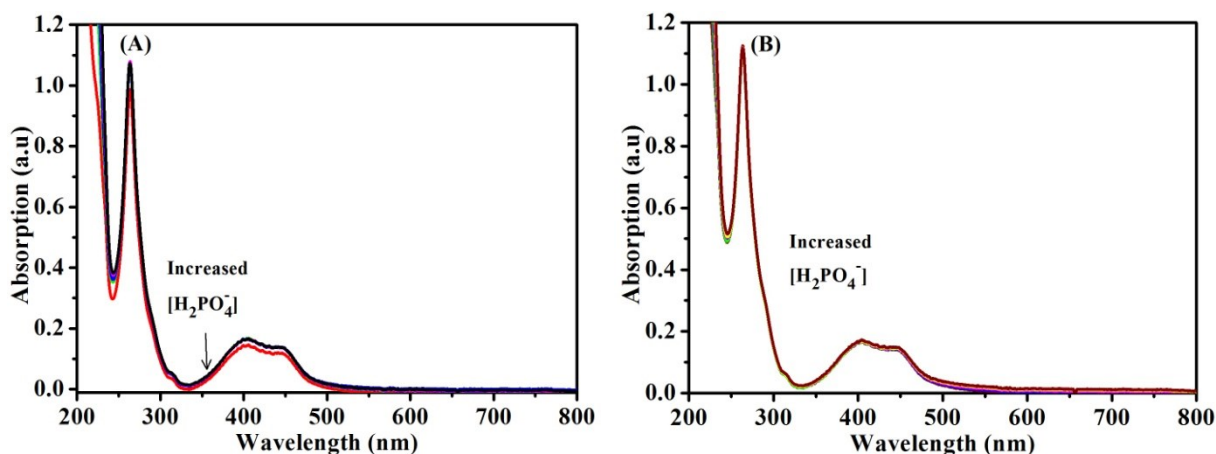
**Fig. S16** HR-ESI Mass spectrum of RtpI (Calculated value for  $[C_{41}H_{28}F_6IN_9O_2PRu]^+ = 1052.0101$ , observed = 1052.0135)



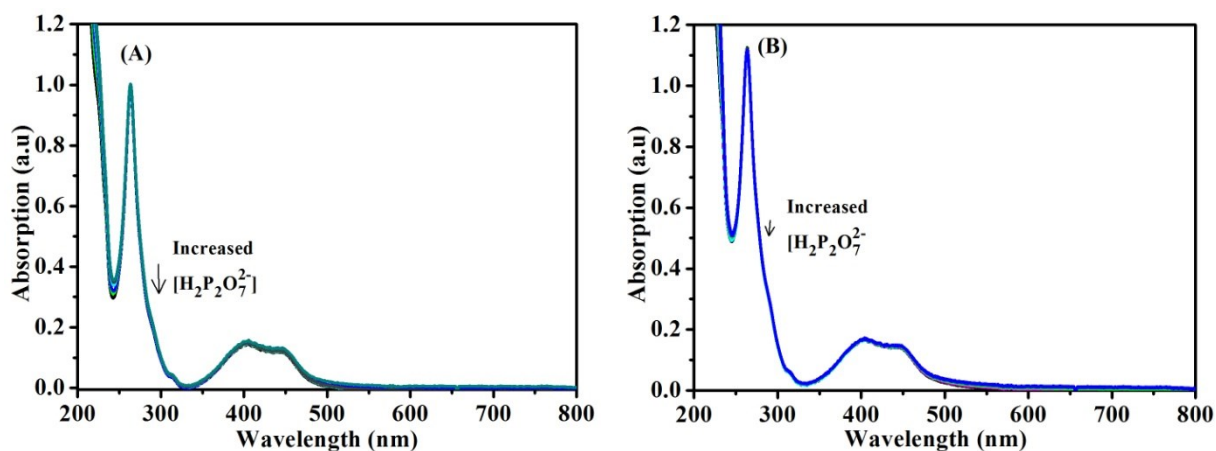
**Fig.S17** FT-IR spectra of TpN<sub>3</sub>,TpH, TpI, RtpH and RtpI.



**Fig.S18** (A) UV-Vis spectra of RtpH (10  $\mu\text{M}$ ) with various anions (5 equiv.) and phosphate anions ( $\text{H}_2\text{PO}_4^-$  and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$ ) (1 equiv.), (B) Uv-Vis spectra of RtpI (10  $\mu\text{M}$ ) with various anions (5 equiv.) and phosphate anions ( $\text{H}_2\text{PO}_4^-$  and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$ ) (1 equiv.) in  $\text{CH}_3\text{CN}$ .

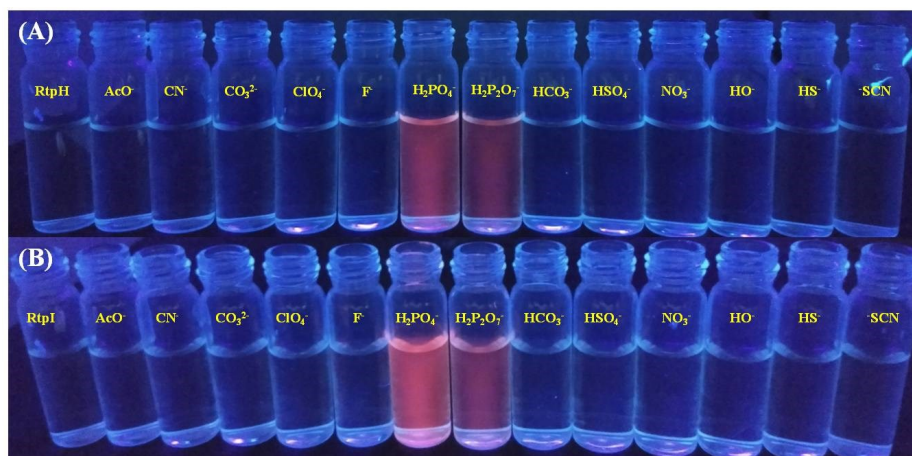


**Fig.S19** (A) UV-Vis spectra of RtpH (10  $\mu\text{M}$ ) with the addition of  $\text{H}_2\text{PO}_4^-$  anions (0-1 equiv.), (B) Uv-Vis spectra of RtpI (10  $\mu\text{M}$ ) with the addition of  $\text{H}_2\text{PO}_4^-$  anions (0-1 equiv.) in  $\text{CH}_3\text{CN}$ .

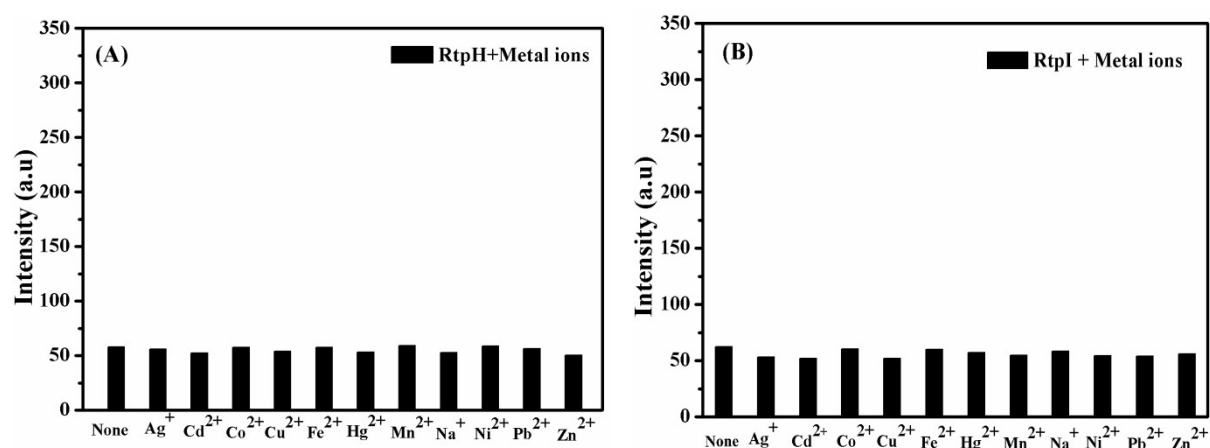


**Fig.S20** (A) UV-Vis spectra of RtpI (10  $\mu\text{M}$ ) with the addition of  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  anions (0-1 equiv.), (B) Uv-Vis spectra of RtpI (10  $\mu\text{M}$ ) with the addition of  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  anions (0-1 equiv.) in  $\text{CH}_3\text{CN}$ .

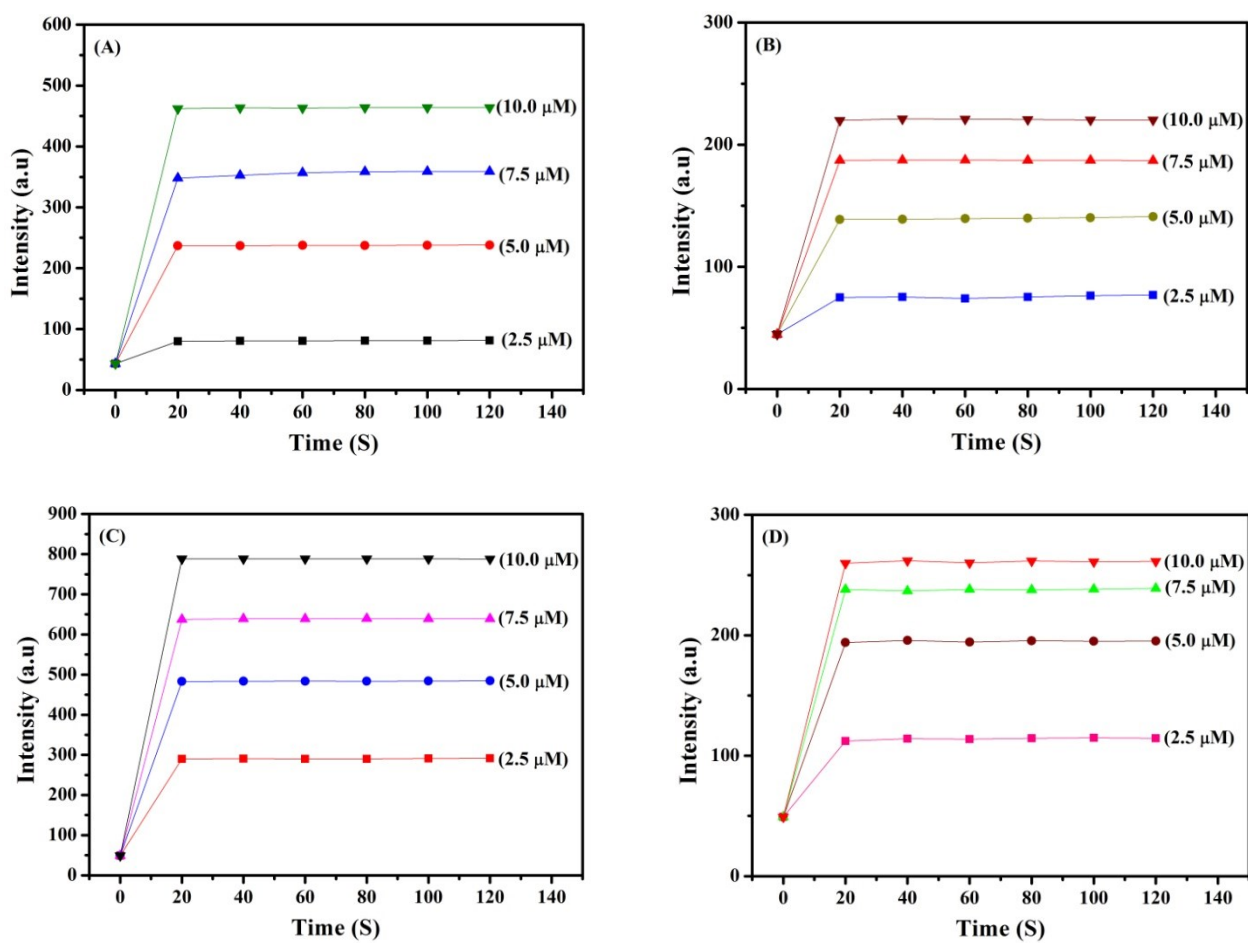




**Fig.S21** Under UV light illumination, (A) RtpH and (B) RtpI with various types of anions in CH<sub>3</sub>CN

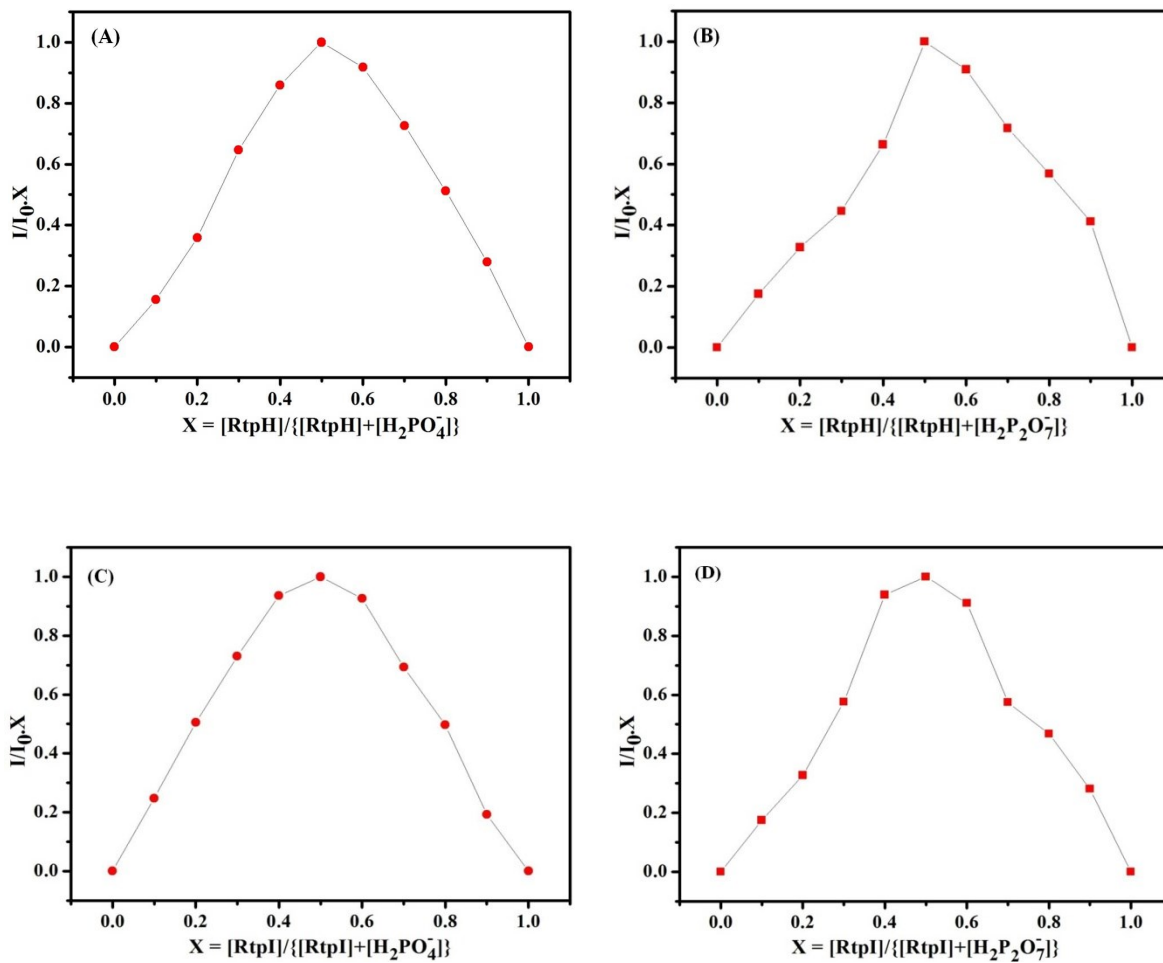


**Fig.S22** (A) Emission intensity of RtpH (10  $\mu$ M) with different metal ions (10 equiv.), (B) Emission intensity of RtpI (10  $\mu$ M) with different metal ions (10 equiv.) in CH<sub>3</sub>CN.

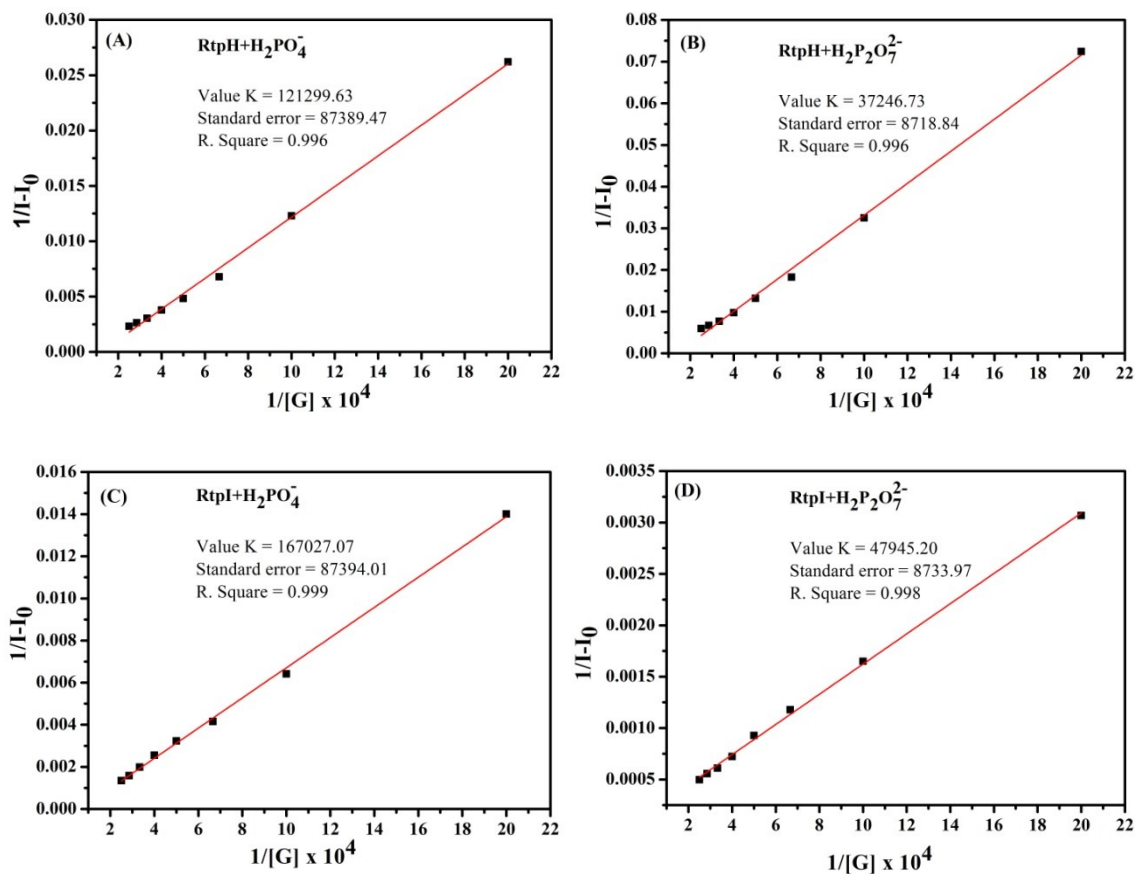


**Fig.S23** (A) Time course study of RtpH (10  $\mu\text{M}$ ) upon addition different equivalent of  $\text{H}_2\text{PO}_4^-$  (B)  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  in  $\text{CH}_3\text{CN}$  and time course study of RtpI (10  $\mu\text{M}$ ) upon addition different equivalent of  $\text{H}_2\text{PO}_4^-$  (C)  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  (D) in  $\text{CH}_3\text{CN}$ .

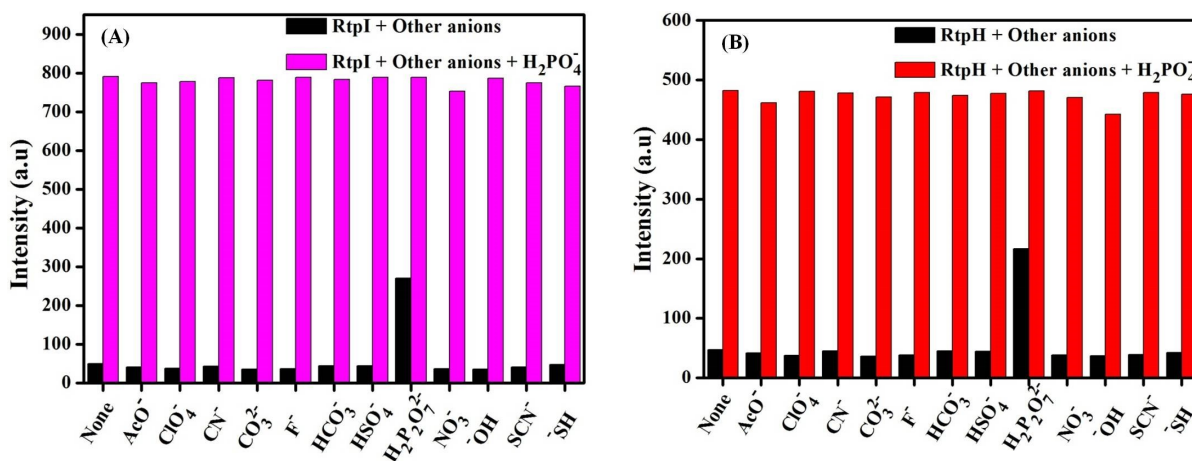




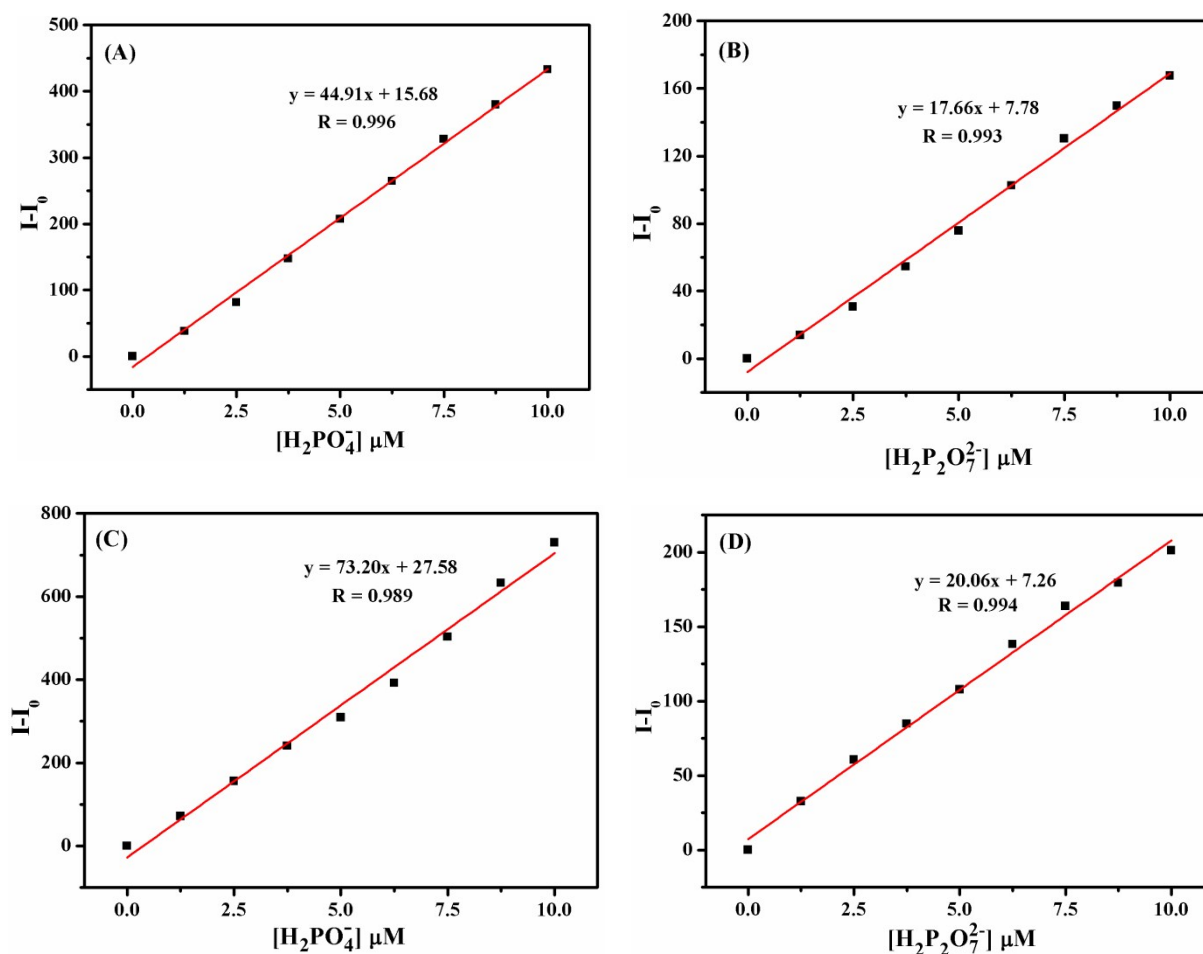
**Fig.S24** Job's plot of RtpH with (A)  $H_2PO_4^-$  (B)  $H_2P_2O_7^{2-}$ , Job's plot of RtpI with (A)  $H_2PO_4^-$  (B)  $H_2P_2O_7^{2-}$



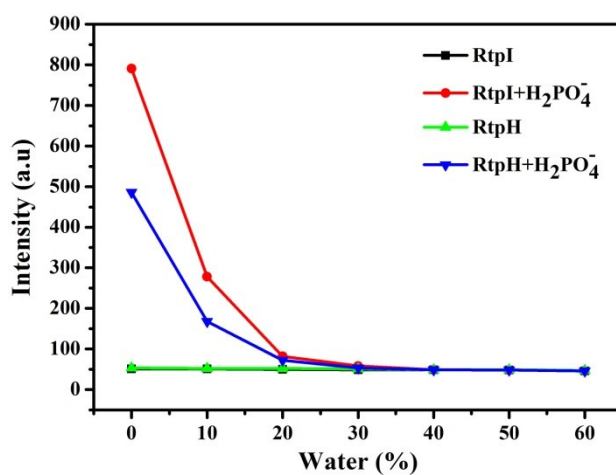
**Fig. S25.** Benesi-Hildebrand plot from emission titration data of RtpH (10  $\mu\text{M}$ ) with  $\text{H}_2\text{PO}_4^-$  (A) and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  (B) and of RtpI with  $\text{H}_2\text{PO}_4^-$  (C) and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  (D).



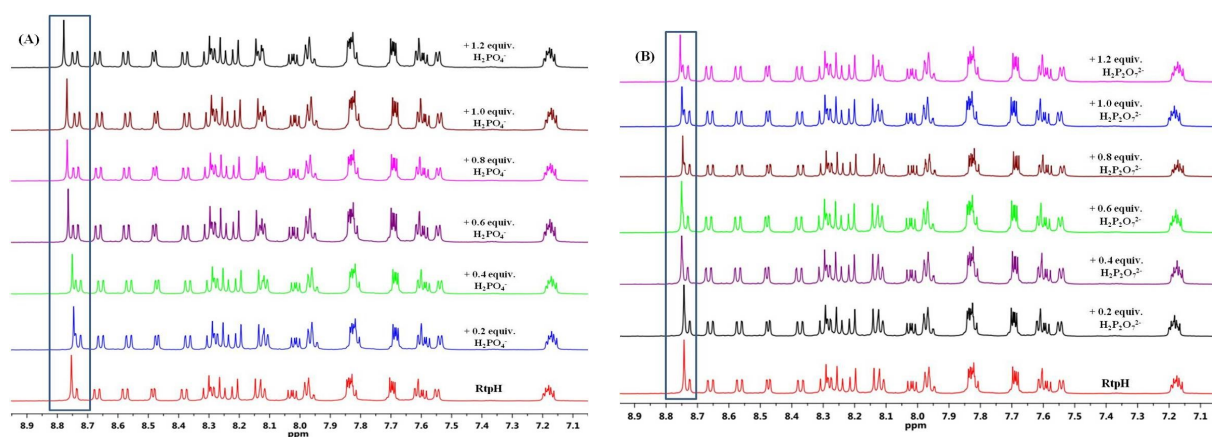
**Fig.S26** (A) Interference study of RtpI (10  $\mu\text{M}$ ) with and without  $\text{H}_2\text{PO}_4^-$  anions in presence of different anions (10 equiv.), (B) Interference study of RtpH (10  $\mu\text{M}$ ) with and without  $\text{H}_2\text{PO}_4^-$  anions in presence of different anions (10 equiv.).



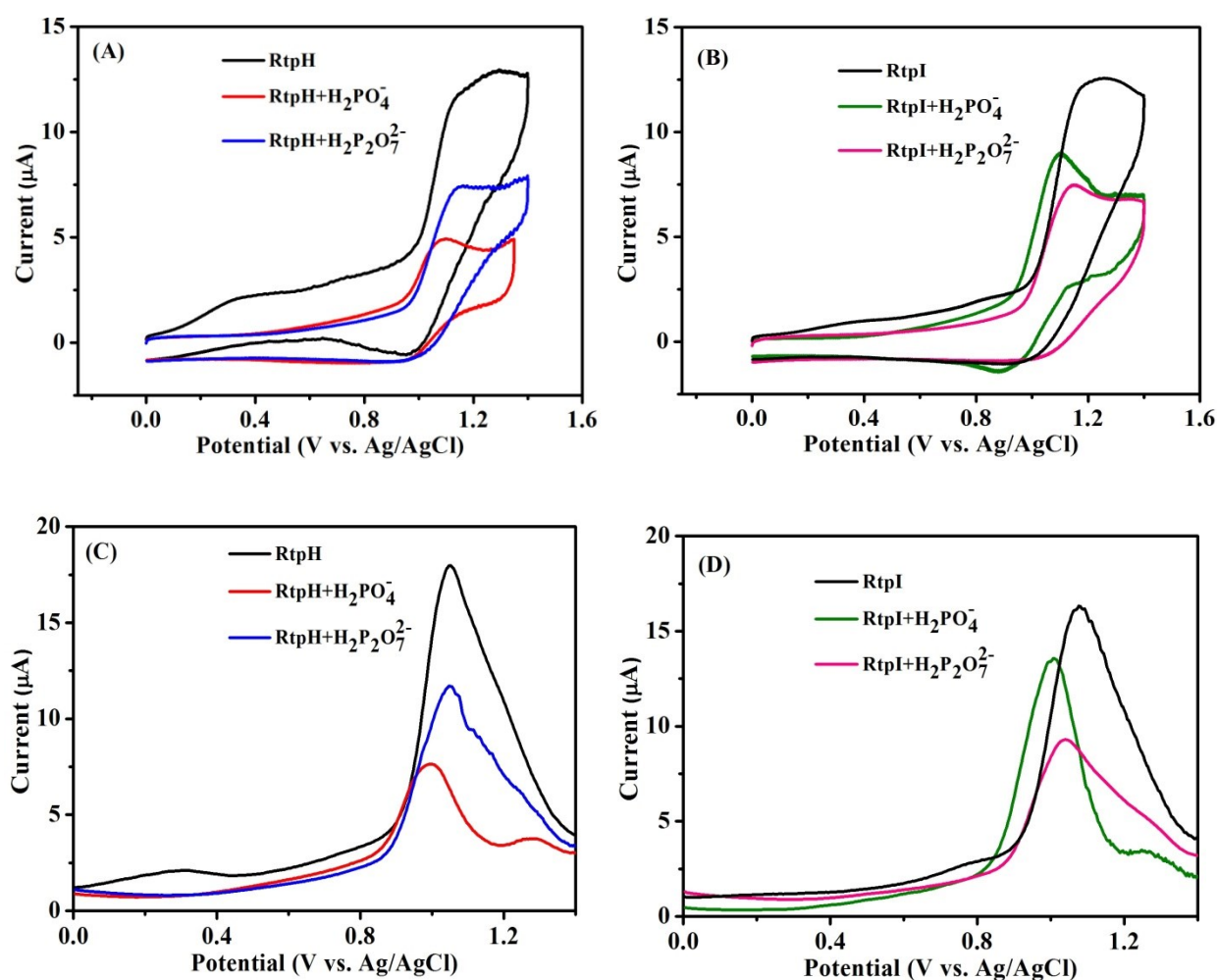
**Fig.S27** A linear plot of RtpH (A, B) and RtpI (C, D) with  $\text{H}_2\text{PO}_4^-$  and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$  in  $\text{CH}_3\text{CN}$ .



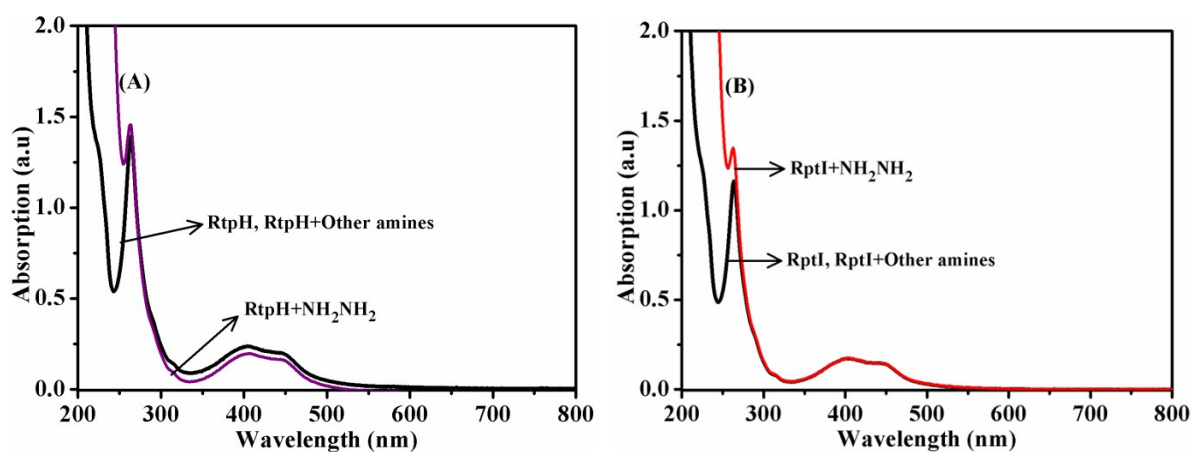
**Fig. S28** The emission intensity of metalloreceptors (RtpH and RtpI, 10  $\mu\text{M}$ ) with phosphate anions ( $\text{H}_2\text{PO}_4^-$  and  $\text{H}_2\text{P}_2\text{O}_7^{2-}$ , 10  $\mu\text{M}$ ) upon varying water percentage with  $\text{CH}_3\text{CN}$ .



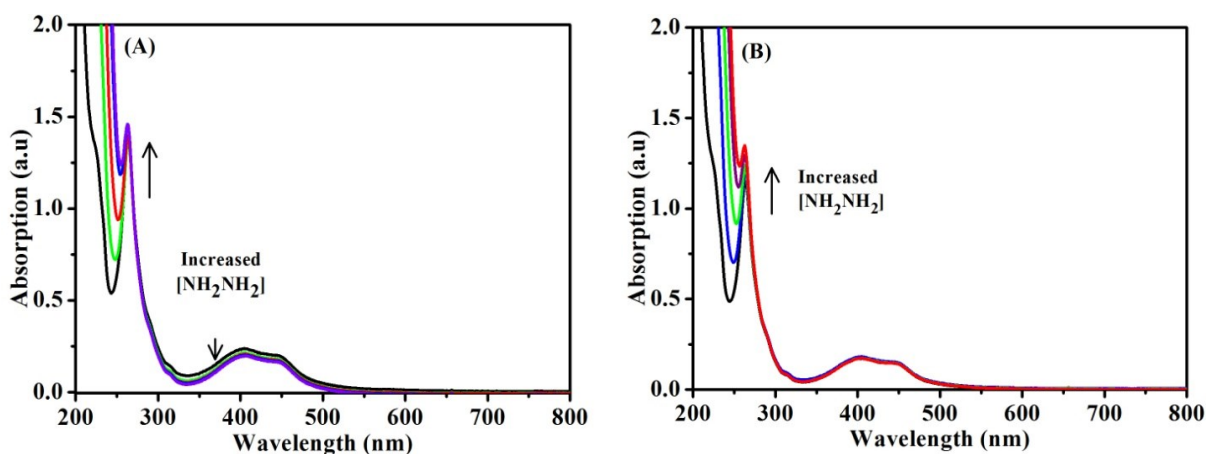
**Fig.S29** (A)  $^1\text{H}$  NMR titration of RtpH (7.5 mM) with  $\text{H}_2\text{PO}_4^-$  (0-1.2 equiv.), (B)  $^1\text{H}$  NMR titration of RtpI (7.5 mM) with  $\text{H}_2\text{PO}_4^-$  (0-1.2 equiv.) in  $\text{CD}_3\text{CN}/\text{D}_2\text{O}$  (9/1, v/v).



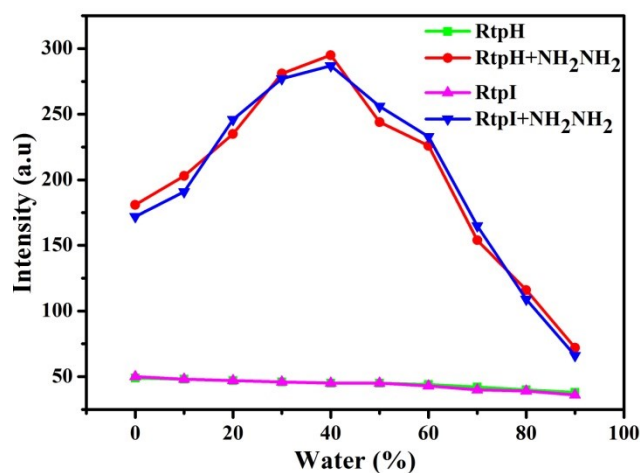
**Fig.S30** (A) Cyclic Voltammetry (A and B) and Differential Pulse Voltammetry (C and D) of metalloreceptors (RtpH and RtpI) with and without phosphate anions.



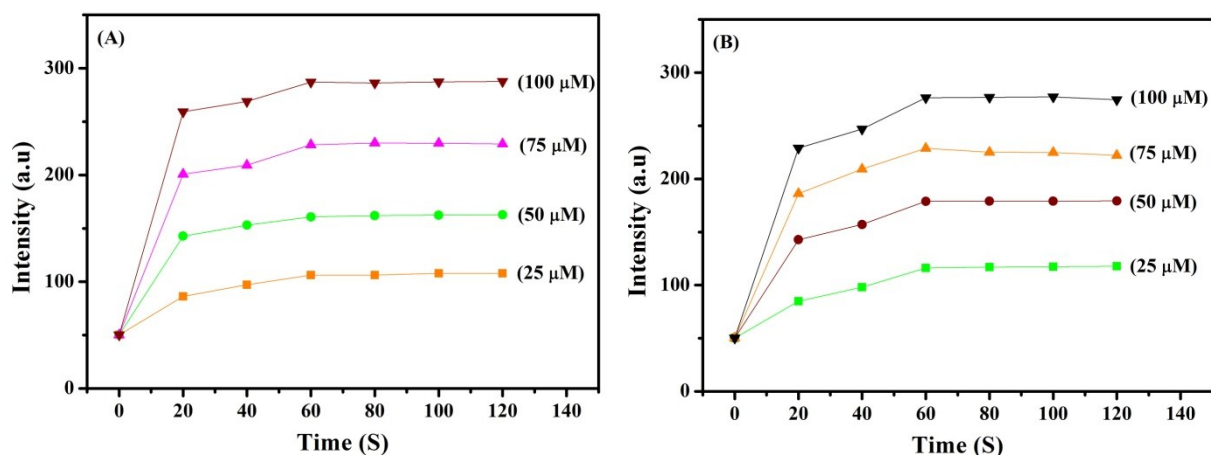
**Fig.S31** (A) UV-Vis spectra of RtpH (25 μM) with various amines (10 equiv.) and hydrazine (4 equiv.), (B) UV-Vis spectra of RtpI (25 μM) with various amines (10 equiv.) and hydrazine (4 equiv.) in CH<sub>3</sub>CN/H<sub>2</sub>O (6/4, v/v).



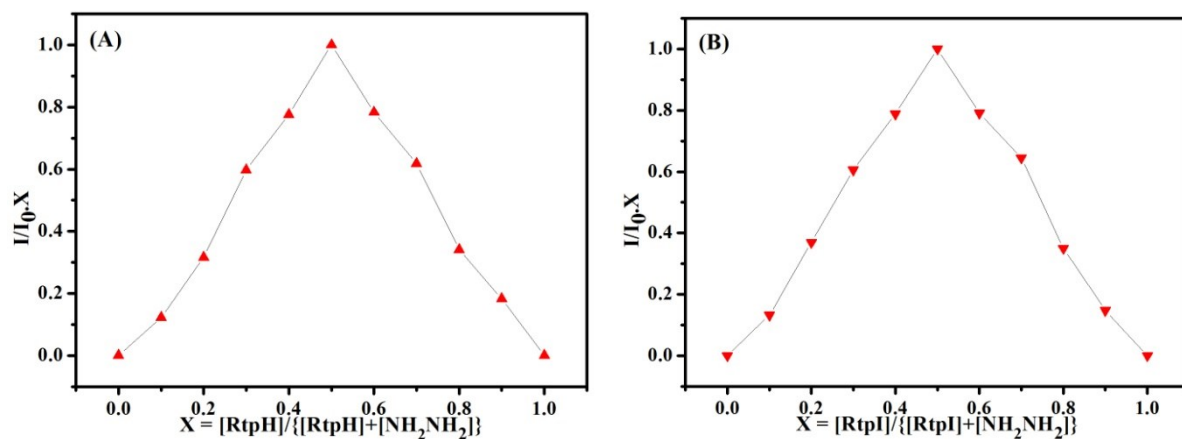
**Fig.S32** (A) UV-Vis spectra of RtpH (25 μM) with the gradual addition of hydrazine (4 equiv.), (B) UV-Vis spectra of RtpI (25 μM) with the gradual addition of hydrazine (4 equiv.) in CH<sub>3</sub>CN/H<sub>2</sub>O (6/4, v/v).



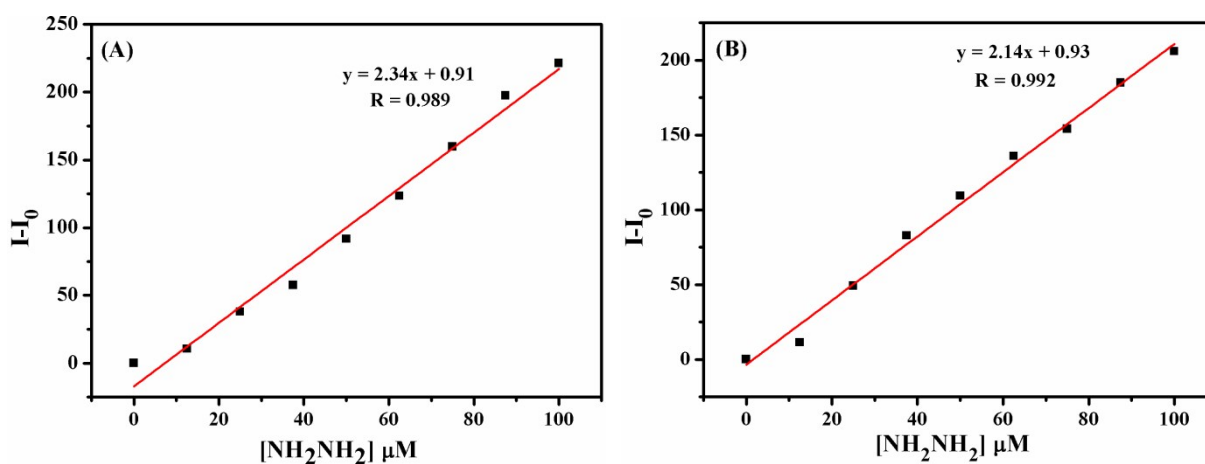
**Fig. S33** The emission intensity of metalloreceptors (RtpH and RtpI, 25 μM) with hydrazine (100 μM) upon varying water percentage with CH<sub>3</sub>CN



**Fig. S34** (A) Time course study of RtpH (25 μM) upon addition different equivalent of hydrazine (B) Time course study of RtpH (25 μM) upon addition different equivalent of hydrazine

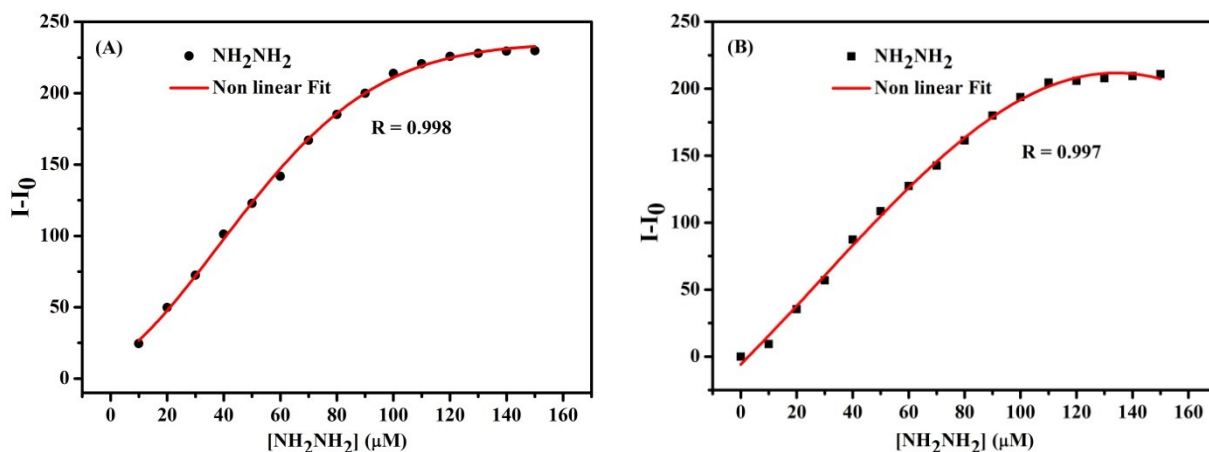


**Fig.S35** (A) Job's plot of RtpH with hydrazine, (B) Job's plot of RtpI with hydrazine in  $CH_3CN/H_2O$  (6/4, v/v)

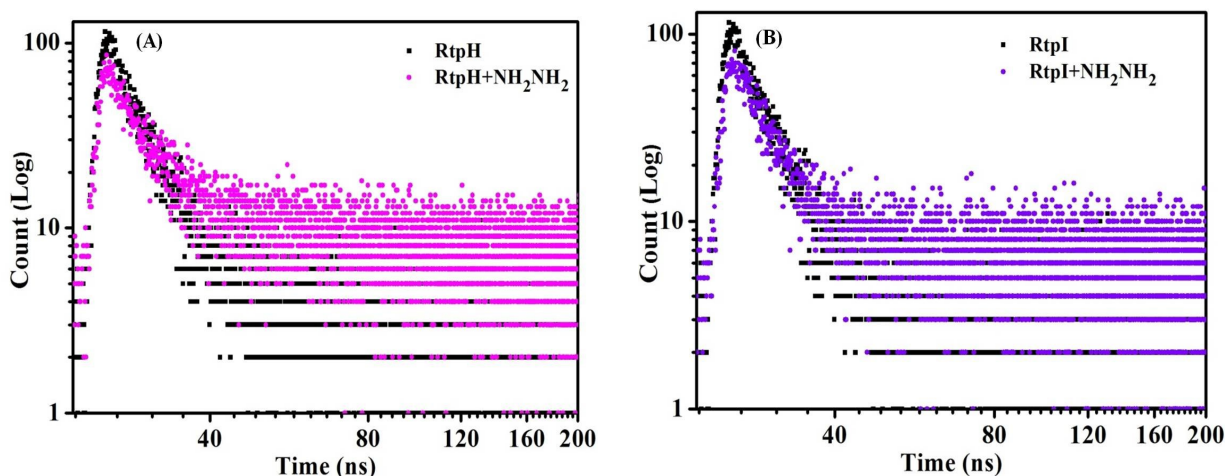


**Fig.S36** (A) A linear plot of RtpH with gradual addition of hydrazine (B) A linear plot of RtpI with gradual addition of hydrazine in  $CH_3CN/H_2O$  (6/4, v/v).

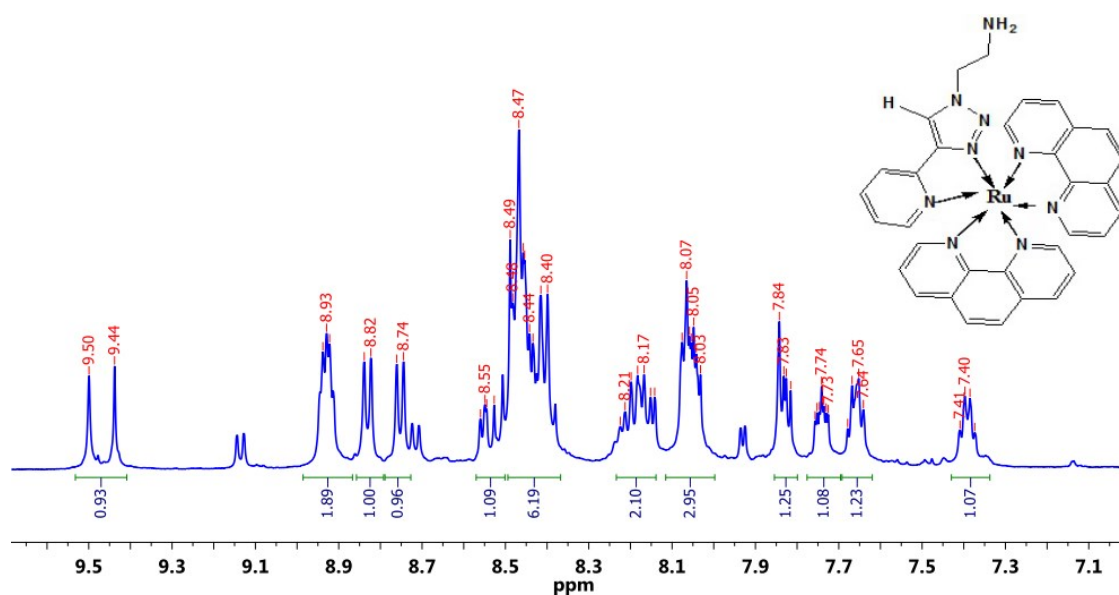




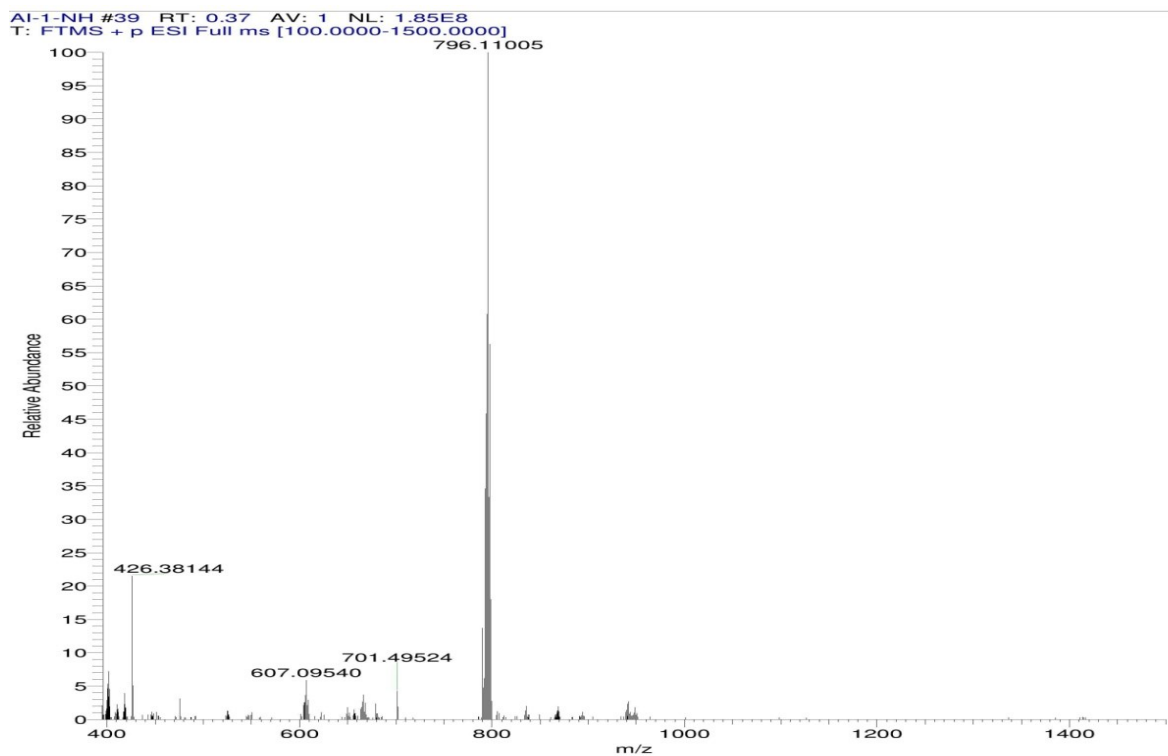
**Fig. S37** (A) Non-linear fitting of RtpH (A)/ RtpI (B) (25  $\mu M$ ) upon addition different concentration of hydrazine



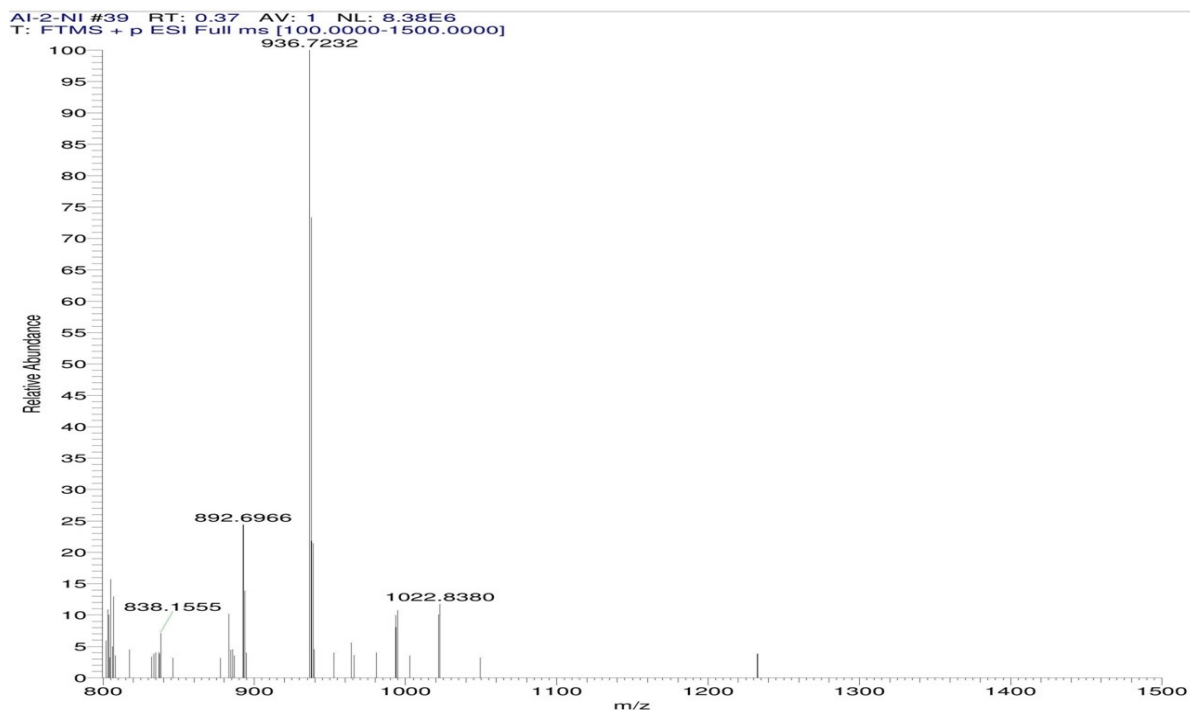
**Fig.S38** (A) Lifetime decay of RtpH (10  $\mu M$ ) with and without hydrazine (4 equiv.), (B) Lifetime decay of RtpI (10  $\mu M$ ) with and without hydrazine (4 equiv.) in  $CH_3CN/H_2O$  (6/4, v/v).



**Fig. S39.** Partial  $^1H$  NMR spectrum of RtpH- $NH_2$  in  $DMSO-d_6$

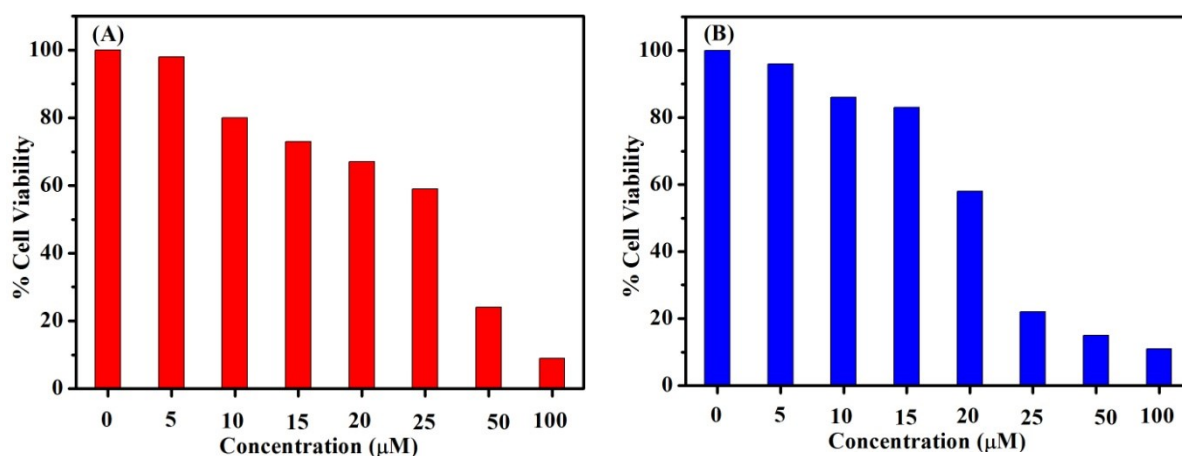


**Fig. S40** HR-ESI Mass spectra of RtpH-NH<sub>2</sub> (Calculated value for [C<sub>33</sub>H<sub>26</sub>F<sub>6</sub>IN<sub>9</sub>PRu]<sup>+</sup> = 796.1080, observed = 796.1100)

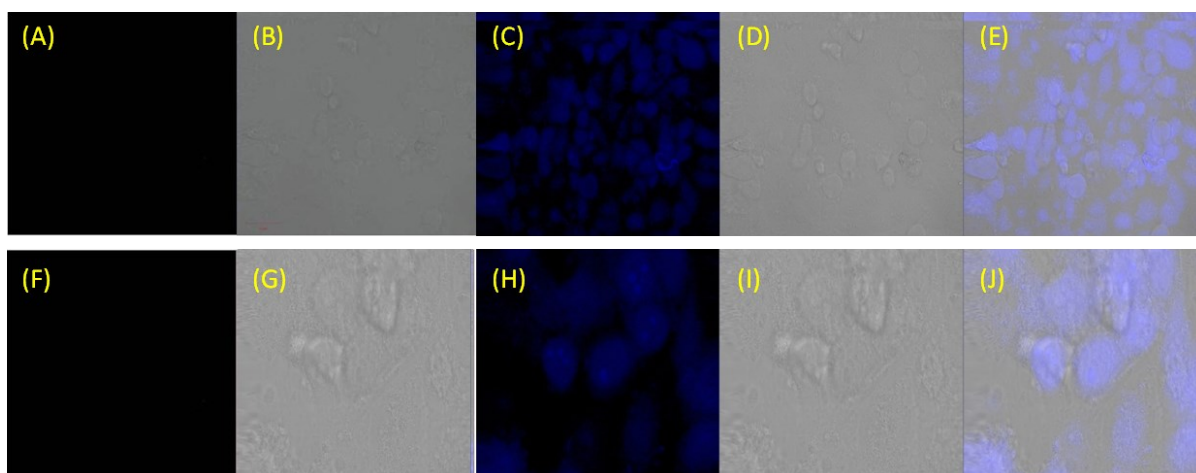


**Fig. S41** HR-ESI Mass spectra of RtpI-NH<sub>2</sub> (Calculated value for [C<sub>33</sub>H<sub>27</sub>F<sub>6</sub>IN<sub>9</sub>PRu]<sup>+</sup> + H + 2Na = 936.7646, observed = 936.7232)

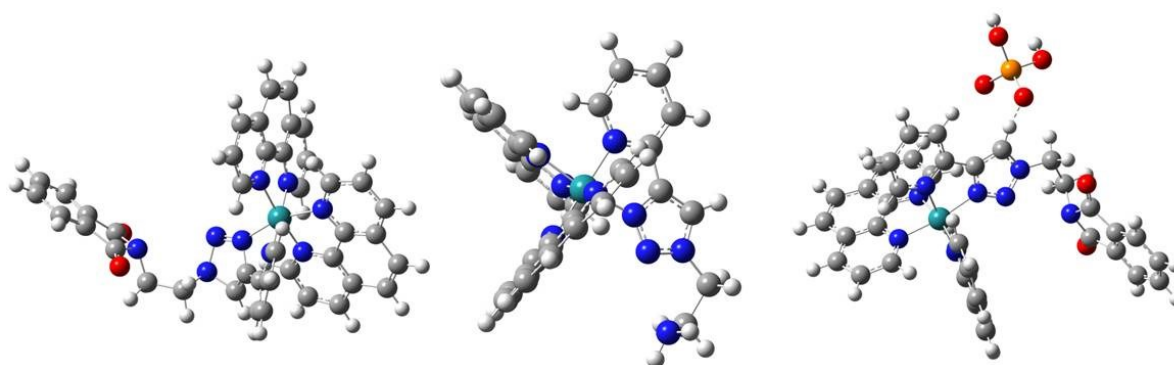




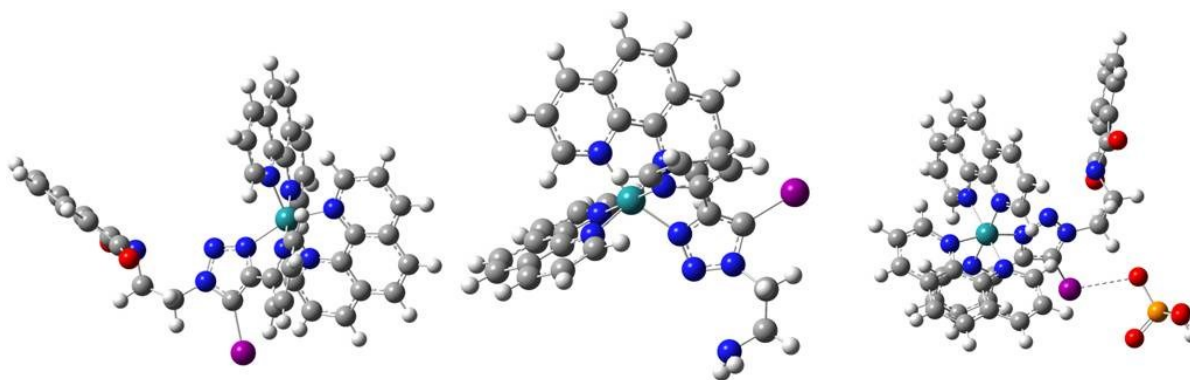
**Fig.S42** Cell viability of HeLa cells in the presence of various concentrations of (A) RtpH and (B) RtpI



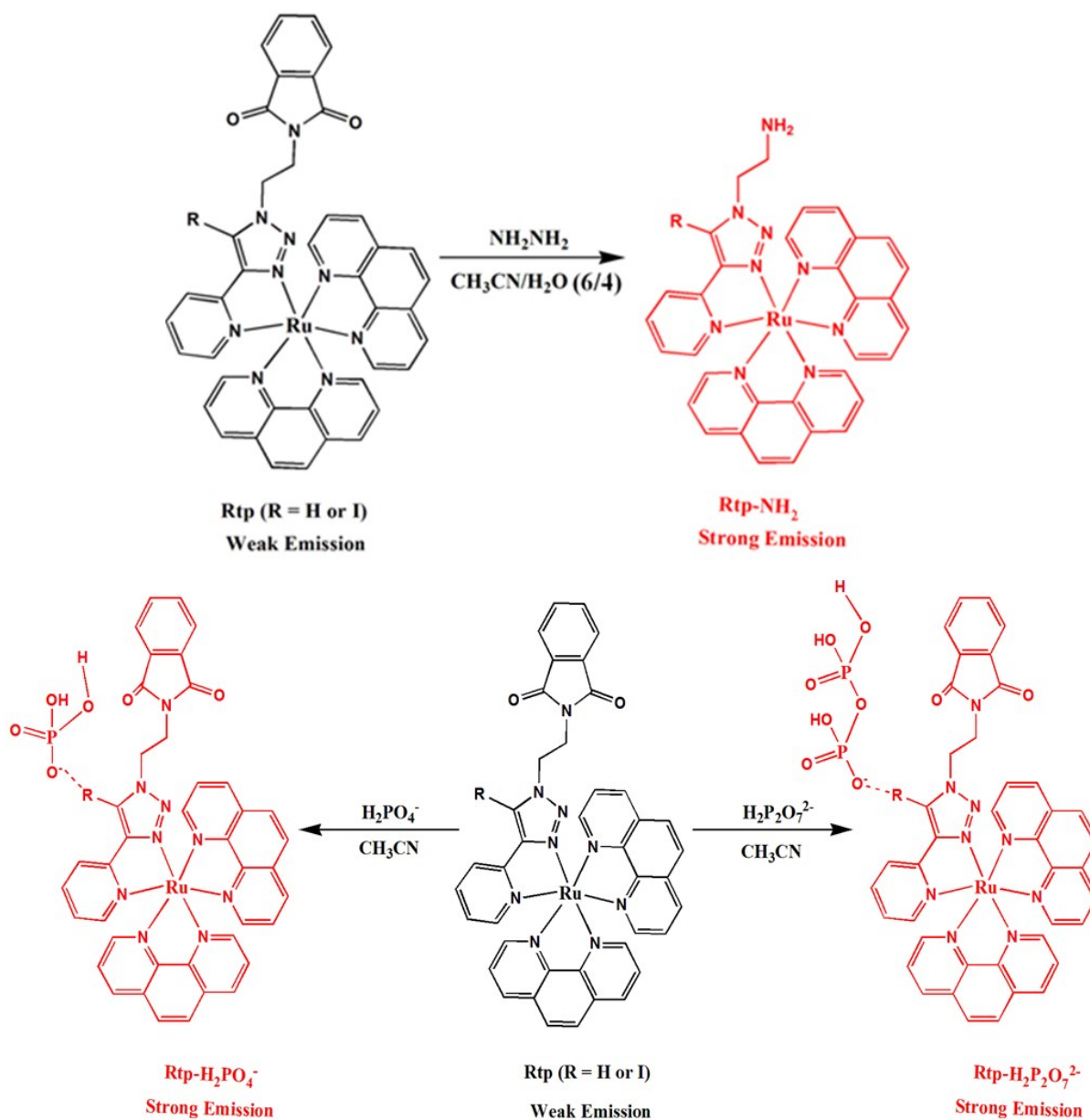
**Fig. S43** Cellular uptake (HeLa) image of  $25\mu\text{M}$  RtpH in dark (A), bright field (B), 4 equiv. of hydrazine treated RtpH (blue filter) (C), 4 equiv. of hydrazine treated RtpH (bright field) (D), 4 equiv. of hydrazine treated RtpH merged with RtpI (E), and  $25\mu\text{M}$  RtpI in dark (F), bright field (G), 4 equiv. of hydrazine treated RtpI (blue filter) (H), 4 equiv. of hydrazine treated RtpI (bright field) (I), 4 equiv. of hydrazine treated RtpI merged with RtpH (J).



**Fig.S44** Optimized structure of RtpH, RtpH-NH<sub>2</sub>, RtpH-H<sub>2</sub>PO<sub>4</sub><sup>-</sup> (from left to right)



**Fig.S45** Optimized structure of RtpI, RtpI-NH<sub>2</sub>NH<sub>2</sub>, RtpI-H<sub>2</sub>PO<sub>4</sub><sup>-</sup> (from left to right)



**Fig.S46** Probable reaction mechanism of metalloceptor (RtpH/RtpI) with hydrazine (above) and phosphate anions (proposed anion binding mechanism, below)

<i>Complex</i>	<i>IC<sub>50</sub> Value (against HeLa cells)</i>
<i>Cis-Platin</i>	<i>7.2 ±1.2 μM</i>
<i>RtpH</i>	<i>28.1 ±3.0 μM</i>
<i>RtpI</i>	<i>22.5 ±1.7 μM</i>

**Table S1.** IC<sub>50</sub> Values for cis-Platin and metalloreceptors against HeLa cells